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A Multiscale Framework to Estimate Water Sorption Isotherms for OPC-based Materials

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ABSTRACT

This paper presents a new multiscale framework to estimate water sorption isotherms (WSI) for ordinary Portland cement (OPC) based materials. This is achieved by integrating: (i) particle packing, (ii) cement hydration kinetics, and (iii) pore network models. The first two models provide pore size distribution for gel and capillary pores. The pore network model takes these as inputs to construct an idealized network of pores connected by so called throats. By invoking appropriate thermodynamic equilibrium laws for the adsorbed and capillary water locally and using an existing percolation algorithm, WSI are estimated via a series of steady-state analysis. A notable feature of the proposed framework is that there is only one geometrical calibration parameter needed in the pore network model, excluding calibration inherent in the cement hydration kinetics model. The capability of the framework is demonstrated by comparing the model predictions with eleven independent experimentally determined WSI, in particular, desorption isotherms. It is shown that the model is able to estimate WSI with coefficient of determination (R^2) value being 0.85 or above for all the cases.

KEYWORDS

Cement hydration, Particle packing, Pore network, Multiscale, Concrete, Water sorption isotherm

2 1 INTRODUCTION

3 The importance of water content *vis-à-vis* the durability of cementitious materials is long
4 established [1, 2]. In particular, water sorption isotherm (WSI) is a key material property that
5 is used extensively in the study of moisture transport of unsaturated cementitious materials with
6 a view to gain improved understanding of many durability problems such as drying shrinkage,
7 creep, corrosion, carbonation, freeze-thaw, etc. The mechanisms affecting WSI are highly
8 complex and depend on concrete composition such as cement type and content, water to cement
9 ratio, aggregates, additives and fillers, moisture content and environmental factors such as
10 ambient temperature [3].

11 A central theme of this study is to explore the possibility of going from the knowledge of mix
12 composition to predicting WSI. Such a framework is of profound importance, for instance, in
13 the study of durability of concrete structures in the existing nuclear power plants for which
14 material properties such as drying shrinkage and creep are unknown. Hence, as far as possible,
15 predictive approach to the determination of WSI is pursued so that any OPC based mixes can
16 be accommodated without the need for an experimental campaign for every mix composition.

17 The principal approach to obtain WSI is via appropriate experimental techniques such as
18 conventional wetting/drying [4, 5], dynamic vapour sorption [6, 7] and centrifuge experiments
19 [8, 9]. For low permeability materials like cementitious materials, it is often time consuming to
20 determine WSI accurately for the complete range of degree of saturation (S_w) [10]. However,
21 once experimental data become available, an analytical model can be fitted in a way that WSI
22 can be predicted for various conditions, for example, different water to cement ratios and
23 porosity. An overview of such analytical models is available in Burgh et al. [11]; a notable work
24 cited in their reference is that of Kumar et al. [12], which explored numerous analytical models
25 to successfully predict WSI for a range of porosity. The accuracy of such models not only
26 depend on their mathematical form but also on access to experimental WSI.

27 A second approach to predict WSI is to derive it from microstructural information such as pore
28 size distribution and the type of water associated with different pore classes. Kelvin's equation
29 is then used to establish an equilibrium relation between water content and capillary pressure,
30 especially for gel and capillary pores. In this regard, Pinson et al. [13] predicted WSI by
31 classifying water into interlayer, gel, capillary and surface adsorbed water. As a first
32 approximation, they assumed a linear relationship between water content and capillary pressure
33 for the interlayer space, i.e. for relative humidity below 15%. For the gel and capillary pores,
34 they extracted pore size distribution from Barrett-Joyner-Halenda (BJH) method and then
35 combined it with Kelvin's equation to derive WSI. For the empty pores in gel and capillary
36 pores, they also considered the effect of adsorbed water via Langmuir and Brunauer–Emmett–
37 Teller (BET) methods. Burgh et al. [11] proposed a model for WSI that also considers
38 interlayer, gel and capillary pore classes as Pinson et al. [13]. They used a modified form of
39 classic Powers and Brownyard hydration model [14] to capture pore volumes. Then a
40 continuous probability distribution is used with shape and scale parameters calibrated with
41 experiments to describe the whole pore size distribution range. With the pore size distribution,
42 Kelvin's equation is used to determine the equilibrium relationship between water content and
43 capillary pressure. More recently, Masoero et al. [15] studied the densification of Calcium-
44 Silicate-Hydrate (C-S-H) gel and its effect on WSI during hydration. Their model combines a
45 C-S-H gel description from nanoscale simulations with evolving capillary pore size
46 distributions using a simple hydration model, however only for C_3S . This information is then
47 coupled with Kelvin's equation to obtain WSI.

48 An interesting and alternative approach to predict WSI is the pore network model pioneered by
49 Mason [16], which conceptualizes a porous material as a network of pores connected by throats
50 (constrictions or windows) and locally invoking Kelvin's equation. He demonstrated the
51 capability of the network model to predict adsorption-desorption hysteresis using a hypothetical

52 network. It was possible to capture hysteresis because pore blocking is a natural consequence
53 of the network model where pore connectivity is inherently defined. In this regard, Islahuddin
54 and Janssen [17] applied the pore network model to numerically explore WSI of Berea
55 sandstone based on topologically equivalent network obtained from micro-CT scanning.

56 In conclusion, with the exception of Masoero et al. [15], the first two approaches strongly rely
57 on experimental WSI for calibrating the model parameters before they can be used for
58 predictions. Although, Masoero et al. [15] do not use experimental WSI, their model is
59 demonstrated for C₃S (Alite) system only. In addition, the first two approaches will need some
60 calibration to incorporate sorption hysteresis unlike the alternative approach. Finally, to the
61 authors' knowledge, the alternative approach, i.e. the pore network model has never been
62 applied to complex microstructures such as cementitious materials. Although the pore network
63 model is applied for other porous materials, it relies **entirely** on experimental pore size
64 distribution as a **key** input.

65 This study therefore proposes a new framework to estimate WSI via a multiscale approach,
66 which integrates particle packing, cement hydration kinetics and pore network models. Thus no
67 experimental data concerning pore size distribution or WSI (except for geometrical parameter)
68 is needed as inputs other than the composition of the material, known microstructural features
69 of high density (HD) and low density (LD) C-S-H phases for OPC based materials and cement
70 hydration kinetics model, which is usually calibrated with hydration experiments. The
71 capability of the framework is demonstrated by comparing the model predictions with a number
72 of experimental desorption WSI. Desorption WSI has been the focus because the ultimate goal
73 of the present study is oriented towards drying shrinkage problems **within the context of**
74 **unsaturated poroelasticity**.

75 **2 MULTISCALE FRAMEWORK**

76 The proposed multiscale framework for water sorption isotherm integrates three models: (i)
77 cement particle packing, (ii) microstructural cement hydration kinetics, and (iii) pore network.
78 For lower (gel) pore size range, which is characterized by HD and LD C-S-H, the particle
79 packing model computes total porosity and pore size distribution relying on a principal
80 assumption that the variability of HD and LD C-S-H characteristics in terms of particle packing
81 is limited for OPC [18]. Based on fundamental inputs such as cement composition and reaction
82 conditions, the microstructural model computes volume fractions of various hydrated phases,
83 capillary porosity and pore size distribution but for a higher capillary pore size range ($> 1 \mu\text{m}$),
84 which is primarily dictated by the limitation of resolution of the model used. However, these
85 two models do not cover the mid pore size range (small capillary pores), i.e. between tens of
86 nm to $1 \mu\text{m}$, as demonstrated in *Figure 1*. Therefore, an approximation to cover this missing
87 pore size range is proposed (Section 2.3) and justified (Section 3). Based on the computed pore
88 size distribution for the full range of pore sizes, the pore network model computes WSI for a
89 given material composition. These steps are illustrated in *Figure 2*.

90 **2.1 PARTICLE PACKING MODEL: nm scale**

91 In order to estimate porosity and pore size distribution of lower (gel) pore size range, a
92 conceptual model that describes the microstructure of C-S-H gel has to be chosen. Various
93 conceptual models such as layered, colloidal and fractal models have been proposed in the
94 recent past [19]. The particle packing model used in this study follows the colloidal model for
95 C-S-H gel proposed by Jennings [20] and illustrated in *Figure 3*. This is a self-consistent model
96 with respect to a number of experimental data, for example, specific surface area and density
97 measurements. The term ‘particle’ used in this study equates to the term ‘C-S-H globules’ in
98 the Jennings’s model. These globules (particles) are packed together to form HD and LD C-S-
99 H and the spacing between the globules are referred to as gel pores. Thus if the packing density

100 for HD and LD C-S-H are known then their respective gel porosities as well as pore size
101 distribution can be geometrically extracted. However, the pore size distribution of pores larger
102 than the gel pores cannot be estimated with the particle packing model. Some assumptions are
103 made to fill this missing information as discussed in Section 2.3.

104 The two types of C-S-H phases, namely, HD C-S-H and LD C-S-H are well-characterized [21-
105 23]. The difference between the two C-S-H phases mainly lies in their gel porosity and packing
106 density; the HD C-S-H being characterized by a gel porosity, ϕ_{HD} , of roughly 0.24, and the LD
107 C-S-H being characterized by a gel porosity, ϕ_{LD} , of roughly 0.37. It is worth noting that the
108 corresponding packing densities, $(1 - \phi_{HD}) = 0.76$ and $(1 - \phi_{LD}) = 0.63$, come very close to the
109 maximum packing densities of mono-sized spherical packing. The packing density of HD C-S-
110 H is of striking similarity to the highest possible density of ordered spheres, known as the face-
111 centred cubic lattice, equal to 0.74, and the packing density of the LD C-S-H is of striking
112 similarity to the so-called random close packing (RCP) of 0.64 [24-27].

