

	This item is	s the	archived	peer-reviewed	author-version	of:
--	--------------	-------	----------	---------------	----------------	-----

Fuels and chemicals from equine-waste-derived tail gas reactive pyrolysis oil : technoeconomic analysis, environmental and exergetic life cycle assessment

Reference:

To cite this reference: http://hdl.handle.net/10067/1455890151162165141

Sorunmu Yetunde E., Billen Pieter, Elkasabi Yaseen, Mullen Charles A., Macken Nelson A., Boateng Akwasi A., Spatari Sabrina.- Fuels and chemicals from equine-waste-derived tail gas reactive pyrolysis oil: technoeconomic analysis, environmental and exergetic life cycle assessment ACS Sustainable Chemistry and Engineering - ISSN 2168-0485 - 5:10(2017), p. 8804-8814 Full text (Publisher's DOI): https://doi.org/10.1021/ACSSUSCHEMENG.7B01609

uantwerpen.be

Fuels and chemicals from equine waste derived tail gas reactive pyrolysis oil: techno-economic analysis, environmental and exergetic life cycle assessment

Yetunde. E. Sorunmu¹, Pieter Billen^{1, 4}, Yaseen Elkasabi², Charles. A. Mullen², Nelson. A. Macken³, Akwasi. A. Boateng², Sabrina Spatari¹*

¹Drexel University, Department of Civil, Architectural and Environmental Engineering, 3141

Chestnut Street, PA 19104, United States

²Eastern Regional Research Center, Agricultural Research Service, U.S. Department of Agriculture, 600 E. Mermaid Lane, Wyndmoor, PA 19038, United States

³Swarthmore College, Department of Engineering, Swarthmore, PA 19081, United States

⁴University of Antwerp, Biochemical Green Engineering and Materials, Salesianenlaan 90, 2660 Hoboken, Belgium

Keywords: Renewable Jet Fuel, Phenolic Compounds, Equine Waste, Life Cycle Assessment, Greenhouse Gas Emissions, Social Cost of Carbon, Exergy Analysis.

^{*} Corresponding author, <u>ss3226@drexel.edu</u>, (+1) 215 571 3557

Abstract

Horse manure, the improper disposal of which, imposes considerable environmental costs, constitutes an apt feedstock for conversion to renewable fuels and chemicals when tail gas reactive pyrolysis (TGRP) is employed. TGRP is a modification of fast pyrolysis that recycles its noncondensable gases and produces a bio-oil low in oxygen concentration and rich in naphthalene. Herein, we evaluate the co-production of phenol as a value-added renewable chemical alongside jet-range fuels within distributed TGRP systems using techno-economic analysis and life cycle assessment. We investigate the metrics global warming potential (GWP), cumulative exergy demand (CExD), and cost for the conversion of 200 dry metric tons per day of horse manure to bio-oil and its subsequent upgrade to hydrocarbon fuel and phenolic chemicals. Assigning credits for the offset of the co-products, the net GWP and CExD of TGRP jet fuel are 10 g of CO₂ eq. and 0.4 MJ per passenger kilometer distance traveled, respectively. These values are considerably lower than the GWP and CExD of petroleum-based aviation fuel. The minimum fuel selling price of the TGRP jet fuel (\$1.35-1.80 L⁻¹) is estimated to be much greater than that of petroleum-based aviation fuel (\$0.42 L⁻¹) except under optimized fuel conversion and co-product market conditions (\$0.53-\$0.79 L⁻¹) when including a market price for carbon.

Introduction

Horse owners in North America pay an estimated \$50 million dollars per month¹ to properly dispose of 7 million tons of equine waste (horse manure and bedding)²⁻³ a cost that reaps no economic value. On the other hand, turning the feedstock into a value-added product could be a cost-effective means of using rather than disposing of the resource. One avenue for recovering value from horse manure is to convert it to biofuels and chemicals using thermochemical processes such as fast pyrolysis⁴⁻⁵. Such a strategy could support a renewable economy using biomass resources⁶⁻⁸, abate climate change, and support the development of rural communities. The successful contribution of biofuels to a renewable energy economy requires that they are energy efficient and cost effective and that they do not introduce additional environmental impacts.9 Currently, resource efficiency and cost limit the commercialization of biofuels. To date, biofuels cannot compete with petroleum-derived products like aviation fuel (\$0.42 L⁻¹)¹⁰, given the historic fluctuations and more recent drop in the price of petroleum, most noticeably demonstrated by gasoline (\$0.87 L⁻¹ in 2008 to \$0.61 L⁻¹ in 2015¹¹). Hence, added value can be sought in the production of biobased chemicals, which may sell for a higher unit price than liquid fuels, alongside of biofuels.⁸

In 2004, the US Department of Energy (DOE) created a single biomass program that merged previous programs for biofuels, bio-power, and biobased products. ¹² The objective of this merger was to promote biorefineries producing multiple products, including higher-value chemicals, along with fuels and power. ¹² Phenol, one of such chemicals present in pyrolysis oil, is examined in this paper. Phenol is a relatively valuable chemical commodity that sells for ~ \$1.3kg⁻¹ ¹³ (compared to aviation fuel at \$0.5 kg⁻¹) and is typically synthesized from crude oil distillates through the

cumene process¹⁴. Additionally, phenolic compounds are used as partial phenol substitutes in the phenol formaldehyde resol resin process ¹⁵⁻¹⁶.

With the goal of co-producing value-added chemicals, Mullen *et al.*^{5, 17} used tail gas reactive pyrolysis (TGRP), a variation of fast pyrolysis in which the non-condensable gases generated during the reaction are recycled into the reactor, to convert horse manure to a bio-oil rich in naphthalenes that could increase the yield of value–added chemicals.¹⁷ Unlike fast pyrolysis, TGRP creates a reductive atmosphere in the pyrolysis reactor, which results in a highly aromatic bio-oil with lower oxygen content than bio-oil produced from fast pyrolysis.¹⁷

The reactive oxygenated compounds present in fast pyrolysis bio-oil render it incompatible with petroleum refinery infrastructure. The oxygenated compounds of TGRP oil are predominantly phenols (Table S1 in SI)¹⁸, which due to their thermal stability, can be distilled and recovered at relatively high purity, compared with fast pyrolysis bio-oil.

The oxygen poor hydrocarbons obtained after distillation of TGRP oil can further undergo hydrodeoxygenation over common catalysts to yield almost completely deoxygenated hydrocarbons that can be directly blended in a petroleum refinery (drop-in) as well as high levels of concentrated phenols that can potentially be separated. Hydrodeoxygenation (HDO) has been widely studied as a method of upgrading fast pyrolysis bio-oil and as such has been the subject of techno-economic analysis (TEA) and life cycle assessment (LCA) studies¹⁹⁻²¹. These include studies on the LCA of technologies and developments for the production and upgrading of fast pyrolysis bio-oil that are focused on upgrading via hydrotreating and hydrocracking with corn stover^{20, 22} and poplar¹⁹ as feedstocks. One study²³ concluded that the majority of the environmental impact is due to direct emissions to air and the production of natural gas and electricity consumed in the process. Another study²² showed that greenhouse gas (GHG) emissions

of biofuel produced from upgraded fast pyrolysis oil may be reduced by 88% relative to gasoline and 94% relative to diesel fuel, exceeding the renewable fuel standard- 2 (RFS-2) requirements which states that the life cycle GHG emission reduction threshold for any biomass-based biofuel is 50%.

The overall objective of the current study is to use LCA and TEA to examine the environmental performance and cost of isolating and extracting value-added chemicals, in this case phenol and its derivatives, and fuel from TGRP oil produced from horse manure, and compare it to the dual stage hydrodeoxygenation (HDO) upgrading process that is required for bio-oil produced from fast pyrolysis. HDO upgrading consists of mild hydrotreating followed by severe hydrotreating, a sequence of unit operations that result in a stable bio-oil²⁴. We use experimental results from Elkasabi, *et al* (2015); in whose study TGRP oil from three feedstocks (horse manure, eucalyptus and switchgrass) was analyzed. Whereas other studies performed comparative LCA of renewable fuels using biomass such as forest residue²⁵, this study exclusively focuses on the use of horse manure due to the favorable quality of its TGRP bio-oil and that the waste feedstock is available in New York State and Pennsylvania^{4, 18, 26}. While horse manure can be disposed by spreading, composting and hauling²⁷, these methods come with high capital and operating costs to comply with regulation and still result in poor water quality²⁸. Herein, we evaluate the conversion of horse manure to fuels as an alternative to disposal.

Methods

Techno-economic analysis and life cycle assessment (ISO 14040)²⁹ were used to investigate the economic and life cycle environmental and exergetic performance of processing horse manure using TGRP and upgrading the TGRP oil to phenol and jet-range fuel. Experimental data from Elkasabi *et al.*¹⁸ were used to construct thermodynamic conversion models using Aspen Plus

software³⁰, which in turn were integrated into a comprehensive life cycle inventory that considers feedstock provision, biomass conversion, TGRP-oil upgrading to phenols and jet-range fuel, and final consumption.

Biomass Supply and Logistics

Biomass supply was investigated to guarantee security of horse manure supply in the production of TGRP-fuel in units of a 200 metric tons per day (MTPD) projected capacity⁸. To illustrate the satisfactory supply potential of horse manure in New York State, we set spatial boundaries using a geographic information system (GIS) map. Data from the USDA 2012 census of agriculture³¹ were merged with an equine waste study in New York State.²⁶

We estimated the total equine waste generated as the product of the number of farms across the state and the quantity of waste generated per farm. The total amount of equine waste available in New York State was estimated as the product of the total number of farms (10,207) taken from the USDA 2012 census 31 and the average dry equine waste generated per farm (0.17 MRPD) taken from Shayya *et al.* 26 .

Our study evaluated the production of TGRP oil, assuming a daily supply of 200 metric tons of horse manure, which could be supported given the available supply in New York State. Horse manure, having a potential higher heating value of 19.5 GJ per dry metric ton ²⁶ (supporting information (SI) - Table S2), comprises about 25% of the total equine waste²⁶, and the supply data on a county level was adapted accordingly with respect to the obtained equine waste generation data.

Furthermore, we used Arc GIS ³² to determine possible locations of the facility. We used countylevel horse manure supply data to predict likely locations for constructing 200 MTPD production facilities; however, more detailed logistics studies would consider additional factors such as road and rail infrastructures for feedstock transport, and access to labor.³³⁻³⁴ We limit feedstock transport to an average distance of 80.5 km (50 mile) radius similar to prior literature on feedstock-biorefinery logistics ^{33, 35}, assuming that beyond this radius the environmental impact and cost of transportation of biomass become disproportionate.

The GIS map for the supply logistics of horse manure (Figure 1) indicates that in most areas of New York, there will be generation of horse manure within a 80.5 km radius average. More details on data calculations for the map are found in the Table S2.

Although TGRP technology may be feasible with other types of manure such as cow or poultry manure ³⁶⁻³⁷ and fast pyrolysis has been used previously on poultry waste, we use horse manure due to its geographical supply in the North East and that it has been tested using TGRP¹⁸ at laboratory scale yielding bio-oil compositions modeled in this paper.

While a 200 MTPD facility is small in scale relative to a conventional crude oil facility, we posit that the TGRP process and subsequent extraction would address waste treatment issues as well as contribute to an existing supply of bio-jet fuel.

