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Fluxes of water, sediments, and biogeochemical 3 compounds in salt marshes

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1 **REVIEW**

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2 **Fluxes of water, sediments, and biogeochemical**  
3 **compounds in salt marshes**

4 Sergio Fagherazzi<sup>1\*</sup>, Patricia L Wiberg<sup>2</sup>, Stijn Temmerman<sup>3</sup>, Eric Struyf<sup>3</sup>, Yong Zhao<sup>4</sup> and Peter A Raymond<sup>4</sup>

5 **Abstract**

6 Tidal oscillations systematically flood salt marshes, transporting water, sediments, organic matter, and  
7 biogeochemical elements such as silica. Here we present a review of recent studies on these fluxes and their effects  
8 on both ecosystem functioning and morphological evolution of salt marshes. We reexamine a simplified model for  
9 the computation of water fluxes in salt marshes that captures the asymmetry in discharge between flood and ebb.  
10 We discuss the role of storm conditions on sediment fluxes both in tidal channels and on the marsh platform. We  
11 present recent methods and field instruments for the measurement of fluxes of organic matter. These methods will  
12 provide long-term data sets with fine temporal resolution that will help scientists to close the carbon budget in salt  
13 marshes. Finally, the main processes controlling fluxes of biogenic and dissolved silica in salt marshes are explained,  
14 with particular emphasis on the uptake by marsh macrophytes and diatoms.

15 **Introduction**

16 The exchange of sediments, silica, and organic matter  
17 between terrestrial ecosystems and the ocean is an important  
18 theme in ecological and biogeochemical studies. As a  
19 widely distributed coastal ecosystem covering 200,000–  
20 400,000 km<sup>2</sup> globally, salt marshes are highly productive  
21 [net ecosystem production (NEP) = 1,585 g C m<sup>-2</sup> year<sup>-1</sup>],  
22 rivaling tropical rainforests, due to abundant solar radiation,  
23 water, and nutrients provided by tidal water (Chmura et al.  
24 2003; Duarte et al. 2005; Cai 2011). Tidal oscillations repeatedly  
25 flood salt marshes, giving rise to fluxes that bring water  
26 and constituents in and out of the system (Fagherazzi  
27 et al. 2012). This review focuses on fluxes of sediments, organic  
28 matter, and silica and their effects on both ecosystem  
29 functioning and morphological evolution of salt marshes.  
30 Particular attention is devoted to the feedbacks between  
31 physical and biological processes.

32 Salt marsh formation and persistence are dependent  
33 on the net flux of suspended sediment onto the marsh  
34 platform (Kirwan et al. 2010; Fagherazzi et al. 2012) and  
35 loss of salt marsh during storms (Mariotti and Fagherazzi  
36 2010; Tonelli et al. 2010). Interactions between tidal flows  
37 and marsh vegetation promote deposition of sediment  
38 delivered to the marsh platform during flooding tides.

Sediment deposition, along with accumulation of autochthonous  
organic material, contributes to the vertical accretion necessary  
for a marsh surface to maintain its elevation above mean sea level  
as sea level rises (Reed 1989; Morris et al. 2002; Mudd et al. 2009).  
Thus, healthy salt marshes must be net sediment sinks.

The buried carbon in the sediment of salt marshes as well as  
other intertidal wetlands, which is called “blue carbon,” is a  
significant carbon sink, sequestering in excess of 4.8–87.2 Tg C  
year<sup>-1</sup> globally according to different estimations (Chmura et al.  
2003; Duarte et al. 2005; McLeod et al. 2011). However, viewed  
in light of the global NEP of salt marshes of 634 Tg C year<sup>-1</sup>, only  
a small portion is retained in marsh sediments, and most of the  
produced biomass decomposes or is exported via lateral tidal  
fluxes (Middelburg et al. 1997; Duarte et al. 2005; Sharitz and  
Pennings 2006; Hopkinson et al. 2012).

In addition to carbon, other nutrients are deposited and recycled  
in coastal salt marshes. Of these, silicon (Si) is an essential  
element for primary production by diatoms in estuaries and  
coastal seas (Struyf and Conley 2009). In this paper we give an  
overview of studies that demonstrate the importance of tidal  
marshes in the associated estuarine Si cycle. In particular, we  
discuss the present state of knowledge regarding Si fluxes  
between tidal marshes and adjacent estuarine and coastal waters.

The paper is organized as follows: in the section “Fluxes of  
water,” we describe fluxes of water in salt marshes and

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67 put forward a simplified model for their computation. In  
 68 “Fluxes of sediments,” we focus on fluxes of sediments  
 69 both in tidal channels and on the marsh platform. In  
 70 “Fluxes of organic material,” we discuss novel techniques  
 71 to measure fluxes of organic matter, and in “Fluxes of silica,”  
 72 we present a review of recent results on fluxes of silica  
 73 in salt marshes. A set of conclusions and future research  
 74 needs closes this review in the section “Conclusions.”

### 75 Fluxes of water

76 The amount of water that enters and exits a marsh in a  
 77 tidal cycle is the tidal prism, which can be computed, to  
 78 a first approximation, as the volume of water contained  
 F1 79 in the marsh between high and low tide (Figure 1A), and  
 80 it can be written as (D’Alpaos et al. 2010):

$$P = \int_A [h_H - \max(z, h_L)] dA \quad (1)$$

81 where  $A$  is the marsh area,  $h_L$  is the water elevation at  
 82 low tide,  $h_H$  the water elevation at high tide, and  $z$  the  
 83 elevation of the marsh surface.

84 When tidal channels are not present, the transport of  
 85 water between marsh and the surrounding coast occurs  
 86 as a sheet flow on the marsh surface with very low water  
 87 depths (French et al. 1995; Temmerman et al. 2005).  
 88 This shallow flow favors the incision of creeks that, once  
 89 scoured, capture more flow in a positive feedback that  
 90 produces a network of dendritic channels dissecting the  
 91 marsh platform (Fagherazzi and Furbish 2001; Fagherazzi  
 92 and Sun 2004; D’Alpaos et al. 2005).

93 The water is thus carried to the marsh during flood  
 94 and subsequently drained during ebb in two steps: dur-  
 95 ing flood, the water first moves at relatively high speed

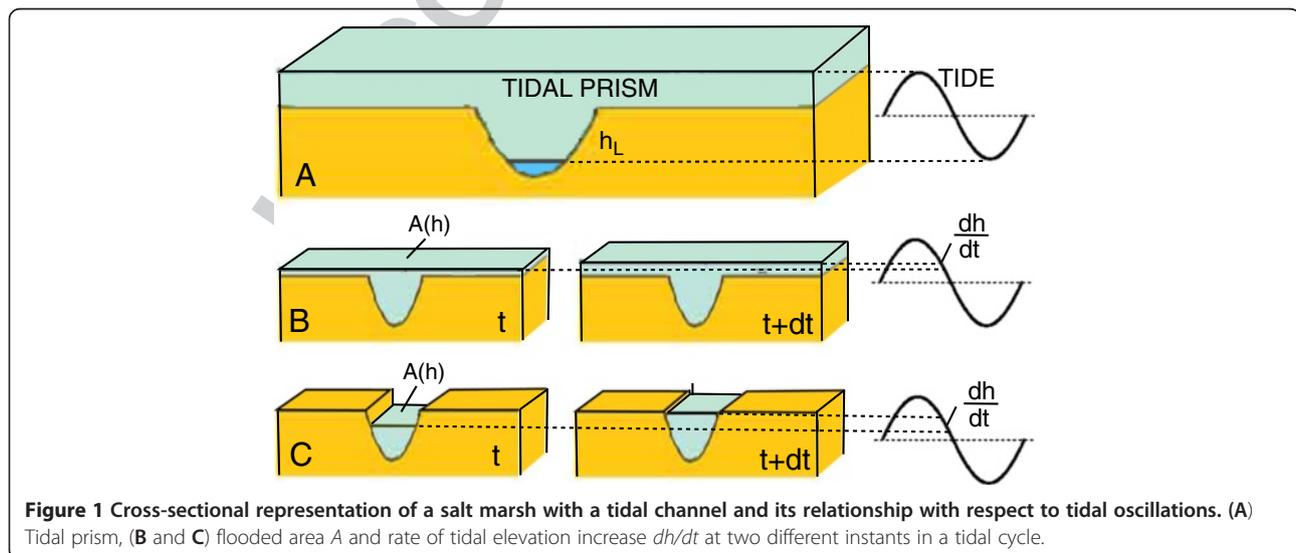
in the channel network and then it spills on the platform 96  
 (e.g., Christiansen et al. 2000; Davidson-Arnott et al. 97  
 2002; Temmerman et al. 2012). During ebb the opposite 98  
 occurs, with the water first drained on the platform and 99  
 then collected in the channels. As a result the morpho- 100  
 logic of the marsh and the distribution of channels 101  
 have a strong influence on water fluxes. 102

If the total volume of water entering or exiting the 103  
 marsh is regulated by the tidal prism, the discharge de- 104  
 pends also on the speed at which this water is delivered 105  
 to or collected from the marsh. The rate of flooding 106  
 (drainage) depends in the first place on the tidal oscilla- 107  
 tion (see Figure 1B,C). In fact, by taking the temporal 108  
 derivative of Equation 1, we obtain: 109

$$Q = A(h) \frac{dh}{dt} \quad (2)$$

where  $A(h)$  is the hypsometric curve (the surface flooded 110  
 at each elevation  $h$ ),  $Q$  is the discharge, and  $dh/dt$  the 111  
 derivative of the tidal oscillation. Here for simplicity we 112  
 assume that the water fluxes in and out of the marsh are 113  
 concentrated in the channels, although a considerable 114  
 volume of water is transported as sheet flow from the 115  
 marsh edge directly on the marsh platform (French et al. 116  
 1995; Temmerman et al. 2005). 117

This equation is defined as the static model (tub 118  
 model) and was first presented by Boon (1975). Since 119  
 then it has been extensively used in salt marsh hydro- 120  
 dynamics (e.g., D’Alpaos et al. 2006, 2007, 2010; Lawrence 121  
 et al. 2004). This model implies that when there is a sharp 122  
 increase in flooded area  $A$ , the discharge surges because 123  
 more water is needed to flood a larger surface (or more 124  
 water must be removed to drain it; see Figure 1B,C). As a 125



**Figure 1** Cross-sectional representation of a salt marsh with a tidal channel and its relationship with respect to tidal oscillations. (A) Tidal prism, (B and C) flooded area  $A$  and rate of tidal elevation increase  $dh/dt$  at two different instants in a tidal cycle.

126 result, two distinct surges are occurring in the channels,  
127 one when the marsh platform is first flooded and one  
128 when the marsh platform is drained. These surges have  
129 been measured in salt marsh channels by several authors  
130 (Myrick and Leopold 1963; Bayliss-Smith et al. 1978;  
131 Healey et al. 1981; French and Stoddart 1992).

132 However, the static model implies that the water is in-  
133 stantaneously delivered to the entire marsh, when in real-  
134 ity it takes a finite time to reach each marsh location,  
135 depending on the travel path both within the channels  
136 and on the marsh platform. A modification of this model  
137 accounting for the delay due to the travel time was pro-  
138 posed by Fagherazzi et al. (2008), with the TIGER model:

$$Q = \int_0^t A(h) \frac{dh}{dt} f(t - \tau) d\tau \quad (3)$$

139 where  $f$  is the distribution of travel time that can be com-  
140 puted by dividing the path length that a parcel of water  
141 needs to travel to reach each location by a characteristic  
F2 142 velocity of the flow (see examples in Figure 2). In particu-  
143 lar, the path followed by the water on the marsh platform  
144 can be computed using the method presented in Rinaldo  
145 et al. (1999).

