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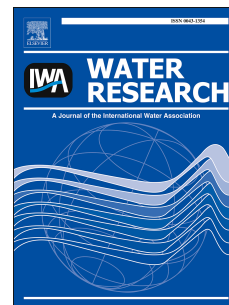
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1 **Supernatant organics from anaerobic digestion after thermal hydrolysis cause direct and/or**
2 **diffusional activity loss for nitrification and anammox**

3

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20 **Abstract**

21 Treatment of sewage sludge with a thermal hydrolysis process (THP) followed by anaerobic digestion
22 (AD) enables to boost biogas production and minimize residual sludge volumes. However, the reject
23 water can cause inhibition to aerobic and anoxic ammonium-oxidizing bacteria (AerAOB & AnAOB),
24 the two key microbial groups involved in the deammonification process. Firstly, a detailed investigation
25 elucidated the impact of different organic fractions present in THP-AD return liquor on AerAOB and
26 AnAOB activity. For AnAOB, soluble compounds linked to THP conditions and AD performance
27 caused the main inhibition. Direct inhibition by dissolved organics was also observed for AerAOB, but
28 could be overcome by treating the filtrate with extended aerobic or anaerobic incubation or with
29 activated carbon. AerAOB additionally suffered from particulate and colloidal organics limiting the
30 diffusion of substrates. This was resolved by improving the dewatering process through an optimized
31 flocculant polymer dose and/or addition of coagulant polymer to better capture the large colloidal
32 fraction, especially in case of unstable AD performance. Secondly, a new inhibition model for AerAOB
33 included diffusion-limiting compounds based on the porter-equation, and achieved the best fit with the
34 experimental data, highlighting that AerAOB were highly sensitive to large colloids. Overall, this paper
35 for the first time provides separate identification of organic fractions within THP-AD filtrate causing
36 differential types of inhibition. Moreover, it highlights the combined effect of the performance of THP,
37 AD and dewatering on the downstream autotrophic nitrogen removal kinetics.

38 **Keywords:** biological nutrient removal, partial nitrification/anammox, water resources recovery facility

39 1 Introduction

40 Thermal hydrolysis processes (THP) are well-established as reliable and energy-efficient ways to
41 intensify solids handling in sewage treatment, by creating a change in the rheology of the waste sludge
42 which allows for increased loading rates resulting in smaller digester volumes (Barber 2016). In
43 addition, this technology lowers the bound water fraction in sludge, thus resulting in an increased cake
44 solids concentration after dewatering irrespective of dewatering equipment (Bader 1978, Jolis 2008,
45 Pickworth et al. 2006). The combination of THP and anaerobic digestion (AD) has been applied
46 worldwide, e.g. at HIAS wastewater treatment plant (WWTP) (Hamar, Norway), Hengelo WWTP
47 (Hengelo, The Netherlands), Oxley Creek Water Reclamation Plant (Australia) and Blue Plains
48 Advanced WWTP (DC Water, Washington, DC, USA) (Dwyer et al. 2008, Pickworth et al. 2006). The
49 rapid decompression due to releasing steam reduces the sludge particle size by 40-50%, while higher
50 levels of colloidal and particulate organics were observed in the returned liquor compared to the
51 returned liquor from conventional AD (Barber 2016, Bougrier et al. 2008, Feng et al. 2014, Haug 1978,
52 Zhang et al. 2016). This can potentially be resolved by adding coagulants which enhance the capture of
53 fine solids or by allowing for enhanced flocculation under optimized combinations of shear and
54 exposure time (Niu et al. 2013, Poon and Chu 1999).

55 Deammonification or partial nitrification/anammox is well established as resource-efficient nitrogen
56 removal process with 100+ full-scale applications, and is particularly suitable to treat dewatering liquor
57 from sludge AD (Lackner et al. 2014). Most observed limitations in performance have been associated
58 with aerobic ammonium-oxidizing bacteria (AerAOB) due to sensitivity towards organics substances,
59 micronutrient limitation or free ammonia inhibition (Gujer 2010, Sinha and Annachhatre 2006).

60 Limitations on anoxic ammonium-oxidizing or anammox bacteria (AnAOB) because of similar factors
61 are mostly compensated by efficient (and enhanced) retention of AnAOB, creating a larger buffer in
62 AnAOB biomass (Han et al. 2016). In recent years the application of deammonification has stepwise
63 been expanded to treat livestock wastewater (diluted), landfill leachate, and THP digestate. All these

64 types of wastewater contain a wider range of inorganic and organic compounds compared to
65 conventional sewage sludge AD effluent, with reported challenges for efficient and stable
66 deammonification treatment (Dapena-Mora et al. 2006, Ganigué et al. 2008, Kindaichi et al. 2016, Lotti
67 et al. 2012, Zhang et al. 2016). For THP, cell lysis under the high temperatures yields AD filtrate higher
68 in organic compounds concentration (Barber 2016), which has shown to be (partially) inhibitory
69 towards AerAOB and AnAOB (Figdore et al. 2011), challenging efficient N removal (Chan, 2015;
70 Zhang et al., 2016). When comparing deammonification treatment of conventional AD filtrate with
71 treatment of THP-AD filtrate in two long-term reactors, decreased AerAOB rates were observed when
72 treating THP-AD filtrate, potentially caused by increased level of organics present in THP-AD filtrate
73 (Zhang et al., 2016). Optimization of key control parameters enabled efficient treatment, including
74 increased dissolved oxygen (DO) set-point (1 mg O₂/L), longer aeration times and selective AnAOB
75 retention through screens (Zhang et al. 2016). When AD instability yielded increased digestate
76 concentrations of volatile fatty acids and other organics, these soluble and colloidal compounds
77 decreased the deammonification performance considerably (Zhang et al., 2016). The link between THP
78 and creation of inhibitory compounds was also observed at full-scale, with the performance of
79 sidestream deammonification decreasing after upstream implementation of THP (Chan 2015).
80 However, to the best of our knowledge, no specific inhibitory factors have yet been identified within
81 THP-AD filtrate, nor have the inhibition mechanisms for AerAOB and AnAOB been unraveled.

82 Monod microbial growth kinetics are the simplest and most popular rate expression, assuming that
83 every single essential substrate is a growth-limiting factor (Beg and Hassan 1987, Han and Levenspiel
84 1988). To include inhibition-related limitations in growth rate, Monod-type models were further
85 developed based on three ways of inhibition (Han and Levenspiel 1988): (i) competitive inhibition,
86 yielding an increased substrate affinity constant (K_S) but not affecting the maximum specific growth
87 rate (μ_{max}), (ii) non-competitive inhibition, lowering μ_{max} but not impacting K_S , and (iii) uncompetitive
88 inhibition, combining aspects of non-competitive and competitive inhibition, i.e. affecting both K_S and

89 μ_{\max} . Interestingly, recent work of Shaw et al. (2015) has shown that K_S is not a constant, but the
90 maximum rate and diffusion influence this parameter with a relationship established based on a linear
91 data fit and the porter-diffusion model. This concept was shown to be widely applicable, as shown by
92 fits for denitrification (biofilm and floc), nitrification and anaerobic digestion (Shaw et al. 2015).
93 Summarizing, inhibitors can be described as (i) direct, causing decreased microbial growth rates hereby
94 assuming inhibitors do not influence the diffusion to the cells (i.e. non-competitive inhibition), by using
95 an 'inverted' Monod factor of the so-called direct inhibitor, and (ii) diffusional (indirect), limiting
96 diffusion represented by an increased K_S value (i.e. competitive inhibition), through including the
97 porter-equation on the affinities for substrate and direct inhibitors. In the case of THP-AD filtrate,
98 diffusion limitation on AerAOB containing flocs caused by the increased particulate and colloidal
99 fractions was suggested to be responsible for decreased ammonium conversion rates (Zhang et al.,
100 2016). Inclusion of this mechanism in a model is hypothesized to improve the predictability of
101 deammonification performance.

102 The first objective of this study was to identify the short-term inhibition effects and mechanisms
103 exerted by the soluble, colloidal and particulate organics in THP-AD filtrate on AerAOB and AnAOB
104 activities. Hereto, batch experiments were performed using non-acclimated deammonification sludge.
105 Tests related to examining whether inhibition was linked to: 1) quality of the THP-AD filtrate, 2)
106 additional pretreatment options, 3) performance of the dewatering stage (flocculant/coagulant polymer
107 dose). The second objective was to develop and evaluate a microbial kinetic model, identifying specific
108 contributions of organic fractions to either direct or diffusional limitations in conversion activities, both
109 for AerAOB and AnAOB.

110

111 **2 Materials and methods**

112 **2.1 THP-AD reactors**

113 Anaerobic sludge for dewatering was obtained from two types of AD systems. Firstly, a semi-technical
114 scale (60 L) reactor was used, fed with thermally hydrolyzed sludge (at 165°C, 30 min) (DC Water,
115 Washington, DC, USA). This digester was stably operated at 15 days solids retention time (SRT), with
116 a feed total solids (TS) concentration of about 10.5% yielding 50% volatile solids (VS) reduction at
117 38.5±1°C. Secondly, one of the four full-scale anaerobic digesters was sampled, which was started up
118 in October 2014 with an increasing feed loading from the same THP process. The digesters were
119 operated around 20 days SRT at 38°C, fed with 9.5% TS with 70% VS reduction.