113 The particle packing model used in this study is similar to the work of Fonseca et al. [19] and
114 Liu et al. [28]. The choice for the size of the particles is based on Jennings's [20] conceptual
115 model, which considers mono-sized particles of size 5.6 nm (Figure 3). With respect to pore
116 size distribution, a representative volume element (RVE) of size $150 \times 150 \times 150 \text{ nm}^3$ is chosen
117 after testing three different RVE sizes (100 nm, 150 nm and 200 nm). Liu et al. [28] also arrived
118 at the same RVE size after testing various RVE sizes. So, Initially, a small number of mono-
119 sized particles each of diameter 5.6 nm are randomly distributed in space, and then they are
120 compacted to the maximum extent without overlapping by moving spheres closer to the centre
121 of the RVE. A new set of particles is added and again compacted. This procedure is repeated
122 until the desired packing density is achieved. One additional step that is taken in comparison to
123 the work of Liu et al. [28] is that after compacting, the compacted globules are further
124 discretized into sub-globules made of mono-sized spheres with size of 2.2 nm, so that the water

125 in the interlayer space (space between 2.2 nm spheres) are also accounted to be consistent with
126 the Jennings' colloidal model. With this, the extraction of pore size distribution is carried out
127 (*Figure 3*). The results obtained from this model corresponds well with the model of Liu et al.
128 [28] for HD C-S-H as shown in *Figure 4*, which is attributed to the fact that the maximum
129 density has to be achieved irrespective of the algorithm used. However, there is some deviation
130 for LD C-S-H, which is attributed to the fact that the required packing density is lower compared
131 to HD C-S-H, which means the results can become sensitive to the particle packing algorithm
132 and pore size distribution extraction method. It is also worth noting that typically 40% of the
133 total volume of RVE of cement paste is comprised of HD and LD C-S-H [1]. Therefore, the
134 particle packing model accounts for majority of the pore size range.

135 **2.2 MICROSTRUCTURAL CEMENT HYDRATION KINETICS MODEL:** 136 **μm scale**

137 All the existing microstructural models are mainly intended to study microstructural properties
138 such as volume fractions of hydrated phases, porosity, percolation threshold and degree of
139 hydration but not pore size distribution. There are currently no cement hydration kinetics
140 models that can estimate reliable pore size distribution. This is a limitation of this framework.
141 However, as a first approximation, such a hydration kinetics model is still relied upon for
142 estimating pore size distribution.

143 A comprehensive review of microstructural modelling and in particular cement hydration
144 kinetics models can be found in Thomas et al. [29]. From the latter as well as author's own
145 experience, it is found that Virtual Cement and Concrete Testing Laboratory (VCCTL) suite
146 offers control over wide range of variables and input parameters and is experimentally better
147 validated compared to other hydration models [30]. Hence, VCCTL is taken forward in this
148 study. In VCCTL, a 3D cement paste microstructure is digitized into a cubic lattice and each
149 volume element or voxel of the cube is assigned to a phase (porosity, C-S-H, CH, etc.). VCCTL

150 simulates the hydration process as an iterative procedure of dissolution, diffusion, and reaction.
151 This iteration evolves the initial lattice cube and forms a reasonable spatial distribution of the
152 anhydrite cement, combined hydration products, and capillary porosity. The minimum voxel
153 size that can be specified is $1\ \mu\text{m}$, which also implies that capillary pores below $1\ \mu\text{m}$ cannot
154 be captured. Hence, only total porosity and pore size distribution of pores greater than $1\ \mu\text{m}$
155 can be captured.

156 Note that one may also consider cement hydration kinetics model that are resolution free or
157 vector-based (e.g. HYMOSTRUC [31]), which have no limitation on the minimum voxel size.
158 However, as shown by Ye [32], the modelled pore size distribution results are not promising.
159 Such resolution free microstructural models are usually used with voxel size larger than $1\ \mu\text{m}$,
160 as capturing the hydration and interaction of the cement particles from nano to micro scale is
161 computationally prohibitive. Moreover, such resolution free models read the spaces existing
162 between the reaction products but not the pores within the C-S-H gel as a product of hydration
163 reaction [33]. This in fact justifies the use of particle packing (Section 2.1) as a necessary
164 component of the framework to estimate gel pore size distribution.

165 **2.3 MISSING PORE SIZE DISTRIBUTION: nm to μm scale**

166 The missing pore size distribution from biggest gel pores (14 nm based on particle packing) to
167 smallest capillary pore imposed by the hydration model ($1\ \mu\text{m}$ by VCCTL) is derived by using
168 the pore size distribution of the capillary pores as a surrogate (Section 2.2). In the absence of
169 any data, this should be treated as a crude approach. The implication of this approximation is
170 discussed in Section 3.1. In this approach, a Weibull distribution function is fitted on the
171 capillary pore size distribution obtained from VCCTL, which is available from one μm to tens
172 of μm . The Weibull distribution function is defined as:

173
$$f(x) = \frac{\beta}{\eta} \left(\frac{x-\gamma}{\eta}\right)^{\beta-1} e^{-\left(\frac{x-\gamma}{\eta}\right)^\beta} \quad (1)$$

174 where x is a pore size, β is the shape factor, η is the scale parameter and γ is the location
175 parameter. Once the Weibull distribution is fitted for the capillary pore size range, the shape is
176 preserved but by manipulating the location and scale parameter the pore size range is stretched
177 to cover a wider range from 14 nm to 1 μ m. This missing pore size range is then added to the
178 capillary pore network, ensuring that the calculated capillary porosity is still respected.

179 **2.4 PORE NETWORK MODEL**

180 The estimation of WSI of cementitious material is carried out using a pore network model with
181 steady state analysis. A pore network is a virtual network of pores and throats, which are
182 connection between the pores [34]. This model can reconstruct WSI on the whole capillary
183 pressure range provided accurate pore/throat size distribution data as well as topology of the
184 network are available, including appropriate physics given the pore size range of a typical
185 cement paste material.

186 There are several techniques to characterize the pore space. Imaging techniques such as
187 producing 3D images by mapping the real interior structure of original material are promising
188 but time consuming and expensive. This mapping can be carried out using destructive approach
189 of cutting and stacking serial 2D sections, followed by confocal laser scanning microscopy,
190 non-destructive X-ray micro-tomography (μ CT), and constructing synthetic 3D images from
191 high resolution 2D thin sections using statistical methods or geological process simulation [34,
192 35]. However, as already described in Sections 2.1 and 2.2, in this study, pore space is
193 characterized based on the particle packing and available microstructural modelling tools
194 (VCCTL here).

195 In this study, the pore space of the cementitious material is characterized by means of three
196 main networks: HD C-S-H, LD C-S-H and capillary porosity. The most important variable to

197 construct a network in this regards is the volume fractions of these three networks. First of all,
198 the volumetric ratio of HD and LD C-S-H network is estimated via Jennings-Tennis's hydration
199 model [36]. Depending on this ratio, the porosity of the whole gel pore space is derived as
200 follows:

$$201 \quad \varphi_{gel} = \varphi_{HD} \cdot V_{HD} + \varphi_{LD} \cdot V_{LD} \quad (2)$$

202 where φ_{HD} is the porosity of HD C-S-H equal to 0.24 [21], V_{HD} is the volume fraction of HD
203 C-S-H in C-S-H gel, φ_{LD} is the porosity of LD C-S-H equal to 0.37 [21], and V_{LD} is the volume
204 fraction of LD C-S-H with $V_{HD} + V_{LD} = 1$.

205 The volume of gel pores is defined as:

$$206 \quad V_{gel} = V_{CSH} \cdot \varphi_{gel} \quad (3)$$

207 where V_{CSH} is the volume fraction of C-S-H gel, which is obtained from the hydration model
208 (section2.2). The volume of capillary pores is also obtained from the hydration model. Once
209 the volume fractions of all the networks are available, a unified network can be generated by
210 merging all these three networks. Each network has its own pore size distribution and size,
211 which comes from the hydration model, particle packing model and aforementioned equations.

212 **2.4.1 GEOMETRICAL CONSIDERATIONS**

213 To ensure that the simulations are reproducible, which means keeping fitting coefficients to a
214 minimum (Eq. (4) and (5)), and to avoid convergence problem, a cubic network with fixed
215 coordination number is constructed for each phase, i.e. HD C-S-H, LD C-S-H and capillary
216 porosity. The pores and throats of the network can have different geometries such as cubes,
217 triangles, spheres, cylinders, etc. To accelerate the simulation, in this study, the pores are
218 defined as spheres and throats as cylindrical pipes. To construct a cubic network of specific
219 size, in addition to pore size, the connectivity or coordination number and throat sizes and
220 lengths are required. The coordination number is defined as the number of connected throats

221 for each pore. Real porous materials might have different coordination numbers based on their
 222 pore structure but to optimize the simulation procedure, the coordination number of the cubic
 223 network is taken as six. This means each pore is connected to its six neighbouring pores. *Figure*
 224 *5* demonstrates a few examples of networks with coordination numbers of 6, 12 and 26 for a
 225 network with 27 pores.