146 It is important to note that the mean path length  
147 (unchannelized length) describes the drainage density of  
148 the tidal network (Marani et al. 2003). Equation 3 there-  
149 fore links water discharge to drainage density. The intro-  
150 duction of a delay results in a peak discharge (surge) in  
151 flood occurring after the platform is flooded; in fact it  
152 takes time for the water to reach the farthest marsh lo-  
153 cations and flood them. Similarly, the ebb surge takes  
154 place after the drainage of the marsh for a water eleva-  
155 tion just below the marsh surface, and it is therefore  
F3 156 confined in the channels (Figure 3). The stage-discharge  
157 and stage-velocity curves therefore present an asymmetry



**Figure 2** Two examples of water paths in a salt marsh, computed on the basis of water surface gradients (e.g., Rinaldo et al. 1999), and related travelling times. During flood, the water moves first in the channels and then on the marsh surface. The travelling times for all marsh locations need to be computed in order to determine the distribution  $f$ .

as observed in the field by many researchers (Pethick 158  
1980; Healey et al. 1981). 159

The introduction of a delay in the static model of 160  
Boon (1975) does not considerably change the magni- 161  
tude of the peak discharge but has a strong impact on 162  
velocities, which are obtained by dividing the discharge 163  
by water depth. If the ebb surge takes place at lower water 164  
depths, then the velocity increases, resulting in the charac- 165  
teristic ebb dominance of marsh creeks (Friedrichs and 166  
Aubrey 1988). Since bottom shear stress scales with the 167  
square of the velocity, this flow asymmetry favors erosion 168  
and incision during ebb. 169

Moreover, the distribution of travel times contains in- 170  
formation about salt marsh morphology and related eco- 171  
systems. For example, deep channels with high drainage 172  
density reduce the travel time of water parcels, while 173  
thick halophyte vegetation on the marsh platform slows 174  
the flow (Mudd et al. 2004; Temmerman et al. 2007). In 175  
fact plant stems and leaves convert kinetic energy into 176  
turbulent kinetic energy, increasing drag and reducing flow 177  
velocity (Nepf 1999). Temmerman et al. (2012) showed 178  
that after cutting marsh vegetation over a relatively large 179  
area (4 ha) the flow velocity on the platform increased by a 180  
factor of 2 to 4, thus decreasing dramatically the travel 181  
time of water parcels. 182

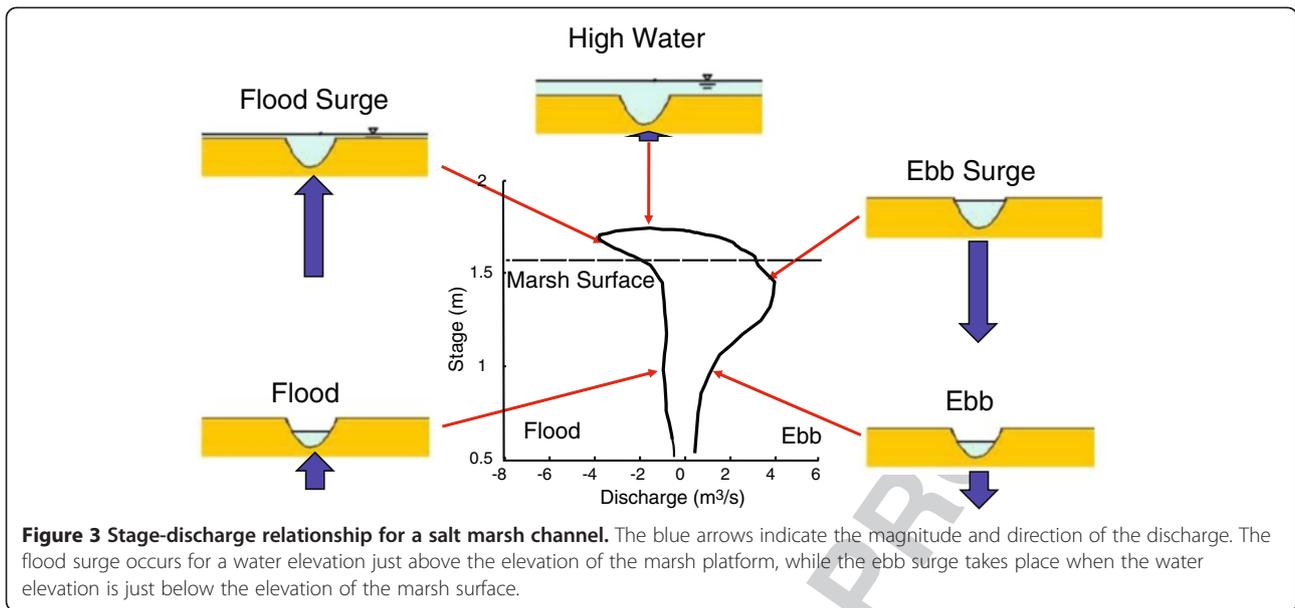
The distribution of travel time is identical to the distri- 183  
bution of residence time of the water in the marsh, and 184  
it is therefore of extreme importance for biogeochemical 185  
cycles (Church 1986; Crump et al. 2004). The TIGER 186  
model (Equation 3 and Figure 4) represents an elegant 187 F4  
method to determine the distribution of residence time 188  
from flow measurements in a channel cross section. It 189  
can also be used to detect shallow subsurface flows 190  
that move under the marsh platform and exit at the 191  
channel banks. In fact the long tail in the distribution 192  
of residence times represents such flows, which can 193  
therefore be isolated from the tidal surface water. These 194  
flows are very important for marsh biogeochemistry 195  
since they are rich in solutes and sediments (Wilson 196  
and Morris 2012). 197

#### Fluxes of sediments 198

In low marsh environments, where mineral sedimenta- 199  
tion is greatest, daily tides inundate the marsh sur- 200  
face. Sediment suspended in the water flooding a low 201  
marsh can be the product of wave-driven resuspension 202  
in adjacent bays or tidally driven resuspension in 203  
tidal creeks. 204

#### Sediment fluxes in tidal channels 205

Most sediments enter the marsh through tidal channels. 206  
Similarly to the water fluxes discussed in the previous 207  
section, we can focus on the sediment fluxes entering 208  
and exiting a channel cross section and use them to 209

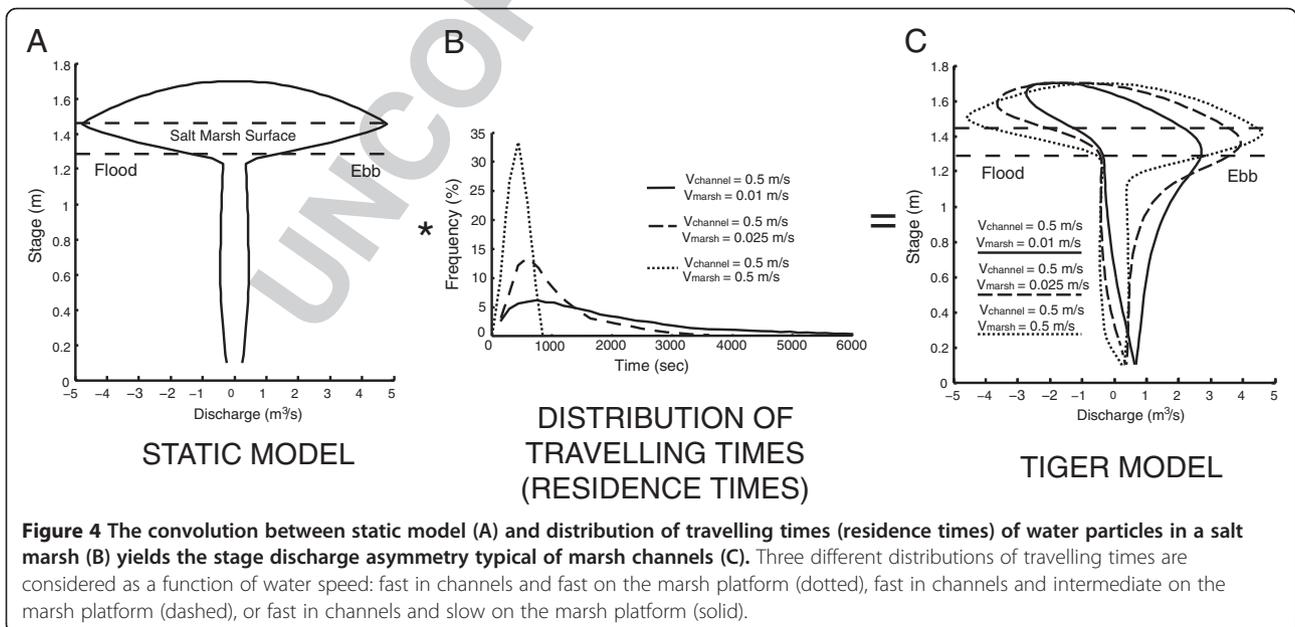


- 210 determine the total sediment budget of a salt marsh.  
 211 The sediment input in a marsh can be expressed as:

$$V_S = \int_{t_{low}}^{t_{high}} C_s Q dt \quad (4)$$

- 212 where  $V_S$  is the total volume of sediments entering the  
 213 channels during flood,  $C_s$  is the sediment concentration  
 214 in the water column,  $t_{low}$  is the instant of low-tide slack  
 215 water and  $t_{high}$  is the instant of high-tide slack water.  
 216 Equation 4 shows that two functions are essential to  
 217 import large amounts of sediments in a marsh: a large  
 218 water discharge during flood and, more importantly,

high concentrations of suspended sediment. These sedi- 219  
 ments need to be in suspension near the channel en- 220  
 trance, so that when the tide is rising they are conveyed 221  
 in the channel and subsequently deposited on the marsh 222  
 platform. Wind waves and tidal currents are ideal pro- 223  
 cesses for sediment resuspension and transport in areas 224  
 adjacent to salt marshes. Waves are particularly effective 225  
 in remobilizing sediments on tidal flats, where shallow 226  
 water depths result in large wave-induced bottom shear 227  
 stresses (Carniello et al. 2005; Fagherazzi et al. 2007). 228  
 Measurements along the muddy Louisiana coastline 229  
 show that the concentration of sediments entering a 230  
 marsh channel is proportional to the significant wave 231



232 height in the bay in front of the channel (Fagherazzi and  
F5 233 Priestas 2010; see Figure 5A).

234 A storm surge therefore represents an ideal event for  
235 sediment input in a marsh (e.g., Cahoon 2006; Reed  
236 1989; Turner et al. 2006): the strong wind associated  
237 with the storm produces waves that resuspend fine sedi-  
238 ments in front of marshes, and the wind and wave setup  
239 increases the maximum tide level and thereby the water  
240 discharge during flood in the channels (by increasing the  
241 tidal prism; see Equations 1, 2). Therefore during a storm  
242 surge both discharge and sediment concentrations of the  
243 entering water are magnified, augmenting the total vol-  
244 ume of sediment imported in the marsh (Equation 4).

245 This mechanism is well accepted in marsh studies, with  
246 several authors indicating storm conditions as ideal for  
247 marsh accretion (e.g., Reed 1989). However, sediment input  
248 during flood represents only half of the sediment budget.  
249 In fact sediments are exiting the marsh through the chan-  
250 nels during the subsequent ebb flow. The same physics  
251 governs the export of sediments, and Equation 4 is still  
252 valid but now the integral is evaluated from  $t_{\text{high}}$  to  $t_{\text{low}}$ .

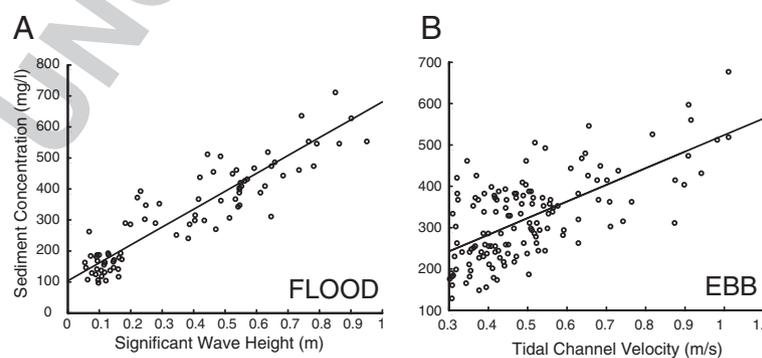
253 The main difference is that the sediment concentration  
254 of the exiting flow is now independent of the hydro-  
255 dynamic conditions in the adjacent bays and tidal flats  
256 but is governed by physical processes mobilizing sedi-  
257 ments within the salt marsh. Wind waves are negligible  
258 on marsh platforms, given the shallow water depths and  
259 the damping effect of vegetation (Möller et al. 1999).  
260 Therefore only tidal currents triggering high velocities  
261 are potentially responsible for sediment mobilization on  
262 the marsh surface during ebb, although field evidence  
263 seems to indicate that this effect is limited. Most of the  
264 sediment is therefore eroded and resuspended from the  
265 channels bottom and banks, where the velocities are  
266 high and vegetation does not protect the substrate. Data  
267 from Louisiana confirm that sediment concentration in  
268 the water exiting a marsh during ebb is proportional to

the tidal velocity in the channel (Figure 5B) (Fagherazzi 269  
and Priestas 2010). 270

271 The discharge asymmetry outlined in the previous sec-  
272 tion enhances sediment erosion during ebb, since the  
273 peak in discharge occurs in the channels for shallow  
274 water depths thus augmenting flow velocity and bottom  
275 shear stresses (Figure 3). Storm surges are therefore ex-  
276 cellent events also for the export of sediments, since all  
277 the water accumulated in the marsh during the surge  
278 will exit the system, giving rise to high ebb velocities and  
279 bottom stresses.