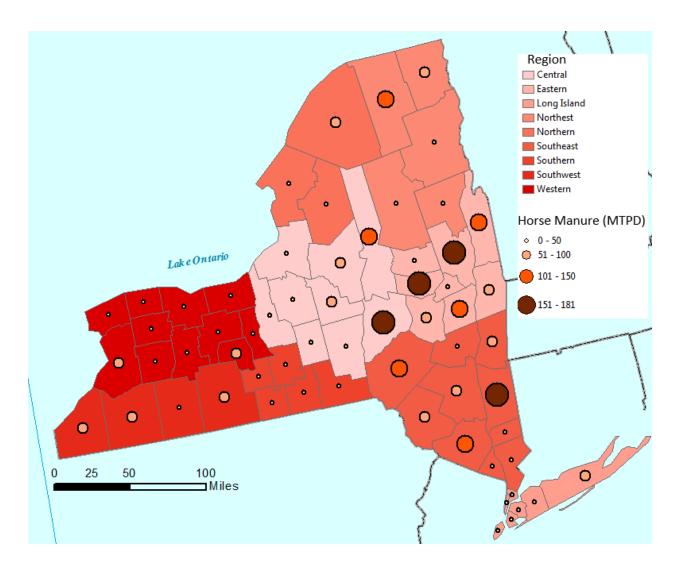


Figure 1. New York State map divided into counties. Color blocks represent the regions in New York State and graduated symbols represent horse manure availability in metric tons per day. Distributed pyrolysis systems scaled up to 200 MTPD would be feasible in multiple locations in New York State.

Biomass Conversion

The TGRP and oil upgrading facility is assumed to receive horse manure feeds of 200 MTPD, a capacity compatible with the small clusters of the equine industry and a scale on par with small to medium-sized fast pyrolysis facilities that could take agricultural residues³⁸ and forestry biomass³⁹.

The TGRP oil yield from horse manure is assumed to be 36.9 wt.%, based on previous experimental work⁵. Major processing steps include biomass preparation, TGRP oil production and TGRP oil upgrade. Biomass preparation and fast pyrolysis have been studied previously⁴, and these results are used to inform the current study. In this study, we build a 200 MTPD biomass conversion model using Aspen Plus³⁰ based on TGRP processing conditions analogous to the work of Hammer *et al.* ⁴ who studied fast pyrolysis at a smaller scale of 5 MTPD, by increasing the process scale, adding a pre-heater, a blower and recycling the stream of non-condensable gases into the fluidized bed (Figure S1).

The Aspen Plus model simulated for the conversion step was based on experiments that treated manure; however, in commercial production equine waste, which consists of both manure and bedding, would be used as feedstock. The LCA feedstock harvest step includes manure and bedding collection but the conversion step assumes only the manure portion is converted to fuels and chemicals, giving a lower bound estimate of production volume. Also, because elemental compositions of horse manure and bedding are similar¹⁸, bio-oil yield would not vary significantly.

The LCA and TEA in this paper used data from previously published bench-scale experiments ^{5, 18} that demonstrated the upgrading of TGRP oil through extraction of bio-oil distillates and separation of phenols.

Aspen Plus ³⁰ was used for process modeling to establish mass and energy balances of the studied TGRP process, scaled to 200 MTPD of horse manure. Furthermore, it was used to estimate utility inputs such as electricity costs for operating pumps and compressors and water utilities for cooling operations. Data for the pre-processing used in the fast pyrolysis Aspen Plus models of Hammer et al.⁴ were applied to this TGRP model. The model was extended with the upgrading process of TGRP oil (distillation, extraction and hydrogenation), using published experimental results^{5, 18}.

In the upgrading process, the thermodynamic models used for the unit operations vary depending on the operation and the input streams. For instance, we used the Peng-Robinson and RK-Soave equations of state in the simulation for major units such as reactors, vessels and coolers because they are suitable for modeling hydrocarbon systems. In the simulation, the condensed TGRP oil was heated to 150 °C from 25 °C using medium pressure steam (Figure S3 in SI). The heated oil was fed into a flash drum as a single-stage distillation process; we estimated a phase-equilibrium separating the vapor phase (VOC1 representing volatiles) from the liquid (Figure S3 in SI). The resulting liquid S10 (volatiles with melting point ≥ 150 °C) stream (Figure S3 in SI) was heated to 420 °C before entering another flash drum. Due to the high melting point of biorenewable coke (modeled as benzopyrene), it exited the flash drum as liquid and was separated from the process.

The vapor phase (VOC1 stream-Figure S3 in SI) was cooled to 90 °C and entered a decanter to remove any excess water from the process. The water-free volatiles stream (VOC2) was mixed with a cooled stream (VOC2B), which exited as a vapor from the second flash drum. After distillation, the distillates underwent a two-step extraction process that separated the phenols from the hydrocarbon stream. The first process step (Figure S4 in SI) was a phenol extraction; NaOH entered the stoichiometric reactor, which was operated at 138 °C, deprotonating the phenols (phenol and cresols) by increasing alkalinity, giving sodium phenoxide and sodium cresolate as phenolic salts. In the second step (shown in Figure S4 in SI), the mixture with phenolic salts was acidified with HCl and decanted, forming phenols in one stream and aqueous NaCl in the other stream. The hydrocarbon stream, consisting of olefins and aromatics, was hydrogenated in two stages (Figure S5 in SI). Both stages were carried out in continuous stirred tank reactors (CSTR) RXTR3, operated at 80 °C, 50 atm, and RXTR4, operating at 200 °C, 75 atm.

Life Cycle Assessment

The primary product we evaluate is the fuel product with the phenol chemical serving as coproduct. The upgraded TGRP-fuel produced is similar in composition to aviation fuel, thus we assume it will serve those markets. Hence, the functional unit was defined as 1 MJ of fuel and 1 person-kilometer (PKM) of travel by aircraft. The co-produced phenol is compared with petroleum based phenol made using the cumene-phenol or hock process, an industrial process used to produce both phenol and acetone from benzene and propylene. A LCA model was constructed with Simapro⁴⁰ software, using input parameters specified in the feedstock harvest and collection, and the mass and energy balances derived from the chemical process model developed in Aspen Plus. The metrics we evaluated using life cycle impact assessment (LCIA) were the 100 year global warming potential (GWP) for CO₂, CH₄, and N₂O according to IPCC⁴¹ and the cumulative exergy demand (CExD).⁴² These metrics were used to describe the products (jet-range fuels and phenol) of the separation and upgrading of condensed TGRP oil. We used the GWP-100 metric in accordance to the benchmark for new biofuel standards given by RFS-2 which states that advanced biofuels must reduce life cycle GHG emissions by 50% according to the advanced fuel designation ⁴³. While many biofuel LCA studies include GWP as a life cycle assessment metric ⁴⁴⁻⁴⁶, CExD has not been as widely used ⁴⁷. Exergy is a thermodynamic measure of the maximum theoretical available work from a substance if it were to achieve equilibrium with the environment. Since exergy is a measure of available work, it represents a more nuanced and complete indicator of resource use compared to cumulative energy consumption (CED). ⁴⁸⁻⁵⁰ Energy can be converted into different forms, but exergy is consumed in all processes. CExD also takes into account the consumption of non-energetic raw materials. However, for non-renewable energy intensive products, results are similar to CED.

The system boundary (Figure 2) of this study is from well-to-flight; whereby the end-of-life of the fuels produced coincides with their use.

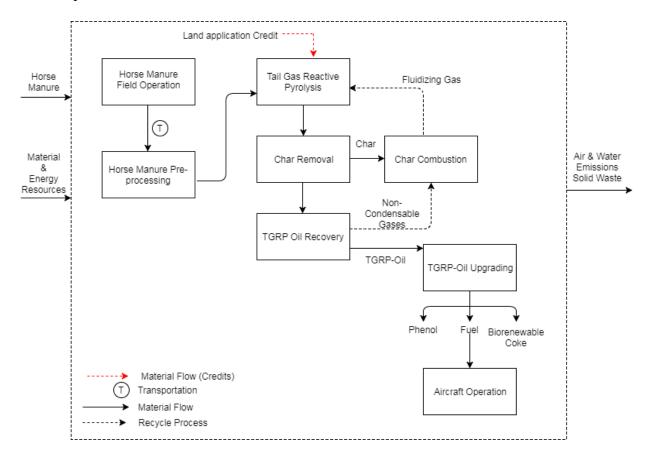


Figure 2. Life cycle system boundary for the production of fuel and phenol from horse manure via TGRP and upgrading through isolation and extraction of TGRP- oil distillates.

Data used for the life cycle inventory model include the material and energy inputs for conversion of horse manure to upgraded biofuel (Figure 2). Inputs such as electricity reflect the region's supply, the Northeast Power Coordinating Council (NPCC) grid. We assumed that the average distance the horse manure is transported from individual sites to the equine facility is 80.5 km ³⁵. The equine facility will be the major site to which the other surrounding farms can deliver their horse manure. After delivery, the horse manure is dried as a preprocessing step (Figure 2).

Allocation of environmental burdens in multiple product systems such as this, in which fuels and phenols are co-produced, is required in LCA. ISO 14040-44 ²⁹ recommends that the allocation methods used shows the physical relationship like mass and energy content or other significant variables such as economic value of the products ⁵¹. In the TGRP process, allocation of the impact between oil and char was done on a mass basis (90% oil, 10% char), since mass allocation is more appropriate for a mixture of chemical and fuel. In the TGRP oil upgrade, we use mass, energy and economic allocation similar to previous studies ⁵²⁻⁵³ because the effect of phenols extraction can vary depending on the allocation method used. Herein, mass fractions, market price and energy values are shown in Table S5 in the SI. Energy values were obtained from Aspen Plus results of HHV and price references are listed in Table S7 in the SI. System expansion was used in the impact calculations because phenols are produced as co-products, hence we seek to capture changes in environmental impacts as a consequence of displacing petroleum based phenols.

Process inputs for the TGRP process (Table S3 in SI) and TGRP oil upgrading (Table S4 in SI) were used to construct the life cycle inventory (LCI). Fuel and coke are produced in the highest quantity followed by phenol and sodium chloride (Table S5 in SI); these values are used to inform the LCI. Fractions of energy, mass, and market price (Table S6 in SI) are used to allocate LCIA results to individual TGRP products. We assume the TGRP-fuel produced will be used in a short haul plane that travels less than 1609 km similar to the Boeing 737-800W currently used by United Airlines-Eco-skies. According to the specifications of the Boeing 737-800W, we assume the aircraft is a 162 seater with fuel efficiency of 1.1 kg.PKM-1.

Finally, we examine the sensitivity of feedstock transportation and credits from the avoided GHG emissions from manure disposal on life cycle GWP.

Techno-Economic Analysis

To calculate the incremental cost of adding a TGRP oil upgrading system to a fast pyrolysis operation, the material and energy flows obtained from Aspen Plus were used to size the major equipment in the upgrade process. We estimated capital costs based on previous reports^{7, 38, 55} and used the Aspen Plus Capital Cost estimator software ⁵⁶. Costs were estimated using a desired rate

of return of 10% and projected in 2013 U.S. dollars using a cost basis from the Aspen Process

Economic Analyzer. All other assumptions made in the TEA are found in the SI.

In addition, a sensitivity analysis was conducted to examine the effects that different model assumptions such as variation in the cost of individual raw materials (hydrogen, catalyst and feedstock), market price of the major co-product (phenol), and the biofuel yield have on the minimum fuel selling price (MFSP). To investigate the relationship between environmental and economic impacts of the processes we use social cost of carbon (SCC), a method that internalizes the cost of GHG emissions (and savings).