226 It is not possible to experimentally obtain throat size distribution as the network of pores and
 227 throats are idealization of complex pore network in cementitious materials. Hence, some
 228 assumptions have to be made. The first rule to define throat size is to make sure that every throat
 229 diameter is smaller than its neighbouring pores. As a first approximation, the throat size,
 230 R_{throat} , is arbitrarily defined as a linear function of the radius of neighbouring pores, R_{np} as
 231 follows:

$$232 \quad R_{throat} = \alpha(\min(R_{np})) \quad (4)$$

233 where α is a factor whose value should be less than 1. For instance, for a configuration of three
 234 pores A, B and C shown in *Figure 6* with radius $R_a \leq R_b \leq R_c$, the throat size for their connecting
 235 throats should be:

$$236 \quad R_{ab} = \alpha R_a, R_{bc} = \alpha R_b \quad (5)$$

237 It is also worth mentioning that the throat length is also defined using a comparable equation as
 238 Eq.(5), where the length of the connecting throat between two pores is defined as:

$$239 \quad L_{throat} = \beta(\max(R_{np})) \quad (6)$$

240 with β being a scaling factor, which should be greater than 1 so that there is no overlapping of
 241 pore neighbouring pore bodies. In this study, the following values have been assigned to the
 242 coefficients: $\alpha = 0.3$ and $\beta = 1 + \alpha$. The α value is calibrated with an arbitrary WSI
 243 (experimental WSI presented in Section 3.2) and is held constant.

244 The following procedure is adopted to generate the overall pore network geometry:

245 a) Assuming that the volume fraction of LD C-S-H is higher than HD C-S-H, the first step
246 in the network construction is to generate an LD C-S-H network as a cubic network of
247 100 pores on each side of the cube, distributed within a $200 \times 200 \times 200 \mu\text{m}^3$ RVE. Since
248 capillary pores, which are usually bigger (up to tens of μm) than the gel pores, are added
249 to the same RVE, the RVE dimensions are considered slightly larger than is necessary
250 for solely gel pores.

251 b) The second step is to generate the HD C-S-H pores and embed them inside the cubic
252 LD C-S-H network until their volume fraction reaches a certain ratio to LD C-S-H
253 volume fraction estimated from the hydration model (*Figure 7(a)*). This is achieved by
254 embedding the HD pores (grey) between the LD pores (green) as illustrated in the inset
255 of *Figure 7(a)*. This step should also ensure that the ratio of HD to LD C-S-H as
256 estimated from the Jennings-Tennis's model is respected. It is worth mentioning that
257 during this merging process the coordination number is continuously checked and if
258 necessary throats are trimmed to meet the coordination number of six for the overall
259 network. Furthermore, note that the HD C-S-H can be as big as the LD C-S-H network
260 or even bigger than the LD C-S-H network depending on its volume fraction. If they are
261 of the same size, it does not matter, which network is constructed first. The sequence
262 should however be reversed in case the HD C-S-H network is larger than the LD C-S-
263 H network. It is worth noting that, by merging these two networks the homogenized C-
264 S-H network will result in a WSI, which falls between the pure HD C-S-H and LD C-
265 S-H networks as illustrated in *Figure 8*.

266 c) The first two steps yield a homogenized C-S-H gel network into which the capillary
267 pore network is embedded in the third (final) step. The capillary pores are added in a
268 different manner compared to the C-S-H pores. Since in terms of number of pores the

269 capillary pores have the smallest population and to avoid over estimation of ink-bottle
270 effect, these pores are randomly distributed in the network rather than starting from the
271 centre. Hence, as shown in *Figure 7(b)*, capillary pores (red) can be spread anywhere in
272 the central region and beyond within the homogenized C-S-H network (grey and green).
273 If the RVE is sufficiently large then the randomness of the capillary pores should not
274 significantly influence the WSI.

275 It is important to note that from the point of view of pore network geometry, RVE basically
276 implies that the domain is sufficiently large enough to cover the entire pore size range as well
277 as to ensure that the random distribution of pores do not significantly affect WSI (converged
278 solution). RVE does not necessarily correspond to the actual RVE size because of the
279 idealization involved in the representation of pores and throats.

280 **2.4.2 WATER SORPTION ISOTHERM**

281 Because of a large variation in pore sizes and their topological arrangement in low permeability
282 materials as is the case with hardened cement paste, a WSI displays hysteresis [37]. In the case
283 of cement paste, this behaviour can be attributed to irreversible processes such as morphological
284 changes in microstructures, cracking or ink-bottle effect. Both adsorption and desorption WSI
285 are of interest to understand the moisture transport behaviour of the material as well as for
286 moisture transport modelling. In this paper, specific attention is paid to estimating desorption
287 isotherms because of its direct relevance to the study of drying shrinkage behaviour and issues.
288 Desorption isotherm also implies that some sort of invasion algorithm is necessary to mimic
289 desorption of the initially saturated pore network.

290 In this study, an idealized pore network is generated using an in-house code developed in the
291 SciPy and NumPy environment. The OpenPNM [34] library is then called upon to compute
292 desorption isotherms. The OpenPNM library uses an “invasion percolation” algorithm based

293 on the work of Wilkinson and Willemsen [38]. However, additional considerations concerning
 294 adsorbed water are introduced. The invasion of moist air will occur only if the applied capillary
 295 pressure (external RH) exceeds the entry pressure of a pore/throat, which is calculated using the
 296 Young-Laplace's equation:

$$297 \quad P_C = \frac{2\sigma \cos(\theta)}{r} \quad (7)$$

298 where σ is the surface tension (N/m), θ is the contact angle ($^\circ$), and r is the pore radius (m) of a
 299 given pore/throat, which comes from the pore/throat size distribution. Since water perfectly
 300 wets the silicate materials and the presence of surface adsorbed water is taken into account, the
 301 contact angle is set to zero in this study [17, 39].

302 The invasion algorithm works in such a way that the pores with lowest entry pressure is first
 303 invaded. The throats that are connected to the invaded pores then become accessible and thus
 304 join the invasion front. This procedure is repeated until all the pores/throats are completely
 305 invaded. Further details of the algorithmic implementation are available in Gostick et al. [34]
 306 and references therein.

307 The invasion of pores/throats by moist air (external RH) does not imply that all the liquid water
 308 is completely evacuated from them because water may still be present in the form of adsorbed
 309 water, which is in equilibrium with the local relative humidity, h_m . Hillerborg [40] defined
 310 adsorbed water as sum of adsorbed and capillary condensation water. He attempted to propose
 311 a relationship between h_m and the radius of curvature of adsorbed water by combining the
 312 standard Kelvin and Young-Laplace equations. However, as it was difficult to define the
 313 curvature of the adsorbed water surface, as an approximation, only the curvature of adsorbed
 314 water was considered, which was determined based on the thickness of adsorbed water, t , via:

$$315 \quad t = \frac{0.525 \times 10^{-8} RH}{(1 - RH/h_m)(1 + RH/h_m + 15RH)} \quad (8)$$

316 where RH is the external relative humidity (boundary condition) and h_m is defined via:

$$317 \quad h_m = \exp\left(\frac{-2\sigma}{R_v T(r-t)}\right) \quad (9)$$

318 where R_v is the gas constant of water vapour (J/kg/K), T is the temperature (K) and r is the radii
319 of pores/throats. Note that Eq. (8) is a modification of the conventional BET model proposed
320 by Hillerborg [40] to account for changes in h_m .

321 To quantify pore wetting/drying mechanism in the pore network model, P_c is considered as the
322 primary variable. An illustration of simulation of free drying of a cube of cement paste using
323 the pore network model is shown in *Figure 9*. In order to mimic real experimental conditions,
324 the pore network is assumed to be initially saturated ($P_c=0$) and a constant boundary condition
325 of $P_c>0$ (or $RH<1$) is prescribed incrementally over the entire RH range (0 to 1). This condition
326 initiates drying of pores and throats until thermodynamic equilibrium is reached between the
327 network and the prescribed external P_c based on the invasion percolation algorithm. The
328 equilibrium S_w is then computed by taking the ratio of volume of elements of pore network that
329 are still wet and the total volume of the network. Adsorbed water (if present) in the pores/throats
330 is also accounted for in the calculation of S_w . Note that the entire computation is purely a steady
331 state analysis for each increment of P_c .

332

333

334 3 VALIDATION

335 The proposed framework is validated against a wide range of measured desorption isotherms
336 available in literature [10, 41-45]. The chosen datasets include plain cement paste, mortar and
337 concrete to examine the capability of the modelling framework to handle different cementitious
338 materials. The W/C of these materials varies from 0.34 to 0.8. The mineral composition and a
339 short description on preparation and measurement methods of the studied cases are presented
340 in *Table 1*. Further details of the experiments can be found in [10, 41-45].

341 Recall from Section 1 that the ultimate goal of the present study is oriented towards
342 understanding of drying shrinkage problems within the context of unsaturated poroelasticity
343 (e.g. [46]), where WSI is a key input data. In unsaturated poroelasticity, apart from S_w , the
344 primary variable is typically capillary pressure instead of relative humidity (RH), which is
345 measured experimentally. Therefore, all experimental RH data used in this study are converted
346 to capillary pressure via Kelvin's equation, which is assumed to be valid irrespective of
347 microstructure or RH.

348 The hydration model considers appropriate curing conditions, including temperature as reported
349 in *Table 2*. In particular, the curing period varies from 56 days to 1 year at which time the rate
350 of change of degree of hydration is expected to be minimal. Accordingly, variations in pore size
351 distribution or porosity is also considered to be minimal and hence neglected beyond the curing
352 period. Similarly, the pore network model considers laboratory temperature reported in *Table*
353 *2* in order to correctly interpret temperature dependent RH changes.