120

121 **2.2 Sludge dewatering: filtrate production**

122 The AD sludge was dewatered including rapid mixing, flocculation and belt press shear, followed by
123 gravity drainage through a belt filter cloth. The remaining solids were mechanically dewatered using a
124 centrifuge at 3000 g for 10 min (Higgins et al., 2014). FLOPAM 4440 (SNF, GA, USA), a
125 polyelectrolyte with a very high molecular weight and median cationic charge (35%) was used as
126 flocculation polymer. The optimal polymer dose (OPD) was determined based on filtrate quality by
127 establishing a polymer dosing response curve, which varies depending on the digester operation
128 conditions.

129

130 **2.3 Description of filtrate applied in the tests**

131 **2.3.1 Filtrate quality at digester stability and instability**

132 The lab-scale digester was stably operated for two years, to compare to the full-scale digesters (Zhang

133 et al., 2016). For the full-scale digesters, representative samples were obtained for unstable and stable
134 behavior. The unstable full-scale AD refers to inconsistent loading and inadequate mixing in the
135 digester one month after reaching its full loading rate. During this event, the digesters experienced
136 foaming, increased volatile fatty acids (VFA) concentrations, and decreased pH, alkalinity, specific
137 gravity and biogas methane content. Four months after the digesters recovered from the unstable period
138 mentioned above, another sludge sample was taken and underwent the same inhibition activity tests,
139 referred to in this study as stable full-scale AD.

140

141 **2.3.3 Filtrate pretreatment procedures**

142 Five procedures were tested to remove specific classes of inhibiting components prior to the activity
143 tests. First, micronutrient addition in the filtrate was tested to check the growth limitation. Secondly, the
144 effect of biodegradable compounds was examined. A one-day aerobic treatment was executed, aerating
145 the full-scale filtrate for one day in the presence of nitrifying activated sludge (3-4 g TSS/L), aiming to
146 remove VFA and some other readily biodegradable organics. Then, long-term aerobic and anaerobic
147 treatment were performed to remove biodegradable COD (bCOD). In the aerobic pretreatment the full-
148 scale filtrate was aerated at a dissolved oxygen (DO) level above 4.0 mg O₂/L for seven days after
149 addition of nitrifying activated sludge (3-4 g TSS/L), until no decrease in soluble COD was observed
150 for the last four days. In the anaerobic pretreatment, sludge was derived from biochemical methane
151 potential tests until only refractory COD remained in the sludge after twelve days inoculation (Stuckey
152 and McCarty, 1984). Thirdly, precipitation with Fe³⁺ and Zn²⁺ was tested. Iron(III) sulfate (1 g Fe/L) or
153 zinc sulfate (1 g Zn/L) was added under pH control, after which precipitates were removed by
154 centrifugation (10 000 g, 10 min). Fourthly, filtrate was added to hexane as extraction solvent on 1:2
155 volume basis, followed by shaking and re-separation in a separatory funnel. Finally, adsorption to
156 activated carbon was investigated, by adding biochar (PA, USA) and granular activated carbon (Donau

157 Chemie, Austria) to the filtrate, stirring the mixture for 30 min, and removing with centrifugation (10
158 000 g, 10 min).

159

160 **2.3.4 Addition of FeCl₃ and polyDADMAC as coagulants**

161 Two conditioning processes were used to alter the COD fractions in the filtrate composition, i.e.
162 flocculation and coagulation-flocculation. As coagulants, FeCl₃ and polyDADMAC were injected
163 before the addition of flocculant polymer at the rapid mixing stage. polyDADMAC (SNF FLOQUAT
164 4520), a very highly charged cationic polymer with a high molecular weight in a liquid form, was
165 added as additional coagulant polymer (SNF, GA, USA).

166

167 **2.4 Microbial inhibition assays**

168 The sludge used in activity test originated from a semi-technical sidestream deammonification
169 sequencing batch reactor (SBR) located in Blue Plains (Zhang et al. 2016). Batch experiments were
170 performed to determine the inhibition factors of THP-AD filtrate on the AnAOB and AerAOB
171 activities. To avoid any impact of high ammonium or free ammonia levels, 2/3 of the ammonium in the
172 reject water was stripped out at an increased pH of 10 (NaOH addition), after which the pH was
173 corrected to 7.5 (H₂SO₄ addition). Chemical oxygen demand (COD) fractionations were measured
174 showing that the difference before and after stripping was less than 10%, i.e. in the range of
175 measurement error. All experiments started with 280 mg NH₄⁺-N/L, 2500 mg CaCO₃/L as alkalinity
176 and 20 mg NO₂⁻-N/L (in anoxic tests). An extra inhibition test was done to see the combined impact of
177 ammonium and COD, thus without ammonia stripping. A synthetic medium of the same osmotic
178 strength and ammonium content (1500 mg NH₄⁺-N/L) as the THP-AD filtrate was prepared, replacing a

179 certain percentage of THP-AD filtrate on a volume basis. The AerAOB and AnAOB inhibition were
180 derived from comparing conversion rates in the treatments with filtrate to the control sets (no filtrate).

181

182 **2.5 COD fractionation, other chemical analyses and calculation methods**

183 Four fractions were considered within total COD (COD_{tot}), based on four COD measurements per
184 sample: COD_{tot} and dissolved COD (COD_{diss}) were directly measured, and particulate COD (COD_{part}),
185 large colloidal COD (COD_{coll-L}), and small colloidal COD (COD_{coll-S}) were indirectly derived based on
186 two additional measurements. COD_{part} ($>1.0 \mu m$) was obtained by subtracting the concentration after
187 direct $1.0\text{-}\mu m$ filtration (Whatman, GE healthcare, UK) of the raw sample. COD_{diss} was considered as
188 the fraction that was not retained after flocculation ($ZnSO_4$) and filtration of colloids over $0.45\text{-}\mu m$, so-
189 called flocculated/filtered COD. COD_{coll-S} (flocculated/filtered - $0.45\text{-}\mu m$) was obtained by subtracting
190 COD_{diss} from the value obtained after direct $0.45\text{-}\mu m$ filtration of the raw sample. COD_{coll-L} ($0.45\text{-}1.0\text{-}$
191 μm) was derived using filtration over $1.0\text{-}\mu m$ and subsequent subtraction of $COD_{diss} + COD_{coll-S}$.

192 Sludge TS, cake solids TS, total suspended solids (TSS) and volatile suspended solids (VSS) were
193 measured according to standard methods (American Public Health Association, 1999). NH_4^+ , NO_2^- ,
194 NO_3^- and COD were measured spectrophotometrically using Hach vial kits (DR-2000, Hach, CO,
195 USA). Volatile fatty acids (VFA) were measured using gas chromatographic (GC) methods (Lu et al.,
196 2015).

197

198 **2.6 Model description and comparison**

199 Two classes of models were used using the results from 97 batch experiments to estimate inhibitor
200 affinities on AerAOB and AnAOB. Model class 1 was based on direct biological inhibition, altering the

201 values of the maximum growth rate (μ_{\max}) and inhibitor affinity (K'_I) to quantify the effects of direct
 202 inhibitors (I), assuming that diffusion limitations are constant, i.e. independent of the inhibitor(s) (Han
 203 and Levenspiel, 1988). Model 1a used the conventional approach.

$$\mu = \mu_{\max} * \left(\frac{S}{S + K'_S} \right) * \left(\frac{K'_I}{I + K'_I} \right) \text{ (eq. 1)}$$

204 In which:

- 205 • μ_{\max} : Maximum specific growth rate (1/day),
- 206 • S: Substrate concentration (mg/L),
- 207 • I: Concentration of direct inhibitors, influencing growth rate but not changing diffusion
 208 resistance (mg/L),
- 209 • K'_S : Apparent affinity coefficient for S (mg/L),
- 210 • K'_I : Apparent affinity coefficient for I (mg/L).

211

212 Model 1b additionally used a minimum function (MIN) to take the individual impact of each fraction
 213 into account, describing the rate-limiting step as the determining factor, following the approach by
 214 Stewart et al. (2017).

$$\mu = \mu_{\max} * \text{MIN} \left[\left(\frac{S}{S + K'_S} \right); \left(\frac{K'_I}{I + K'_I} \right) \right] \text{ (eq. 2)}$$

215 Model class 2 additionally included a mechanistic underpinning of diffusional limitations caused by
 216 diffusional inhibitors (I_d). To predict the differential inhibition effect by direct and diffusional inhibitor
 217 types, respectively I and I_d , a porter-diffusion formula was included adjusting apparent affinities for

218 substrate (K'_S) (eq.3) and direct inhibitors (K'_I) (eq.4) with diffusivities (D'_S and D'_I) determined by
 219 inhibition switching functions (eq.1) (Shaw et al. 2015) (see supplementary material S1).

$$K'_S = K_{S0} + R_{\max} * \frac{r_0^2}{3 * \phi * Sh * D'_S}; \text{ with } D'_S = D_S * \frac{K_{I_{d1}}}{K_{I_{d1}} + I_{d1}} * \frac{K_{I_{d2}}}{K_{I_{d2}} + I_{d2}} * \dots \text{ (eq. 3)}$$

$$K'_I = K_{I0} + R_{\max} * \frac{r_0^2}{3 * \phi * Sh * D'_I}; \text{ with } D'_I = D_I * \frac{K_{I_{d1}}}{K_{I_{d1}} + I_{d1}} * \frac{K_{I_{d2}}}{K_{I_{d2}} + I_{d2}} * \dots \text{ (eq. 4)}$$

220 In which:

- 221 • I_{d_i} : Concentration of diffusional inhibitor I, changing diffusion resistance but not affecting
 222 growth rate (mg/L),
- 223 • K_{S0} : Intrinsic affinity coefficient for S; near zero in comparison to diffusion resistance, thus
 224 usually neglected (mg/L),
- 225 • K_{I0} : Intrinsic affinity coefficient for I; near zero in comparison to diffusion resistance, thus
 226 usually neglected (mg/L),
- 227 • K'_S : Apparent affinity coefficient for S (mg/L), already including porter equation,
- 228 • K'_I : Apparent affinity coefficient for I (mg/L), already including porter equation,
- 229 • $K_{I_{di}}$: Diffusion inhibition affinity coefficient caused by I_{d_i} (mg/L),
- 230 • D_S : Diffusivity of S within the given biomass characteristics (m^2/s),
- 231 • D'_S : Modified diffusivity of S due to I_{d_i} (m^2/s),
- 232 • D_I Diffusivity of I within the given biomass characteristics (m^2/s),
- 233 • D'_I : Modified diffusivity of I due to I_{d_i} (m^2/s),
- 234 • R_{\max} : Maximum volumetric growth rate (mg/L/d),
- 235 • r_0 : single cell radius (m),
- 236 • Sh: Sherwood number, to account for convective mass transport compared to diffusive mass

237 transport (-), assumed to be 1.0 in all cases,

238 • \emptyset : Shape factor, to non-spherical cell (-), assumed to be 1.0.