Results and Discussion

Process Yield, Energy Use, and Production Volume

In the modeled TGRP facility, 200 MTPD of horse manure (8,300 kg.h⁻¹) are fed into the system, producing about 74 metric tons of TGRP oil per day (3075 kg.h⁻¹), corresponding to a yield of 36.9% on a mass basis. The heat for the pyrolysis process is transferred from the char combustor by fluidization sand. Fuels, phenols, sodium chloride and biorenewable coke are separated or formed during the TGRP oil upgrade. In the upgrade process, energy is internally generated by enthalpies of solution and exothermic reactions such as the cooling of volatiles; serving as a heating source for the flash drums, heaters, and separators.

The analysis shows that available horse manure supply in New York State could yield about 4.3 million gallons per year of bio-jet fuel (SI Table S2). Although we do not base our study on a regional scale, if we were to include equine farms in the North-East region, bio-jet fuel supply from Pennsylvania alone would add an additional 5.7 million gallons per year (SI Table S2). Given that the 4.3 million gallons produced would be 100% renewable jet fuel and that bio-refineries may blend 25% bio-based and 75% petroleum-based fuel in the near term (c. 2020), ⁵⁷ New York State's equine waste jet fuel production would supply 5.5 billion gallons of 25%/75% bio-jet fuel to the market. Also, 4.3 million gallons per year is a conservative estimate since we only consider horse manure (25% of the total equine waste) and not horse bedding. If all equine waste were converted with the same bio-oil yield for horse manure and bedding, then New York State could produce a total of 172 million gallons per year and meet about 3% of total U.S aviation fuel demand ⁵⁸.

Life Cycle Metrics

The life cycle GWP and CExD were calculated for the production of fuel and its co-product (phenols) from the upgrade of TGRP oil, using functional units of 1 MJ of fuel (Figures S6 and S8 in SI) and 1 PKM (Figures 3a and 3d). Due to the different values of the mass, price and energy content of the co-products, the allocation results found in the supporting information vary by method. The high market price of phenol (\$1.28 kg⁻¹) compared to fuels (\$0.5 kg⁻¹) translates into a low economic allocation for both the GWP and CExD of fuel.

Compared to other allocation methods, the fuel product has a higher GWP when allocated by mass, because of its high mass fraction among all product flows. (Figure S6 in SI). Moreover, regardless of the allocation method, the TGRP-fuel has a lower GWP than aviation fuel on a well-to-wheel basis. Herein, we assumed similar GHG emissions from the distribution distance of

TGRP fuel and aviation fuel, due to similar distances from refinery to pump. The impact of horse manure biofuel is almost entirely caused by its cradle-to-gate emissions, whereas the major share of the GWP of petroleum based aviation fuel comes from operation, i.e. CO₂ emissions during aviation fuel combustion. This is because the TGRP upgraded fuel is assumed to have zero net GHG emissions during combustion, as its carbon is biogenic. Overall, the analysis suggests (Figure 3a) that the TGRP-fuel emits less than 10% (10 g of CO₂ eq. per PKM) of the GHGs of aviation fuel (110 g of CO₂ eq. per PKM). The GWP of the TGRP upgraded fuel product shown in Figure 3b meets the RFS-2⁴³ standards; hence, the TGRP fuel product is environmentally preferable relative to aviation fuel on a life cycle basis. When the GWP of the TGRP process is allocated according to economic value, the GWP of phenol is slightly higher (9%) than that of petroleum based phenol (30g CO₂ eq. per PKM of fuel) ⁴⁰ (excluding the end-of-life impact of phenol applications) due to its high economic value, but lower using mass and energy allocations (Figure S6 in SI). This validates the results of other studies ⁵⁹⁻⁶⁰ that have argued that different allocation methods determine the outcome of the environmental impacts on the products and co-products. Also, this suggests that it is more beneficial to separate the phenols in the TGRP upgrade process than to produce only fuels (as is the conventional hydrogenation (HDO) method)¹⁹. Using system expansion yields a higher net GWP for TGRP-fuel than when it is allocated on an economic, mass or energy basis (Figure S6 in SI). This is mostly because in all of the allocation methods, a large share of the GWP is attributed to the byproducts, mainly phenol and coke. The total impact of the TGRP process is identical for all allocation methods, but a reduced GWP for TGRP fuel induced by shifting the impact to phenol or coke is artificial. Therefore, in the interest of the functional unit defined, i.e. the production of 1 MJ equivalent of TGRP-fuel or 1 PKM of distance traveled, we opt to discuss only system expansion hereafter ^{25, 61-63}. This means that the entire GWP of the

47

48

49

50

51

52

53

54

55

56

57

58

59

60

61

62

63

64

65

66

67

68

TGRP process is allocated to the fuel, and afterwards corrected for the credits achieved by phenol and coke production, offsetting phenol from the cumene process and coke as byproducts from the production of liquid fuels from crude oil. Nonetheless, the entire GWP of the TGRP process (10 g CO₂ eq.), producing the fuel, coke and phenols, is still significantly lower than the GWP of petroleum based aviation fuel (110 g CO₂ eq.) ⁴⁰ combined with phenol from the cumene process (30 g CO₂ eq.), per PKM of distance traveled. A previous study⁶⁴ determined that the GWP (Figure 3b) of the conversion of popular biomass to drop-in bio-jet fuel via the bioconversion platform ranges from 30 g CO₂ eq. to 70 g CO₂ eq. per MJ of fuel burned, depending on the hydrogen generation method used in the hydrogenation steps. Bio-jet fuel (LG) represents a fuel for which the hydrogen for upgrading comes from ligningasification (LG) and natural gas is used for heat and steam. Bio-jet (LG-HF) represents a fuel produced from a process that uses lignin gasification for hydrogen generation and hog fuel (HF) for heat and steam. When comparing this range to the GHG emissions from the current process (TGRP oil upgrade) without the credits from the co-products (40 g CO₂ eq. per MJ), and the various process contributions, the GHG credit for phenol (30 g CO₂ eq.) is larger than the extra impact of the separation process (Figure 3a). In addition, we explore the effects of not disposing the manure and including the GHG emissions as a credit (Figure 3b). These results show that the average GHG emissions from land application reduce the TGRP fuel GHG emissions to 4.4 g CO₂ eq. The results for CExD assessment are given in Figures S8, S9, 3c and 3d. Previous studies used exergy assessment to evaluate the production of biofuels such as biodiesel and ethanol 49, 65-68. Most relevant for our study is the work of Keedy, et al. ⁴⁹, who used CExD as an assessment metric for evaluating the sustainability of bio-oil production via fast pyrolysis using three feedstocks, one of which was horse manure. When analyzing the CExD metric, the trend in allocation results

70

71

72

73

74

75

76

77

78

79

80

81

82

83

84

85

86

87

88

89

90

91

(Figure S8) is similar to the GWP results; whereby the phenol contributes the highest CExD in economic allocation compared to the other co-products. According to the Ecoinvent database⁶⁹⁻⁷⁰ from which data were drawn, the CExD for the production and combustion of aviation fuel is 1.8 MJ per MJ of fuel. This value is approximately 28% greater than the CExD of the TGRP-fuel produced in this study without co-product credits (1.3 MJ per MJ of fuel). Figure 3c and 3d present the life cycle environmental impact of CExD per person kilometer (PKM) of travel by aircraft passenger (as in Figure 3a for GWP). These figures compare CExD of TGRP fuel with aviation fuel. Credits are also displayed for the TGRP fuel. The dashed bar for TGRP fuel indicates the net CExD (including credits). Figure 3c includes the contribution of CExD by resource for the net amount. Figure 3d indicates the TGRP contribution by process and details credits. Both figures indicate the cumulative exergy destruction per passenger kilometer is much higher for aviation fuel compared to TGRP fuel. Figure 3d indicates that this is true even without credits. The positive TGRP area on Figure 3d represents the CExD for total production. Of the individual process contributions shown on Figure 3d, the results indicate that the majority of the CExD in the overall process is attributed to horse manure pre-processing. The bulk of the energy consumed in the pre-processing step is from electricity used to operate milling, drying, and conveying equipment. This indicates that the material and energy needs of the pre-processing step could be targeted for CExD reduction. Figure 3c provides a breakdown of the net CExD by resource. The majority of the resource inputs used to produce aviation fuel are derived from fossil fuels (petroleum, natural gas, coal). A significant contribution to the CExD of the TGRP fuel is attributable to electricity production. The NPCC grid resources are about 50% fossil (including petroleum, natural gas) and 31% nuclear on an energy basis. However, the CExD percentages are 75% nuclear and 24% fossil. This results in the larger nuclear contribution to CExD on Figure 3c

93

94

95

96

97

98

99

100

101

102

103

104

105

106

107

108

109

110

111

112

113

114

(compared to the fossil contribution). Note the fossil contribution is considered a credit because of the production of phenol and coke.



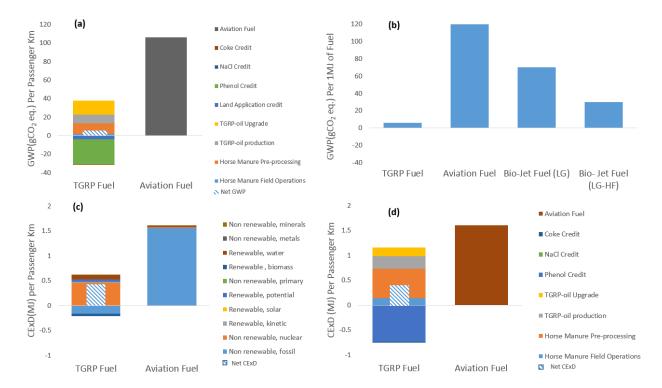


Figure 3. Life cycle environmental impact of TGRP fuel using system expansion. a) GWP comparing TGRP fuel to aviation fuel per passenger km. b) GWP comparing TGRP fuel to aviation fuel and other bio-jet fuels produced from the bioconversion of poplar biomass per 1MJ of fuel produced ⁶⁴. c) CExD comparing TGRP fuel to aviation fuel per PKM by natural resource contribution. d) CExD comparing TGRP fuel to aviation fuel per PKM by process type. LG refers to lignin gasification and LG-HF refers to lignin gasification and hog fuel.

SI Figure S9 compares the CExD of phenol extracted in the TGRP process with the phenol produced by the cumene process. The former demands either more or less exergy than the latter, depending on the allocation method used. With energy and mass allocations, the CExD of phenol

is smaller than the CExD of the cumene process whereas with economic allocation, the CExD of phenol is greater than the CExD in the cumene process. This difference due to allocation method choice is a result of phenol having a higher market value than fuel and the other co-products.

Global Warming Potential Sensitivity

Diverting horse manure from agricultural lands for fuel production may avoid GHG emissions. Manure is often land applied in agriculture to recycle nutrients. When manure is disposed by land, it releases GHGs predominantly in the form of N₂O and CH₄ owing to the presence of inorganic N, microbic available sources of carbon and water. ⁷¹ These emissions are in the range of 16,400-33,500 g CO₂ eq. per ton of manure ⁷² depending on the size of the farm. By using thermochemical conversion to dispose of the manure, we can reduce CO₂ emissions from land application by 88% to 94% (as shown in SI- Figure S12). A scenario that includes the GHG emissions released from land application as a credit (Figure 3b) shows that the average GHG emissions from the TGRP fuel process reduces to 4.4 g CO₂ eq.

In addition, we examined the sensitivity of GWP on feedstock transportation by doubling the horse manure transportation distance to the equine facility. This increases the GHG emission by 0.5g CO2 eq. per passenger kilometers (5% increase in the total GHG emissions), a small amount that aligns with prior literature ^{62, 73} on the transport of biomass feedstocks for biofuels.

