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Enhancing swine manure treatment : a full-scale techno-economic assessment of nitrogen recovery, pure oxygen aeration and effluent polishing

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- Enhancing swine manure treatment: a full-scale techno-economic assessment of nitrogen recovery,
- pure oxygen aeration and effluent polishing
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- **List of acronyms, abbreviations and definitions**
- AN ammonium nitrate
- AS ammonium sulphate
- BOD Biological oxygen demand
- CAPEX capital expenditure
- COD Chemical oxygen demand
- CW Constructed wetland
- EU European Union
- FA Free ammonia
- FPR Fertilising Products Regulation
- LF Liquid fraction
- MLSS Mixed Liquor Suspended Solids
- NDN Nitrification denitrification
- NVZ Nitrate Vulnerable Zones
- OPEX operational expenditure
- PO Pure oxygen aeration
- Re Recovery efficiency
- RENURE Recovered nitrogen from manure
- Rm- Removal efficiency
- Se Separation efficiency
- SF Solid fraction
- SS Stripping-scrubbing
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#### **ABSTRACT**

 In regions with intensive livestock production, managing the environmental impact of manure is a critical challenge. This study, set in Flanders, Belgium, evaluates the effectiveness of integrating process intensification measures into the treatment of piggery manure to mitigate nitrogen surplus issues. The research investigates the techno-economic benefits of implementing three key interventions: pure oxygen (PO) aeration, ammonia (NH3) stripping-scrubbing (SS) pretreatment, and tertiary treatment using constructed wetlands (CW), within the conventional nitrification-denitrification (NDN) process.

 Conducted at a full-scale pig manure treatment facility, our analysis employs steady-state mass balances for nitrogen (N) and phosphorus (P) to assess the impact of these process intensification strategies. Findings indicate that the incorporation of advanced treatment steps significantly enhances the efficiency and cost-effectiveness of the manure management system. Specifically, the application of PO aeration is shown to reduce overall treatment costs by nearly 4%, while the addition of an NH3 SS unit further decreases expenses by 1 to 2%, depending on the counter acid utilized. Moreover, the implementation of a CW contributes an additional 4% in cost savings.

 Collectively, these measures offer substantial improvements in processing capacity, reduction of by-product disposal costs, and generation of additional revenue from high-quality fertilizing products. The study highlights the potential of advanced treatment technologies to provide economically viable and environmentally sustainable solutions for manure management in livestock-dense regions, emphasizing the cumulative economic benefit of a holistic approach to process intensification.

#### **1. INTRODUCTION**

 Since global meat consumption and production has grown considerably over the last century, a large amount of manure is produced contributing to a significant part of anthropogenic N emissions and the exceedance of the 'planetary boundaries' capacity (Campbell et al., 2017). Annually about 1.3–1.8 billion tonnes of manure are produced in the European Union (EU) (De Vrieze et al. 2019), mainly concentrated in densely populated regions where the Nitrates Directive (EU) 676/1991 70 application limit of 170 kg N ha<sup>-1</sup> y<sup>-1</sup> is in effect, limiting application of animal manure and hence resulting in its excess. Strict manure management regulations in Flanders (Belgium), exemplifying a livestock-dense region with a manure surplus, enforce the processing of manure. Around half of the produced pig manure in Flanders is processed by the conventional treatment line consisting of a centrifuge for separation and an activated sludge tank to remove organic and inorganic contaminations (Coppens et al., 2016). The excess manure is first separated into a solid (SF) and a liquid fraction (LF) by centrifugation, whereafter the SF is dried and exported. The LF is subsequently processed by biological nitrification-denitrification (NDN) treatment followed by a clarifying step to produce an effluent with a reduced N content that can be applied on agricultural land (Brienza et al., 2023). The NDN process in Flanders removes 21 kilotonnes of N from manure 80 by converting reactive N into N gas  $(N_2)$  (Vingerhoets et al., 2021). Nitrosomonas sp. bacteria 81 primarily limit the rate of ammonia ( $NH<sub>3</sub>$ ) oxidation to nitrite ( $NO<sub>2</sub>$ ), consequently restricting the 82 process capacity of the treatment line to the critical N load in the NDN tank (Yu et al., 2020). N 83 loads above the rate-limiting nitrification reaction capacity induces accumulation of NH<sub>4</sub><sup>+</sup> which 84 can inhibit nitrification and further reduce the whole processing rate of the NDN tank (Nehmtow et al., 2016). Increasing the capacity of the NDN tank by expanding the reactor's dimensions is associated with high capital costs or infeasible because of space constraints (Derden and Dijkmans, 2020).