239 The actual diffusivities (D'_s and D'_l) are immediately impacted by I_d during short-term exposure, and
240 this work focused on the change of K'_s and K'_l due to particulates and large colloids. All other
241 parameters such as r_0 , Sh and \emptyset were considered constant as activity tests were performed in a short-
242 term and were cancelled out in the model fitting (See supplementary material S1). The values for S , I
243 and K_S (not impacted by diffusion) were set using values in the batch tests, while μ_{max} and K_I were
244 fitted. For AerAOB, the oxygen affinity coefficient K_{O_2} was set at 0.25 mg O_2 /L (Al-Omari et al. 2015),
245 and $S_{O_2} = 3.5$ mg O_2 /L during the test. For AnAOB, the nitrite affinity coefficient $K_{NO_2^-}$ was set at 0.5
246 mg NO_2^- -N/L (Al-Omari et al. 2015), and $S_{NO_2^-} = 20$ mg NO_2^- -N/L during the test. All other non-
247 limiting substrates, e.g. $NaHCO_3$ and NH_4^+ , were not taken into account in the tests.

248 As all COD fractions were interlinked, principal component analysis (PCA) was firstly used for
249 statistical analysis to describe and summarize the dataset by reducing the dimensionality and providing
250 deeper understanding the correlation between variables using 'FactoMineR' package in R software
251 (Francois Husson et al. 2018). Secondly, 'minpack.lm' and 'nlstools' packages in R software were used
252 to find the best fit and standard error of values of each model (Florent Baty et al. 2015, Timur V.
253 Elzhov et al. 2016). Lastly, the goodness of the model fitting was quantified in three ways, by
254 comparing the total squared error (TSE), the mean squared error (MSE) and a statistical F-test. The last
255 indicates whether the fit of different models was significantly different, shown by a probability number
256 (p-value) smaller than or equal to 0.05.

257

258 3 Results

259 The fractionation of organics in THP-AD filtrate was as follows (belt filter press operation from January
260 to June 2015): 12±8% particulate (620 ± 402 mg COD_{part}/L), 15±5% colloidal, of which 5±3% large
261 colloids and 12±6% small colloids (211 ± 155 mg COD_{colloidal-l}/L and 458 ± 198 mg COD_{colloidal-s}/L), and
262 73±8% soluble organics (3575 ± 930 mg COD_{diss}/L). To elucidate the inhibitory impacts of different COD
263 fractions within the THP-AD filtrate, samples were taken from different AD systems (lab & full-scale),
264 operational periods (stable & unstable), with different pretreatment methods and different dewatering
265 conditions.

266

267 3.1 Impact of anaerobic digester performance: VFA and other biodegradable organics

268 Three different types of filtrate from the lab-scale (stable) and full-scale (stable and unstable) ADs were
269 used in the tests to evaluate the impact of digester performance. The filtrate composition of the samples is
270 shown in Table 1. Although COD_{diss} was similar between the lab- and full-scale system during stable
271 operation, the full-scale AD had slightly higher particulate and colloidal COD concentrations. The
272 comparison showed no significant difference in AerAOB inhibition under stable anaerobic digestion (Fig.
273 1a). An additional test was performed at the original ammonium concentration from the stable full-scale
274 AD filtrate (1500 mg N/L instead of constant 280 mg N/L), allowing higher volume exposures to be
275 tested. This showed a similar inhibition degree as the one with lower ammonium levels (Fig. 1a). Overall,
276 AerAOB lost 50% of its activity when exposed to 40% of lab-scale AD filtrate, 46% of full-scale AD
277 filtrate (lower ammonium/free ammonia concentration) and 30% of full-scale AD filtrate (higher
278 ammonium/free ammonia concentration) respectively (Fig. 1a). In contrast for AnaAOB, significantly
279 higher inhibition was observed when exposed to filtrate from full-scale AD resulting in no measurable

280 AnAOB activity at 25% volume exposure of filtrate, while in case of the lab-scale digester still $54\pm 5\%$
281 AnAOB activity remained (Fig. 1b).

282 When the digester showed instability as characterized by VFA accumulation and decreased biogas
283 production, all COD concentrations increased (Table 1 I). The largest increase was in COD_{diss} , of which
284 25% was due to the increase of acetate as the main VFA present (from < 100 mg COD/L to 1383 mg
285 COD/L). The increase of COD due to the digester instability resulted in statistically higher inhibition on
286 AerAOB and AnAOB (Fig. 1a;b). Residual AnAOB activity was less than 10% when exposed to 10% of
287 the unstable filtrate (Fig. 1b), indicating that AnAOB were highly sensitive to the digester performance.
288 AerAOB activity decreased initially only slightly by $11\pm 5\%$ at 10% of filtrate volume exposure. However,
289 activity loss was more substantial at higher exposure ($30\pm 4\%$ of residue activity at 46% volume exposure)
290 (Fig. 1c). Since a considerable acetate level was observed in the unstable AD filtrate, an extra test was
291 performed spiking acetate to stable AD filtrate (10%), reaching a similar final concentration of 1250 mg
292 acetate-COD/L. This yielded statistically the same result (Fig. 1c;d), indicating the acetate did not
293 contribute to the additional inhibition observed in unstable AD filtrate.

294

295 **3.2 Impact of pretreatment**

296 The first pretreatment tests examined the biodegradability of the inhibitory organics. The filtrate was
297 treated aerobically for 1-day to verify the impact of short-term aeration, removing mainly VFA and some
298 readily biodegradable organics (COD_{diss} decreased from 3574 to 1948 mg COD/L). The results showed
299 that inhibitions on AerAOB and AnAOB were not significantly reduced compared to non-treated full-
300 scale AD filtrate (stable) (Fig. 1c;d). To investigate the impact of all biodegradable organics, filtrate
301 underwent a more extended biological treatment, both aerobic and anaerobic. This aimed at evaluating if

302 the observed inhibitory impacts shown in Fig. 1 (a-d) were related to the recalcitrant or biodegradable
303 fractions.

304 Comparing organics to full-scale unstable filtrate, all COD fractions decreased after 4-day aerobic
305 pretreatment and 12-day anaerobic treatment (compositions shown in Table 1 I and II). For AerAOB, no
306 significant inhibition was observed with the pretreated filtrate, indicating the original loss of AerAOB
307 activity was most likely related to slowly biodegradable COD (Fig. 1.a;e). For AnAOB, however, there
308 was no clear relation between biodegradable COD and activity loss, as AnAOB were still totally inhibited
309 when exposed to more than 30% filtrate (Fig. 1b;f).

310 To further identify the potential inhibitors on AerAOB and possible remedies to avoid such compounds,
311 additional pretreatments were performed, based on the biological (1+2) or chemical nature (3+4+5) of the
312 activity-limiting compounds: (1) micronutrient addition, (2) six-month filtrate storage (4 °C), (3)
313 coagulation/precipitation through iron or zinc addition, (4) extraction using hexane, (5) sorption of
314 specific organics on biochar and activated carbon. Adding extra micronutrients in the filtrate did not
315 mitigate the activity loss (Table 2), indicating that the filtrate composition did not cause growth limitation
316 for the AerAOB. After six-month storage, the filtrate showed an inhibition of $39.0 \pm 0.5\%$, similar to the
317 fresh filtrate indicating that biodegradable organic fraction underwent few changes during storage. As for
318 the pretreatments chemically removing the particular organic fraction, no significant impact was observed
319 for the precipitation with iron or zinc, extraction with solvent or sorption on biochar (Table 2).

320 Interestingly, sorption on activated carbon showed a clear potential of removing the inhibitory compounds
321 for AerAOB. Furthermore, the impact of activated carbon on alleviating AnAOB inhibition was also
322 demonstrated, decreasing the inhibition significantly from $42 \pm 6\%$ to $2 \pm 6\%$ (at 20% of volume exposure).

323

324 3.3 Impact of dewatering efficiency

325 To understand the impact of colloidal and particulate COD fractions, and to examine the optimal
326 dewatering conditions, changes in flocculant polymer (FLOPAM) dose were tested (Table 1 III), along
327 with dosage variations in polyDADMAC or FeCl_3 as coagulants (Table 1 IV and V).