Techno-economic Analysis

We estimate the capital and operating costs of TGRP bio-oil upgrading operations from the Aspen Plus simulations, and compare these incremental costs to fast pyrolysis to produce bio-oil from prior research³⁸ (Table 1). Since TGRP and fast pyrolysis use the same equipment, are identical in scale, and the only difference between the two is that the TGRP process recycles the

non-condensable gases whereas fast pyrolysis uses the gases for process energy, we assume the differences in costs between the two are negligible. The additional capital cost of bio-oil upgrading operations that produce value-added products (jet fuel, phenols and green coke) is one third (\$7.7 million) the cost of fast pyrolysis³⁸ (\$24.7 million), which produce bio-oil alone (Table 1). The project capital cost was determined based on 5% per year working capital ⁷⁴, which is a percentage of total capital expense per period showing the amount required to operate the facility until the revenue from product sales is sufficient to cover costs ⁵⁶.

An economic life span of 10 years was chosen for this study for a side by side comparison with a similar 200 MTPD fast pyrolysis study ³⁸. The cost of raw materials and market prices of products are shown in Table S7 in SI. The labor cost is estimated from the number of full time equivalents required for operating the TGRP upgrade process for 7920 hours per year. We use two operators per shift at a unit cost of \$20 per operator per hour and one supervisor per shift with a unit cost of \$35 per supervisor per hour. The total additional fixed operating costs include facility overhead and operating charges.

Total revenue consists of sales of the main product of the upgrade and by-products. The price of fuel used in this analysis is based on the current price of aviation fuel; consequently, the outcome of this comparative study may change based on future market conditions. Sodium chloride and water are considered waste products of the process; hence, they have a negative monetary value and incur waste treatment costs of \$0.002 kg⁻¹ 75 and \$0.074 kg⁻¹ 74, respectively.

Table 1. Capital and operating costs of fast pyrolysis and the incremental cost of TGRP upgrading.

	Fast Pyrolysis ³⁸	TGRP Upgrade
Total Project Capital Costs (\$)	24,700,000	7,680,000
Annual Operating Costs (in \$/year)	9,360,000	4,510,000

MFSP of bio-oil from fast pyrolysis and subsequent TGRP upgrade (\$/L)	1.1ª-1.8	b
Sales from Fuels		2,600,000
Co-product Sales		3,020,000
Total Revenue (\$/year)		5,620,000
Depreciation (10 years, straight line)	2,470,000	768,000
Total Additional Fixed Operating Cost	712,000	423,000
Total Operating Labor & Maintenance Costs	1,130,000	1,060,000
Utilities	768,000	313,000
Raw material	4,280,000	1,950,000

a. MFSP for Option B (Free Feedstock)

Using the economic analysis of Pourhashem *et al.* ³⁸ with the incremental costs of TGRP upgrade, the MFSP of the final upgraded TGRP-fuel is estimated to be \$1.8 L⁻¹ of fuel (MFSP Option B-Table 1), which is comparable to a similar study by Carracso, *et al* ⁵⁵, who investigated the conversion of forest residue to bio-diesel through fast pyrolysis and catalytic upgrading. The MFSP for Option A in Table 1, assumes the agricultural residue feedstock cost from Pourhashem *et al.* ³⁸, \$0.055 kg⁻¹ to approximate the cost of collecting and hauling properly haul the horse manure to the conversion facility and contrasts it with the Option B MFSP, a lower limit in which horse manure is assumed to be given freely⁴ by the farmers as an alternative to paying for disposal and thus is \$0 kg⁻¹. Option A represents a conservative approximation of feedstock cost given that the cost of a waste like horse manure is expected to be low, or possibly zero or negative (if the horse keeper has to pay to have the manure hauled offsite for treatment). However, horse manure is an agricultural waste product with no value, thus, the assumed feedstock cost of \$0 kg⁻¹ (MFSP Option B-Table 1) is plausible, reducing the MFSP to \$1.1 L⁻¹. An investment analysis over a

b. MFSP for Option A (Feedstock price \$0.055 kg⁻¹)

project life of 10 years results in a net present value of \$1.6 million; hence, the project is economically feasible under the assumptions made.

Jones *et al* ⁷ evaluated the production of fuels via pyrolysis and HDO upgrade and found a positive net present with a minimum fuel selling price (MFSP) of \$0.54 L⁻¹, which is far below the MFSP for the TGRP oil upgrade process; \$1.1 L⁻¹. This variation in MFSP can be attributed to the significant difference in scale between the two processes; the biomass supply needed for the HDO process is 10 times higher than the distributed TGRP-fuel process.

Sensitivity Analysis

The MFSP is most sensitive to feedstock cost and yield, and to a smaller extent on the market prices of phenol and coke, but hydrogen and catalyst costs do not have a significant effect on the MFSP because they are used in small quantity compared to the other raw materials in the process (Figure S10). The MFSP value showing sensitivity analysis to feedstock cost depicts a significant reduction from \$1.8 L⁻¹ to \$1.1 L⁻¹ (Table S9 in SI) when the feedstock is treated as a waste (\$0 kg⁻¹), hence making TGRP fuel more economically competitive compared to the base case. If the price of feedstock were to double, assuming it has to be purchased from the farmers, the MFSP increases significantly to \$2.4 L⁻¹. This can be the case if equine waste is in high demand and its supply is limited. Also, because the market price of green coke can change drastically ⁷⁶, it is important to see the effect of this change on the MFSP. However, the tornado plot shows little variation in MFSP due to changes in the price of coke; hence, the MFSP is not too sensitive to the price of coke.

If the yield increases by 25% ^{20, 77}, the MFSP decreases significantly to \$1.4 L⁻¹. Likewise, if the market selling price of phenol increases, the MFSP decreases, and if the selling price decreases,

the MFSP increases. This implies that increasing process yields, increasing the price of co-products and reducing the price of feedstocks can render the process economically feasible.

211

212

213

214

215

216

217

218

219

220

221

222

223

224

225

226

227

228

229

230

231

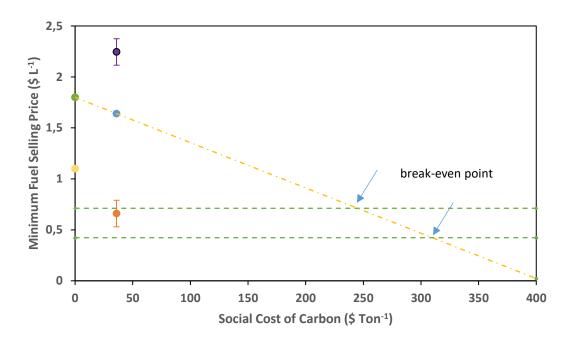
210

209

Environmental and Economic Implications for Advanced Fuel Development and Policy

To fully account for the climate change mitigation benefits of biofuels and support policy decisions in their investment, it is important to include both internal and external costs and measure both direct and indirect costs of a process or product. In this section, we estimate all internal costs associated with TGRP and aviation fuel production and consider the social cost of carbon (SCC), an external cost⁷⁸. The United States Environmental Protection Agency (U.S. EPA)⁷⁹ uses SCC to estimate the climate benefits of set standards. The SCC is a monetary value estimating the economic damages caused by a marginal increase in carbon dioxide emissions in a defined year. In order to overcome the high MFSP, SCC is expressed as a revenue equivalent to the difference in the monetary value of GHG emissions between the renewable fuel and petroleum-based aviation fuel. Using the 2007 year SCC value of \$36 with a 3% discount rate⁷⁹, the TGRP SCC value is \$930,000 yr⁻¹, a quantity representing the reduction in economic damages associated with CO₂. Using Option A (Table 1-MFSP) while including the SCC, the MFSP decreases to \$1.64 L⁻¹. When varying the discount rate (2.5%, 3% and 5%) and statistics (average and 95th percentile)⁸⁰ applied to the SCC, the MFSP will vary from \$1.35 L⁻¹ to \$1.74 L⁻¹. In order to break even with the MFSP of aviation fuel (\$0.42 L⁻¹), the SCC would need to be as high as \$354 per metric ton based on the assumption that the original SCC of \$36 has been underestimated. Finally, to understand how sensitive the MFSP is to variability in the SCC, a sensitivity analysis (Figure 4) was carried out using scenarios reflecting the highest and lowest MFSP. When the cost of raw materials (feedstock, hydrogen and catalyst) is high, yield is low and co-product (phenol

and coke) selling prices are low, the MFSP ranges from \$2.11 to \$2.38 L⁻¹. On the other hand, a scenario that includes the lowest price of raw materials, higher yields and high selling price of coproducts reduces the MFSP to a range of \$0.53-\$0.79 L⁻¹, which could be competitive with the price of petroleum-based aviation fuel on the market. If we add a manure management fee of \$0.008/L, this would have a small effect on the MFSP, reducing it from \$0.66/L to \$0.65/L. Therefore, when all fuel conversion parameters are optimized as would be expected of n^{th 81} biofuel conversion facilities, and when external costs are reflected in the market price of energy, it is possible to economically produce low carbon jet-range fuels and value-added chemicals through TGRP technology.



241242

232

233

234

235

236

237

238

239

240

243

245

246

Option A with SCC of \$36/ton ⁶⁷, baseline scenario and feedstock Price of \$0.055/kg

Option A with SCC of \$36/ton ⁶⁷, upper bound conditions from the sensitivity analysis (tornado plot) and high cost of raw materials, low yield, low selling price of co-products

 Option B with SCC of \$36/ton ⁶⁷, low cost of raw materials, high yield, low selling price of coproducts

Option A without SCC, baseline scenario and feedstock price of \$0.055/kg

Option B without SCC, baseline scenario and feedstock price of \$0/kg

--- MFSP for option A at varying SCC; curve shows break-even points (arrows) with petroleum-based aviation fuel

----Price range of petroleum based aviation fuel (2014-2017)

247	
248	
249	
250	Figure 4. The effect of social cost of carbon on the minimum fuel selling price.
251	Associated Content:
252	Supporting Information:
253	Supporting life cycle inventory data, co-product allocation parameters and results, bio-oil
254	characteristics, feedstock logistics and supply, and economic data are summarized; process flow
255	diagrams from Aspen Plus simulations show details of feedstock conversion and fractionation to
256	fuels and phenolic compounds (PDF).
257	Corresponding Author:
258	S. Spatari, Tel.: +1-215-571-3557. E-mail: ss3226@drexel.edu
259	ORCID:
260	Yetunde E. Sorunmu: 0000-0003-2074-6371
261	Charles A Mullen: <u>0000-0001-5739-5451</u>
262	Akwasi A. Boateng: <u>0000-0002-8496-2201</u>
263	Sabrina Spatari: <u>0000-0001-7243-9993</u>
264	Acknowledgments
265	Funding for this work was provided by the U.S. Department of Agriculture under USDA-NIFA
266	2012-10008-20271. We thank four anonymous reviewers, whose comments improved the
267	manuscript.