354 3.1 INFLUENCE OF CAPILLARY PORES

355 Recall from Section 2.3 that a crude approach was proposed to account for the missing capillary
356 pore size distribution in the range 14 nm to 1 μm . To justify this approach, the influence of
357 capillary pore size range on WSI is examined in this section with the help of a series of virtual

358 experiments. A network including C-S-H gel pores with evenly distributed HD and LD pores
359 (50% HD C-S-H, 50% LD C-S-H) is generated, following which a capillary network is added.
360 Influence of variation of the mean pore size of the added capillary network and its volume
361 fraction are then studied. The volume fraction of the capillary network varies from 40% of the
362 total porous phase (Eq.1-2) to 70% and four different mean pore sizes are considered: 0.1 μm ,
363 0.5 μm , 1 μm and 10 μm . The predicted WSI for such networks with 40%, 50%, 60% and 70%
364 of volume fraction of capillary porosity is presented in *Figure 10*.

365 It is seen that as the volume fraction of capillary porosity increases, the effect of its pore size
366 range becomes slightly more pronounced. For example, the network with 40% capillary pores
367 and mean capillary pore size of 10 μm has a maximum deviation of 6% compared to a network
368 with mean capillary pore size of 0.1 μm . Whereas, the same comparison for a network with
369 70% capillary pores has 9% deviation. Although the difference of 3% does not seem to be a big
370 deviation, these networks are constructed in an ideal sense where the proportion of HD and LD
371 C-S-H is fixed. In reality, higher porosity means either higher W/C or lower hydration degree
372 and consequently higher percentage of LD than HD C-S-H, which shifts the whole retention
373 curve.

374 However, an important conclusion from *Figure 10* is that the capillary pore size distribution
375 and capillary pore size range ($> 14 \text{ nm}$) do not significantly affect WSI. At least three reasons
376 can be attributed to this insensitivity:

377 a) The population of capillary pores are at least two orders of magnitude lower than gel
378 pores to have any pronounced effect. For example, for the material CP5, the calculated
379 volume fraction of C-S-H gel is 47% (of which 72% is LD and 28% is HD C-S-H) and
380 that of capillary porosity is 37% (*Table 2*). The 47% C-S-H gel fraction with
381 aforementioned LD and HD C-S-H ratios results in 13% of LD C-S-H gel pores and 3%
382 of HD C-S-H gel pores. This distribution results in 1 million LD C-S-H pores and 1.28

383 million HD C-S-H pores. In contrast, the calculated capillary pores are only 23,552 in
384 the total pore space.

385 b) From Eq. (10), the relative humidity at which a pore radius of 20 nm ($T=293$ K) is
386 invaded is 0.95. Whereas, a big capillary pore with a radius of 10 μm is invaded at a
387 relative humidity of 0.99. Because of this narrow range of relative humidity covering
388 the whole range of capillary pores, the invasion sequence is not considerably affected
389 to significantly influence WSI.

$$390 \quad r_p = - \frac{2\gamma}{\left(\frac{RT}{M}\right) \ln RH} \quad (10)$$

391 c) In addition, because of the random distribution of capillary pores as well as its smaller
392 population in the overall pore network, the invasion sequence is not considerably
393 affected. For example, not all the capillary pores are exposed to air once the relative
394 humidity reaches 0.95 because some of them are trapped between the gel pores.

395 Therefore, the crude approach proposed in Section 2.3 to handle the missing pore size
396 distribution in the pore size range of 14 nm to 1 μm is justified, at least for predicting WSI.

397 **3.2 WSI PREDICTIONS**

398 Based on the material composition defined in *Table 1*, predicted phase fractions of C-S-H and
399 the porosity of LD C-S-H, HD C-S-H and capillary pores are presented in *Table 2*, which form
400 basic inputs for generating pore networks. The results show that the capillary porosity and
401 volume fraction of LD C-S-H increase with increase in W/C and the volume fraction of HD C-
402 S-H shows opposite trend (*Table 2*). The results also show that the capillary porosity of cement
403 paste is larger than that of concrete for the same W/C. All these are qualitatively consistent with
404 the known behaviour of OPC [37].

405 *Figure 11 to Figure 13* show comparisons of experimental WSI against the pore network model
406 results for the materials mentioned in Table 2. It is seen that for all materials the model provides
407 reasonably good correlation. The accuracy in the lower saturation regime is considerably higher
408 than the rest of the curve, revealing that the C-S-H gel has a reproducible behaviour and its
409 variation is more limited than capillary pores. However, there are some deviations for materials
410 CP2, Co1 and Co2, which may be attributed to uncertainties in the composition of the materials
411 used in the experiments as well as the idealizations and assumptions in the particle packing and
412 pore network model. Nevertheless, it is reasonable to conclude that the particle packing model
413 provides reliable inputs for pore network modelling.

414 For the materials CP1, CP3, CP5, M2, Co1 and Co2, the model slightly underestimates WSI in
415 the mid-range saturation regime. Part of the reasoning is the same as mentioned above for lower
416 S_w , but in addition, the reliability of the pore size distribution extracted from the hydration
417 kinetics model is also a major factor. Recall from Section 2.2 that the existing hydration kinetics
418 models are not intended to capture pore size distribution and is admitted only as a first
419 approximation in this study.

420 Furthermore, note that CP5 and M2 are from the same dataset where a standard for cement type
421 used is reported but XRD information of the chemical composition is not available. Co1 and
422 Co2 are concrete (cement paste + aggregate) and the ability of hydration models to account for
423 the presence of aggregates is yet to be examined. Finally, CP1 is a one year old sample, which
424 means self-desiccation would have occurred, which is not considered in the present framework.
425 The above issues add further uncertainties, which are reflected in the marginal deviations
426 observed in the predicted WSI.

427 **Table 3 shows that R^2 value (coefficient of determination [47]) is 0.85 or above for the studied**
428 **cases.** Given the available tools and assumptions, the overall performance of the multiscale
429 framework is highly encouraging.

430 4 CONCLUSIONS

431 A largely predictive multiscale framework for the estimation of WSI is proposed. This is
432 achieved by integrating the following models: (i) particle packing, (ii) cement hydration
433 kinetics, and (iii) pore network. The need for predicting pore size distribution necessitates the
434 use of particle packing and cement hydration models, including Jennings-Tennis's model to
435 obtain additionally the HD-LD C-S-H ratio. Based on the predicted (complete) pore size
436 distribution of the material, the merging of HD C-S-H, LD C-S-H and capillary pore networks
437 has been proposed, which actually leads to a very large pore network with few millions of pore
438 bodies and throats. Both capillary water and adsorbed water are considered in estimating water
439 content for WSI calculations. The pore network is implemented in the SciPy and NumPy
440 computational environment with recourse to an existing library called OpenPNM, which uses
441 an efficient invasion algorithm. A quasi-static analysis of the network using Kelvin's equation
442 is then invoked in pore bodies and throats to estimate desorption WSI, the latter because the
443 ultimate goal of this study is to address drying shrinkage problem.

444 The capability of the framework has been demonstrated with eleven independent experiments
445 available from literature. A reasonably good correlation with experimental WSI has been
446 demonstrated for all the experiments, with deviations explained. **In fact, the R^2 value is equal**
447 **to or above 0.85 for any of the comparisons.** It is important to note that calibrations primarily
448 stem from the cement hydration kinetics and Jennings-Tennis's model, which are usually based
449 on hydration experiments for OPC based materials. Secondly, for the particle packing model,
450 the knowledge of the particle size and packing density is required, which also comes from
451 experimental investigations. From the pore network model perspective, only one calibration
452 parameter exists, which is related to the throat size, and can be calibrated on any one
453 experimental WSI. This framework therefore allows a greater flexibility to study WSI
454 behaviour of arbitrary OPC based materials. Note that if experimental pore size distribution is

455 available then the pore network model is solely sufficient to estimate WSI. One of the main
456 advantages of using pore network modelling is that it is not only useful for WSI (including
457 hysteresis) but also for estimating liquid permeability and vapour diffusivity of cementitious
458 materials. Hence, the proposed work is a promising approach to address many hydraulic
459 parameters for moisture transport studies within a single framework.

460 Finally, it is acknowledged that there are shortcomings in the models used in the proposed
461 framework. For instance, uncertainties associated with the chosen cement hydration kinetics
462 model, the missing pore size distribution because of the limitation of model resolution of the
463 cement hydration kinetics model and the applicability of Jennings-Tennis's model for arbitrary
464 OPC based materials.

465

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Table 1. Materials studied (CP: cement paste; M: Mortar; Co: concrete).

Material	C ₃ S	C ₂ S	C ₃ A	C ₄ AF	W/C	Aggregate/Cement	Experimental conditions	Experimental technique
CP1	56.5	18	6.3	11.4	0.45	-	Immersion in limewater for 56 days	Drying progressively for 270 days using ASTM C157, T=25 ± 0.2
CP2	57.28	23.98	3.3	7.6	0.34	-	1 year old specimen without water exchange, vacuum rewetted for drying	Drying controlled by saturated salt solutions, T=21±0.5°C
CP3	44.61	37.14	10.56	7.7	0.35	-	Endogenous curing conditions for 1 year	Drying controlled by saturated salt solutions, T=20°C
CP4	65.84	22.72	3.25	8.19	0.45	-	Same as CP3	Same as CP3
CP5	17.53	54.67	13.4	14.4	0.5	-	Immersion in limewater for 5 months	Climatic chamber, T=20±1 °C
CP6	17.53	54.67	13.4	14.4	0.8	-	Same CP5	Same CP5
M1	62.74	17.39	7.92	11.95	0.5	3.03	Same as CP3	Same as CP3
M2	17.53	54.67	13.4	14.4	0.5	3	Same as CP5	Same as CP5
M3	17.53	54.67	13.4	14.4	0.8	3	Same as CP5	Same as CP5
Co1	74.27	9.66	0.74	15.33	0.45	4.5	Same as CP3	Same as CP3
Co2	32.43	44.54	12.39	10.55	0.40	4.46	Same as CP3	Same as CP3

Table 2. Results from VCCTL and Jennings-Tennis's hydration model (CP: cement paste, Co: concrete, M: Mortar): All % values are rounded.