328 OPD of the flocculant polymer was around 10 g FLOPAM/kg sludge-TS and no improvement in
329 dewatering was observed with a further increase in polymer dose (Fig. 2a). A coagulant (polyDADMAC
330 or FeCl_3) was added to differentially capture particulate and colloidal fractions. Dosing polyDADMAC
331 and FeCl_3 in addition to 10 g FLOPAM/kg sludge-TS significantly improved the filtrate quality by
332 capturing more particulate and colloidal COD (Fig. 2 b,c). PolyDADMAC captured the large colloids
333 efficiently, reducing $\text{COD}_{\text{coll-L}}$ from around 500 to 155 mg COD/L, but could not efficiently capture the
334 small colloids (Fig. 2b, Table 1 IV). FeCl_3 captured both large and small colloids, resulting around 200
335 mg COD/L of small colloids. From a dose of 2 g polyDADMAC/kg sludge-TS and 70 g FeCl_3 /kg sludge-
336 TS onwards, no additional improvement was observed (Fig. 2 b,c). The biological impact of the
337 presence/absence of these coagulant dosages was therefore tested in the subsequent microbial activity
338 tests.

339 For AerAOB, dewatering without coagulants, i.e. using only FLOPAM during dewatering, yielded a
340 filtrate composition with around 1200 mg $\text{COD}_{\text{part}}/\text{L}$, 500 mg $\text{COD}_{\text{coll-L}}/\text{L}$ and 400 mg $\text{COD}_{\text{coll-S}}/\text{L}$ (Fig.
341 2a). The lowest AerAOB inhibition ($13 \pm 8\%$) was obtained at an optimal FLOPAM dose of 10 g
342 FLOPAM/kg sludge-TS (Fig. 3 a). Underdosing of polymer (7 g FLOPAM/kg sludge-TS) led to an
343 increase in particulate and colloidal COD content in the filtrate and an AerAOB inhibition of $36.7 \pm 8.9\%$.
344 However, overdosing (14 g FLOPAM/kg Sludge-TS) also lowered AerAOB rates by $31.4 \pm 7.9\%$, which
345 might have been caused by the presence of residual polymer potentially creating oxygen uptake limitation
346 due to diffusion resistance (Fig. 3a).

347 Adding coagulant only improved the AerAOB activity in the case of polyDADMAC. The addition of
348 polyDADMAC efficiently captured the large colloidal fraction, as indicated above, lowering the AerAOB

349 inhibition to a minimum for filtrate from unstable full-scale THP-AD operation (Fig. 3c, Table 1 IV, V).
350 A similar experiment was performed under stable AD performance as shown in Fig. 3c. As a lower large
351 colloidal fraction was automatically obtained under stable AD (Table 1 IV, V), AerAOB rates were not
352 improved significantly (Fig. 3c). Though FeCl_3 captured both large and small colloids, no clear
353 improvement in the AerAOB activity was observed (Fig. 3c), which might relate to the potential impact
354 of excess FeCl_3 in the filtrate. Therefore, assuming 10% of the dosed FeCl_3 would remain in the filtrate,
355 an additional batch test was performed by adding 333 mg FeCl_3/L to filtrate only dewatered by flocculant
356 polymer dose. Higher AerAOB activity loss at FeCl_3 dosing was observed; $34\pm 10\%$ compared to $21\pm 4\%$
357 activity loss when exposed to filtrate only dosed by FLOPAM, confirming the hypothesis of harmful
358 residual FeCl_3 .

359 As the previous experiments optimized FLOPAM dose before coagulant dose (polyDADMAC), this was
360 reverted in an additional experiment. Results in Fig. 2b pointed at 2 g polyDADMAC/kg sludge-TS as a
361 minimum to improve the capture efficiency of large colloidal COD. Additional dewatering tests showed
362 that the presence of this coagulant level increased the capture efficiency for particulate and colloidal
363 matter increased, enabling to lower the flocculant dose (Fig. 4 a,b). Indeed, the capture efficiency of large
364 colloids increased to 70-80% at total polymer dose of 6.2 g polymer/kg sludge-TS, i.e. containing 4.2 g
365 FLOPAM/g sludge-TS (Fig. 4b). However, in the absence of polyDADMAC, about 9 g FLOPAM/kg
366 sludge-TS was needed to achieve a similar effect (Fig. 4a).

367 For AnAOB, in all dewatering tests, no substantial improvements on AnAOB rates were achieved (Fig.
368 3b;d), indicating that particulate and colloidal COD potentially did not play a major role in AnAOB
369 inhibition.

370

371 **3.3 Model-based mechanisms of lowered activities**

372 In order to identify the mechanisms lowering AerAOB and AnAOB activities, different model types were
373 fitted to the data obtained from the batch experiments. As only short-term testing was performed in this
374 study, the focus was on identification of direct inhibition impacts (K'_I impacts) and diffusion impacts (K'_S
375 impacts).

376 Model class 1 was based on direct inhibitors (I), assuming diffusion limitations were independent of the
377 inhibitor: COD fractionating was not considered: $I = \text{COD}_{\text{tot}}$. The conventional Monod approach was used
378 in model 1a (S.eq. 1), while model 1b incorporated a minimum function yielding full impact of the rate-
379 limiting step (S.eq. 2) (Supplementary material S2 and S3). Model 1a estimated K'_I values of 1519 ± 155
380 and 269 ± 32 mg $\text{COD}_{\text{tot}}/\text{L}$ for AerAOB and AnAOB, respectively (Table 3, Fig. 5). With Model 1b,
381 slightly lower K'_I values were obtained of 1282 ± 122 and 258 ± 30 mg $\text{COD}_{\text{tot}}/\text{L}$ for AerAOB and AnAOB,
382 respectively. Comparing the goodness of fit of both models, the conventional approach (model 1a) was
383 found superior to the minimum model (model 1b), as the latter yielded higher TSE, MSE and RSE (Table
384 3).

385 In model class 2, next to direct inhibitors (I), also diffusional inhibitors (I_d) were taken into account, as
386 compounds influencing the diffusion of substrates and direct inhibitors. To perform a preliminary
387 screening as to which COD fraction corresponds to which inhibitor type, PCA was performed using all
388 batch test data. This revealed that the four COD fractions could be categorized in two or three classes
389 with a differential impact on AerAOB and AnAOB activities: either $\text{COD}_{\text{diss}} + \text{COD}_{\text{coll-S}}$ vs. $\text{COD}_{\text{coll-L}} +$
390 COD_{part} , or $\text{COD}_{\text{diss}} + \text{COD}_{\text{coll-S}}$ vs. $\text{COD}_{\text{coll-L}}$ vs. COD_{part} (Fig. S.1). Soluble and small colloidal COD
391 were grouped as one fraction and considered to only cause direct inhibition ($I = \text{COD}_{\text{diss}} + \text{COD}_{\text{coll-S}}$). For
392 large colloidal and particulate COD, an impact of the diffusion resistance was anticipated, either lumped
393 into one inhibitor in model 2a ($I_d = \text{COD}_{\text{part}} + \text{COD}_{\text{coll-L}}$; S.eq. 3) or treated as separate inhibitors in model
394 2b ($I_{d-1} = \text{COD}_{\text{part}}$; $I_{d-2} = \text{COD}_{\text{coll-L}}$; S.eq. 4) (Supplementary material S4 and S5).

395 When implementing model 2a (S.eq. 3) for AerAOB, the estimated parameters indicated a major
396 sensitivity to compounds causing diffusional resistance, as the K'_I value was considerably lower
397 compared to model 1, and as K'_{I-d} was considerably lower than K'_I (Table 3). However, despite a slightly
398 lower TSE, model 2a did not render a better fit than model 1, as MSE increased, and the F-test showed no
399 significant improvement ($p > 0.05$). For model 2b, discriminating between the individual contributions by
400 particulate and large colloidal organics, a relatively high K'_{I-d} value was obtained for COD_{part} and a low
401 value for COD_{coll-L} , indicating that the latter was mainly causing the diffusion resistance. Model 2b
402 provided the best fit for AerAOB inhibition, as obtained TSE and MSE were the lowest, and the F-test
403 showed significant improvement compared to models 1 and 2a ($p \leq 0.05$). Another model fitting was
404 performed by considering small colloidal COD as a diffusional impactor (data not shown). The results
405 predicted a K'_I for COD_{coll-S} of 2744 ± 405 mg COD/L, a value considerably above the maximum observed
406 concentration of small colloidal COD in THP-AD filtrate. It can, therefore, be concluded that COD_{coll-S}
407 did not limit substrate diffusion.

408 For AnaAOB, the direct-only inhibition models did not show improvement of including a minimum
409 function (model 1b), as TSE and MSE did not lower compared to model 1a (Table 3). Furthermore, the
410 inclusion of diffusional limitation in model 2 yielded no significant improvement in fit, as shown by the
411 F-tests ($p > 0.05$; Table 3). Models 2a and 2b estimated relatively low and high values for K_{Id} and K_{Ind} ,
412 respectively, indicating indeed that diffusion resistance was limited (Table 3).