268 References

- 1. Hallbar Consulting, Turning Horse Manure into Biomass Renewable Energy.
- https://www.hallbarconsulting.com/horse-manure-into-biomass-renewable-energy/ (accessed March 3, 2017)
- American Horse Council Foundation, Economic Impact of the United States Horse Industry.
 American Horse Council Foundation: Washington, DC, 2005,
- http://www.horsecouncil.org/economics/.
- 3. Smith, C. *Horse Manure Management*; Virginia Cooperative Extension: May 1, 2009, 2009, https://www.pubs.ext.vt.edu/406/406-208/406-208.html
- 4. Hammer, N. L.; Boateng, A. A.; Mullen, C. A.; Wheeler, M. C., Aspen Plus® and economic modeling of equine waste utilization for localized hot water heating via fast pyrolysis.
- Journal of Environmental Management **2013**, 128, 594-601. DOI:
- 280 http://dx.doi.org/10.1016/j.jenvman.2013.06.008
- 5. Elkasabi, Y.; Mullen, C. A.; Boateng, A. A., Distillation and Isolation of Commodity Chemicals from Bio-Oil Made by Tail-Gas Reactive Pyrolysis. *ACS Sustainable Chemistry* & Engineering **2014**, 2 (8), 2042-2052. DOI: 10.1021/sc5002879
- 284 6. Jenkins, T., Toward a biobased economy: examples from the UK. *Biofuels, Bioproducts and Biorefining* **2008**, *2* (2), 133-143. DOI: 10.1002/bbb.62
- 7. Jones, S. B.; Holladay, J.; Valkenburg, C.; Stevens, D.; Walton, C.; Kinchin, C.; Elliott, D.; Czernik, S. *Production of Gasoline and Diesel from Biomass via Fast Pyrolysis*,
- 288 Hydrotreating and Hydrocracking: A Design Case PNNL-18284; 2009; p 76,
- http://www.pnl.gov/main/publications/external/technical_reports/PNNL-18284.pdf
- 8. Jong, E. d.; Higson, A.; Walsh, P.; Wellisch, M., IEA Bioenergy- Task 42 Biorefinery. 2012; p 36. http://www.ieabioenergy.com/publications/bio-based-chemicals-value-added-products-
- 292 from-biorefineries/
- Bioenergy, Bioenergy–a sustainable and reliable energy source. *International Energy Agency Bioenergy, Paris, France* 2009, http://www.ieabioenergy.com/wp-

- content/uploads/2013/10/MAIN-REPORT-Bioenergy-a-sustainable-and-reliable-energy-source.-A-review-of-status-and-prospects.pdf.
- 297 10. IATA-Methodology Fuel Price Analysis http://www.iata.org/publications/economics/fuel-298 monitor/Pages/price-analysis.aspx
- 299 11. Energy Information Administration, U.S Average Monthly Retail Gasoline and Crude oil
 300 prices 2008-2014 Monthly, P. M., Ed. 2015,
 301 http://www.eia.gov/energyexplained/index.cfm?page=gasoline fluctuations.
- 302 12. Bozell, J. J., Holladay. J. E, Johnson, D, White, J.F. *Top Value Added Chemicals from* 303 *Biomass* Pacific Northwest National Laboratory: Oak Ridge, TN, October 2007, 2007; p 79,
 304 http://www.pnl.gov/main/publications/external/technical_reports/PNNL-16983.pdf
- 305 13. ICIS Editorial Europe *Phenol*; ICIS Pricing: 10th January, 2014, 2014; p 3, https://www.icis.com/chemicals/phenol/europe/
- 307 14. Effendi, A.; Gerhauser, H.; Bridgwater, A. V., Production of renewable phenolic resins by thermochemical conversion of biomass: A review. *Renewable and Sustainable Energy* 309 *Reviews* **2008**, *12* (8), 2092-2116. DOI: http://dx.doi.org/10.1016/j.rser.2007.04.008
- 15. Pan, H., Synthesis of polymers from organic solvent liquefied biomass: A review. *Renewable and Sustainable Energy Reviews* 2011, *15* (7), 3454-3463. DOI: http://dx.doi.org/10.1016/j.rser.2011.05.002
- 313 16. Pizzi, A., Recent developments in eco-efficient bio-based adhesives for wood bonding:
 314 opportunities and issues. *Journal of Adhesion Science and Technology* 2006, 20 (8), 829-846.
 315 DOI: 10.1163/156856106777638635
- 17. Mullen, C. A.; Boateng, A. A.; Goldberg, N. M., Production of Deoxygenated Biomass Fast
 Pyrolysis Oils via Product Gas Recycling. *Energy & Fuels* 2013, 27 (7), 3867-3874. DOI:
 10.1021/ef400739u
- 18. Elkasabi, Y.; Mullen, C. A.; Boateng, A. A., Aqueous Extractive Upgrading of Bio-Oils
 Created by Tail-Gas Reactive Pyrolysis To Produce Pure Hydrocarbons and Phenols. *ACS* Sustainable Chemistry & Engineering 2015, 3 (11), 2809-2816. DOI:
 10.1021/acssuschemeng.5b00730
- 19. Hsu, D. D., Life cycle assessment of gasoline and diesel produced via fast pyrolysis and hydroprocessing. *Biomass and Bioenergy* 2012, *45*, 41-47. DOI: http://dx.doi.org/10.1016/j.biombioe.2012.05.019
- 20. Zhang, Y.; Brown, T. R.; Hu, G.; Brown, R. C., Techno-economic analysis of two bio-oil upgrading pathways. *Chemical Engineering Journal* **2013**, 225, 895-904. DOI: http://dx.doi.org/10.1016/j.cej.2013.01.030
- Wright, M. M.; Brown, R. C.; Boateng, A. A., Distributed processing of biomass to bio-oil for subsequent production of Fischer-Tropsch liquids. *Biofuels, Bioproducts and Biorefining* 2008, 2 (3), 229-238. DOI: 10.1002/bbb.73
- 22. Zhang, Y.; Hu, G.; Brown, R. C., Life cycle assessment of the production of hydrogen and transportation fuels from corn stover via fast pyrolysis. *Environmental Research Letters* **2013**, *8* (2), 025001. DOI: 10.1088/1748-9326/8/2/025001
- 23. Iribarren, D.; Hospido, A.; Moreira, M. T.; Feijoo, G., Benchmarking environmental and
 operational parameters through eco-efficiency criteria for dairy farms. *Science of the Total*

- 337 Environment **2011**, 409 (10), 1786-1798. DOI:
- 338 https://doi.org/10.1016/j.scitotenv.2011.02.013
- 24. Elliott, D. C., Historical developments in hydroprocessing bio-oils. *Energy and Fuels* 2007,
 21 (3), 1792-1815. DOI: 10.1021/ef070044u
- 341 25. Zaimes, G. G.; Soratana, K.; Harden, C. L.; Landis, A. E.; Khanna, V., Biofuels via Fast
- Pyrolysis of Perennial Grasses: A Life Cycle Evaluation of Energy Consumption and Greenhouse Gas Emissions. *Environmental Science & Technology* **2015**, *49* (16), 10007-
- 344 10018. DOI: 10.1021/acs.est.5b00129
- Shayya, W.; Ballard, B.; Boateng, A., Assessment of equine waste as a biomass resource in
 New York State. Extension Publications. Morrisville State College: 2016, pp 1-18.
- 27. Equine Facilities Assistance Program *Land Application of Horse Manure*; June 2003, 2003,
- http://www.marincounty.org/~/media/files/departments/pw/mcstoppp/residents/land_application.pdf
- 28. Pascoe, E., Solve the Horse Manure Pile Problem. *Practical Horseman Magazine* December 2011, 2011, http://practicalhorsemanmag.com/article/horse-manure-management-11633
- 352 29. International Organization for Standardization, ISO 14041: Environmental Management:
- Life Cycle Assessment- Goal and Scope Definition and Inventory Analysis. Geneva, 2006, https://www.iso.org/standard/23152.html.
- 355 30. AspenTechnology Aspen Plus, Bedford, Massachusetts, V8.6,
- 356 31. United States Department of Agriculture, 2012 Census Volume 1, Chapter 1: State Level
- Data- New York. Agriculture, U. S. D. o., Ed. United States Department of Agriculture: 2012,
- https://www.agcensus.usda.gov/Publications/2012/Full_Report/Volume_1,_Chapter_1_State Level/New York/.
- 361 32. ESRI, ArcGIS desktop: release 10. Environmental Systems Research Institute, CA 2011.
- 33. Liu, W.; Wang, J.; Richard, T. L.; Hartley, D. S.; Spatari, S.; Volk, T. A., Economic and life cycle assessments of biomass utilization for bioenergy products. *Biofuels, Bioproducts and Biorefining* **2017**, *11* (4), 633-647. DOI: 10.1002/bbb.1770
- 34. Nguyen, L.; Cafferty, K.; Searcy, E.; Spatari, S., Uncertainties in Life Cycle Greenhouse Gas
 Emissions from Advanced Biomass Feedstock Logistics Supply Chains in Kansas. *Energies* 2014, 7 (11), 7125. http://www.mdpi.com/1996-1073/7/11/7125
- 368 35. Aden, A., Ruth, M., Ibsen, K., Jechura, J., Neeves, K., Sheehan, J., Wallace, B.
- 369 Lignocellulosic Biomass to Ethanol Process Design and Economics Utilizing Co-Current
- 370 Dilute Acid Prehydrolysis and Enzymatic Hydrolysis for Corn Stover; NREL/TP-510-32438;
- National Renewable Energy Laboratory Golden, Colorado, 2002; p 154,
- https://www.nrel.gov/docs/fy02osti/32438.pdf
- 36. Lima, I. M.; Boateng, A. A.; Klasson, K. T., Pyrolysis of Broiler Manure: Char and Product
- Gas Characterization†† The mention of firm names or trade products does not imply that
- 375 they are endorsed or recommended by the US Department of Agriculture over other firms or

- similar products not mentioned. *Industrial & engineering chemistry research* **2008**, *48* (3), 1292-1297. DOI: 10.1021/ie800989s
- 37. Mullen, C. A.; Strahan, G. D.; Boateng, A. A., Characterization of various fast-pyrolysis biooils by NMR spectroscopy. *Energy & Fuels* **2009**, *23* (5), 2707-2718. DOI: 10.1021/ef801048b
- 38. Pourhashem, G.; Spatari, S.; Boateng, A. A.; McAloon, A. J.; Mullen, C. A., Life Cycle Environmental and Economic Tradeoffs of Using Fast Pyrolysis Products for Power Generation. *Energy & Fuels* **2013**, 27 (5), 2578-2587. DOI: 10.1021/ef3016206
- 384 39. Fan, J.; Kalnes, T. N.; Alward, M.; Klinger, J.; Sadehvandi, A.; Shonnard, D. R., Life cycle assessment of electricity generation using fast pyrolysis bio-oil. *Renewable Energy* **2011**, *36* (2), 632-641. DOI: https://doi.org/10.1016/j.renene.2010.06.045
- 387 40. SimaPro *SimaPro V. 8.1.1.16*, 8.1.1.16; PRé Consultants: The Netherlands, 2014,
- 388 41. IPCC Climate Change 2007: Synthesis report; Geneva, Switzerland, 2007; p 104,
- 42. Goedkoop M; Oele M; A, d. S.; M, V., Simapro Database Manual: Methods Library (Amersfoort: PRe Consultants). 2008, http://www.pre-sustainability.com/manuals.
- 43. Schnepf, R.; Yacobucci, B. D. In *Renewable fuel standard (RFS): overview and issues*, CRS
 Report for Congress, 2010, https://www.ifdaonline.org/IFDA/media/IFDA/GR/CRS-RFS-Overview-Issues.pdf
- 44. Sills, D. L.; Paramita, V.; Franke, M. J.; Johnson, M. C.; Akabas, T. M.; Greene, C. H.;
 Tester, J. W., Quantitative uncertainty analysis of life cycle assessment for algal biofuel
 production. *Environmental science & technology* 2012, 47 (2), 687-694. DOI:
 10.1021/es3029236
- 45. Vasudevan, V.; Stratton, R. W.; Pearlson, M. N.; Jersey, G. R.; Beyene, A. G.; Weissman, J.
 C.; Rubino, M.; Hileman, J. I., Environmental performance of algal biofuel technology
 options. *Environmental science & technology* 2012, 46 (4), 2451-2459. DOI:
 10.1021/es2026399
- 402 46. Evans, S. G.; Ramage, B. S.; DiRocco, T. L.; Potts, M. D., Greenhouse gas mitigation on marginal land: a quantitative review of the relative benefits of forest recovery versus biofuel production. *Environmental science & technology* **2015**, *49* (4), 2503-2511. DOI: 10.1021/es502374f
- 47. Frischknecht, R.; Jungbluth, N.; Althaus, H.-J.; Doka, G.; Dones, R.; Heck, T.; Hellweg, S.;
 Hischier, R.; Nemecek, T.; Rebitzer, G., The ecoinvent database: Overview and
 methodological framework (7 pp). *The International Journal of Life Cycle Assessment* 2005,
 10 (1), 3-9. DOI: 10.1065/lca2004.10.181.1
- 48. Bösch, M. E.; Hellweg, S.; Huijbregts, M. A.; Frischknecht, R., Applying cumulative exergy demand (CExD) indicators to the ecoinvent database. *The International Journal of Life Cycle Assessment* **2007**, *12* (3), 181-190. DOI: http://dx.doi.org/10.1065/lca2006.11.282
- 49. Keedy, J.; Prymak, E.; Macken, N.; Pourhashem, G.; Spatari, S.; Mullen, C. A.; Boateng, A. A., Exergy Based Assessment of the Production and Conversion of Switchgrass, Equine