280 Moreover, given the low settling velocity of the fine  
281 material typical of marshlands, some of the sediment  
282 that was imported during flood might remain in suspen-  
283 sion and exit the system during the following ebb. This  
284 is particularly true when the residence time of the water  
285 in the marsh is low, such as in marshes of small dimen-  
286 sions with a high drainage density (limited water paths),  
287 or when the water velocity is high and thus the travel time  
288 is short (large advection velocity). Again storm surges, by  
289 increasing flow velocities, decrease the likelihood of par-  
290 ticle settling during high tide, reducing the potential accu-  
291 mulation of sediments on the marsh platform.

292 Best conditions for sediment accumulation are moder-  
293 ate storms that increase sediment resuspension near  
294 salt marshes, but don't trigger fast flows in the channels  
295 (Fagherazzi and Priestas 2010). On the contrary, for  
296 sediment export the worst conditions occur during me-  
297 teorological low tides (setdowns), when wind blows  
298 water away from the coastline and a large volume of  
299 water exits the marsh system. The setdown leads to  
300 shallow water depths, which, combined with large dis-  
301 charges, results in velocities that erode sediments in the  
302 channels. The low water levels also trigger large gradi-  
303 ents in groundwater pressure leading to seepage and  
304 piping at channel banks (Gardner 2005; Howes and  
305 Goehringer 1994).



**Figure 5 Relationships between sediment concentration and (A) significant wave height during flood and (B) tidal channel velocity during ebb in a marsh channel in Louisiana, USA.** The wave height was measured in the bay just in front of the tidal channel (after Fagherazzi and Priestas 2010).

### 306 Sediment fluxes on the marsh platform

307 Regardless of how large the wave and tidal bed shear  
308 stresses are in the adjacent bays and creeks, interactions  
309 between the flow and marsh vegetation reduce the ve-  
310 locities and turbulence in the flooding flows within a  
311 short distance of entering the vegetation (Leonard and  
312 Luther 1995; Christiansen et al. 2000), creating condi-  
313 tions favorable for deposition of the sediment in sus-  
314 pension. In addition, reductions in near-bed velocities  
315 and drag around plant stems lead to skin friction  
316 stresses that are almost always insufficient to entrain  
317 sediment from the marsh surface. Therefore the dy-  
318 namics of sediment transport on the marsh surface are  
319 dominated by slow advection during tidal inundation,  
320 accompanied by particle settling and interception by  
321 vegetation stems.

322 A very simple model is useful for illustrating the fac-  
323 tors controlling suspended sediment concentrations and  
324 deposition on the marsh platform. The model can be  
325 obtained by considering an advection-dispersion equa-  
326 tion governing the dynamics of suspended sediments  
327 over the marsh platform and by neglecting the advective  
328 and dispersive terms so that the variation in time of the  
329 concentration equals the settling rate. Assuming that the  
330 water depth and flow velocity are constant in space and  
331 time and that the water column is well mixed, the evolu-  
332 tion in concentration,  $C_s$ , within the flooding tidal water  
333 can be expressed as:

$$C_s = C_{s0} \exp[-(w_s/h)t] = C_{s0} \exp[-w_s x/(Uh)] \quad (5)$$

334 where  $w_s$  is settling velocity,  $h$  is flow depth,  $C_{s0}$  is initial  
335 concentration,  $t$  is time,  $x$  is distance from the creek  
336 bank or marsh edge, and  $U$  is average flow velocity on  
337 the marsh. Deposition rate is given by the divergence of  
338 sediment flux,  $-d(C_s U h)/dx$ . It is clear from Equation 5  
339 that concentration and hence deposition rate decrease  
340 along the flow path on the marsh platform and therefore  
341 with distance from the marsh edge (e.g., French et al.  
342 1995; Leonard 1997; Reed et al. 1999; Temmerman et al.  
343 2003). In fact the flow path on the marsh platform often  
344 coincides with the direction perpendicular to the marsh  
345 edge (Rinaldo et al. 1999; Marani et al. 2003). Peak flow  
346 velocities on the marsh surface tend to be small and do  
347 not vary much with tidal elevation or location on the  
348 marsh. Therefore settling velocity and flow depth are the  
349 primary controls of how much of the sediment entering  
350 the marsh is deposited.

351 These processes can be illustrated with an example  
352 from a microtidal (mean tidal range = 1.2 m) low marsh  
353 site in coastal Virginia, USA (Christiansen et al. 2000).  
354 Peak flow velocities at the site average  $\sim 0.2$  cm/s during  
355 rising and falling tides. For tidal amplitudes between  
356 0.95 and 1.05 m ( $h_{\max} \approx 0.6$  m), average peak suspended

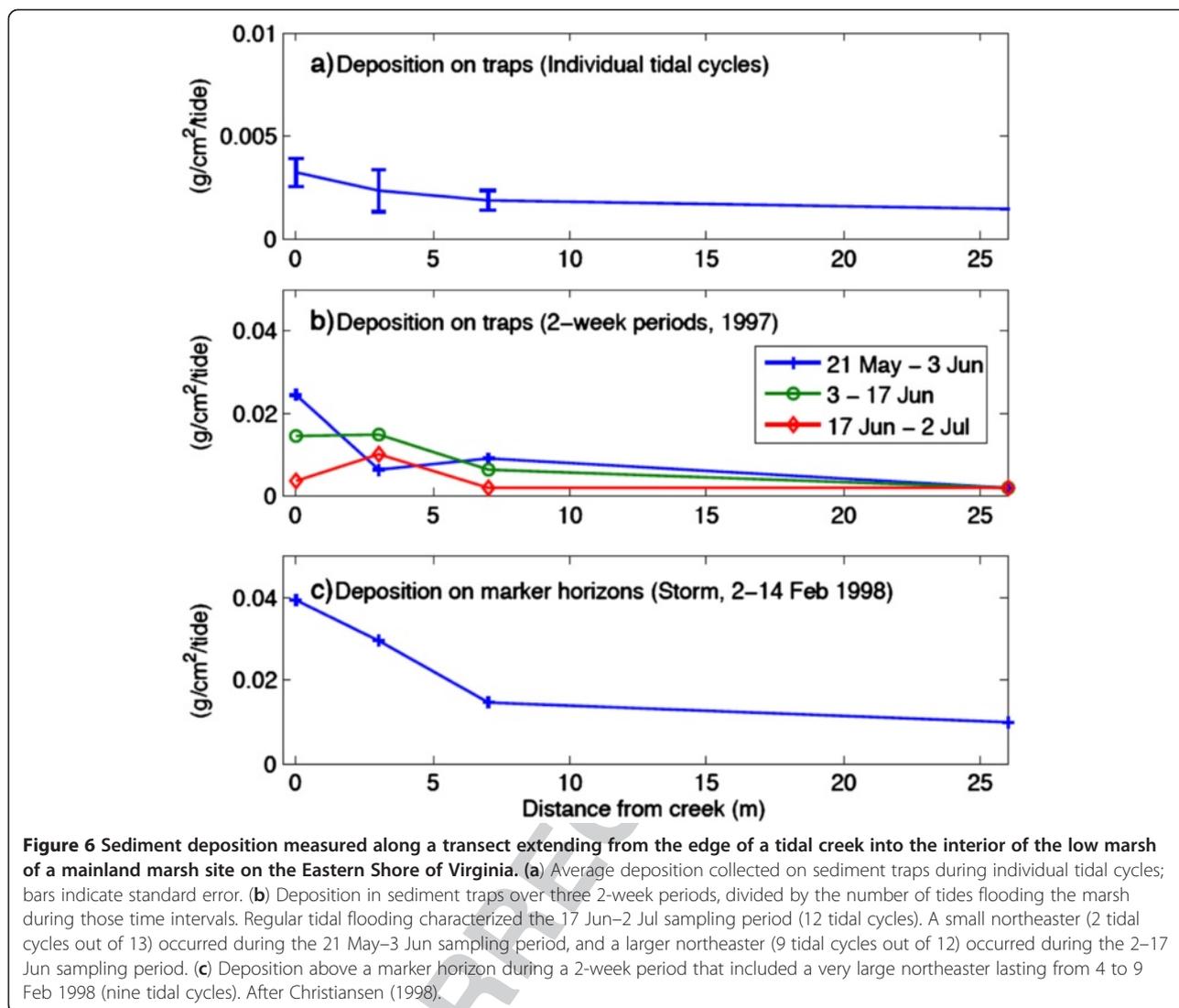
357 sediment concentration entering the marsh platform 357  
358 during flood is 85 mg/L, which decreases by roughly 358  
359 20% 3 m into the marsh (Christiansen et al. 2000). This 359  
360 rate of decrease is consistent with an average settling 360  
361 velocity of roughly 0.1 mm/s, the Stoke's settling rate for 361  
362 particles of  $\sim 12$   $\mu\text{m}$  diameter. Grain size distributions of 362  
363 marsh surface sediment near the marsh edge have a 363  
364 modal size of 15–20  $\mu\text{m}$  (Christiansen et al. 2000). 364

365 The apparent agreement between modal bed grain size 365  
366 and estimated settling rate indicated above proves some- 366  
367 what misleading when size distribution and its evolution 367  
368 across the marsh surface are taken into consideration. 368  
369 First, analysis of size distribution indicates that 70–75% 369  
370 of the sediment deposited within 10 m of the creek bank 370  
371 is deposited as flocs (Christiansen et al. 2000). Second, 371  
372 there was no significant grain size fining with distance 372  
373 from the creek bank, as would be expected from a 373  
374 simple application of Equation 1 to a disaggregated grain 374  
375 size distribution. Only if deposition is dominated by flocs 375  
376 can the grainsize distribution of deposited sediment 376  
377 remain largely unchanged along the flow path. 377

378 Voulgaris and Meyers (2004) measured settling veloci- 378  
379 ties and particle characteristics in a tidal channel in 379  
380 South Carolina, USA, and on the adjacent marsh. They 380  
381 found that suspended particles in the channel were pre- 381  
382 dominantly flocculated, with average settling velocities 382  
383 of about 0.1 mm/s during neap tides (mean diameter of 383  
384 50  $\mu\text{m}$ ) and 0.25 mm/s during spring tides (estimated 384  
385 mean diameters of 50–150  $\mu\text{m}$ ) in the tidal channel and 385  
386 an average settling velocity of 0.24 mm/s on the marsh. 386  
387 Moskalski and Sommerfield (2012) measured particle 387  
388 characteristics on a salt marsh in Delaware and also 388  
389 found that flocs with median grain sizes of 50–150  $\mu\text{m}$  389  
390 predominated in marsh suspended sediment. 390

391 Sediment deposition measured on a Virginia coastal 391  
392 marsh site (Christiansen 1998) averaged 0.004 g cm<sup>-2</sup> 392  
393 per tide near the tidal creek over a range of non-storm 393  
394 tidal conditions (Figure 6). This value is a little lower 394  
395 than deposition measured near a creek bank along a 395  
396 Delaware salt marsh of  $\sim 0.01$  g cm<sup>-2</sup> per tide (Moskalski 396  
397 and Sommerfield 2012, assuming two tides per day). 397

398 Wind-enhanced tidal and wave-generated bed shear 398  
399 stresses result in higher concentrations of suspended 399  
400 sediment in the water flooding the marsh (Christiansen 400  
401 et al. 2000; Lawson et al. 2007; Fagherazzi et al. 2010) 401  
402 and longer inundation times. As a result, deposition 402  
403 rates are higher and deposition extends further into 403  
404 the marsh during storm tides (e.g., Cahoon 2006; Reed 404  
405 1989; Turner et al. 2006). For example, measured de- 405  
406 position near the creek bank at the Virginia coastal 406  
407 marsh site increased by roughly an order of magnitude 407  
408 during a period that included a significant northeaster 408  
409 (Christiansen 1998; Figure 6). At that site, northeasterly 409  
410 winds occurred during 11% of the tidal cycles but were 410



**Figure 6** Sediment deposition measured along a transect extending from the edge of a tidal creek into the interior of the low marsh of a mainland marsh site on the Eastern Shore of Virginia. (a) Average deposition collected on sediment traps during individual tidal cycles; bars indicate standard error. (b) Deposition in sediment traps over three 2-week periods, divided by the number of tides flooding the marsh during those time intervals. Regular tidal flooding characterized the 17 Jun–2 Jul sampling period (12 tidal cycles). A small northeaster (2 tidal cycles out of 13) occurred during the 21 May–3 Jun sampling period, and a larger northeaster (9 tidal cycles out of 12) occurred during the 2–17 Jun sampling period. (c) Deposition above a marker horizon during a 2-week period that included a very large northeaster lasting from 4 to 9 Feb 1998 (nine tidal cycles). After Christiansen (1998).