413

414 **4. Discussion**

415 **4.1 Organics inhibiting AerAOB**

416 THP-AD filtrate caused direct AerAOB inhibition linked to biodegradable dissolved organic compounds,
417 excluding acetate. The model however indicated AerAOB were less sensitive to soluble compounds than

418 AnAOB, with an estimated K'_1 for $\text{COD}_{\text{diss+coll-S}}$ about 4 times above the one for AerAOB. The AerAOB
419 inhibition could be mitigated by extended biological treatment (aerobic or anaerobic) or activated carbon
420 adsorption. The latter process is widely used to adsorb polar compounds, for instance in soil remediation,
421 removing hydrophobic organic contaminants and polychlorinated biphenyls (Hale et al. 2013, Marchal et
422 al. 2013, Vasilyeva et al. 2006). It is unclear why biochar could not overcome AerAOB inhibition though
423 it has similar properties. In theory, each of the three options could be implemented to lower the
424 biodegradable COD content in THP-AD filtrate, (i) by treating the filtrate aerobically, e.g. in a trickling
425 filter, (ii) by extending the SRT in AD, e.g. with a post-digester, or (iii) by installing an activated carbon
426 process. However, such pretreatment approaches require an additional unit process and are not practically
427 feasible or cost-effective. Preferentially the inhibition can be prevented or cured by altering the operation
428 of the existing processes, e.g. by increasing the aerobic SRT or the dilution of the filtrate. Additional
429 clues are expected from a deeper analytical characterization of differences in dissolved organics in THP-
430 AD vs. conventional AD filtrate.

431 Besides the direct effect, the filtrate's organics also indirectly lowered the AerAOB activity, with
432 particulates and large colloids putatively limiting the diffusion of substrates. A clear relationship between
433 the dewatering performance and AerAOB activities was shown, similar to earlier studies showing the
434 important influence of diffusion on nitrification kinetics in practice, for instance in case of an increased
435 solids load to the system during storm events (Armstrong, 2008; Shaw et al., 2015; Stenstrom and
436 Poduska, 1980). Optimizing flocculant polymer dose to reduce residue solids or polymer in THP-AD
437 filtrate diminished the AerAOB inhibition (Fig. 3.a). This is in line with observations in full-scale
438 operation of sidestream deammonification, where a high TSS inflow led to decreased turnover rates of
439 AerAOB and nitrite oxidizing bacteria (NOB), and restoring of the AerAOB activity was achieved after
440 doubling the aeration time and increasing the DO concentration of up to 1 mg O_2/L (Joss et al. 2011). In
441 contrast to the under-dosed filtrate, the polymer residue in the over-dosed filtrate can cause foaming
442 issues and potentially diffusion limitations, as observed in 30% of the plants (Lackner et al. 2014). In

443 contrast to the under-dosed filtrate, the polymer residue in the over-dosed filtrate can cause foaming
444 issues and potentially diffusion limitations, as observed in 30% of the plants (Lackner et al. 2014). The
445 impact of foaming on reactor performance created substrate intake limitation and the recovery usually
446 takes about 1-5 days confirming that the system is not inhibited.

447 The AerAOB inhibition from large colloids could be significantly prevented, although not completely
448 removed, by optimizing the dewatering stage, i.e. adding coagulant to better capture this fraction,
449 especially during unstable digestion (Fig. 3c). Model 2b showed that separating the particle sizes (0.45 to
450 $1\ \mu\text{m}$ vs. $> 1\ \mu\text{m}$) led to a relatively low K'_1 value for large colloidal COD ($7.6 \pm 1.3\ \text{mg COD}_{\text{col-L}}/\text{L}$),
451 pinpointing this fraction as a key inhibitor class to AerAOB. Compared to the typical content of large
452 colloids in THP-AD filtrate ($211 \pm 151\ \text{mg COD}_{\text{col-L}}/\text{L}$), the low K'_1 indicated that the large colloidal
453 fraction was the rate-limiting factor. Under high levels of colloidal matter, oxygen transfer efficiency
454 decreased, resulting in limited DO availability and consequently a poor nitrification performance
455 (Germain et al. 2007, Wu et al. 2013). Li et al. (2018) observed the decrease of oxygen availability
456 lowered substrate conversion in biofilms due to the high loadings of particulate organics in wastewater,
457 which resulted in 20-70% of dissolved oxygen flux reduction from the biofilm surface into biofilm matrix.
458 Coagulants neutralize the negative electric charge on particles and destabilize the forces keeping colloids
459 apart (Poon and Chu 1999). The large colloidal fractions, in the case of THP-AD filtrate, had more impact
460 on AerAOB flocs possible due to the relative small AerAOB floc size compared to the AnAOB granules
461 (Han et al. 2016, Vlaeminck et al. 2010), and The diffusion impact on the AerAOB-containing flocs was
462 relatively dominant, creating diffusion limitation of substrates, e.g. oxygen or ammonium, rather than
463 directly impacting the cell's metabolism or growth rate negatively. The curative strategy of increasing the
464 DO level helps to overcome diffusion limitation and improve ammonium removal rates as confirmed by
465 Zhang et al. (2016). Additionally, unlike the inhibition related to the soluble fractions, the lowered
466 AerAOB rates caused by diffusion limitation are usually transient, reversible and easy to notice.

467 Adequate trace nutrient concentrations, including metal ions and vitamins, are necessary to have a diverse
468 microbial community thrive in an activated sludge plant, especially when treating certain types of
469 industrial wastewater (Burgess et al. 1999, Jefferson et al. 2001). Klein et al. (2013) and Nifong et al.
470 (2013) observed a full-scale sidestream deammonification system instability due to the micronutrient
471 limitations. The activity test in this study showed that the biomass inhibition was not related to acute
472 micronutrient deficiency (Table 2). Long-term operation is, however, necessary to validate if adding extra
473 bioavailable micronutrients might be helpful to compensate for the decreased growth rates and/or
474 increased decay rates.

475

476 **4.2. Organics inhibiting AnAOB**

477 For AnAOB, the kinetic tests indicated that soluble organics in THP-AD filtrate were the primary source
478 of inhibition, potentially related to the nonbiodegradable fraction. Concentrations of dissolved organics
479 were tightly related to THP and AD operation stability. However, due to the multitude and complexity of
480 the organics, it is an analytical challenge to identify the specific inhibitors. Dwyer et al. (2008) reported
481 an increased presence of recalcitrant organic compounds in THP-AD filtrate was related to the
482 pretreatment at an excessively high temperature (higher than 170 – 190°C) which led to significant
483 changes in a generation of recalcitrant N and COD. Meanwhile, these soluble inhibitory organics were not
484 easily removed via aerobic and anaerobic treatment as the remaining AnAOB activity was lower than 80%
485 (Fig. 1f). Gupta et al. (2015) showed that organic matter in the THP-AD filtrate could not be efficiently
486 removed by aerobic biological treatment. The biodegradable organics related to the digester instability
487 increased inhibition and can be caused by variable (e.g. shock) loading, resulting in partial sludge
488 degradation and associated accumulation of surface active substances and subsequent foaming issues
489 (Ganidi et al. 2009). The total nitrogen removal rate in a lab-scale sidestream deammonification system
490 decreased with 90% resulting in substantial nitrite accumulation ($> 20 \text{ mg NO}_2^- \text{-N/L}$) when fed with

491 filtrate containing increased soluble organics (5934 ± 680 mg COD/L) caused by digester overloading
492 (Zhang et al. 2016). An increase in VFA concentration usually occurs as an indicator and consequence of
493 unstable AD (Chen et al. 2008, Iv 2003), as also seen in this study with an increase of soluble COD from
494 2762 mg COD/L to 6600 mg COD/L in the filtrate, with around 35% of the increase due to acetate.
495 However, actual VFA concentrations in unstable filtrate were a lot lower than the inhibitory levels for
496 AnAOB observed in previous studies (Dapena-Mora et al. 2004, Molinuevo et al. 2009). In comparison to
497 our acetate value of 1.3 g COD/L in unstable THP-AD filtrate, Kindaichi et al. (2016) observed that
498 concentrations up to 3.2 g COD/L of acetate had no effects on AnAOB. The acetate spike test at 1.3 g
499 COD/L also confirmed no impact on the AnAOB activity (Fig. 1d).

500 Model runs confirmed the sensitivity of AnAOB towards soluble COD. The estimated K'_I of 250 mg
501 COD/L is 4 times lower than the toxicity level at 800 mg COD/L observed in the long-term SBR
502 operation (Zhang et al. 2016). This higher level can relate to an underloaded AnAOB inventory resulting
503 from efficient AnAOB retention in the SBR. However, once inhibition has been observed in sidestream
504 deammonification, it usually takes the biomass a month to recover. Therefore, longer SRT and good
505 AnAOB retention are necessary to compensate for the decreased growth rates. Furthermore, even with
506 actions taken in THP and AD, improvements in filtrate composition lag because of the long SRT in AD.
507 Thus, monitoring the digesters' performance is important, and upon instability, manifested by less
508 methane, foaming, more VFA and ammonia, etc., it is necessary to dilute the filtrate more. The full-scale
509 deammonification system at DC Water is designed to treat THP-AD at a 1:1 dilution ratio with water
510 (Zhang et al. 2016). The potential drawback of a higher dilution is the temperature loss, especially in
511 winter time.

512

513 5. Conclusion

514 This paper investigated four organic fractions in the filtrate from anaerobic digestion preceded by thermal
515 hydrolysis, to identify specific inhibition compounds and mechanisms for AerAOB and AnaAOB.
516 Furthermore, the findings were confronted with a global mechanistic inhibition model. AerAOB were
517 inhibited directly, from dissolved organics, and indirectly, with particulate and colloidal fractions causing
518 diffusion limitation. The latter can be resolved through optimization of the dewatering process in terms of
519 flocculant and coagulant polymer dosing, to obtain a better capture of the colloidal fraction to the sludge
520 cake. An inhibition model including diffusion resistance based on the porter-equation showed the best fit
521 with the experimental data, and highlighted that AerAOB were highly sensitive to large colloids. For
522 AnaAOB, it was found activity was mainly impacted by dissolved organics in the filtrate, tightly linked to
523 the THP and digester performance, not by changes in the dewatering process.