Material	W/C	C-S-H in cement paste (%)	HD C-S-H in C-S-H gel (%)	LD C-S-H in C-S-H gel (%)	HD C-S-H Porosity (%)	LD C-S-H Porosity (%)	Capillary Porosity (%)
CP1	0.40	51	54	46	7	9	21
CP2	0.34	45	64	36	7	6	10
CP3	0.35	45	70	30	8	5	15
CP4	0.45	44	40	60	4	10	25
CP5	0.5	47	28	72	3	13	37
CP6	0.8	49	22	78	3	14	47
M1	0.5	40	28	72	3	11	25
M2	0.5	44	28	72	3	12	16
M3	0.8	39	28	72	3	10	18
Co1	0.45	40	42	58	4	9	10
Co2	0.40	45	43	57	5	9	14

Table 3. R^2 (coefficient of determination) values for the predicted desorption curves.

Material	R^2 value
CP1	0.94
CP2	0.88
CP3	0.92
CP4	0.95
CP5	0.92
CP6	0.95
M1	0.94
M2	0.92
M3	0.94
Co1	0.93
Co2	0.85

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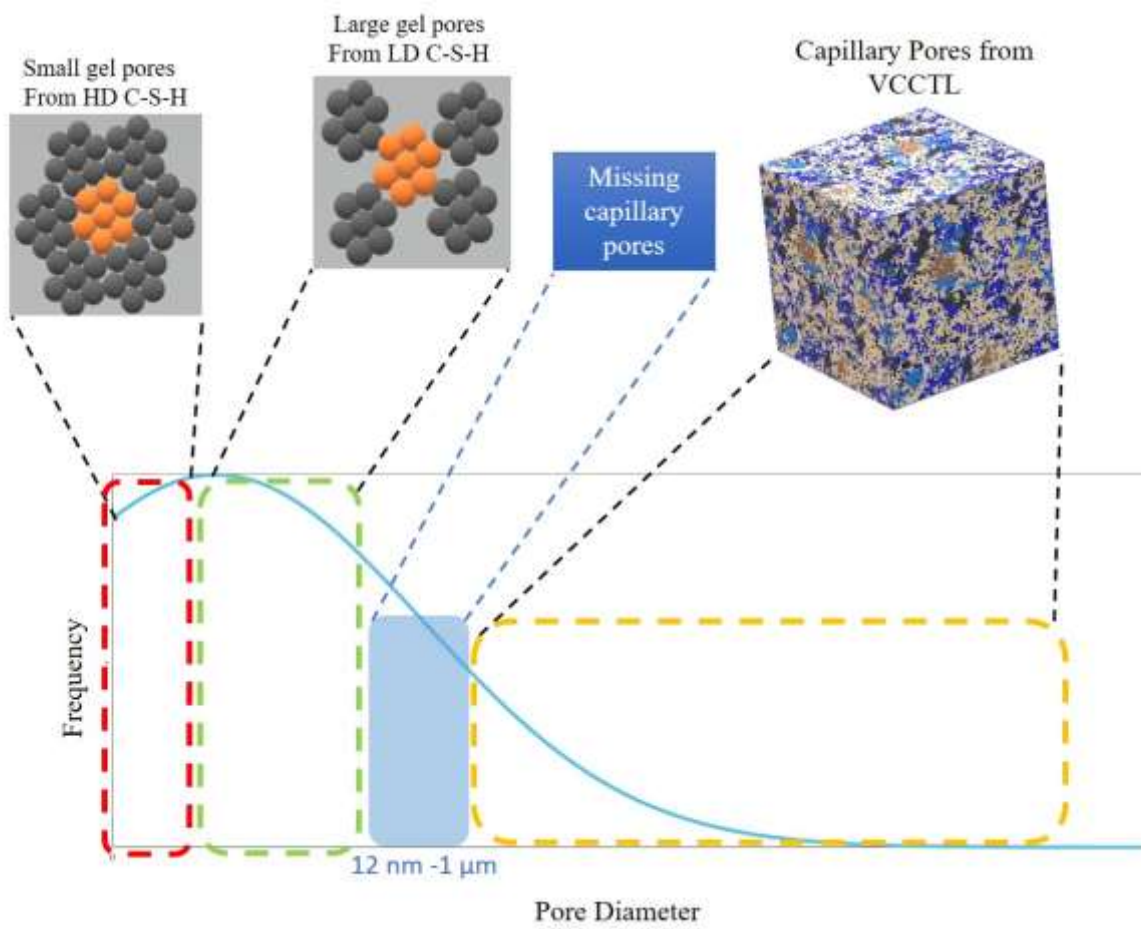


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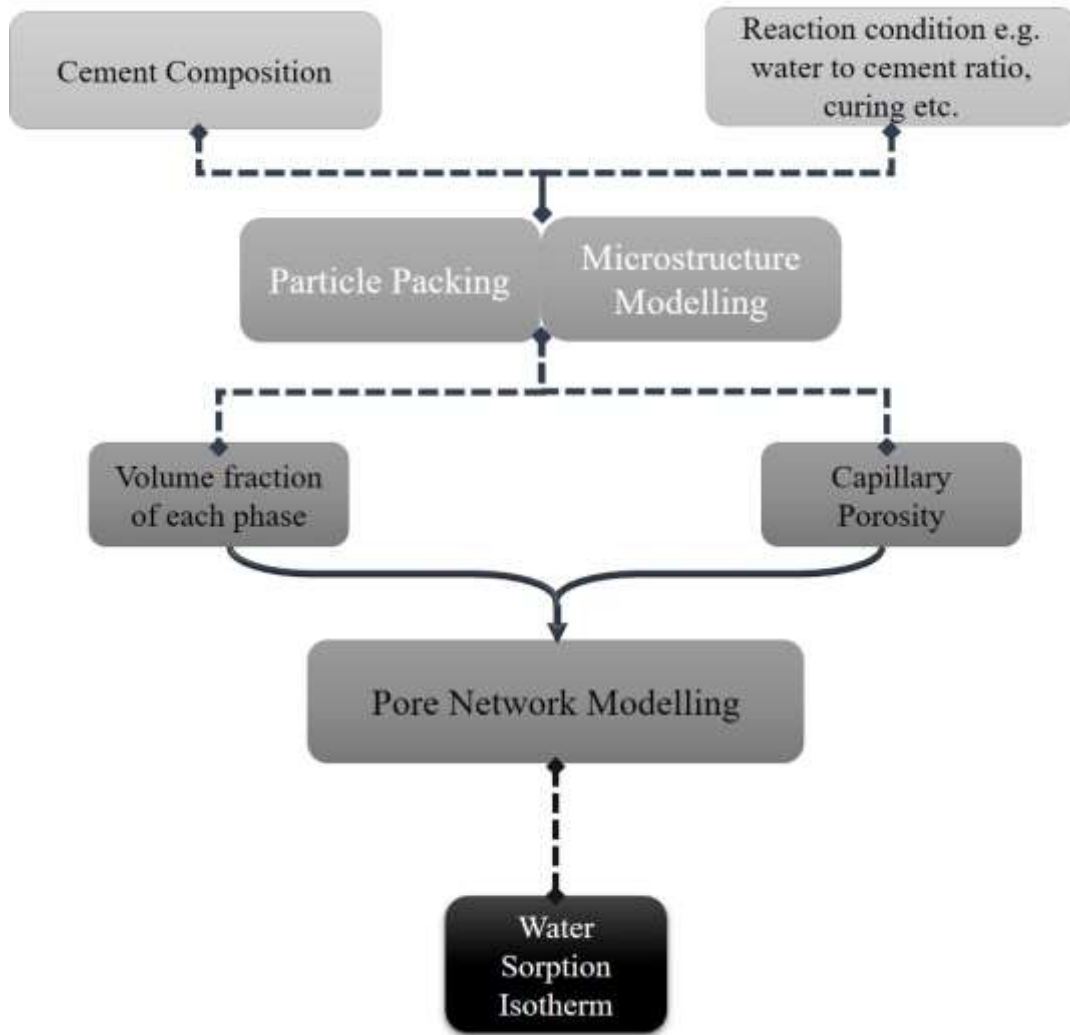