411 estimated to contribute 27% of the annual deposition  
 412 (Christiansen 1998).

413 Estimates of annual accretion based on short-term mea-  
 414 surements of marsh deposition in the Virginia salt marsh  
 415 site ( $\sim 1 \text{ g cm}^{-2} \text{ year}^{-1}$ , equivalent to about 1 cm/year as-  
 416 suming a bed porosity of about 0.65) exceed rates based on  
 417 longer-term measurements (e.g., SETs) and rates of sea-  
 418 level rise, suggesting that the marsh surface is subject to  
 419 erosion as well as deposition (Christiansen 1998). Because  
 420 bed shear stresses due to tidal flows on the marsh are too  
 421 low to erode sediment, it is likely that erosion occurs  
 422 via another process. Studies in a South Carolina marsh  
 423 (Mwamba and Torres 2002; Torres et al. 2003) showed  
 424 that rainfall can mobilize large quantities of sediment on a  
 425 saturated, exposed marsh surface and suggest that low-tide  
 426 rainfall runoff driven by marsh topography may be the pri-  
 427 mary erosional process on intertidal salt marshes.

#### Fluxes of organic material

428  
 429 In the past years several studies have focused on the  
 430 storage of organic carbon in salt marshes (Morris and  
 431 Bowden 1986; Chmura et al. 2003; Hussein et al. 2004;  
 432 Choi and Wang 2004). Mudd et al. (2009) presents a  
 433 combined framework including both refractory and la-  
 434 bile organic fractions, as well as root growth and decay.  
 435 The model shows that the amount of carbon stored  
 436 in salt marshes depends on the competition between  
 437 mineral sediment deposition and organic-matter accu-  
 438 mulation. Direct warming of the marsh due to climate  
 439 change favors aboveground vegetation production, while  
 440 sea level rise leads to more accommodation space,  
 441 thus increasing rates of carbon burial (Kirwan and  
 442 Mudd 2012).

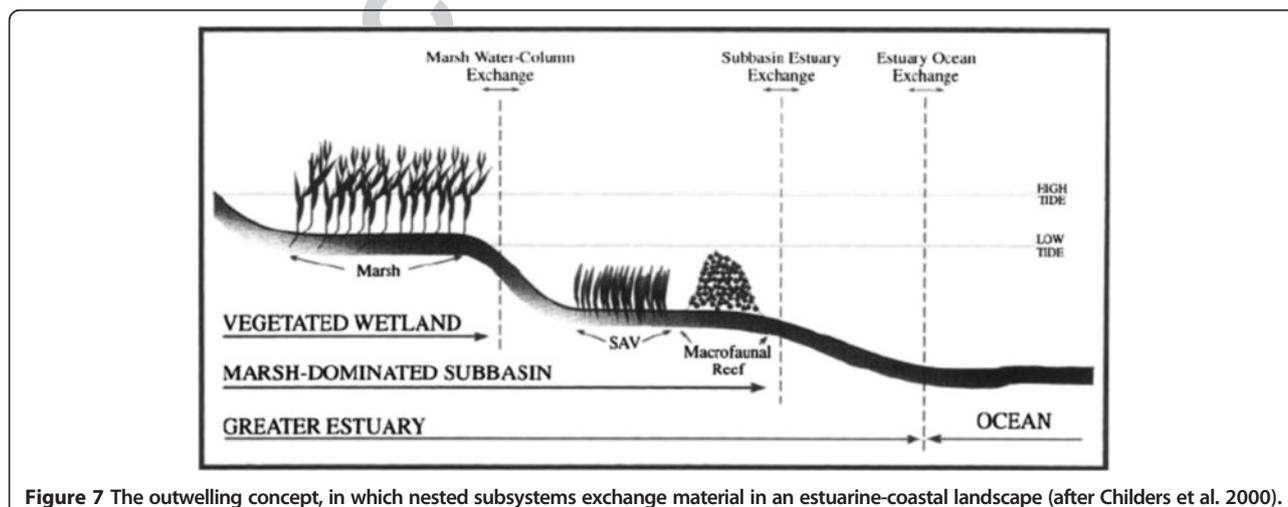
443 However, little is known about tidal fluxes of dissolved  
 444 and particulate carbon, which could dramatically affect

445 the carbon budget of salt marshes. Tidal inundation  
446 brings estuarine organic matter to salt marshes via tidal  
447 creeks, and through biological, biogeochemical, and  
448 physical processes, the exported organic matter is altered  
449 quantitatively and qualitatively (Hemminga et al. 1992,  
450 1993). Initial studies on lateral fluxes of organic carbon  
451 mainly focused on the volume of carbon exchanged be-  
452 tween salt marsh and estuary. Teal (1962) assessed the  
453 energy budget of Sapelo Island salt marsh and found a  
454 45% loss of production to the estuarine water. Although  
455 Teal's estimate was not based on direct flux measure-  
456 ments, ecologists have shown great interest in this  
457 hypothesis and many studies have been conducted to as-  
458 sess and develop this idea. Odum (1968) described the  
459 export of nutrients and organic detritus from salt  
460 marshes to coastal areas for supporting biotic activities  
F7 461 as "outwelling" (Figure 7). During the four decades since  
462 1970, more than 50 papers have been published develop-  
463 ing this hypothesis. Among the studies available, only a  
464 few of them directly measured the flux of organic car-  
465 bon transported by tidal water due to the difficulty of  
466 sampling water flow and constituent concentrations in  
467 tidal channels if the estuary is multi-channeled and  
468 over-flooded in spring tide (Dankers et al. 1984; Dame  
469 et al. 1986; Childers et al. 2000). In addition, in-situ  
470 continuous measurements of concentration were not  
471 available for many important constituents until field  
472 UV spectrometer and fluorometers were developed in  
473 recent years to measure nitrate, fluorescing dissolved or-  
474 ganic matter (FDOM) colored dissolved organic matter  
475 (CDOM), chlorophyll, and turbidity. So far the largest  
476 data set available is from a long-term material flux mea-  
477 surement in Oyster Landing, North Inlet, South Carolina,  
478 USA (Gardner and Kjerfve 2006). The data set started in  
479 1993 and has meteorological data at 30 min intervals as  
480 well as 13 water samples taken at 2 h intervals every 20

481 days over a period of 13 years. The finding that dissolved  
482 organic nitrogen (DON) and dissolved organic carbon  
483 (DOC) are significantly exported is consistent with sev-  
484 eral previous flux studies at North Inlet.

485 There is general consensus that lateral fluxes of or-  
486 ganic carbon from salt marshes to the coastal ocean are  
487 significant (Childers et al. 2000; Odum 2000; Valiela  
488 et al. 2000). Globally, export of organic carbon from salt  
489 marshes ranges from 27 to 1,052 g C m<sup>-2</sup> year<sup>-1</sup> (Alongi  
490 1998). Coastal geomorphology, geophysics, and hydrology  
491 are important determinants of organic matter ex-  
492 change between coastal wetlands and estuaries (Bianchi  
493 2007). In addition, from the perspective of stable isotope  
494 measurements and modeling, it has been shown that  
495 marsh DOC outwelling dominates particulate organic  
496 carbon (POC) outwelling (Eldridge and Cifuentes 2000).  
497 This conclusion is consistent with isotopic analysis sug-  
498 gesting that old, refractory organic carbon is dominant  
499 in riverine POC while salt marshes export more labile  
500 organic matter (Raymond and Bauer 2001; del Giorgio  
501 and Davis 2003).

502 Compared to quantitative studies of organic matter ex-  
503 change between marsh and ocean, a small number of or-  
504 ganic matter quality studies have been conducted so far.  
505 The quality of material flux could be interpreted as the  
506 identity (source) and functionality (fate) of the compo-  
507 nents within the pool of estuarine organic material. To  
508 identify the source of organic matter, researchers have  
509 used radioactive carbon and stable isotopic signatures of  
510 carbon, nitrogen, and sulfur (Peterson and Howarth  
511 1987; Peterson et al. 1994; Raymond and Bauer 2001;  
512 Zhou et al. 2006), stoichiometry (Hopkinson and Vallino  
513 2005; Lønborg et al. 2009), and biomarkers (Xu et al. 2006;  
514 Volkman et al. 2008). The studies of the fate of organic  
515 matter are based on biodegradability by using pyrolysis,  
516 gas chromatography–mass spectrometry (Hemminga et al.



Q19

Figure 7 The outwelling concept, in which nested subsystems exchange material in an estuarine-coastal landscape (after Childers et al. 2000).

517 1992, 1993; Klap et al. 1996), and laboratory bioassays with  
 518 dark incubation (del Giorgio and Davis 2003; Moran and  
 519 Covert 2003). Photodegradation studies also track another  
 520 major pathway of DOM consumption and transformation  
 521 (1997; Del Vecchio and Blough 2002; Osburn et al. 2009).  
 522 Spectroscopic techniques such as UV absorption and  
 523 fluorescence detection [3D excitation-emission matrices  
 524 (EEMs)] are now used both in source and fate studies  
 525 (Goldberg 1990; Her et al. 2003; Alberts et al. 2004;  
 526 Vignudelli et al. 2004; Jaffe et al. 2008).  
 527 A static model of aquatic organic matter mixing in salt  
 528 marshes can be revealed by two-dimensional stable iso-  
 529 tope plots of components in trophic levels, which are  
 F8 530 the organic matter providers (Figure 8). However, to  
 531 understand the ecological processes at play in these dy-  
 532 namic coastal systems, we have to look into the quantity  
 533 and quality of organic matter flux at different temporal  
 534 and spatial scales. For example, analysis of 3D-EEM  
 535 samples in a salt marsh creek in Plum Island Sound,  
 536 Massachusetts, USA, show that at high tide more re-  
 537 cently created DOM is available (likely microbial or  
 F9 538 algal; Figure 9 A-F). On the contrary at mid and low  
 539 tide, when the tidal water is exported, old and refractory  
 540 DOM is flushed away. A typical DOC flux pattern in  
 541 the same tidal channel shows that the largest DOC flux  
 542 happens during mid-tide. Therefore, a more significant

543 signal of salt marsh organic matter should be present in  
 544 the estuary at mid and low tidal levels.

545 Two important environmental controls on the quan-  
 546 tity and quality of salt marsh outwelling are light avail-  
 547 ability and extent of inundation (Hemminga et al. 1992,  
 548 1993; Kathilankal et al. 2008). The inundation of the salt  
 549 marsh determines the extent of the surface that may  
 550 interact and provide DOC to estuarine water, while light  
 551 availability impacts both in-situ production by auto-  
 552 trophs and transformations through photo-oxidation.  
 553 Soil decomposition could also change in terms of rates  
 554 and pathways both because of diminished oxygen avail-  
 555 ability and because of the flushing of metabolic waste  
 556 material into the water (Boon 2006). In the global theme  
 557 of climate change, sea-level rise will increase the areal  
 558 extent and frequency of inundations, and this could fur-  
 559 ther affect the quantity and quality of organic matter  
 560 mobilized in a tidal marsh estuary.

561 Lateral fluxes of organic matter are also an essential  
 562 component of the short-term salt marsh carbon budget.  
 563 Several studies have already determined the vertical car-  
 564 bon exchange between marshes and the atmosphere  
 565 with eddy flux towers (Heinsch et al. 2004; Kathilankal  
 566 et al. 2008; Guo et al. 2009, 2010; Polsenaere et al.  
 567 2012), but they were never coupled to lateral organic  
 568 and inorganic carbon flux measurements with similar  
 569 temporal and spatial resolution. Currently, field-ready  
 570 FDOM fluorometers or UV spectrometers are a potential  
 571 solution with a fine temporal resolution for in situ  
 572 measurement of DOC, which represents 90% of primary  
 573 production released from marshes to estuaries and to the  
 574 sea (Eldridge and Cifuentes 2000). In addition, since in  
 575 situ pCO<sub>2</sub> and pH probes are available to measure DIC  
 576 concentrations in tidal waters, it is now possible to close  
 577 the carbon budget of a salt marsh.