524

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529

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648 **Legends**

649 Figure 1. Impact of different dilutions of raw and treated THP-AD filtrate on AerAOB (a, c, e) and
650 AnAOB activities (b, d, f), originating from different digesters and performance status (a, b, c, d; Table
651 1 I), linked to the impact of acetate dosing (c, d) and biological pre-treatment: 1-day aerobic treatment
652 (c, d); 7-day aerobic treatment and 12-day anaerobic treatment (e, f).

653 Figure 2. Impact of chemical dosing in THP-AD sludge dewatering (filtration) on COD fractionation
654 and TSS levels: dose of flocculant FLOPAM (a-1, a-2) and variation in coagulant doses at optimal
655 flocculant dose (10 g FLOPAM/kg sludge-TS), using polyDADMAC (b-1, b-2) or FeCl_3 (c-1, c-2). The
656 selected filtrate used for the subsequent activity tests (Figure 3) was marked with *.

657 Figure 3. Impacts on AerAOB (a, c) and AnAOB (b, d) activities of chemicals added in full-scale THP-
658 AD sludge dewatering (filtration): dose of flocculant FLOPAM (a, b; Table 1 III; 30% filtrate based on
659 volume) and variation in coagulant at optimal flocculant dose (10 Gflopam/kg Sludge-TS), using
660 polyDADMAC or FeCl_3 , linked to digester stability (c, d; Table 1 IV,V; 20% filtrate based on volume).
661 Significantly improved activity was marked with * ($p \leq 0.05$).

662 Figure 4. Impact of the cationic polymer dose on the capture efficiencies of suspended solids (TSS),
663 particulate organics (COD_{part}) and large colloidal organics ($\text{COD}_{\text{coll-L}}$) to the cake fraction during
664 digestate dewatering: flocculant polymer (FLOPAM) dosing in absence of polyDADMAC (a) and
665 presence of 2 g polyDADMAC/kg sludge-TS.

666 Figure 5. Model fitting comparisons of direct inhibition models (Model 1 a, b) with direct and
667 diffusional inhibition models (Model 2 a, b) for AerAOB (a) and AnAOB (b) (model parameters
668 displayed in Table 3).

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670

671 Table 1. Composition of filtrate from anaerobic digestion (AD) of sewage sludge pre-treated with a thermal hydrolysis process (THP) as a function of
 672 its origin and post-treatment (aerobic/anaerobic, flocculation, coagulation) based on 97 batch tests. a. particulate COD; b. large colloidal COD; c.
 673 small colloidal COD; d. soluble COD, e. volatile fatty acid, f. measurement including $COD_{coll-s} + COD_{diss}$, g. not available, h. not detectable

#	Test description	Composition (No dilution)						Refer to Figure		
		Polymer dose g/kg sludge-TS	NH_4^+ mg N/L	COD_{part}^a mg/L	COD_{coll-L}^b mg/L	COD_{coll-S}^c mg/L	COD_{diss}^d mg/L	VFA ^e mg COD/ L		
I	Different THP-AD sludge at different phases	Lab-scale AD	14	1910	158	399		2830 ^f	70	Fig. 1
		Full-scale AD (stable)	10	1882	381	243	219	2762	50	
		Full-scale AD (unstable, start-up phase)	10	1925	550	450	825	6600	1502	
II	Biodegradable COD	1-day aerobic treatment (25°C)	14	N.A. ^g	56	366		3896 ^f		Fig. 1
		7-days aerobic treatment (25°C)	10	593	535	120	415	3175	N.D. ^h	
		12-day anaerobic treatment (38°C)	17	2235	325	230	200	2680	N.D.	
I	Optimizing flocculant polymer dose	Under-dose (FLOPAM)	7	1883	587	937		3454 ^f	68	Fig. 3
		Optimum dose (FLOPAM)	10	1885	218	76		3670 ^f	61	
		Over-dose (FLOPAM)	14	1800	148	32		3758 ^f	77	
I V	Coagulant addition to unstable full-scale digester sludge	FLOPAM	0+10	2217	1275	515	415	3770	N.A.	Fig. 3
		PolyDADMAC +FLOPAM	2+10	2205	895	155	425	3775	N.A.	
		FeCl ₃ +FLOPAM	0.07+10	2170	685	220	170	3395	N.A.	
V	Coagulant addition to stable full-scale digester sludge	FLOPAM	0+10	1793	478	222	394	2694	55	Fig. 3
		PolyDADMAC +FLOPAM	2+10	1688	286	114	502	2684	45	
		FeCl ₃ +FLOPAM	0.07+10	1752	240	68	134	2640	44	

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675 Table 2. Overview of the AerAOB inhibition caused by raw and treated THP-AD filtrate. A statistically
 676 improved activity compared to the reference was marked with * ($p \leq 0.05$).

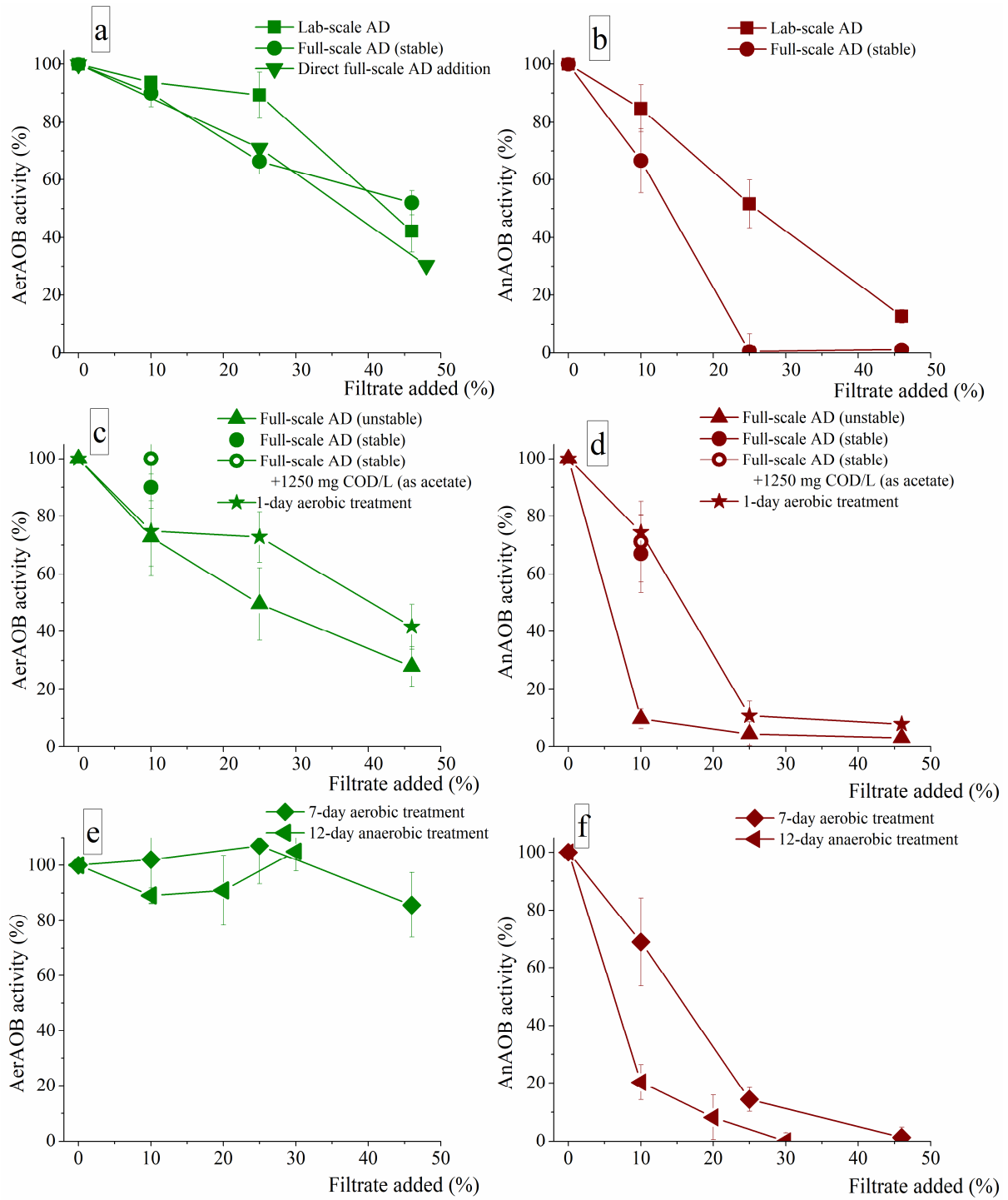
Test category	Filtrate description or treatment	AerAOB activity (%)	
		25% filtrate	30% filtrate
Reference	Raw filtrate (stable full-scale AD)		61±3*
Growth limitation	Micronutrient addition to raw filtrate		60±1
Presence of VFA and other biodegradable compounds	Raw filtrate (unstable AD, 4000 mg VFA-COD/L)	50±13	
	Raw filtrate (stable AD); acetate spike (1250 mg COD/L)	0±11	
Removal of biodegradable compounds	1-days aerobic treatment (25°C)	37±3	
	7-day aerobic treatment (25°C)	100±7*	
	12-day anaerobic treatment (38°C)		100±9*
	6-month storage of raw filtrate (4°C)		39±1
Coagulation of organics /precipitation of anions	Fe(III) dosage (1 g/L); centrifuged		38±11
	Zn(II) dosage (1g/L); centrifuged		97±1
Removal of (apolar) organics	Extraction with hexane		42±1
	Sorption with biochar (40 g/L); centrifuged		39±1
	Sorption with granular activated carbon; centrifuged		94±12*