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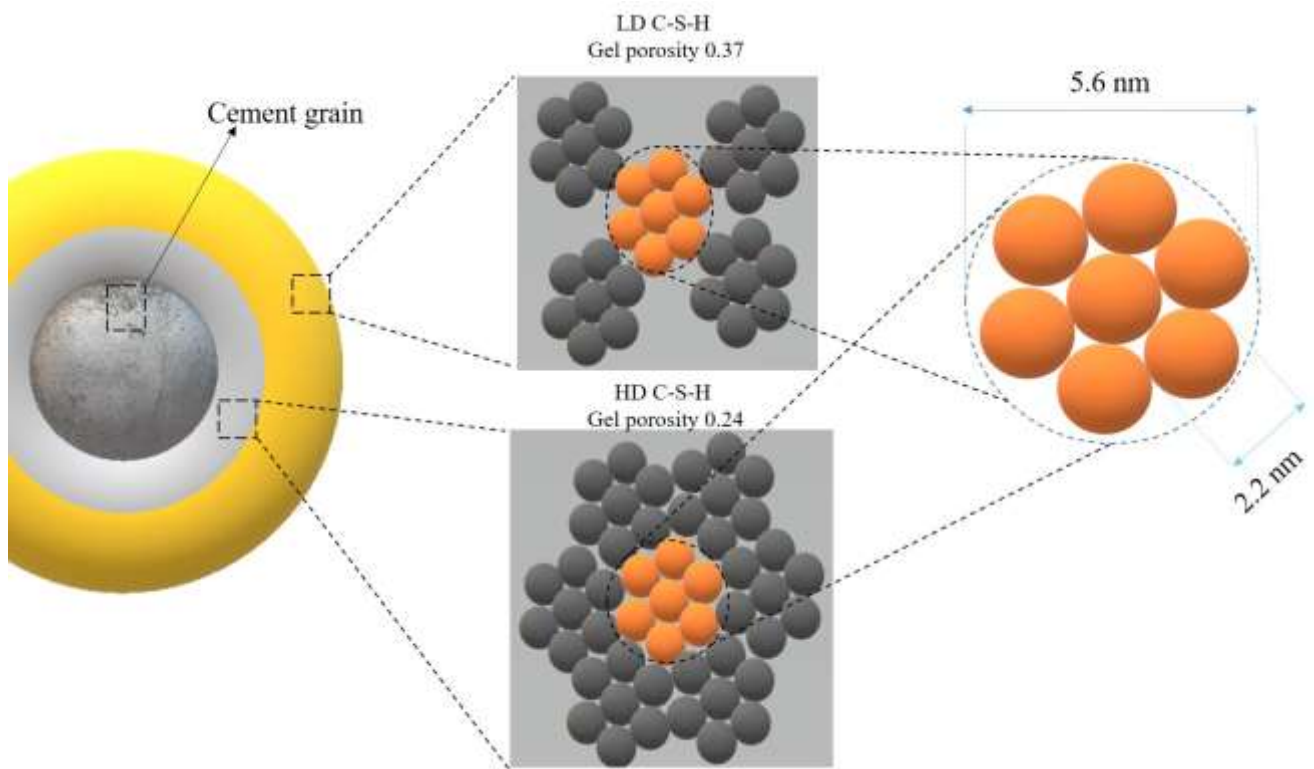


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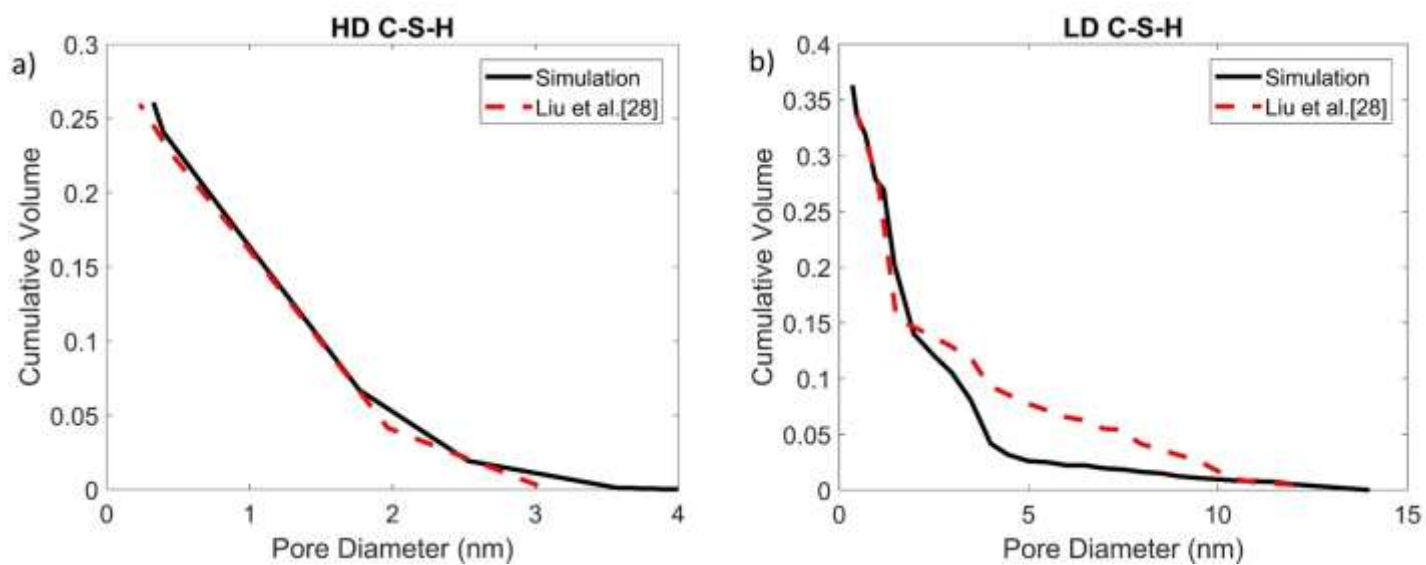


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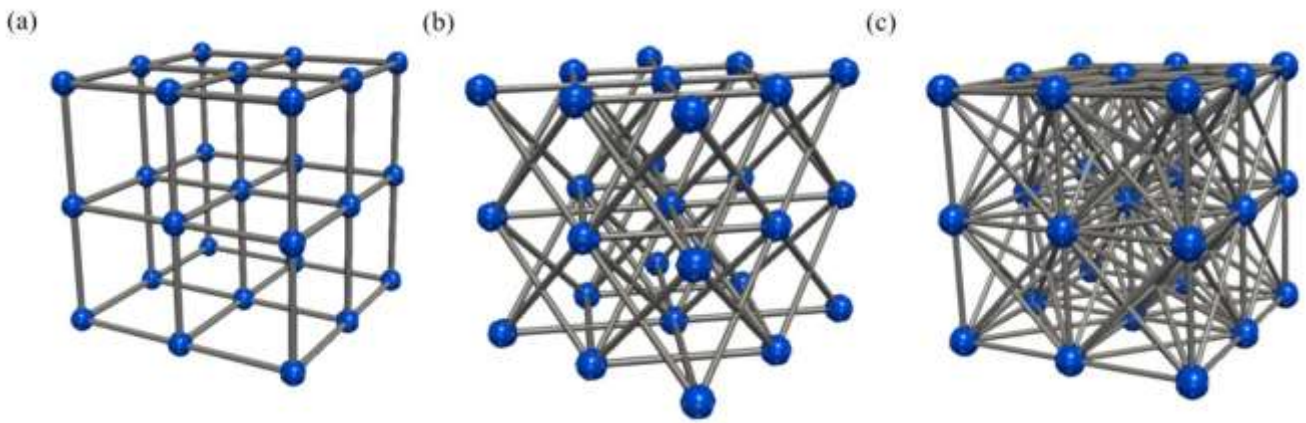


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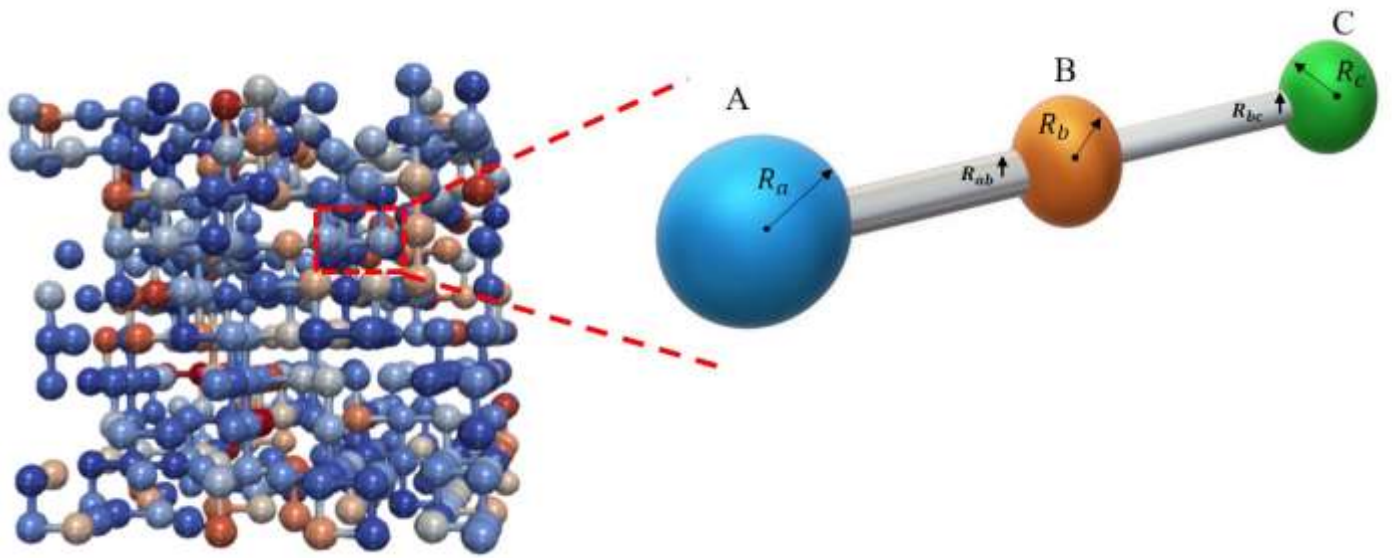