 Elevating NDN processing capacity by process intensification can resolve this problem. The usage of pure oxygen (PO) instead of air during the aeration process increases the dissolved oxygen concentration in the wastewater, which improves the rate of the nitrification process and potentially increases the capacity of the NDN system (Rodriguez et al., 2012), but will also increase cots associated with the installation and maintenance of a PO tank. Skouteris et al. (2020) reviewed 25 studies evaluating PO aeration instead of air aeration for different wastewaters including municipal, industrial, leachate, mill effluent, food processing, petrochemical and synthetic. This review

 suggests a research gap on PO in swine manure treatment, indicating a broader lack of available studies or data on the use of PO aeration in swine manure treatment. Therefore, this study provides a first (to the authors knowledge) dataset to evaluate the economic feasibility of PO aeration for swine manure treatment at full scale.

 To enhance the processing capacity of the NDN unit, another pathway involves reducing the N load entering the NDN tank. This is achieved by recovering N before NDN treatment through the addition of an extra recovery step, without compromising the amount of treated manure. This is particularly relevant for long-stored manure, where the elevated concentration of Ammonium N (NH4-N), following separation, poses a challenge to the efficiency of subsequent biological treatment steps due to its inherent toxicity (Hollas et al., 2021). However, this issue can be effectively addressed through the application of physio-chemical recovery technologies, where NH4-N is easily targeted. By reducing the N load introduced into the NDN system, a substantial 107 decrease in nitrous oxide  $(N_2O)$  emissions associated with manure treatment can be achieved. Current estimates place these emissions in the range of 0.035-1.1% of the total input into the NDN (de Haas and Andrews, 2022; Kampschreur et al., 2009 and Ravi et al., 2023). N removal in 110 wastewater treatment therefore contributes almost 5% of the global  $N_2O$  emissions (Olivier et al., 2017). As arable farming and horticulture in Flanders require 81 kilotonne additional N in the form of mineral fertilizers (Vingerhoets et al., 2023b), which are produced through the energy-intensive 113 Haber-Bosch process converting atmospheric  $N_2$  to  $NH_3$ , N recovery as a mineral N fertiliser equivalent also counters the additional need for N in the form of mineral fertilisers and reduce costs of purchasing N mineral fertilisers (Zarebska et al., 2015). Membrane filtration (e.g., nanofiltration and reverse osmosis) and physicochemical processes (e.g., NH<sup>3</sup> SS and chemical precipitation) are alternative treatment configuration with a high technology readiness level and adoption potential (Brienza et al., 2023). The implementation of membrane filtration involves high investment costs, intensive maintenance, and thorough pre-treatment whereas NH<sup>3</sup> SS is well suited for small-scale implementation and more concentrated wastewaters (van Puffelen et al., 2022). The rollout of full-scale NH3 SS technology installations at manure treatment facilities located in livestock-dense regions to complement the current NDN treatment process could reduce the mineral fertilizer demand with 8% (Vingerhoets et al., 2023). NH<sup>3</sup> SS towers utilize a two-step process, where the initial step involves inducing NH3 volatilization from the liquid phase into the gas phase by 125 circulating ventilation air over the LF sprayed in a packed tower with inert material.  $NH<sub>3</sub>$  in the recirculation air is absorbed by sending the NH3-rich gas over a counter acid solution in the 127 scrubber unit, which recovers  $NH_3$  in the form of ammonium ( $NH_4$ ) salts.  $NH_3$  SS installations 128 have employed sulfuric acid  $(H_2SO_4)$  and nitric acid  $(HNO_3)$  for this purpose, as reported by 129 Brienza et al. (2021), which results in the production of ammonium sulphate ( $(NH_4)_2SO_4$  AS) and ammonium nitrate (NH<sub>4</sub>NO<sub>3</sub>, AN) solutions respectively. The recovery of N in the form of NH<sub>4</sub><sup>+</sup> salts presents a promising opportunity for the integration of manure-derived products into the mineral fertiliser market and thereby combines two key nutrient management strategies: circularity and system's efficiency (Spiller et al. 2024). In particular, these products are likely to conform to the recently established RENURE (REcovered Nitrogen from manURE) product criteria, which dictate the suitability of manure-derived products as mineral fertilisers in Nitrate Vulnerable Zones (NVZs) under the same regulations as synthetic fertilisers (Huygens et al., 2020). These criteria are in line with the existing Fertilising Products Regulation (FPR) (EU) 1009/2019 regulation. To this end, this study seeks to evaluate the economic viability of two scrubbing acids (AS and AN) and assess their compliance with the RENURE and FPR criteria.