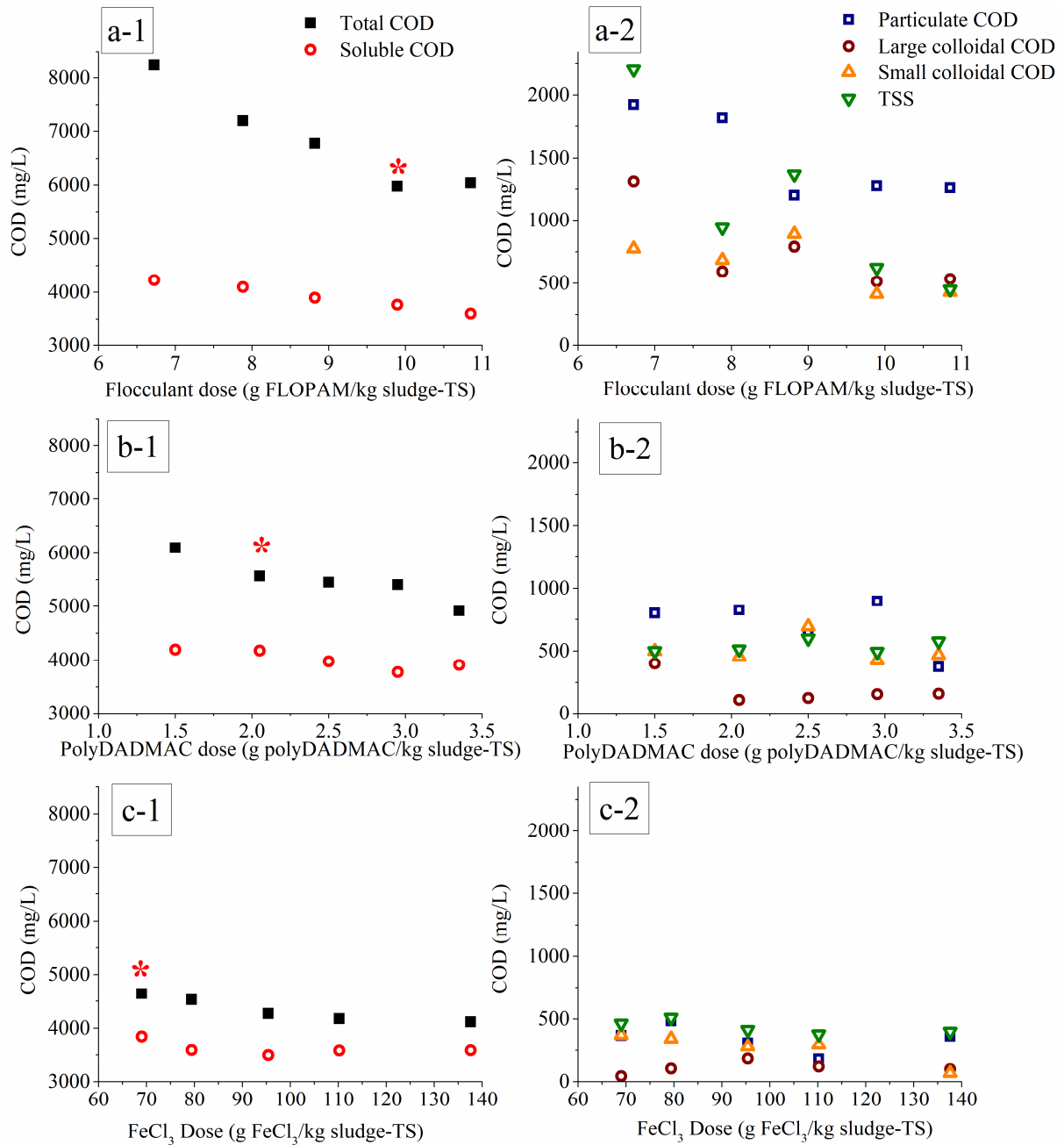
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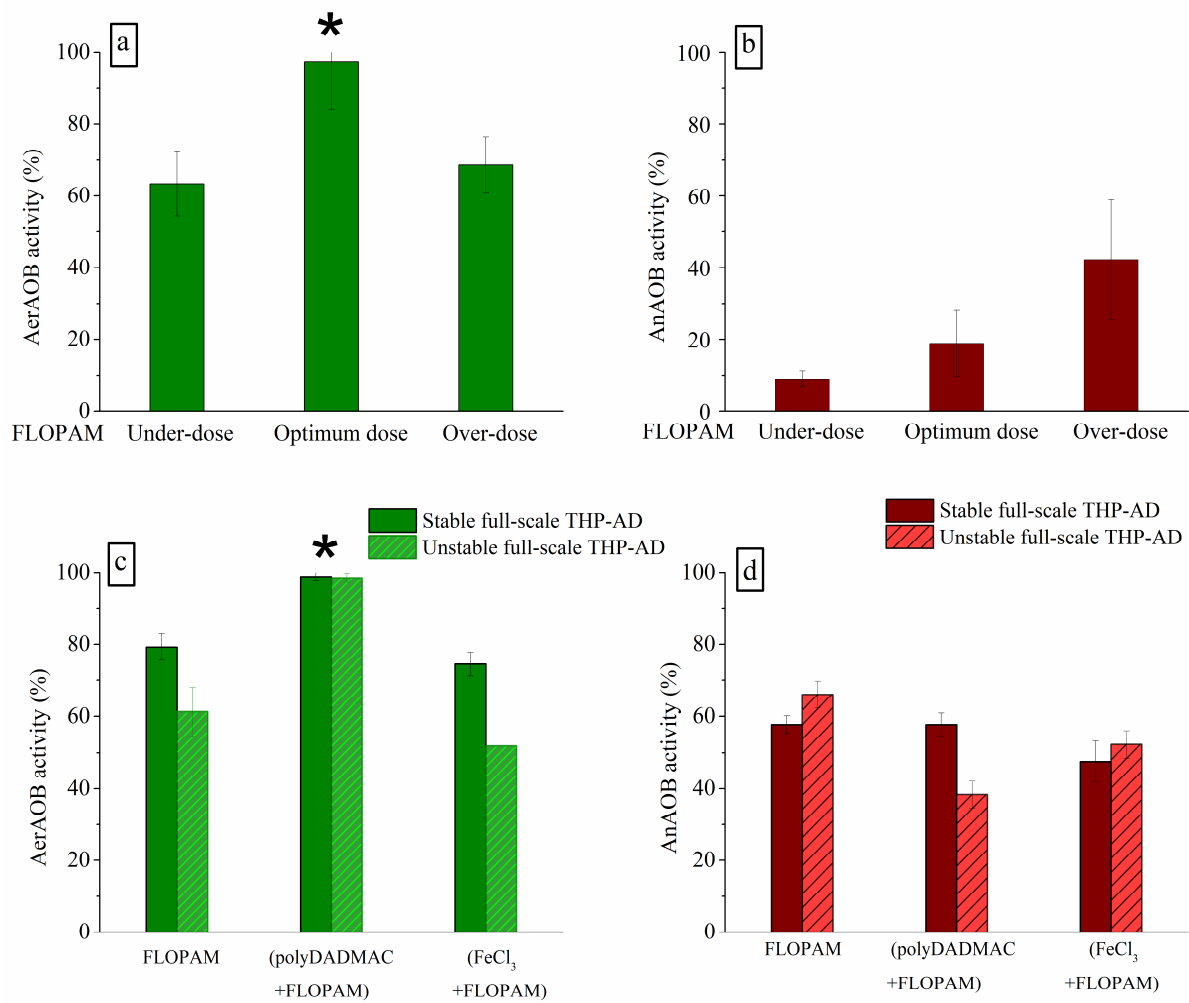
678 Table 3. Overview of two approaches for AerAOB and AnAOB parameter estimation. For AerAOB,
 679 the apparent oxygen half-saturation without addition change due to I_d was set at 0.25 mg O_2/L , and
 680 similarly for AnAOB the apparent nitrite half-saturation was set at 0.5 mg NO_2^-/L . The two direct
 681 inhibition models (Models 1a and b) do not include diffusion limitation. Model 2 incorporates
 682 diffusion limitation as additional inhibition mechanism. Firstly particulate COD and large colloidal
 683 COD were grouped together as one inhibitory compound (I_d) affecting the diffusion resistance, while
 684 soluble COD was considered as inhibitory compound without diffusion impacts (I_{nd}) (Model 2a). A
 685 second parameter estimation was performed by considering particulate and large colloidal COD as
 686 two individual diffusion-affecting compounds (I_{d1} and I_{d2} , respectively) (Model 2b). Lower values
 687 for total and mean summed errors (TSE and MSE) indicate a better goodness of fit, and significant
 688 parameter estimation improvements are obtained with F-test p values ≤ 0.05 (*).