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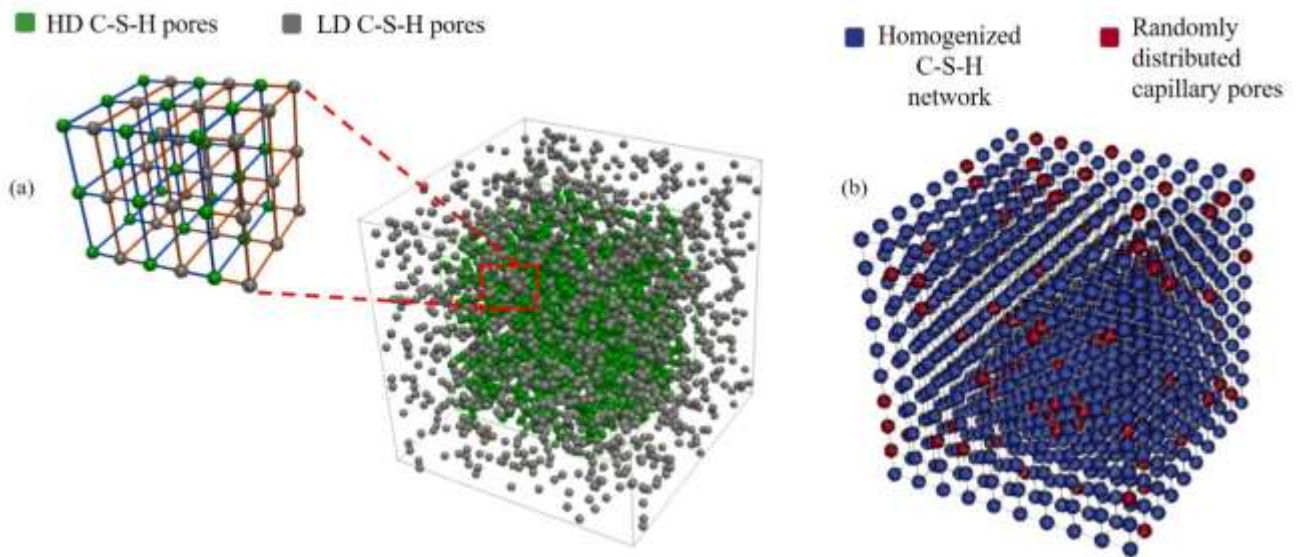


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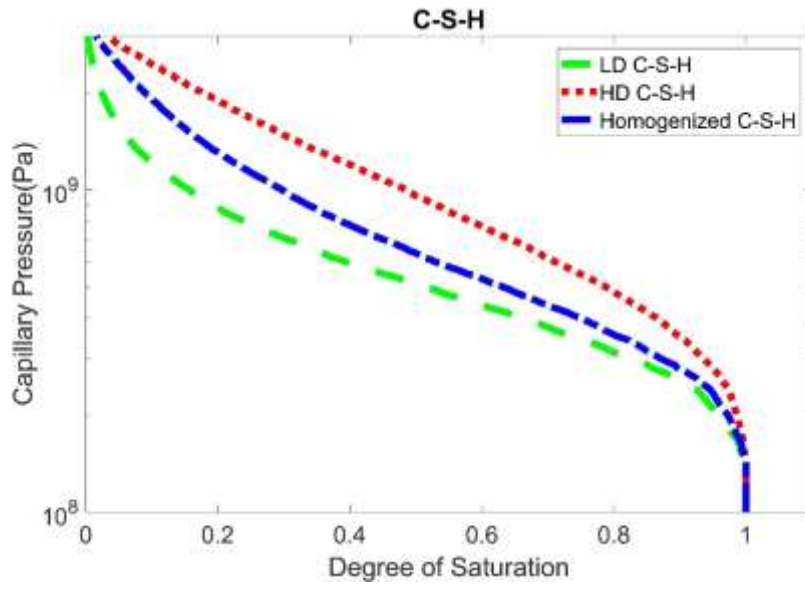
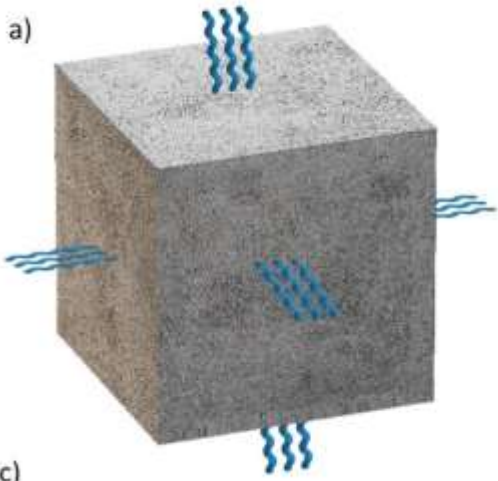


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The real material, Drying from all the sides



Invasion from all the sides

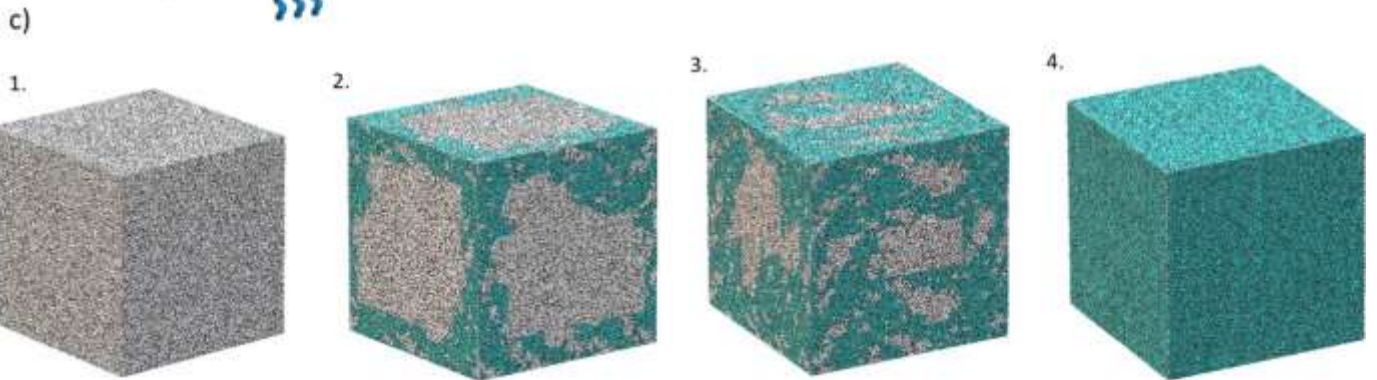
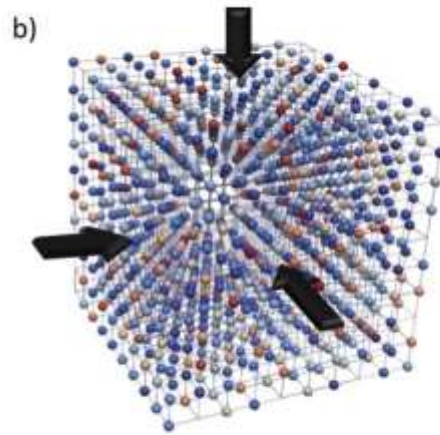


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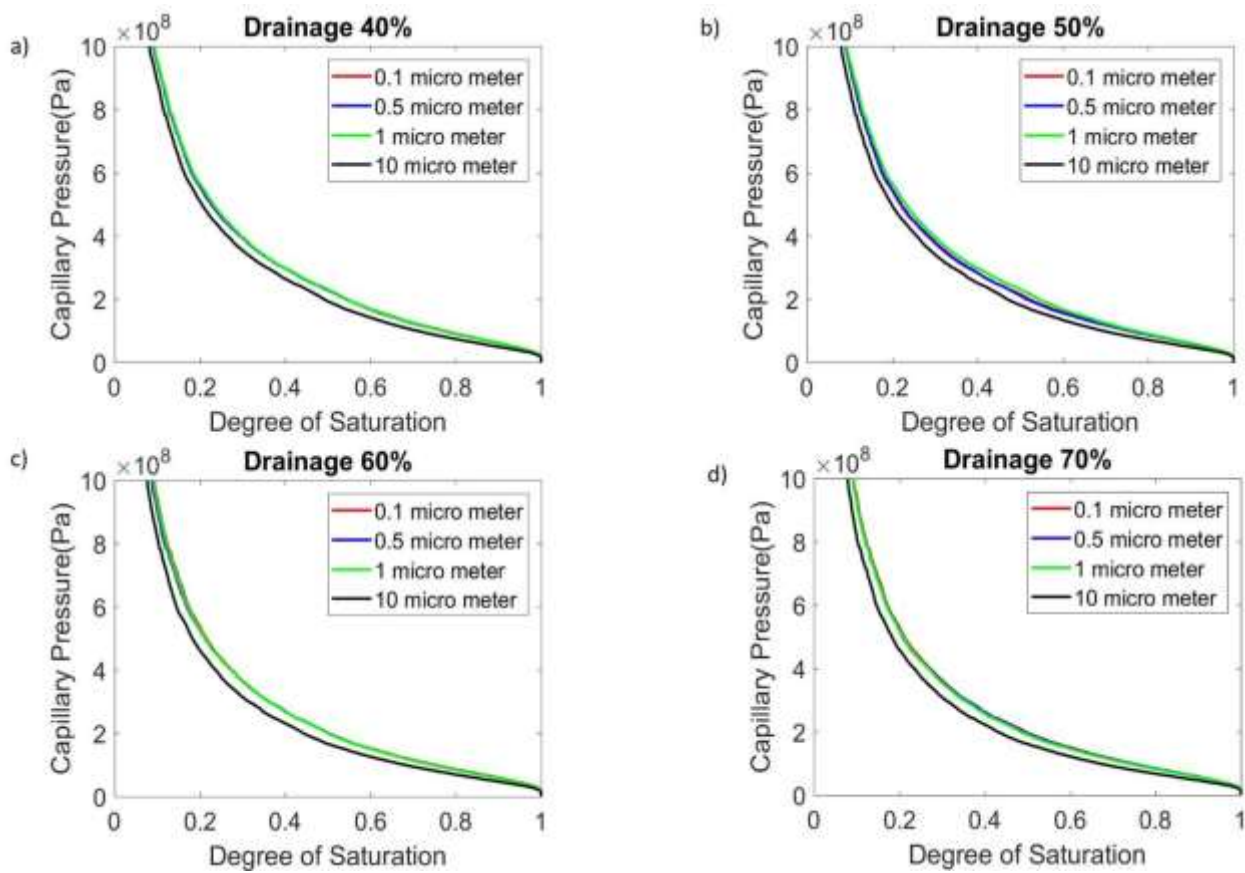