As the effluent of NDN treatment tank does not comply with the Flemish discharge limits, transport and land spreading of the produced effluent also contributes to significant part of the manure processing costs (Willeghems et al., 2016). To comply with Flemish discharge limits, Donoso et al. (2015) and Meers et al. (2008) proposed the use of CW structed wetlands as an economically and ecologically beneficial solution for the tertiary treatment of NDN effluent. Therefore, in our study the hypothesis that the implementation of this in-situ post-treatment step can further increase economic profitability and sustainability of manure processing plants as it reduces costs associated to transport and gate fees was assessed.

The primary focus of this study is to conduct an economic analysis of intensive manure management strategies and assess the extent to which the integration of process intensification measures into the swine manure treatment configuration proves economically advantageous as an addition to existing manure treatment facilities. The economic assessment of each configuration (i.e. implementation of an additional measure) was based on technological assessment with steady-state mass balances, including mass, N and P, derived from a monitoring campaign executed for this study. By analysing the mass balances, this study determined operational expenditure (OPEX), capital expenditure (CAPEX), product quality and revenues from the final products for five configurations, which enables the determination of the most economical set of intensification measures for a full-scale pig manure treatment facility.

#### **2. MATERIAL AND METHODS**

- 2.1. Bio Sterco site description and employed nutrient recovery configurations
- The Bio Sterco farm (Hooglede, Belgium) has a capacity to raise 454 sows, 5 boars and 5524 and
- 162 exploits its own manure treatment installation (current maximum capacity t y<sup>-1</sup>) which has
- been operational since 2011. The Bio Sterco farm implemented the following configurations from
- 2011 to 2022: (i) configuration 1: Centrifugation + NDN, (ii) configuration 2: Centrifugation +
- NDN with PO aeration, (iii) configuration 3: Centrifugation + NH3 SS + NDN with PO aeration,
- 166 and (iv) configuration 4: Centrifugation +  $NH<sub>3</sub>$  SS + NDN with PO aeration + CW (Figure 1).



168 Figure 1 Different configurations of the treatment line of the Bio Sterco manure processing plant (1) Raw swine manure, (2) LF of manure, (3) SF of manure, (4) NDN tank, (5) Sludge, (6) Effluent settling, (7) Effluent 169 manure, (3) SF of manure, (4) NDN tank, (5) Sludge, (6) Effluent settling, (7) Effluent SS, (8) NH4+ salt, (9) P-precipitate, (10) 170 Influent of the wetland, (11) Effluent of the wetland Influent of the wetland , (11) Effluent of the wetland

#### 2.1.1 Configuration 1: Centrifugation + NDN

- 
- Configuration 1 follows the treatment path of a conventional manure processing system including
- a centrifuge for mechanical separation, an activated sludge tank focused on NDN and a settling
- tank to remove the activated sludge from the effluent. In this processing cascade, pig manure is

first separated into a LF and SF fraction by centrifugation (Westfalia decanter AG) with a polymer dosing (cationic polyacrylamide) unit to improve separation efficiencies into SF and thereby increase the clarity of the LF. The SF is subsequently composted while the N-rich LF of manure is 180 biologically treated through a NDN system. The NDN system consists of a tank of  $3393m<sup>3</sup>$  with four submerged ejectors aerators developed by the Bio Armor. The oxygenation cycle consists of intermittent aeration (20−30 min oxygenation followed by 15−20 min anoxic period). The NDN 183 effluent is sent to a settling tank of  $320 \text{ m}^3$  to separate the produced sludge from the effluent, which are both transported and disposed on land. The NDN unit is the crucial factor determining processing capacity in the whole treatment line. By applying a negative feedback-loop in the NDN 186 unit, the input quantity of LF of manure is regulated based on the prevailing  $NH<sub>4</sub>$ <sup>+</sup> concentration in the NDN tank which in its turn manages the amount of raw manure processed by the centrifuge. This affects the manure level in the raw manure storage and subsequently the raw manure acceptance rate (Figure 2).