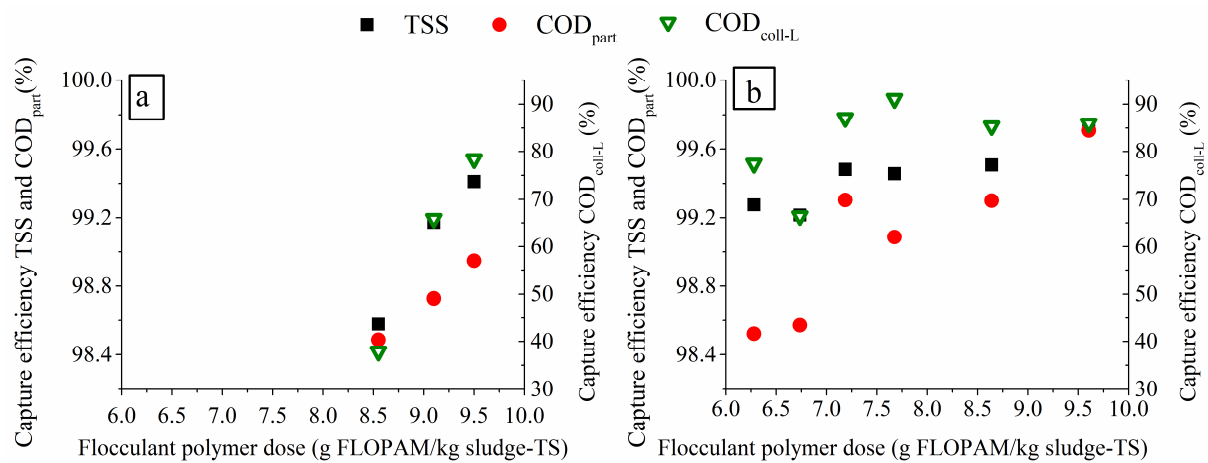
	Estimated Parameter	Value	Data range	TSE	MSE	F-test p value
AerAOB						
1 Direct inhibition model: $I = COD_{tot}$						
1a Conventional	R_{max} (mg N/g VSS/h)	25.7±0.6	4.3-30.6	900	9.4	
	K'_{I_1} (mg COD _{tot} /L)	1519±155	0-3876			
1b Minimum	R_{max} (mg N/g VSS/h)	25.3±0.7	4.3-30.6	918	9.6	
	K'_{I_1} (mg COD _{tot} /L)	1282±122	0-3876			
2 Direct and diffusional inhibition model: $I = COD_{diss} + COD_{coll-S}$						
2a Lumped diffusional inhibitor: $I_d = COD_{part} + COD_{coll-L}$	R_{max} (mg N/g VSS/h)	26.2±0.7	4.3-30.6	889	9.4	0.367* (2a vs. 1a)
	K'_{I_1} (mg COD _{diss+coll-S} /L)	826±136	0-3416			
	$K'_{I_{d1}}$ (mg COD _{part+coll-L} /L)	25.0±4.1	0-460			
2b Separate diffusional inhibitors: $I_{d1} = COD_{part}$; $I_{d2} = COD_{coll-L}$	R_{max} (mg N/g VSS/h)	26.1±1.0	4.3-30.6	813	8.6	0.035 (2b vs. 1a) 0.001 (2b vs. 2a)
	K'_{I_1} (mg COD _{diss+coll-S} /L)	890±199	0-3416			
	$K'_{I_{d1}}$ (mg COD _{part} /L)	5 E+8±4 E+21	0-253			
	$K'_{I_{d2}}$ (mg COD _{coll-L} /L)	7.6±1.3	0-207			
AnAOB						
1 Direct inhibition model: $I = COD_{tot}$						
1a Conventional	R_{max} (mg N/g VSS/h)	10.0±0.5	0-11.6	174	2.1	
	K'_{I_1} (mg COD _{tot} /L)	270±32	0-3876			
1b Minimum	R_{max} (mg N/g VSS/h)	10.0±0.5	0-11.6	176	2.1	
	K'_{I_1} (mg COD _{tot} /L)	258±30	0-3876			
2 Direct and diffusional inhibition model: $I = COD_{diss} + COD_{coll-S}$						
2a Lumped diffusional inhibitor: $I_d = COD_{part} + COD_{coll-L}$	R_{max} (mg N/g VSS/h)	10.1±0.5	0-11.6	166	2.0	0.051 (eq.6 compared to eq.4)
	K'_{I_1} (mg COD _{diss+coll-S} /L)	238±27	0-3416			
	$K'_{I_{d1}}$ (mg COD _{part+coll-L} /L)	1. E09±2. E 22	0-460			
2b Separate diffusional inhibitors: $I_{d1} = COD_{part}$; $I_{d2} = COD_{coll-L}$	R_{max} (mg N/g VSS/h)	10.1±0.9	0-11.6	166	2.0	0.124 (eq.7 compared to eq.4) 1.000 (eq.7 compared to eq.6)
	K'_{I_1} (mg COD _{diss+coll-S} /L)	238±39	0-3416			
	$K'_{I_{d1}}$ (mg COD _{part} /L)	5. E±9±4. E+22	0-253			
	$K'_{I_{d2}}$ (mg COD _{coll-L} /L)	3. E+10±2. E±21	0-207			

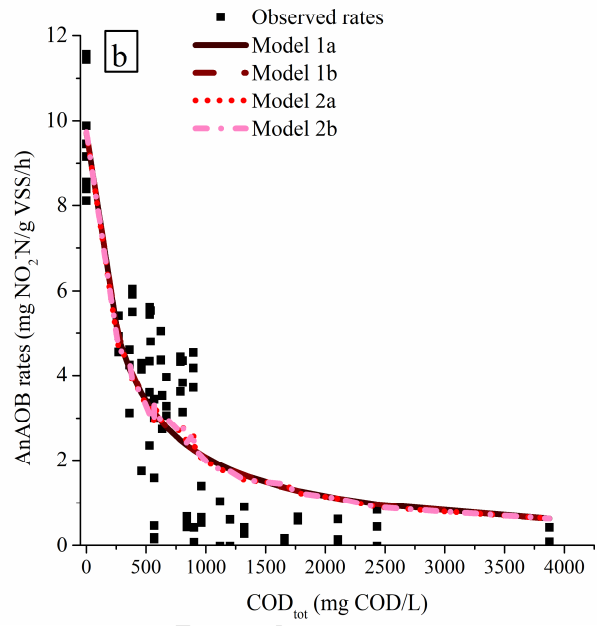
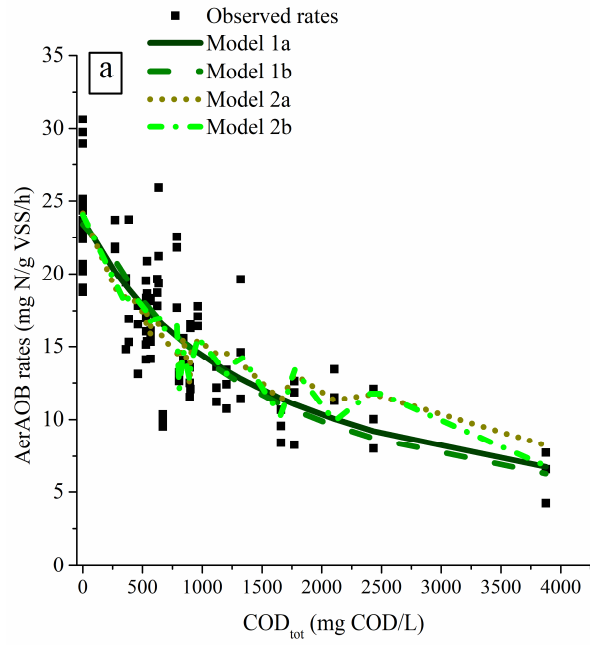
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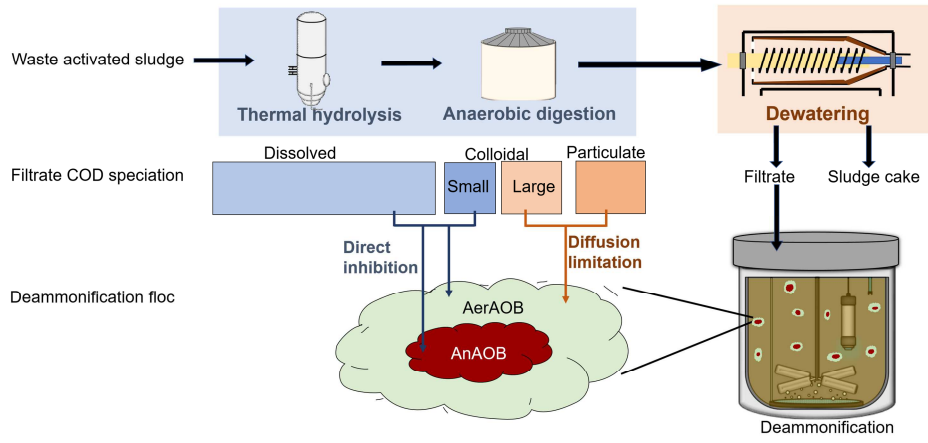












- Four organic fractions quantified in digester filtrate after thermal hydrolysis
- Organics inhibit aerobic and anoxic ammonium-oxidizing bacteria (AerAOB and AnAOB)
- Dissolved organics and small colloids directly inhibit AerAOB and AnAOB
- Large colloids create diffusion resistance indirectly limiting AerAOB activity
- Digester stability and chemically optimized dewatering facilitate deammonification