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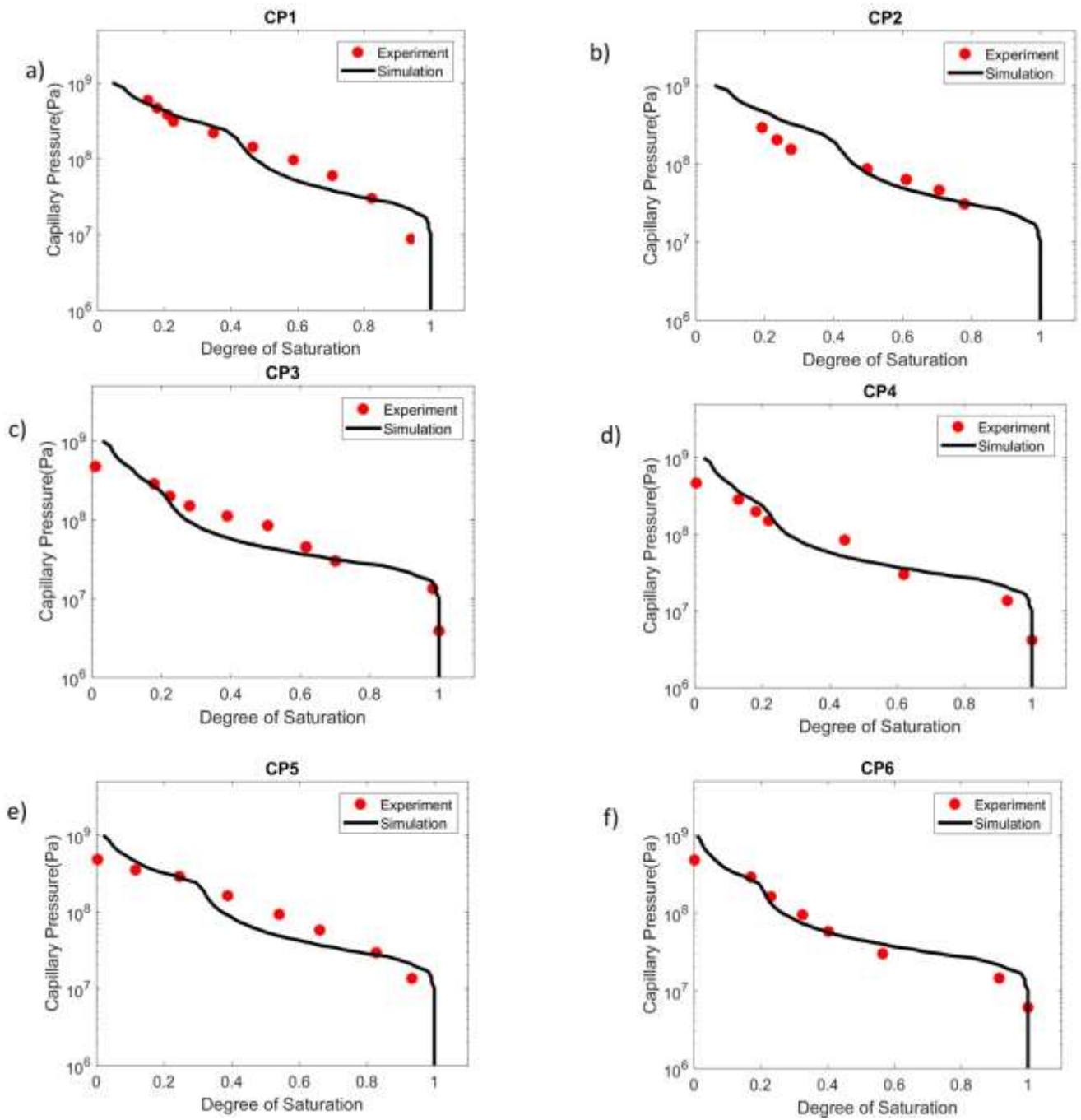


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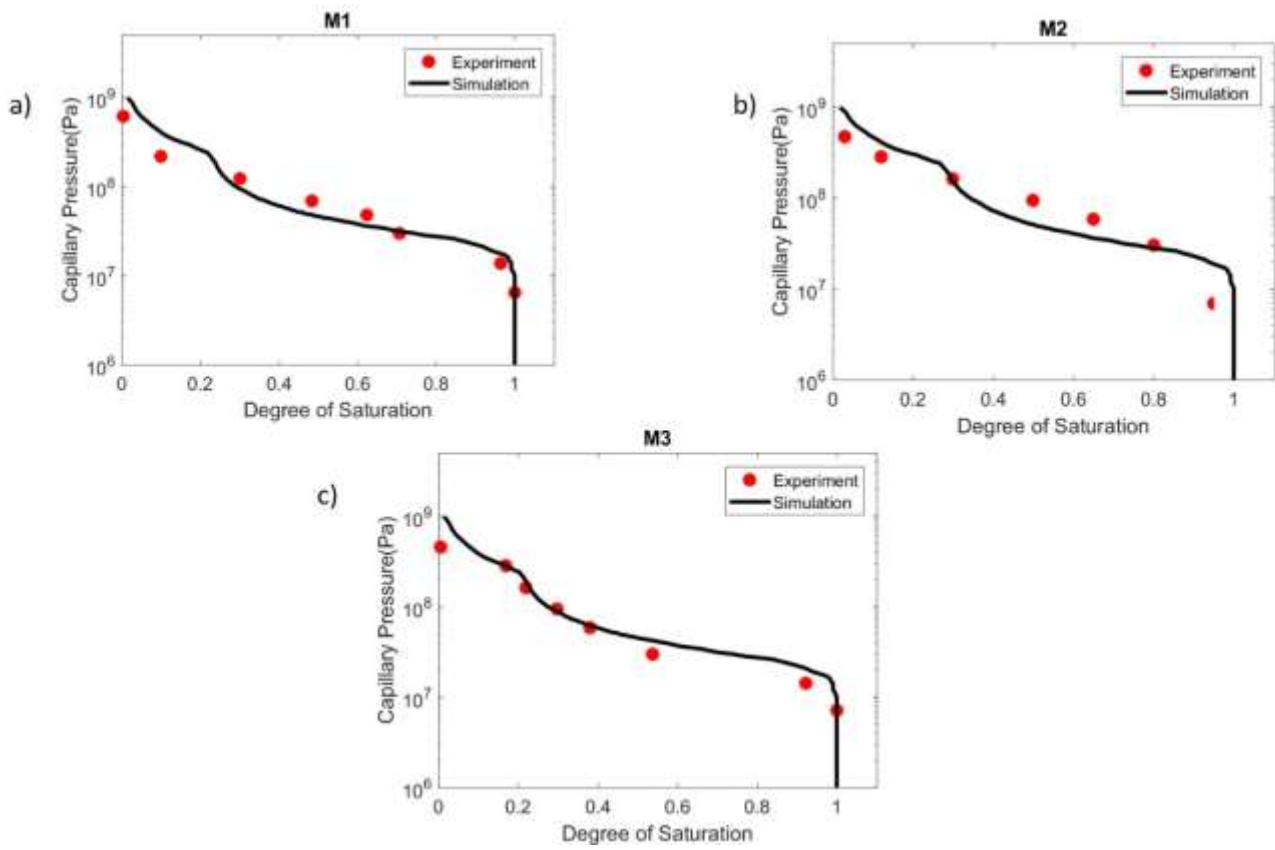


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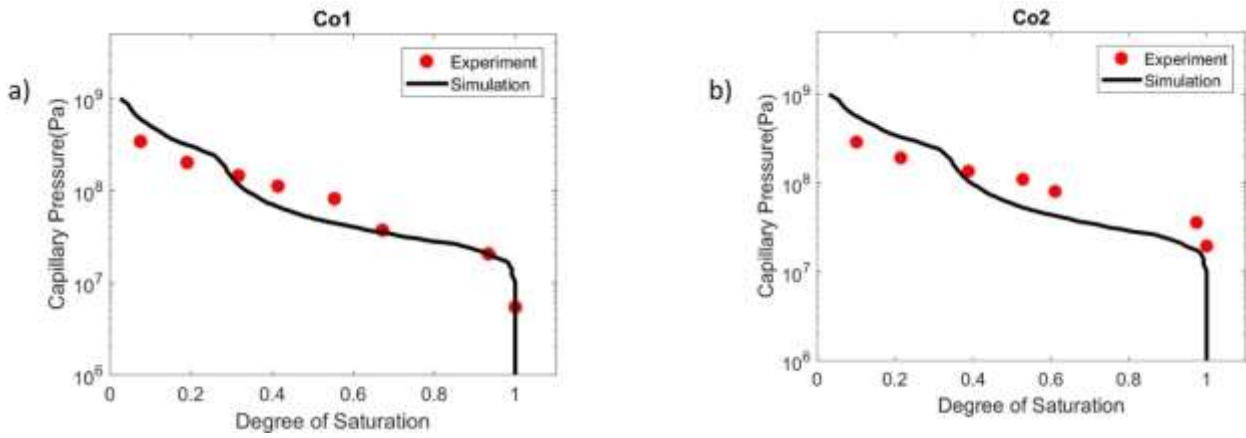


Figure 13. Experimental observations vs. simulation using the pore network model. (a) Co1, and (b) Co2.

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