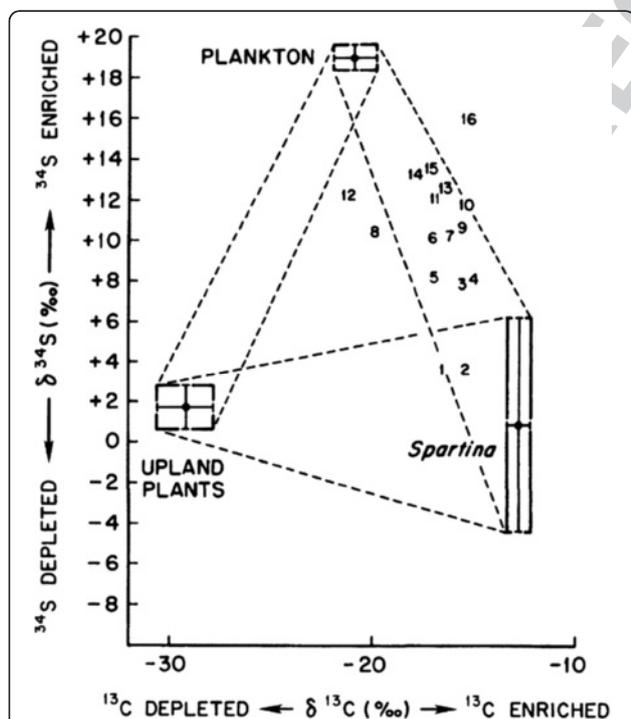
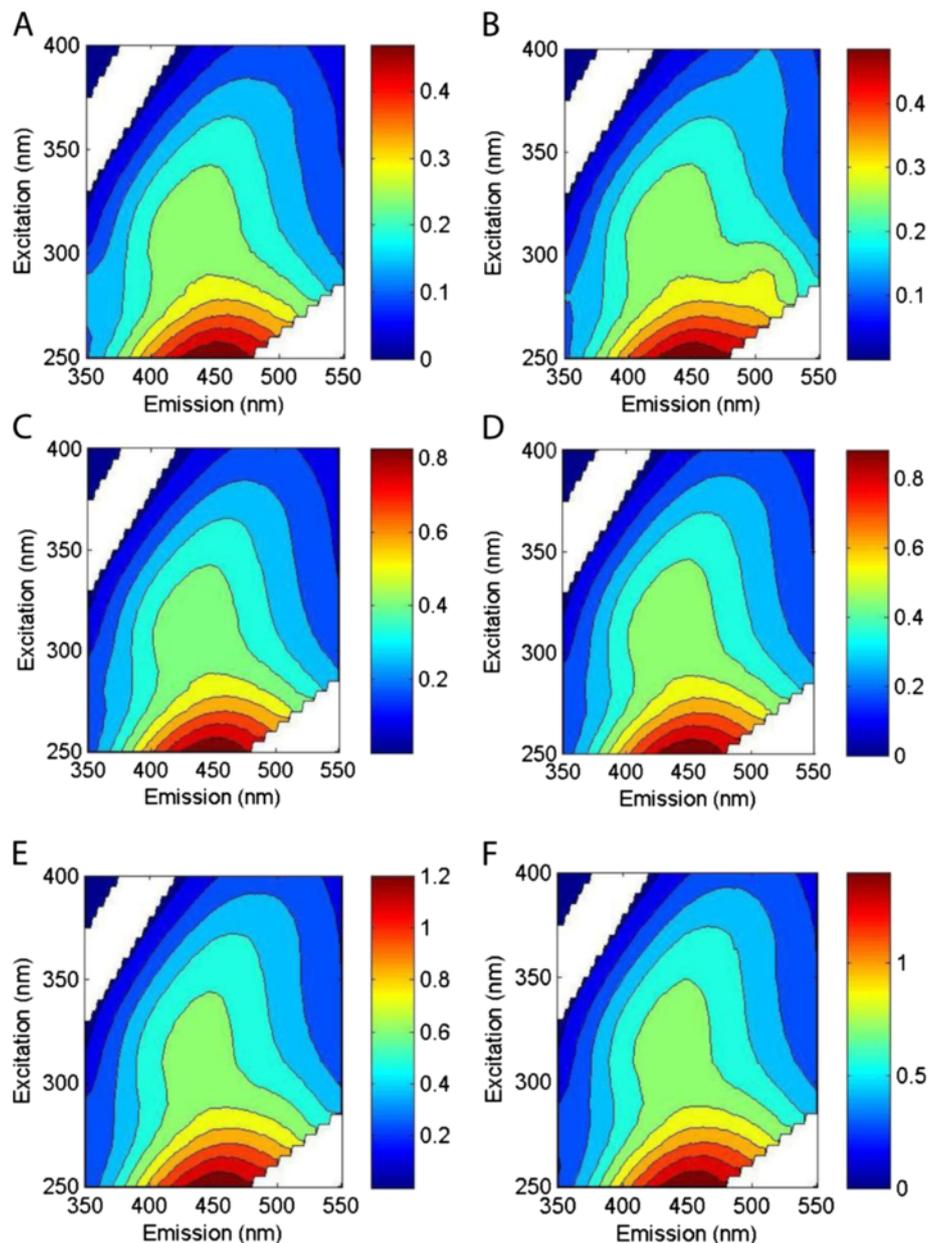


Figure 8 Sulfur isotopic ratio of marsh consumers as a function of stable isotopic ratio of carbon in the potential organic matter sources in Sapelo marshes. Different number indicates different species of consumers (after Peterson and Howarth 1987).

#### Fluxes of silica

578 Salt marshes accumulate large amounts of biogenic sili-  
 579 con [BSi, biogenically deposited hydrated amorphous  
 580 SiO<sub>2</sub>.nH<sub>2</sub>O, present in plants, sponges, and diatoms, also  
 581 referred to as amorphous Si (ASi); Struyf and Conley  
 582 2012] in sediments and vegetation. This can be recycled  
 583 to dissolved Si [(DSi), H<sub>4</sub>SiO<sub>4</sub>], the dissolved form of  
 584 Si available for biogenic uptake, also referred to as silicic  
 585 acid], mainly in pore water and surface puddles. During  
 586 ebb tide, this DSi is exported again to the estuary, where  
 587 it can be an important source of Si for pelagic estuarine  
 588 diatoms during times of Si limitation.  
 589

590 Most studies on Si cycling in tidal marshes have actu-  
 591 ally focused on tidal freshwater marshes (mostly in the  
 592 Scheldt estuary; see Struyf and Conley 2009). However,  
 593 in recent years, researchers have started investigating  
 594 fluxes in salt marshes, and general mechanisms and

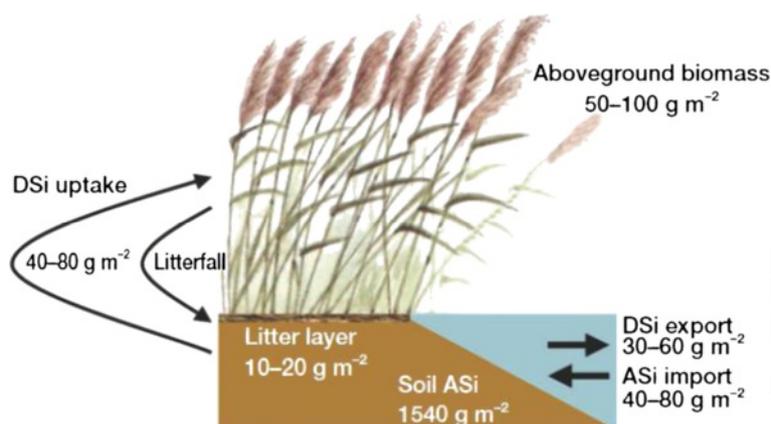


**Figure 9** DOM quantity and quality dynamics of Plum Island Ecosystem (PIE) salt marsh. (A-F) 3D-EEM contours of water samples collected in a tidal creek during high, middle, and low tide on 1 and 31 Aug 2012, respectively.

595 processes as described below are similar (e.g., Querné  
F10 596 et al. 2012; Vieillard et al. 2011) (Figure 10).

597 Where does the build-up of BSi originate from? As  
598 discussed in the previous section, marshes receive large  
599 amounts of suspended matter during tidal flooding. To-  
600 gether with suspended matter, BSi is imported (e.g.,  
601 Struyf et al. 2005a). Few studies have actually investi-  
602 gated whether this BSi mainly originates from pelagic  
603 diatom production within nearby coastal or estuarine  
604 waters or from terrestrial inputs of BSi (e.g., plant Si par-  
605 ticles called phytoliths) through rivers discharging into

the estuary. This distinction probably depends on the 606  
season: in spring and summer, estuarine production of 607  
diatoms is high, as reflected in high BSi concentrations 608  
in coastal water, coinciding with low DSi due to diatom 609  
uptake (e.g., Struyf et al. 2006; Carbonnel et al. 2009). In 610  
winter, suspended solids concentrations are often higher 611  
within estuaries (e.g., Fettweis et al. 1998; Lesourd et al. 612  
2003), but BSi content of the suspended matter (SPM) is 613  
low (Struyf et al. 2007a), which suggests that BSi mainly 614  
originates from external terrestrial inputs (e.g., soil run- 615  
off), reflected in high concurrent DSi in the estuary (no 616



**Figure 10** Stocks and yearly fluxes of DSi and ASi within a Belgian freshwater marsh. Sediment stocks are for the upper 30 cm. Litter fall and DSi uptake by vegetation equal each other on an annual time scale (from Struyf and Conley 2009).

617 diatom uptake). In a study in the Scheldt estuary, it was  
 618 found that absolute import of BSi into marshes is  
 619 highest in winter, as the higher import of SPM offsets  
 620 the lower BSi content in the SPM (Struyf et al. 2007a).

621 Tidal import of BSi is not the only source for BSi ac-  
 622 cumulation in marshes. A second, very scarcely studied  
 623 factor is the autochthonous production of diatoms in the  
 624 marshes. Hackney et al. (2000) indicated that high di-  
 625 atom production in tidal marshes and the food web  
 626 dependent on this production could actually be impor-  
 627 tant contributors to estuarine production (as earlier sug-  
 628 gested by Sullivan and Moncreiff 1990). Macrophytes  
 629 are another factor in autochthonous production of BSi;  
 630 aquatic and wetland plants can contain significant  
 631 amounts of BSi (Schoelynck et al. 2010). In freshwater  
 632 marshes, *Phragmites australis* is particularly rich in BSi  
 633 (Struyf et al. 2007b; up to 5% of the dry biomass in dead  
 634 reed culms is BSi in saltwater and mesohaline marshes,  
 635 the dry biomass of *Spartina alterniflora* and *Juncus*  
 636 *roemerianus* contains up to ±0.5% of BSi (Norris and  
 637 Hackney 1999; Querné et al. 2012). The relative contri-  
 638 bution of autochthonous production vs. tidal import to  
 639 the total BSi stock in tidal marshes certainly warrants  
 640 further investigation.

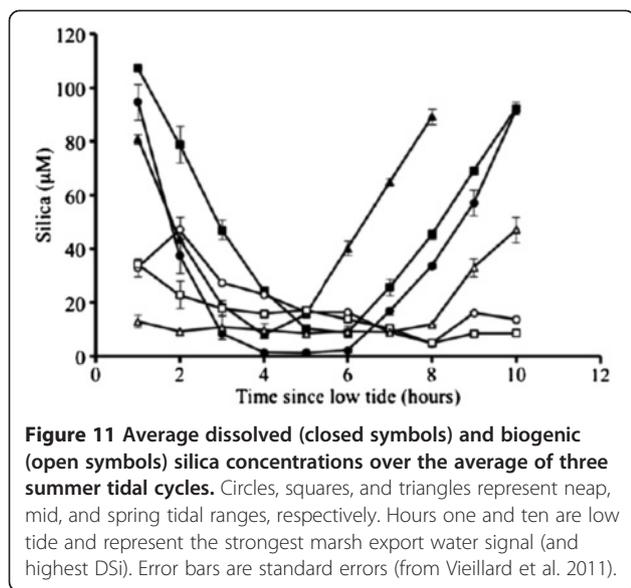
641 The high BSi import and autochthonous BSi production  
 642 in tidal marshes result in high concentrations of DSi in  
 643 marsh pore water and tidal pools (100–600 μM; e.g.,  
 644 Norris and Hackney 1999; Jacobs et al. 2008; Struyf et al.  
 645 2005b; Querné et al. 2012) Tidal pools can show a  
 646 strong decrease in DSi due to diatom production. Plants  
 647 could have a similar effect on pore water DSi concentra-  
 648 tions, but the limited available studies do not confirm  
 649 this hypothesis (Querné et al. 2012). Due to its high  
 650 solubility in comparison to mineral silicates, sediment  
 651 and soil BSi can exert strong control on DSi concentra-  
 652 tions in wetland pore water (e.g., Struyf et al. 2009).