- Waste, and Forest Residue to Bio-Oil Using Fast Pyrolysis. *Industrial & Engineering Chemistry Research* **2014**, *54* (1), 529-539. DOI: 10.1021/ie5035682
- 50. Szargut, J.; Morris, D. R., Cumulative exergy consumption and cumulative degree of perfection of chemical processes. *International journal of energy research* **1987,** *11* (2), 245-261. DOI: 10.1002/er.4440110207
- 51. Guinee, J. B., Handbook on life cycle assessment operational guide to the ISO standards. *The International Journal of Life Cycle Assessment* 2002, 7 (5), 311. DOI: https://doi.org/10.1007/BF02978897.
- 52. Fritsche, U. R.; Hennenberg, K. J.; Wiegmann, K.; Herrera, R.; Franke, B.; Köppen, S.; Reinhardt, G.; Dornburg, V.; Faaij, A.; Smeets, E., Bioenergy environmental impact analysis (bias): analytical framework. *FAO: Food and Agricultural Organisation of the United Nations Rome* **2010**, http://www.fao.org/3/a-am303e.pdf.
- 53. Ko, C. H.; Park, S. H.; Jeon, J.-K.; Suh, D. J.; Jeong, K.-E.; Park, Y.-K., Upgrading of biofuel by the catalytic deoxygenation of biomass. *Korean Journal of Chemical Engineering* 2012, 29 (12), 1657-1665. DOI: 10.1007/s11814-012-0199-5
- 430 54. United Airlines Alternative Fuels. https://www.united.com/web/en 431 US/content/company/globalcitizenship/environment/alternative-fuels.aspx
- 55. Carrasco, J. L.; Gunukula, S.; Boateng, A. A.; Mullen, C. A.; DeSisto, W. J.; Wheeler, M. C., Pyrolysis of forest residues: An approach to techno-economics for bio-fuel production. *Fuel* **2017**, *193*, 477-484. DOI: http://dx.doi.org/10.1016/j.fuel.2016.12.063
- 56. AspenTechnology, Aspen Capital Cost Estimator. Aspen Technology: Burlington, MA, 2012.
- 57. Wang, W.-C.; Tao, L.; Markham, J.; Zhang, Y.; Tan, E.; Batan, L.; Warner, E.; Biddy, M.
 Review of Biojet Fuel Conversion Technologies; NREL (National Renewable Energy
 Laboratory (NREL), Golden, CO (United States)): 2016,
 https://www.nrel.gov/docs/fy16osti/66291.pdf
- 58. Energy Information Administration, Production Capacity of Operable Petroleum Refineries.
 U.S. Energy Information Administration: 2017,
 https://www.eia.gov/dnav/pet/pet_pnp_capprod_dcu_nus_a.htm.
- 59. Canter, C. E.; Dunn, J. B.; Han, J.; Wang, Z.; Wang, M., Policy Implications of Allocation Methods in the Life Cycle Analysis of Integrated Corn and Corn Stover Ethanol Production. BioEnergy Research 2016, 9 (1), 77-87. DOI: 10.1007/s12155-015-9664-4
- 60. Murphy, C. W.; Kendall, A., Life cycle analysis of biochemical cellulosic ethanol under multiple scenarios. *GCB Bioenergy* **2015**, *7* (5), 1019-1033. DOI: 10.1111/gcbb.12204
- 449 61. Spatari, S.; Bagley, D. M.; MacLean, H. L., Life cycle evaluation of emerging lignocellulosic
 450 ethanol conversion technologies. *Bioresource technology* **2010**, *101* (2), 654-667. DOI:
 451 10.1016/j.biortech.2009.08.067
- 452 62. Spatari, S.; Zhang, Y.; MacLean, H. L., Life Cycle Assessment of Switchgrass- and Corn 453 Stover-Derived Ethanol-Fueled Automobiles. *Environmental Science & Technology* **2005**, *39* 454 (24), 9750-9758. DOI: 10.1021/es048293+
- 455 63. Zaimes, G. G.; Beck, A. W.; Janupala, R. R.; Resasco, D. E.; Crossley, S. P.; Lobban, L. L.; Khanna, V., Multistage torrefaction and in situ catalytic upgrading to hydrocarbon biofuels:

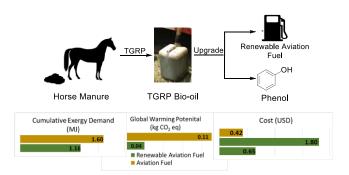
- 457 analysis of life cycle energy use and greenhouse gas emissions. *Energy & Environmental*458 *Science* **2017**, *10* (5), 1034-1050. DOI: 10.1039/C7EE00682A
- 459 64. Budsberg, E.; Crawford, J. T.; Morgan, H.; Chin, W. S.; Bura, R.; Gustafson, R.,
 460 Hydrocarbon bio-jet fuel from bioconversion of poplar biomass: life cycle assessment.
 461 Biotechnology for Biofuels 2016, 9 (1), 170. DOI: 10.1186/s13068-016-0582-2
- 462 65. Dewulf, J.; Van Langenhove, H.; Van De Velde, B., Exergy-Based Efficiency and
 463 Renewability Assessment of Biofuel Production. *Environmental Science & Technology* 2005,
 464 39 (10), 3878-3882. DOI: 10.1021/es048721b
- 465 66. Hovelius, K.; Hansson, P.-A., Energy-and exergy analysis of rape seed oil methyl ester
 466 (RME) production under Swedish conditions. *Biomass and Bioenergy* 1999, 17 (4), 279-290.
 467 DOI: 10.1016/S0961-9534(99)00047-1
- 468 67. Ojeda, K. A.; Sánchez, E. L.; Suarez, J.; Avila, O.; Quintero, V.; El-Halwagi, M.; Kafarov,
 469 V., Application of computer-aided process engineering and exergy analysis to evaluate
 470 different routes of biofuels production from lignocellulosic biomass. *Industrial & Engineering Chemistry Research* 2010, 50 (5), 2768-2772. DOI: 10.1021/ie100633g
- 472 68. Peiró, L. T.; Lombardi, L.; Méndez, G. V.; i Durany, X. G., Life cycle assessment (LCA) and exergetic life cycle assessment (ELCA) of the production of biodiesel from used cooking oil (UCO). *Energy* **2010**, *35* (2), 889-893. DOI: https://doi.org/10.1016/j.energy.2009.07.013
- 69. Ecoinvent. Duebendorf, Switzerland: Swiss Center for life cycle inventories; 2009.
- 476 70. Wernet, G., Bauer, C., Steubing, B., Reinhard, J., Moreno-Ruiz, E., and Weidema, B., The 477 ecoinvent database version 3 (part I): overview and methodology. . *The International Journal* 478 of Life Cycle Assessment **2016**, http://link.springer.com/10.1007/s11367-016-1087-8, 13.
- Thorman, R.; Fangueiro, D.; Cardenas, L.; Amon, B.;
 Misselbrook, T., Manure management: Implications for greenhouse gas emissions. *Animal Feed Science and Technology* 2011, *166*, 514-531. DOI: http://dx.doi.org/10.1016/j.anifeedsci.2011.04.036
- 483 72. Aguirre-Villegas, H. A.; Larson, R. A., Evaluating greenhouse gas emissions from dairy 484 manure management practices using survey data and lifecycle tools. *Journal of Cleaner* 485 *Production* **2017**, *143*, 169-179. DOI: https://doi.org/10.1016/j.jclepro.2016.12.133
- T3. Sheehan, J.; Aden, A.; Paustian, K.; Killian, K.; Brenner, J.; Walsh, M.; Nelson, R., Energy and Environmental Aspects of Using Corn Stover for Fuel Ethanol. *Journal of Industrial Ecology* 2003, 7, 117. DOI: 10.1162/108819803323059433
- 74. Phillips, S.; Aden, A.; Jechura, J.; Dayton, D.; Eggeman, T. *Thermochemical ethanol via indirect gasification and mixed alcohol synthesis of lignocellulosic biomass*; National
 Renewable Energy Laboratory (NREL), Golden, CO.: 2007, https://www.nrel.gov/docs/fy07osti/41168.pdf
- 493 75. Sedivy, V. M. In *Economy of salt in chloralkali manufacture*, National Salt Conference, 494 2008, http://www.salt-

- 495 partners.com/pdf/Gandhidham2008/Economy%20of%20Salt%20in%20Chloralkali%20Man 496 ufacture.pdf
- 76. Edwards, L., The History and Future Challenges of Calcined Petroleum Coke Production and Use in Aluminum Smelting. *JOM* **2015**, *67* (2), 308-321. DOI: 10.1007/s11837-014-1248-9
- 77. Kim, S.; Dale, B. E., All biomass is local: The cost, volume produced, and global warming
 impact of cellulosic biofuels depend strongly on logistics and local conditions. *Biofuels*,
 Bioproducts and Biorefining 2015, 9 (4), 422-434. DOI: 10.1002/bbb.1554
- 78. Jasinski, D.; Meredith, J.; Kirwan, K., A comprehensive review of full cost accounting
 methods and their applicability to the automotive industry. *Journal of Cleaner Production* 2015, 108, Part A, 1123-1139. DOI: http://dx.doi.org/10.1016/j.jclepro.2015.06.040
- 79. Environmental Protection Agency, Technical Support Document: Technical Update of the
 Social Cost of Carbon for Regulatory Impact Analysis. Under Executive Order 12866
 Interagency Working Group on Social Cost of Carbon, U. S. G., Ed. May 2013, Revised July
 2015, https://www.epa.gov/sites/production/files/2016 12/documents/sc co2 tsd august 2016.pdfp 21.
- 510 80. Greenstone, M.; Kopits, E.; Wolverton, A. *Estimating the social cost of carbon for use in us*511 *federal rulemakings: A summary and interpretation*; National Bureau of Economic Research:
 512 2011. DOI:10.3386/w16913.
- 81. Worley, M.; Yale, J. Biomass Gasification Technology Assessment: Consolidated Report;
 National Renewable Energy Laboratory (NREL), Golden, CO.: 2012,
 https://www.nrel.gov/docs/fy13osti/57085.pdf

516

518 For Table of Contents Use Only

Table of Contents (TOC) Graphic:



- 521 LCA and TEA are used to evaluate renewable jet-range fuels and phenolic compounds by tail gas
- reactive pyrolysis of equine waste.