Concentrations of DSi in marsh water are two- to five- 653  
 fold higher in comparison to maximum concentrations 654  
 observed in the upper reaches of estuaries: the average 655  
 riverine DSi concentration is 100 μM (Conley 1997), 656  
 but higher concentrations (up to 250 μM) have been 657  
 observed in the freshwater tidal reaches of the Scheldt 658  
 estuary (Belgium) (Carbonnel et al. 2009). These are, 659  
 however, maximum concentrations: in summer, diatom 660  
 production can reduce DSi concentrations, even in the 661  
 upstream tidal reaches, to concentrations near 5 μM (e.g., 662  
 Cox et al. 2009; Carbonnel et al. 2009). In general, estua- 663  
 rine DSi concentrations also decrease with increasing 664  
 salinity, due to conservative mixing with seawater and 665  
 diatom uptake, reaching concentrations below 20 μM at 666  
 higher salinity (e.g., Van Damme et al. 2005). There is thus 667  
 a strong enrichment of both easily soluble BSi and DSi in 668  
 tidal marshes compared to adjacent estuarine and coastal 669  
 waters. As a result, ebb water flowing out of tidal marshes 670  
 is enriched with DSi compared to inflowing water, espe- 671  
 cially in the seepage water (i.e., soil pore water flowing out 672  
 of the tidal marsh soil when estuarine tidal height has 673  
 dropped below marsh elevation) (e.g., Struyf et al. 2006; 674  
 Vieillard et al. 2011) (Figure 11). Especially in summer 675  
 months, when concentrations of DSi in estuaries are re- 676  
 duced due to diatom uptake, and coastal DSi concentra- 677  
 tions are minimal, this DSi export from tidal marshes 678  
 could be an important factor in supporting estuarine and 679  
 coastal diatom production (Figure 12). 680

F11

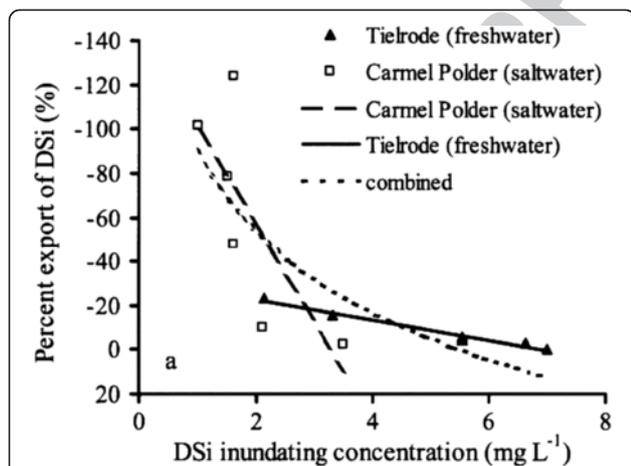
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Mass-balance studies show that the magnitude of ob- 681  
 served export fluxes of DSi is similar to modeled and ex- 682  
 perimental results for BSi recycling fluxes (Struyf et al. 683  
 2007a, b). There is thus a strong indication that export 684  
 of Si from the marsh to the estuary is a biologically con- 685  
 trolled mechanism, despite large amounts of mineral Si 686  
 (e.g., clays) that are imported along with suspended mat- 687  
 ter into the marsh at flood tide. This is because mineral 688



689 Si is orders of magnitude less soluble than BSi. Finally,  
 690 the large accumulation of BSi in salt marshes points to a  
 691 strong role of biology in the Si cycle.

692 Marshes therefore potentially increase the resilience of  
 693 coastal waters to dissolved Si limitation and associated  
 694 phytoplankton shifts to non-diatom communities. N-P-Si  
 695 ratios are important in controlling estuarine and coastal  
 696 phytoplankton composition; increased inputs of N and P  
 697 due to human activities have led to shifts from diatom-  
 698 dominated communities (which essentially need DSi) to a  
 699 prevalence of other phytoplankton species (e.g., Smayda



1997) that are less available to higher trophic levels. In  
 701 comparison to studies on carbon, nitrogen, and phosphorus,  
 702 results on silicon cycling in tidal marshes are still limited.  
 703 The general importance of BSi accumulation and DSi  
 704 delivery by tidal marshes has been confirmed in a  
 705 growing number of studies (e.g., Struyf et al. 2005a, 2006;  
 706 Jacobs et al. 2008; Vieillard et al. 2011).

### Conclusions

707 Tidal oscillations systematically flood salt marshes trans-  
 708 porting water, sediments, organic matter, and biogeo-  
 709 chemical elements such as silica. To measure water  
 710 fluxes in a salt marsh drained by tidal channels, three  
 711 different approaches can be adopted, depending on what  
 712 hydrodynamic quantities need to be computed:  
 713

- The tidal prism (Equation 1) provides an estimate of  
 714 the total volume of water exchanged by a marsh in a  
 715 tidal cycle. 716
- If the value of the instantaneous discharge is needed,  
 717 the static model of Boon (Equation 2) is simple and  
 718 adequate. 719
- If we seek the flow velocity or the distribution of  
 720 residence times, the use of the TIGER model is  
 721 recommended (Equation 3). 722

723 These simplified approaches are particularly suitable  
 724 for the study of fluxes of biogeochemical compounds in  
 725 salt marshes and could be easily adopted by researchers  
 726 working in this area.

727 Sediment fluxes to and from the marsh strongly de-  
 728 pend on sediment resuspension by waves and currents.  
 729 Recent results show that:

- During flood the input of sediment to the marsh  
 730 depends on resuspension of sediments in adjacent  
 731 areas by wind waves and tidal currents. 732
- During ebb the export of sediments is strongly  
 733 affected by the magnitude of water velocity in the  
 734 channels draining the marsh. 735
- Storm surges import large volumes of sediment to the  
 736 marsh, but they also export a comparable amount  
 737 during the subsequent ebb due to the high velocities  
 738 in the channels and related bottom shear stresses. 739
- Moderate storm conditions with limited surges  
 740 maximize the sediment import to the marsh. 741

742 Knowledge of internal marsh processes and mecha-  
 743 nisms affecting both biogenic silicon and organic matter  
 744 accumulation and turnover is scarce. It is likely that  
 745 more and more measurement methods will be available  
 746 in the future to allow us to investigate ecological pro-  
 747 cesses at a finer temporal, spatial, and chemical reso-  
 748 lution and at a broader scale.

749 Regarding lateral flux studies of salt marsh organic  
750 matter, the following key topics for future studies are  
751 suggested:

- 752 – Long-term data sets with fine temporal resolution of  
753 lateral organic and inorganic matter exchange  
754 should be coupled to eddy flux tower data sets to  
755 close the carbon budget in salt marshes.
- 756 – Quality of organic matter should also be measured  
757 in tidal fluxes, together with its interactions at  
758 different trophic levels.
- 759 – Linkages between organic matter fluxes and salt  
760 marsh physical and physiological processes should  
761 be established.
- 762 – Recent studies on fluxes of organic matter have  
763 focused more on the entire estuarine system,  
764 including fluxes from the ocean and nearby fluvial  
765 watersheds. More research is clearly needed to  
766 determine the mechanisms controlling fluxes of  
767 organic matter within the marsh system, for  
768 example from the marsh platform to the tidal  
769 channels and vice versa.

770 To move knowledge beyond the general pattern of BSi  
771 accumulation and DSi delivery, several key topics can be  
772 defined:

- 773 – How does the balance between BSi accumulation  
774 and DSi delivery change along gradients of marsh  
775 age, salinity, and seasons? Scarce studies so far have  
776 indicated net accumulation in tidal freshwater  
777 marshes in the Scheldt estuary over a decadal  
778 timescale (burial of about 40% of imported BSi  
779 without dissolution to DSi). This balance is crucial  
780 to understanding the long-term role of tidal marshes  
781 in coastal Si cycling: net annual accumulation of BSi  
782 could be equally as important as (or even more  
783 important than) seasonal delivery of DSi.
- 784 – What is the importance of Si-rich vegetation and  
785 diatom production in tidal marshes for coastal food  
786 webs?
- 787 – What processes potentially impact BSi accumulation  
788 and DSi recycling in tidal marshes through their  
789 impact on sediment accumulation and vegetation  
790 and diatom growth, such as grazing management,  
791 sea level rise, invasive species, and hydro-  
792 engineering?

#### 793 Competing interests

794 The authors declare that they have no competing interests.

#### 795 Authors' contributions

796 SF wrote the paragraph on Fluxes of Water, PW and SW wrote the  
797 paragraph on Fluxes of Sediments, YZ and PR wrote the paragraph on Fluxes  
798 of Organic Material, ES and ST wrote the paragraph on Fluxes of Silica.  
799 All authors read and approved the final manuscript.

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#### References

- Alberts JJ, Takacs M, Shalles M (2004) Ultraviolet-visible and fluorescence spectral evidence of natural organic matter (NOM) changes along an estuarine salinity gradient. *Estuaries* 27(2):296–310
- Alongi D (1998) Coastal ecosystem processes. CRC Press, Boca Raton, p 448
- Bayliss-Smith TP, Healey R, Lailey R, Spencer T, Stoddart DR (1978) Tidal flows in salt marsh creeks. *Estuarine Coastal Mar Sci* 9:235–255
- Bianchi TS (2007) Biogeochemistry of estuaries. Oxford University Press, New York
- Boon JD (1975) Tidal discharge asymmetry in a salt marsh drainage system. *Limnol Oceanogr* 20:71–80
- Boon PI (2006) Biogeochemistry and bacterial ecology of hydrologically dynamic wetlands. In: Batzer DP, Sharitz RR (ed) Ecology of freshwater and estuarine wetlands. University of California Press, Berkeley, p 568
- Cahoon DR (2006) A review of major storm impacts on coastal wetland elevations. *Estuaries Coasts* 29:889–898
- Cai WJ (2011) Estuarine and coastal ocean carbon paradox: CO<sub>2</sub> sinks or sites of terrestrial carbon incineration? *Annu Rev Mar Sci* 3(3):123–145
- Carbonnel V, Lionard M, Muylaert K, Chou L (2009) Dynamics of dissolved and biogenic silica in the freshwater reaches of a macrotidal estuary (The Scheldt, Belgium). *Biogeochemistry* 96:49–72
- Carniello L, Defina A, Fagherazzi S, D'Alpaos L (2005) A combined wind wave-tidal model for the Venice lagoon, Italy. *J Geophys Res* 110(F4):F04007. doi:10.1029/2004JF000232
- Childers DL, Day JW, Mckeller HN (2000) Twenty more years of marsh and estuarine flux studies: revisiting Nixon (1980). In: Weinstein MP, Kreeger DA (ed) Concepts and controversies in tidal marsh ecology, 5. Springer, New York, pp 391–423
- Chmura GL, Anisfeld SC, Cahoon DR, Lynch JC (2003) Global carbon sequestration in tidal, saline wetland soils. *Global Biogeochem Cycles* 17(4):1111
- Choi YH, Wang Y (2004) Dynamics of carbon sequestration in a coastal wetland using radiocarbon measurements. *Global Biogeochem Cycles* 18(4):GB4016. doi:10.1029/2004GB002261
- Christiansen T (1998) Sediment deposition on a tidal salt marsh. Ph.D. Dissertation. University of Virginia, Charlottesville, p 114
- Christiansen T, Wiberg PL, Milligan TG (2000) Flow and sediment transport on a tidal salt marsh surface. *Estuarine Coastal Shelf Sci* 50:315–331
- Church TM (1986) Biogeochemical factors influencing the residence time of microconstituents in a large tidal estuary, Delaware Bay. *Mar Chem* 18(2–4):393–406
- Conley DJ (1997) Riverine contribution of biogenic silica to the oceanic silica budget. *Limnol Oceanogr* 42:774–777
- Cox T, Maris T, Soetaert K, Conley DJ, Van Damme S, Meire P, Middelburg JJ, Vos M, Struyf E (2009) A macro-tidal freshwater ecosystem recovering from hyper-eutrophication: the Schelde case study. *Biogeosciences* 6:2935–2948
- Crump BC, Hopkinson CS, Sogin ML, Hobbie JE (2004) Microbial biogeography along an estuarine salinity gradient: combined influences of bacterial growth and residence time. *Appl Environ Microbiol* 70(3):1494–1505
- D'Alpaos A, Lanzoni S, Marani M, Fagherazzi S, Rinaldo A (2005) Tidal network ontogeny: channel initiation and early development. *J Geophys Res-Earth Surface* 110:F02001. doi:10.1029/2004JF000182
- D'Alpaos A, Lanzoni S, Mudd SM, Fagherazzi S (2006) Modeling the influence of hydroperiod and vegetation on the cross-sectional formation of tidal channels. *Estuarine Coastal Shelf Sci* 69(3–4):311–324
- D'Alpaos A, Lanzoni S, Marani M, Rinaldo A (2007) Landscape evolution in tidal embayments: modeling the interplay of erosion, sedimentation, and vegetation dynamics. *J Geophys Res* 112:F01008. doi:10.1029/2006JF000537