SUPPORTING INFORMATION

The supporting information consists of 18 pages, including cover page, containing 12 figures and 9 tables.

Fuels and chemicals from equine waste derived tail gas reactive pyrolysis oil: techno-economic analysis, environmental and exergetic life cycle assessment

Yetunde. E. Sorunmu¹, Pieter Billen^{1, 4}, Yaseen Elkasabi², Charles. A. Mullen², Nelson. A. Macken³, Akwasi. A. Boateng², Sabrina Spatari¹*

¹Drexel University, Department of Civil, Architectural and Environmental Engineering, 3141 Chestnut Street, PA 19104, United States

²Eastern Regional Research Center, Agricultural Research Service, U.S. Department of Agriculture, 600 E. Mermaid Lane, Wyndmoor, PA 19038, United States

³Swarthmore College, Department of Engineering, Swarthmore, PA 19081, United States

⁴University of Antwerp, Biochemical Green Engineering and Materials, Salesianenlaan 90, 2660 Hoboken, Belgium

SUPPORTING FIGURES

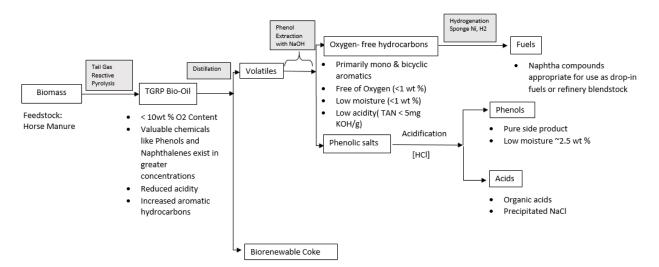


Figure S1. Process flow diagram for tail gas reactive pyrolysis of horse manure to bio-oil and further upgrade through distillation and extraction of the bio-oil distillates, forming value-added chemicals and fuels.

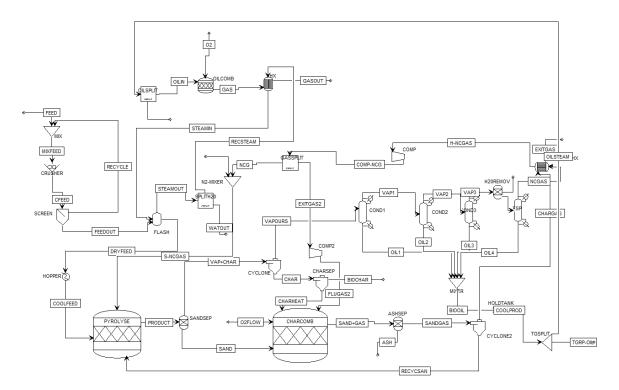


Figure S2. TGRP -Oil production

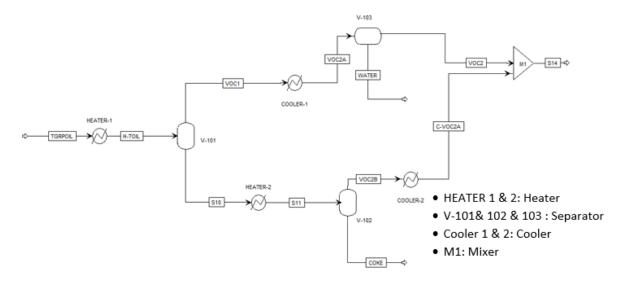


Figure S3. Aspen Plus process flow diagram showing a TGRP oil distillation process that separates the oxygenated oil into volatiles and bio-renewable coke. The volatiles are further upgraded by isolation and extraction of TGRP-oil, while the bio-renewable coke can be used as green coke.

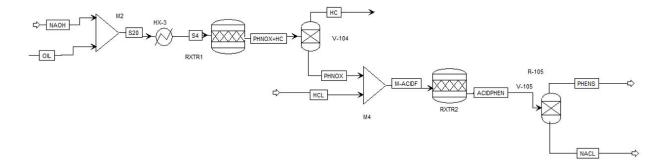
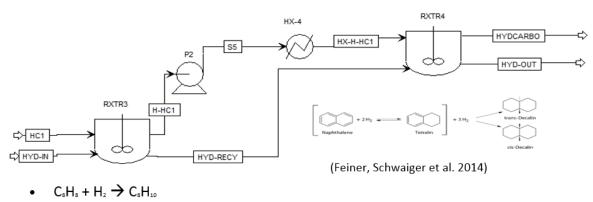


Figure S4. Phenol Extraction



• $C_8H_{10} + H_2 \rightarrow C_8H_{12}$ (Lee, Yung et al. 2013)

Figure S5. Two-step hydrogenation procedure consisting of (1) olefin hydrogenation at 80 °C 1 and (2) aromatics hydrogenation at 200 °C 2 . RXTR 3 and RXTR 4 represent the stoichiometric reactors used, P2 represents the pump and HX-4 is for the heater.

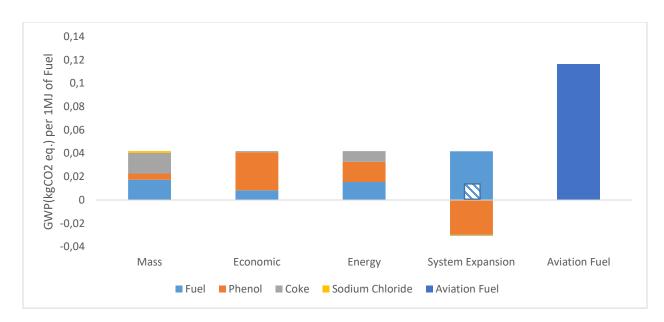


Figure S6. Contribution of the products and co-products to global warming potential per 1MJ of Fuel produced based on the economic, mass and energy allocations. The figure compares these allocations to the GWP of Low –Sulfur diesel extracted from Simapro ³. Total GWP for economic, mass and energy allocations are the same but the percent of GWP allocated to each product is different.

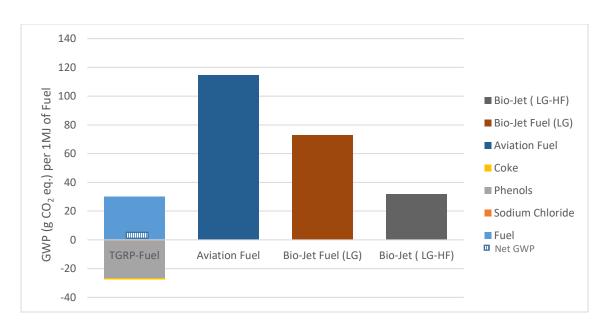


Figure S7. Global warming potential of TGRP fuel compared to aviation fuel and bio-jet fuel from other bio-fuel processes using system expansion.

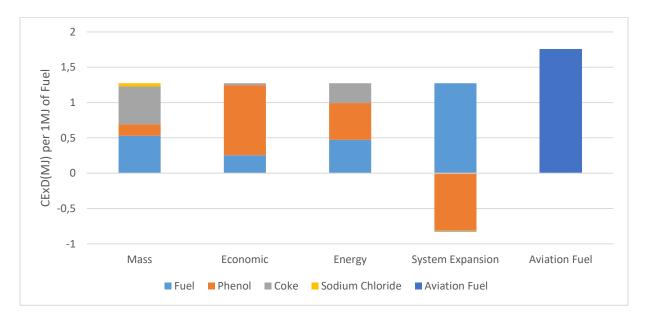


Figure S8. Contribution of the products and co-products to cumulative exergy demand per 1MJ of Fuel produced based on the economic, mass and energy allocations. The figure compares these allocations to the CExD of Low –Sulfur diesel extracted from Simapro ³. Total CExD for

economic, mass and energy allocations are the same but the percent of CExD allocated to each product is different.

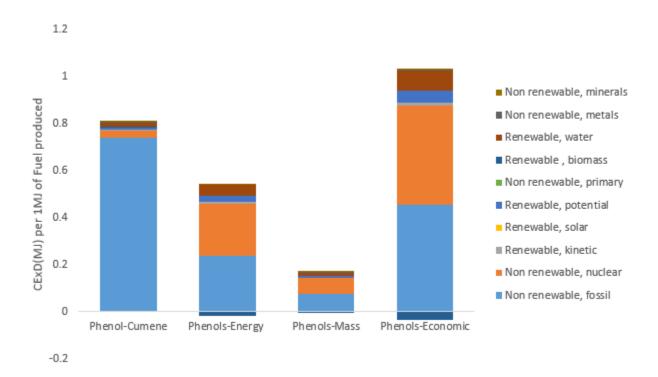


Figure S9. Cumulative Exergy demand (MJ) of phenol per 1MJ of Fuel produced. Phenol form the cumene process is compared with phenol form the TGRP process using mass, energy and economic allocations.

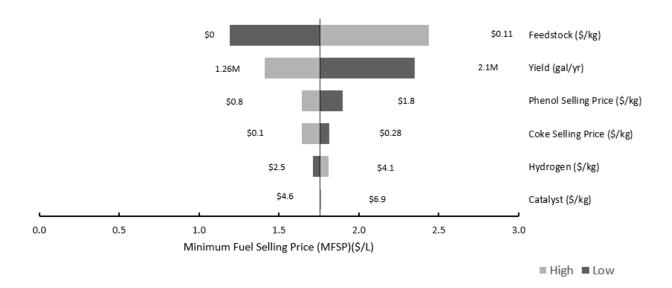


Figure S10. Tornado plot that shows the financial sensitivities of the TGRP upgrade process. The economic analysis includes the TGRP oil production and its upgrade to fuels and phenols.

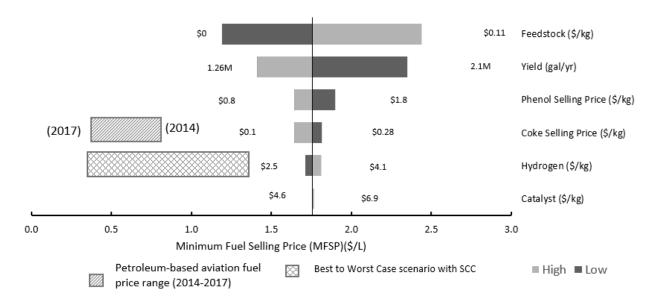


Figure S11. Aviation price range and minimum fuel selling price of best to worst case scenario.

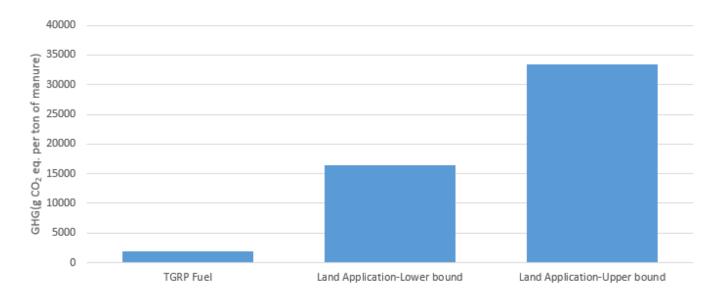


Figure S12. GHG emissions per ton of manure from TGRP fuel and land application

The literature shows that depending on the practice and farm size, GHG emissions per ton of manure range from 16,400 to 33,500 g CO_2 -eq 4 for land-application. Although the cited literature does not give sufficient breakdown for the N_2O and CH_4 emissions, it details that the major source of emissions is N_2O from land application. We calculated the amount of GHG emissions from the TGRP fuel per ton of horse manure and this value is between 88% - 94% less than the GHG emissions from land application.

SUPPORTING TABLES

Table S1. Composition of horse manure biomass and oils from Tail Gas Reactive Pyrolysis and Fast Pyrolysis ⁵

Process(Feedstock)	Horse Manure	TGRP oil (Manure)	Traditional Fast Pyrolysis Oil
	(Dry Basis)		(Manure)
Wt.%			
Phenols/Cresols		4.31	4.04
Naphthalenes		16.83	0.34
Acetic acid/Acetol		0.23	8.04
Wt. % ⁶			
C	48.43	79.53	67.35
Н	5.98	5.75	6.82
N	1.21	3.54	2.31
0	38.43	11.18	23.52
S	0.07		
Moisture (wt %)	0	2.08	7.31

Table S2. Equine waste generated in New York State. The dry residue estimates were based on an average of 41.7lb/stall/day (on dry residue basis. Potential energy value estimates based on samples Average higher heating value (dry) of 19.5MJ/kg⁷. The moisture content of horse manure is about 48 wt. % and the manure is assumed stored in open storage areas at the end of the barn for a very brief period. In the pre-processing step, the manure is dried in an oven at 150°F. In addition, the manure is ground in the pre-processing step and this process of grinding removes moisture via heat transfer.

Equine ¹ per farm	No. of farms	Total No. of equine	Average No. of equine per farm	Dry residue to be generated (lb./d)	Potential energy value (MJ/day)
1 to 24	9,514	59,083	6	259	2,293
25 to 29	529	17,329	33	1365	12,097
55 to 99	131	8,267	63	2629	23,305
>= 100	33	5,478	166	6916	61,303
Total	10,207	90,157	9	368	3,262
Total Horse mar	nure (lb/day)		1	,126,963	
Total Horse man	nure (kg/hr)			21,299	
Bio-Fuel Produc	ced (kg/hr)			1,948	
Bio-Fuel Densit	y (lb/cu.ft)			59	
Bio-Fuel Produc	ced in New Yo	ork State (gallons	s/yr) 4	,311,283	
Bio-Fuel Produc	ced in Pennsy l	lvania (gallons/yı	:)		

¹Equine includes horses and ponies. More details on location are provided in the equine waste study ⁷.

Table S3. Inventory data for TGRP oil Production per 1MJ of fuel produced

Item	Amount	Unit	Ecoinvent Database Module
Outputs			
Tail Gas Reactive Pyrolysis Oil	2620	kg	
Biochar	1040	kg	
Resources			
Water	55600	kg	Water, process, surface
Oxygen	33300	kg	Oxygen
Nitrogen	134000	kg	Nitrogen
Materials & Fuels			
Preheated Horse manure	8300	kg	User defined
Nitrogen	188	kg	Nitrogen, liquid, at plant/RER U
Ash	420	kg	Ash I
Electricity & Heat			
Electricity for Pyrolysis	65500	MJ	Electricity, medium voltage {NPCC, US only} market for Alloc Def, U
Emissions to air			
Oxygen	31200	kg	Oxygen
Nitrogen	134000	kg	Nitrogen, atmospheric
Carbon dioxide	4500	kg	Carbon dioxide, biogenic
Waste			
Ash	420	kg	Coal ash in landfill U
Water	57200	kg	Wastewater, average ⁹ market for Conseq, U

Table S4. Inventory data for TGRP oil Upgrade per 1MJ of Fuel produced

Item	Amount	Unit	Ecoinvent Database Module
Outputs			
Hydrocarbons(Fuels)	760	kg	
Hydrogen	27	kg	
Phenols	230	kg	Phenol {RoW} production Alloc Def, U
Sodium Chloride	66	kg	Sodium chloride, powder {RoW} production Alloc Def, U
Biorenewable coke	770	kg	Coke {RoW} coking Alloc Def, U
Resources			
Water	1.81	m^3	Water, unspecified natural origin, US
Materials and Fuels			
TGRP oil	2620	kg	User Defined
Hydrogen Chloride	45	kg	Hydrochloric acid, without water, in 30% solution state {RoW} hydrochloric acid production, from the reaction of hydrogen with chlorine Alloc Def, U
Hydrogen	47	kg	Hydrogen (cracker) E
Nickel (Catalyst)	1.2	kg	Nickel, 99.5% ⁹ nickel mine operation, sulfidic ore Alloc Def, U
Sodium Hydroxide	45	kg	Sodium hydroxide, without water, in 50% solution state ⁹ market for Alloc Def, U
Waste			
Water	1830	kg	Wastewater, average ⁹ market for Conseq, U
Utilities			
Electricity	32	MJ	Electricity, medium voltage {NPCC, US only} market for Alloc Def, U
Steam	970	kg	Steam, for chemical processes, at plant/RER U

Table S5. Mass, Price and Energy values of all products and co-products in the TGRP upgrade process. The mass and energy values were obtained from the Aspen Plus Simulation and price sources are cited in Table S7.

	Mass (kg)	Price (\$/kg)	Energy (MJ/kg)
Fuel	759	0.5	46.3
Sodium Chloride	66	0.0*	0.0
Phenols	231	1.3	51.5
Coke	767	0.16	27.6

^{*}Sodium Chloride is considered \$0 since it is in small quantities and will be disposed.

Table S6. Allocation Percentages Used

	Mass (%)	Price (%)	Energy (%)
Fuel	41.6%	20.0%	37%
Sodium Chloride	3.6%	0.0%	0%
Phenols	12.7%	77.8%	41%
Coke	42.1%	2.2%	22%

 Table S7. Pricing References

	Price	Reference
Hydrogen(\$/kg)	3.33	10
Process water(\$/MT)	0.032	10
Electricity(\$/kWh)	0.061	10
Natural Gas(\$/MMBTU)	5	10
Steam (\$/1000lb)	5.25	10
Phenol(\$/kg)	1.28	11
Catalyst(\$/kg)	4.61	12
HCl(\$/kg)	0.22	11
NaOH(\$/kg)	0.4	11
Coke(\$/kg)	0.036	11
NaCl(\$/kg)	0.2	13

Table S8. Techno economic analysis assumptions for the Tail gas reactive pyrolysis upgrade process. Assumptions are based on literature, Pourhashem et al, 2013 model, PNNL 18284 and Aspen Process Economic Analyzer. Utility pricing are shown in Table S7.

Period Description	Year		
Number of weeks per period	Weeks/Period	52	
Number of period for analysis		20	
Tax Rate	Percent/Period	39	
Interest Rate/Desired Rate of Return	Percent/Period	20	
Economic Life of Project	Period	10	
Salvage value	Percent	0	
Depreciation Method		Straight Line	
Escalation Parameter			
Project Capital Escalation	Percent/Period	5	
Products Escalation	Percent/Period	5	
Raw Material Escalation	Percent/Period	3.5	
Operating and Maintenance Labor Escalation	Percent/Period	3	
Utilities Escalation	Percent/Period	3	
Project Capital Parameters			
Working Capital Percentage	Percent/Period	15	
Operating Costs Parameters			
Operating Supplies	Cost/Period	25	
Laboratory Charges	Cost/Period	25	
Operating Charges	Percent/Period	25	
Plant Overhead	Percent/Period	30	
General and Administrative Expenses	Percent/Period	8	

Facility Operation Parameters		
Facility Type		Specialty Chemical Processing Facility
Operating Mode		Continuous Process- 24hrs
Length of Start-up Period	Weeks	20
Operating Hours per Period	Hours/Period	7,920
Process Fluids		Liquids and Gases
Operating Unit Costs		
Labor Unit Costs		
Operator	cost/operator/H	20
Supervisor	cost/Supervisor/H	30
Utility Unit Costs		
Electricity	cost/KWH	0.07
Portable Water	cost/M3	0
Fuel	cost/MEGAWH	27

Table S9. Itemized cost in dollar per liter of fuel produced.

	Fast Pyrolysis ¹⁴ + Upgrade (\$ L ⁻¹)		
Cost	Option A	Option B	
Raw materials	0.98	0.31	
Utilities	0.17	0.17	
Total labor cost	0.35	0.35	
Total additional fixed operating costs	0.18	0.18	
Depreciation	0.52	0.52	
Co-product credit	(0.44)	(0.44)	
MFSP (\$ L ⁻¹)	1.8	1.1	

References

- 1. Lee, J.-Y.; Yung, T.-Y.; Liu, L.-K., The microwave-assisted ionic liquid nanocomposite synthesis: platinum nanoparticles on graphene and the application on hydrogenation of styrene. *Nanoscale Research Letters* **2013**, *8* (1), 414-414.
- 2. Feiner, R.; Schwaiger, N.; Pucher, H.; Ellmaier, L.; Derntl, M.; Pucher, P.; Siebenhofer, M., Chemical loop systems for biochar liquefaction: hydrogenation of Naphthalene. *RSC Advances* **2014**, *4* (66), 34955-34962.
- 3. SimaPro SimaPro V. 8.1.1.16, 8.1.1.16; PRé Consultants: The Netherlands, 2014.
- 4. Aguirre-Villegas, H. A.; Larson, R. A., Evaluating greenhouse gas emissions from dairy manure management practices using survey data and lifecycle tools. *Journal of Cleaner Production* **2017**, *143*, 169-179.
- 5. Elkasabi, Y.; Mullen, C. A.; Boateng, A. A., Distillation and Isolation of Commodity Chemicals from Bio-Oil Made by Tail-Gas Reactive Pyrolysis. *ACS Sustainable Chemistry & Engineering* **2014**, *2* (8), 2042-2052.
- 6. McKendry, P., Energy production from biomass (part 2): conversion technologies. *Bioresource Technology* **2002**, *83* (1), 47-54.
- 7. Shayya, W.; Ballard, B.; Boateng, A. A., Assessment of equine waste as a biomass resource in New York State. Extension Publications. Morrisville State College: 2016; pp 1-18.
- 8. United States Department of Agriculture, 2012 Census Volume 1, Chapter 1: State Level Data- New York. Agriculture, U. S. D. o., Ed. United States Department of Agriculture: 2012.
- 9. Ma, W.; Chen, J.; Wang, T.; Chen, G.; Ma, L.; Zhang, Q., Catalytic cracking mechanism of bio-oil model compounds. *Nongye Gongcheng Xuebao/Transactions of the Chinese Society of Agricultural Engineering* **2013**, *29* (9), 207-213.

- 10. EIA, U., Annual energy outlook 2011 with projections to 2035. Washington DC: Energy Information Administration, United States Department of Energy 2011.
- 11. ICIS, Chemical Industry News & Chemical Market Inteligence. 2014.
- 12. Johnson-Matthey, Personal Communication. July 1, 2016. Patil, P., Ed.
- 13. Sedivy, V. M. In *Economy of salt in chloralkali manufacture*, National Salt Conference, 2008.
- 14. Pourhashem, G.; Spatari, S.; Boateng, A. A.; McAloon, A. J.; Mullen, C. A., Life Cycle Environmental and Economic Tradeoffs of Using Fast Pyrolysis Products for Power Generation. *Energy & Fuels* **2013**, *27* (5), 2578-2587.