Q1

Q2

Q3

Q4

- 868 D'Alpaos A, Lanzoni S, Marani M, Rinaldo A (2010) On the tidal prism-channel  
 869 area relations. *J Geophys Res* 115:F01003. doi:10.1029/2008JF001243
- 870 Dame R, Chrzanowski T, et al. (1986) The outwelling hypothesis and North Inlet,  
 871 South-Carolina. *Mar Ecol Prog Ser* 33(3):217-229
- 872 Dankers N, Binsbergen M, Zegers K, Laane R, Vanderloeff MR (1984)  
 873 Transportation of water, particulate and dissolved organic and inorganic  
 874 matter between a salt-marsh and the Ems-Dollard Estuary, the Netherlands.  
 875 *Estuarine Coastal Shelf Sci* 19(2):143-165
- 876 Davidson-Arnott RGD, Van Proosdij D, Ollerhead J, Schostak L (2002)  
 877 Hydrodynamics and sedimentation in salt marshes: examples from a  
 878 macrotidal marsh, Bay of Fundy. *Geomorphology* 48:209-231
- 879 del Giorgio PA, Davis J (2003) Patterns of dissolved organic matter lability and  
 880 consumption across aquatic ecosystems. In: Findlay SEG, Sinsabaugh RL (ed)  
 881 *Aquatic ecosystems: interactivity of dissolved organic matter*. Academic  
 882 Press, San Diego, pp 399-424
- 883 Del Vecchio R, Blough NV (2002) Photobleaching of chromophoric dissolved  
 884 organic matter in natural waters: kinetics and modeling. *Mar Chem*  
 885 78(4):231-253
- 886 Duarte CM, Middelburg JJ, Caraco N (2005) Major role of marine vegetation on  
 887 the oceanic carbon cycle. *Biogeosciences* 2(1):1-8
- 888 Eldridge PM, Cifuentes LA (2000) A stable isotope model approach to estimating  
 889 the contribution of organic matter from marshes to estuaries. In: Weinstein  
 890 MP, Kreeger DA (ed) *Concepts and controversies in tidal marsh ecology*.  
 891 Springer, New York, pp 495-513
- 892 Fagherazzi S, Furbish DJ (2001) On the shape and widening of salt marsh creeks.  
 893 *J Geophys Res* 106(C1):991-1005
- 894 Fagherazzi S, Priestas AM (2010) Sediments and water fluxes in a muddy  
 895 coastline: interplay between waves and tidal channel hydrodynamics.  
 896 *Earth Surf Process Landforms* 35(3):284-293
- 897 Fagherazzi S, Sun T (2004) A stochastic model for the formation of channel  
 898 networks in tidal marshes. *Geophys Res Lett* 31:L21503. doi:10.1029/  
 899 2004GL020965
- 900 Fagherazzi S, Palermo C, Rulli MC, Carniello L, Defina A (2007) Wind waves in  
 901 shallow microtidal basins and the dynamic equilibrium of tidal flats.  
 902 *J Geophys Res* 112:F02024. doi:10.1029/2006JF000572
- 903 Fagherazzi S, Hannion M, D'Odorico P (2008) Geomorphic structure of tidal  
 904 hydrodynamics in salt marsh creeks. *Water Resour Res* 44:W02419.  
 905 doi:10.1029/2007WR006289
- 906 Fagherazzi S, Mariotti G, Porter JH, McGlathery KJ, Wiberg PL (2010) Wave energy  
 907 asymmetry in shallow bays. *Geophys Res Lett* 37:L24601
- 908 Fagherazzi S, Kirwan ML, Mudd SM, Guntenspergen GR, Temmerman S, D'Alpaos  
 909 A, van de Koppel J, Rybczyk JM, Reyes E, Craft C, Clough J (2012) Numerical  
 910 models of salt marsh evolution: ecological and climatic factors. *Rev Geophys*  
 911 50(1). doi:10.1029/2011RG000359
- 912 Fettweis M, Sas M, Monbaliu J (1998) Seasonal, neap-spring and tidal variation of  
 913 cohesive sediment concentration in the Scheldt Estuary, Belgium. *Estuarine  
 914 Coastal Shelf Sci* 47:21-36
- 915 French JR, Stoddart DR (1992) Hydrodynamics of salt-marsh creek systems:  
 916 implications for marsh morphological development and material exchange.  
 917 *Earth Surf Processes Landforms* 17(3):235-252
- 918 French JR, Spencer T, Murray AL, Arnold NS (1995) Geostatistical analysis of  
 919 sediment deposition in two small tidal wetlands, Norfolk, United Kingdom.  
 920 *J Coastal Res* 11:308-321
- 921 Friedrichs CT, Aubrey DG (1988) Non-linear tidal distortion in shallow well-mixed  
 922 estuaries: a synthesis. *Estuarine Coastal Shelf Sci* 27:521-545
- 923 Gardner LR (2005) Role of geomorphic and hydraulic parameters in governing  
 924 pore water seepage from salt marsh sediments. *Water Resour Res* 41:  
 925 W07010. doi:10.1029/2004WR003671
- 926 Gardner LR, Kjerfve B (2006) Tidal fluxes of nutrients and suspended sediments at  
 927 the North Inlet - Winyah Bay National Estuarine Research Reserve. *Estuarine  
 928 Coastal Shelf Sci* 70(4):682-692
- 929 Goldberg MC (1990) Determination of the transport and change in composition  
 930 of fluorescent materials in hydrologic systems by use of Eem-spectroscopy.  
 931 *Abstr Pap Am Chem Soc* 199:24-Envr
- 932 Guo HQ, Noormets A, et al. (2009) Tidal effects on net ecosystem exchange of  
 933 carbon in an estuarine wetland. *Agr Forest Meteorol* 149(11):1820-1828
- 934 Guo HQ, Zhao B, Chen JQ, Yan YE, Li B, Chen JK (2010) Seasonal changes of  
 935 energy fluxes in an estuarine wetland of Shanghai, China. *Chin Geogr Sci*  
 936 20(1):23-29
- 937 Hackney CT, Cahoon LB, Prestos C, Norris A (2000) Silicon is the link between  
 938 tidal marshes and estuarine fisheries: a new paradigm. In: Weinstein MP,  
 Kreeger DA (ed) *Concepts and controversies in tidal marsh ecology*. Kluwer,  
 Amsterdam, pp 543-552
- Healey RG, Pye K, Stoddart DR, Bayliss-Smith TP (1981) Velocity variations in salt  
 marsh creeks, Norfolk, England. *Estuarine Coastal Shelf Sci* 13:535-555
- Heinsch FA, Heilman JL, et al. (2004) Carbon dioxide exchange in a high marsh  
 on the Texas Gulf Coast: effects of freshwater availability. *Agr Forest Meteorol*  
 125(1-2):159-172
- Hemminga MA, Klap VA, Vansoelen J, Deleeuw J, Boon JJ (1992) Shifts in Seston  
 characteristics after inundation of a European coastal salt-marsh. *Limnol  
 Oceanogr* 37(7):1559-1564
- Hemminga MA, Klap VA, Vansoelen J, Boon JJ (1993) Effect of salt-marsh  
 inundation on estuarine particulate organic-matter characteristics. *Mar Ecol  
 Prog Ser* 99(1-2):153-161
- Her N, Amy G, McKnight D, Sohn J, Yoon YM (2003) Characterization of DOM as  
 a function of MW by fluorescence EEM and HPLC-SEC using UVA, DOC, and  
 fluorescence detection. *Water Res* 37(17):4295-4303
- Hopkinson CS, Vallino JJ (2005) Efficient export of carbon to the deep ocean  
 through dissolved organic matter. *Nature* 433(7022):142-145
- Hopkinson CS, Cai WJ, Hu XP (2012) Carbon sequestration in wetland dominated  
 coastal systems - a global sink of rapidly diminishing magnitude. *Curr Opin  
 Environ Sustain* 4(2):186-194
- Howes BL, Goehring DD (1994) Porewater drainage and dissolved organic  
 carbon and nutrient losses through the intertidal creek banks of a  
 New England salt marsh. *Mar Ecol Prog Ser* 114:289-301
- Hussein AH, Rabenhorst MC, Tucker ML (2004) Modeling of carbon sequestration  
 in coastal marsh soils. *Soil Sci Soc Am J* 68(5):1786-1795
- Jacobs S, Struyf E, Maris T, Meire P (2008) Spatio-temporal aspects of silica  
 buffering in restored tidal marshes. *Estuarine Coastal Shelf Sci* 80:42-52
- Jaffe R, McKnight D, et al. (2008) Spatial and temporal variations in DOM  
 composition in ecosystems: the importance of long-term monitoring of  
 optical properties. *J Geophys Res-Biogeosciences* 113(G4):G04032
- Kathilankal JC, Mozdzer TJ, Fuentes JD, D'Odorico P, McGlathery KJ, Zieman JC  
 (2008) Tidal influences on carbon assimilation by a salt marsh. *Environ Res  
 Lett* 3(4):044010
- Kirwan ML, Mudd SM (2012) Response of salt-marsh carbon accumulation to  
 climate change. *Nature* 489(7417):550. doi:10.1038/nature11440
- Kirwan ML, Guntenspergen GR, D'Alpaos A, Morris JT, Mudd SM, Temmerman S  
 (2010) Limits on the adaptability of coastal marshes to rising sea level.  
*Geophysical Res Lett* 37:L23401. doi:10.1029/2010GL045489
- Klap VA, Boon JJ, Hemminga MA, vanSoelen J (1996) Assessment of the  
 molecular composition of particulate organic matter exchanged between  
 the Saefinghe salt marsh (southwestern Netherlands) and the adjacent  
 water system. *Mar Chem* 54(3-4):221-243
- Lawrence DSL, Allen JRL, Havelock GM (2004) Salt marsh morphodynamics:  
 an investigation of tidal flows and marsh channel equilibrium. *J Coastal Res*  
 20(1):301-316
- Lawson SE, Wiberg PL, McGlathery KJ, Fugate DC (2007) Wind-driven sediment  
 suspension controls light availability in a shallow coastal lagoon. *Estuaries  
 Coasts* 30:102-112
- Leonard LA (1997) Controls on sediment transport and deposition in an incised  
 mainland marsh basin, southeastern North Carolina. *Wetlands* 17:263-274
- Leonard LA, Luther ME (1995) Flow hydrodynamics in tidal marsh canopies.  
*Limnol Oceanogr* 40:1474-1484
- Lesourd S, Lesueur P, Brun-Cottan JC, Garnaud S, Poupinet N (2003) Seasonal  
 variations in the characteristics of superficial sediments in a macrotidal  
 estuary (the Seine inlet, France). *Estuarine Coastal Shelf Sci* 58:3-16
- Lönborg C, Álvarez-Salgado XA, Davidson K, Miller AEJ (2009) Production of  
 bioavailable and refractory dissolved organic matter by coastal heterotrophic  
 microbial populations. *Estuarine Coastal Shelf Sci* 82(4):682-688
- Marani M, Belluco E, D'Alpaos A, Defina A, Lanzoni S, Rinaldo A (2003) On the  
 drainage density of tidal networks. *Water Resour Res* 39(2):1040. doi:10.1029/  
 2001WR001051
- Mariotti G, Fagherazzi S (2010) A numerical model for the coupled long-term  
 evolution of salt marshes and tidal flats. *J Geophys Res* 115:F01004.  
 doi:10.1029/2009JF001326
- McLeod E, Chmura GL, et al. (2011) A blueprint for blue carbon: toward an  
 improved understanding of the role of vegetated coastal habitats in  
 sequestering CO<sub>2</sub>. *Front Ecol Environ* 9(10):552-560
- Middelburg JJ, Nieuwenhuize J, Lubberts RK, van de Plassche O (1997) Organic  
 carbon isotope systematics of coastal marshes. *Estuarine Coastal Shelf Sci*  
 45(5):681-687

Q11

Q12

Q13

Q14

- 1010 Möller I, Spencer T, French JR, Leggett DJ, Dixon M (1999) Wave transformation  
1011 over salt marshes: a field and numerical modelling study from North Norfolk,  
1012 England. *Estuarine Coastal Shelf Sci* 49(3):411–426
- 1013 Moran MA, Covert J (2003) Photochemically mediated linkages between  
1014 dissolved organic matter and bacterioplankton. In: Findlay SEG, Sinsabaugh  
1015 RL (ed) *Aquatic ecosystems: interactivity of dissolved organic matter*.  
1016 Academic Press, San Diego
- 1017 Morris JT, Bowden WB (1986) A mechanistic, numerical model of sedimentation,  
1018 mineralization, and decomposition for marsh sediments. *Soil Sci Soc Am J*  
1019 50(1):96–105
- 1020 Morris DP, Hargreaves BR (1997) The role of photochemical degradation of  
1021 dissolved organic carbon in regulating the UV transparency of three lakes on  
1022 the Pocono Plateau. *Limnol Oceanogr* 42(2):239–249
- 1023 Morris JT, Sundareswar PV, Nietch CT, Kjerfve B, Cahoon DR (2002) Responses of  
1024 coastal wetlands to rising sea level. *Ecology* 83:2869–2877
- 1025 Moskalski SM, Sommerfield CK (2012) Suspended sediment deposition  
1026 and trapping efficiency in a Delaware salt marsh. *Geomorphology*  
1027 139–140:195–204
- 1028 Mudd SM, Fagherazzi S, Morris JT, Furbish DJ (2004) Flow, sedimentation, and  
1029 biomass production on a vegetated salt marsh in South Carolina: toward a  
1030 predictive model of marsh morphologic and ecologic evolution. In:  
1031 Fagherazzi S, Marani M, Blum LK (ed) *The ecogeomorphology of tidal  
1032 marshes*. Coastal and Estuarine Studies, vol. 59. AGU, Washington DC,  
1033 pp 165–187
- 1034 Mudd SM, Howell SM, Morris JT (2009) Impact of dynamic feedbacks between  
1035 sedimentation, sea-level rise, and biomass production on near surface  
1036 marsh stratigraphy and carbon accumulation. *Estuarine Coastal Shelf Sci*  
1037 82(3):377–389
- 1038 Mwamba MJ, Torres R (2002) Rainfall effects on marsh sediment redistribution,  
1039 North Inlet, South Carolina. *Mar Geol* 189:267–287
- 1040 Myrick RM, Leopold LB (1963) Hydraulic geometry of a small tidal estuary.  
1041 *US Geol Surv Prof Pap* 422–B. US GPO, Washington DC
- Q15 1042 Nepf HM (1999) Drag, turbulence, and diffusion in flow through emergent  
1043 vegetation. *Water Resour Res* 35(2):479. doi:10.1029/1998WR900069
- 1044 Norris AR, Hackney CT (1999) Silica content of a mesohaline tidal marsh in North  
1045 Carolina. *Estuarine Coastal Shelf Sci* 49:597–605
- 1046 Odum EP (1968) A research challenge: evaluating the productivity of coastal and  
1047 estuarine water. *Proceeding of Second Sea Grant Conference*. University of  
1048 Rhode Island, Kingston
- 1049 Odum EP (2000) Tidal marshes as outwelling/pulsing systems. In: Weinstein MP,  
1050 Kreeger DA (ed) *Concepts and controversies in tidal marsh ecology*.  
1051 Springer, New York, pp 3–7
- 1052 Osburn CL, Retamal L, Vincent WF (2009) Photoreactivity of chromophoric  
1053 dissolved organic matter transported by the Mackenzie River to the Beaufort  
1054 Sea. *Mar Chem* 115(1–2):10–20
- 1055 Peterson BJ, Howarth RW (1987) Sulfur, carbon, and nitrogen isotopes used to  
1056 trace organic-matter flow in the salt-marsh estuaries of Sapelo Island,  
1057 Georgia. *Limnol Oceanogr* 32(6):1195–1213
- 1058 Peterson B, Fry B, Hullar M, Saupe S, Wright R (1994) The distribution and stable  
1059 carbon isotopic composition of dissolved organic carbon in estuaries.  
1060 *Estuaries* 17(1B):111–121
- 1061 Pethick JS (1980) Velocities, surges and asymmetry in tidal channels. *Estuarine  
1062 Coastal Mar Sci* 11:331–345
- 1063 Polsenaere P, Lamaud E, et al. (2012) Spatial and temporal CO<sub>2</sub> exchanges  
1064 measured by eddy covariance over a temperate intertidal flat and  
1065 their relationships to net ecosystem production. *Biogeosciences*  
1066 9(1):249–268
- 1067 Querné J, Ragueneau O, Poupart N (2012) In situ biogenic silica variations in the  
1068 invasive salt marsh plant, *Spartina alterniflora*: a possible link with  
1069 environmental stress. *Plant Soil* 352:157–171
- 1070 Raymond PA, Bauer JE (2001) Use of C-14 and C-13 natural abundances for  
1071 evaluating riverine, estuarine, and coastal DOC and POC sources and cycling:  
1072 a review and synthesis. *Org Geochem* 32(4):469–485
- 1073 Reed DJ (1989) Patterns of sediment deposition in subsiding salt marshes,  
1074 Terrebonne Bay, Louisiana: the role of winter storms. *Estuaries* 12(4):222–227
- 1075 Reed DJ, Spencer T, Murray AL, French JR, Leonard L (1999) Marsh surface  
1076 sediment deposition and the role of tidal creeks: implications for created and  
1077 managed coastal marshes. *J Coast Conserv* 5:81–90
- 1078 Rinaldo A, Fagherazzi S, Lanzoni S, Marani M, Dietrich WE (1999) Tidal networks 2.  
1079 Watershed delineation and comparative network morphology. *Water Resour  
1080 Res* 35(12):3905–3917
- Schoelynck J, Bal K, Backx H, Okruszko T, Meire P, Struyf E (2010) Silica uptake in  
1081 aquatic and wetland macrophytes: a strategic choice between silica, lignin  
1082 and cellulose? *New Phytol* 186:385–391
- 1083  
1084 Shariit RR, Pennings SC (2006) Development of wetland plant communities. In:  
1085 Batzer DP, Shariit RR (ed) *Freshwater and estuarine wetlands*. University of  
1086 California Press, Berkeley, p 567
- 1087 Smayda TJ (1997) Harmful algal blooms: their ecophysiology and general relevance  
1088 to phytoplankton blooms in the sea. *Limnol Oceanogr* 42:1137–1153
- 1089 Struyf E, Conley DJ (2009) Silica: an essential nutrient in wetland  
1090 biogeochemistry. *Front Ecol Environ* 7(2):88–94
- 1091 Struyf E, Conley DJ (2012) Emerging understanding of the ecosystem silica filter.  
1092 *Biogeochemistry* 107:9–18
- 1093 Struyf E, Van Damme S, Gribsholt B, Meire P (2005a) Freshwater marshes as  
1094 dissolved silica recyclers in an estuarine environment (Schelde estuary,  
1095 Belgium). *Hydrobiologia* 540:69–77
- 1096 Struyf E, Van Damme S, Gribsholt B, Middelburg JJ, Meire P (2005b) Biogenic  
1097 silica in freshwater marsh sediments and vegetation. *Mar Ecol Prog Ser*  
1098 303:51–60
- 1099 Struyf E, Dausse A, Van Damme S, Bal K, Gribsholt B, Boschker HTS, Middelburg JJ,  
1100 Meire P (2006) Tidal marshes and biogenic silica recycling at the land-sea  
1101 interface. *Limnol Oceanogr* 51(2):838–846
- 1102 Struyf E, Temmerman S, Meire P (2007a) Dynamics of biogenic Si in freshwater  
1103 tidal marshes: Si regeneration and retention in marsh sediments (Scheldt  
1104 estuary). *Biogeochemistry* 82:41–53
- 1105 Struyf E, Van Damme S, Gribsholt B, Bal K, Beauchard O, Middelburg JJ, Meire P  
1106 (2007b) Phragmites australis and Si cycling in tidal wetlands. *Aquat Bot*  
1107 87:134–140
- 1108 Struyf E, Opdekamp W, Backx H, Jacobs S, Conley DJ, Meire P (2009) Vegetation  
1109 and proximity to the river control amorphous Si storage in a riparian  
1110 wetland (Bierbza National Park, Poland). *Biogeosciences* 6:623–631
- 1111 Sullivan MJ, Moncreiff CA (1990) Edaphic algae are an important component of  
1112 salt marsh food webs: evidence from multiple stable isotope analyses. *Mar  
1113 Ecol Prog Ser* 62:149–159
- 1114 Teal JM (1962) Energy flow in the salt marsh ecosystem of Georgia. *Ecology*  
1115 43(4):614–624
- 1116 Temmerman S, Govers G, Wartel S, Meire P (2003) Spatial and temporal factors  
1117 controlling short-term sedimentation in a salt and freshwater tidal marsh,  
1118 Scheldt estuary, Belgium, SW Netherlands. *Earth Surf Process Landforms*  
1119 28:739–755
- 1120 Temmerman S, Bouma TJ, Govers G, Lauwaet D (2005) Flow paths of water and  
1121 sediment in a tidal marsh: relations with marsh developmental stage and  
1122 tidal inundation height. *Estuaries* 28(3):338–352. doi:10.1007/BF02693917
- 1123 Temmerman S, Bouma TJ, Van de Koppel J, Van der Wal D, De Vries MB, Herman  
1124 PMJ (2007) Vegetation causes channel erosion in a tidal landscape. *Geology*  
1125 35:631–634. doi:10.1130/G23502A.1
- 1126 Temmerman S, Moonen P, Schoelynck J, Govers G, Bouma TJ (2012) Impact of  
1127 vegetation die-off on spatial flow patterns over a tidal marsh. *Geophys Res  
1128 Lett* 39:L03406
- 1129 Tonelli M, Fagherazzi S, Petti M (2010) Modeling wave impact on salt marsh  
1130 boundaries. *J Geophys Res* 115:C09028. doi:10.1029/2009JC006026
- 1131 Torres R, Mwamba MJ, Goni MG (2003) Properties of intertidal marsh sediment  
1132 mobilized by rainfall. *Limnol Oceanogr* 48:1245–1253
- 1133 Turner RE, Baustian JJ, Swenson EM, Spicer JS (2006) Wetland sedimentation from  
1134 Hurricanes Katrina and Rita. *Science* 314:449–452
- 1135 Valiela I, Cole ML, et al. (2000) Role of salt marshes as part of coastal landscapes.  
1136 In: Weinstein MP, Kreeger DA (ed) *Concepts and controversies in tidal  
1137 marsh ecology*, 23. Springer, New York, p 38
- 1138 Van Damme S, Struyf E, Maris T, Ysebaert T, Dehairs F, Tackx M, Heip C, Meire P  
1139 (2005) Spatial and temporal patterns of water quality along the estuarine  
1140 salinity gradient of the Scheldt estuary (Belgium and The Netherlands):  
1141 results of an integrated monitoring approach. *Hydrobiologia* 540:29–45
- 1142 Vieillard AM, Fulweiler RW, Hughes ZJ, Carey JC (2011) The ebb and flood of  
1143 silica: quantifying dissolved and biogenic silica fluxes from a temperate salt  
1144 marsh. *Estuarine Coastal Shelf Sci* 95:415–423
- 1145 Vignudelli S, Santinelli C, Murru E, Naincinini L, Seritti A (2004) Distributions of  
1146 dissolved organic carbon (DOC) and chromophoric dissolved organic matter  
1147 (CDOM) in coastal waters of the northern Tyrrhenian Sea (Italy). *Estuarine  
1148 Coastal Shelf Sci* 60(1):133–149
- 1149 Volkman JK, Revill AT, Holdsworth DG, Fredericks D (2008) Organic matter  
1150 sources in an enclosed coastal inlet assessed using lipid biomarkers and  
1151 stable isotopes. *Org Geochem* 39(6):689–710

- 1152 Voulgaris G, Meyers ST (2004) Temporal variability of hydrodynamics, sediment  
1153 concentration and sediment settling velocity in a tidal creek. *Cont Shelf Res*  
1154 24:1659–1683
- 1155 Wilson AM, Morris JT (2012) The influence of tidal forcing on groundwater flow and  
1156 nutrient exchange in a salt marsh-dominated estuary. *Biogeochemistry* 108:27–38
- 1157 Xu YP, Mead RN, Jaffe R (2006) A molecular marker-based assessment of  
1158 sedimentary organic matter sources and distributions in Florida Bay.  
1159 *Hydrobiologia* 569:179–192
- 1160 Zhou J, Wu Y, Zhang J, Kang Q, Liu Z (2006) Carbon and nitrogen composition  
1161 and stable isotope as potential indicators of source and fate of organic  
1162 matter in the salt marsh of the Changjiang Estuary, China. *Chemosphere* 65  
1163 (2):310–317

1164 doi:10.1186/2192-1709-2-3  
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