 Figure 2 Control system and its feedback loops in a manure treatment plant

2.1.2 Configuration 2: Centrifugation + NDN with PO aeration

Like configuration 1, configuration 2 includes treatment steps such as a centrifuge for mechanical separation, an NDN tank for removing organic compounds and N, and a settling tank to eliminate activated sludge from the effluent. However, the NDN treatment was intensified by implementing 196 PO aeration tank (Air products) with a volume of  $22.7 \text{ m}^3$  and a maximal storage pressure of 22.2 bar. The aeration by PO can improve the dissolved oxygen concentration in the wastewater, which 198 reduces the vulnerability to  $NH_4$ <sup>+</sup> toxicity and thereby potentially increases the capacity of the NDN system. This treatment set-up was active from 2015 till 2022.

## 201 2.1.3 Configuration 3: Centrifugation +  $NH<sub>3</sub>$  SS + NDN with PO aeration

202

203 In the processing cascade of configuration 3, pig manure is first separated into a LF and SF by 204 centrifugation. A part of the N-rich LF of manure is treated in the  $NH<sub>3</sub>$  SS unit to reduce its NH<sub>4</sub>-205 N content before being fed to the NDN system. The NH<sup>3</sup> SS unit, developed by Detricon bvba 206 (Belgium) has a capacity to process 20,000 t  $y^{-1}$  and consists of two vertical acrylate stripping 207 columns (2.5m x 3m x 2m) and scrubbing column (2.5m x 3m x 2.5m). The stripping columns are 208 equipped with 9 spraying nozzles and are filled with pall rings as packing material. The ventilation 209 flow of 1,440 m<sup>3</sup> per hour, with an air speed of 0.2 - 0.8 m/s, is divided equally between the two 210 stripping columns. The NH3-rich air is then directed to the scrubber column, where it encounters a 211 diluted counter acid solution to produce a  $NH_{4+}$  salt solution as  $NH_3$  is absorbed by this solution. 212 Two sub-configurations (configuration 3a and 3b) were employed to evaluate the disparity in the 213 usage of counter acids  $(HNO<sub>3</sub>$  and  $H<sub>2</sub>SO<sub>4</sub>$ ), with NH<sub>3</sub> being recovered as AN or AS, respectively. 214 The resulting stripped LF was then mixed with the non-stripped LF and biologically treated through 215 a NDN system with PO aeration, whereafter the effluent was spread on land. From 2022 onwards, 216 configuration 3 a/b and 4 were active simultaneously. whereby on average almost 79 t  $d^{-1}$  was 217 following the treatment line of configuration 3 and 5 t  $d^{-1}$  was processed by configuration 4.

218

## 219 2.1.4 Configuration 4: Centrifugation +  $NH<sub>3</sub>$  SS + NDN with PO aeration + CW

 Configuration 4 followed the same treatment line as in configuration 3b. However, a tertiary treatment line to polish the effluent of the NDN treatment to dischargeable water was added. To reduce the remaining P content of the NDN effluent before entering the wetland, P precipitation is 223 induced by the addition of 4.5 l  $H_2SO_4$  (55%) and 3.5 l of FeCl<sub>3</sub> (40%).in a settling tank. The 224 acidification induced by  $H_2SO_4$  establishes more optimal conditions for the formation of FePO<sub>3</sub> precipitates, thereby reducing the demand for FeCl3, a crucial factor given the stringent discharge limits for chloride (Cl) in Flanders. Besides Fe precipitation in combination with pH reduction, the precipitation of P can also be achieved by elevating the pH and forming Ca complexes through the 228 introduction of  $Ca(OH)_2$ . In this study only Fe precipitation in combination with acidification by  $H<sub>2</sub>SO<sub>4</sub>$  addition was tested as the authors believed that this is the most economic option for the studied treatment line. After P removal, the effluent is polished in a CW of 1268 ha to comply with 231 the surface water discharge norms. The CW system is divided into nine beds/lagoons including one

232 underwater system, one U-turn wetland, four horizontal flow fields, two vertical percolation fields

233 and one lagoon (Figure Supplementary Material (SM) A).

234

235 2.2. Sampling and physio-chemical analysis of process streams

236 We monitored the different configurations of the manure processing plant over the following 237 periods for this study:

<sup>238</sup> • Configuration 1: from February 12<sup>th</sup>, 2012, to April 21<sup>th</sup>, 2015

- <sup>239</sup> Configuration 2: from April  $16<sup>th</sup>$ , 2015, to February  $1<sup>th</sup>$ , 2022
- <sup>240</sup> Configuration 3a: from February  $1<sup>th</sup>$  2022 to March  $15<sup>th</sup>$  2022
- <sup>241</sup> Configuration 3b: from March  $16<sup>th</sup> 2022$  to March  $16<sup>th</sup> 2023$
- $\bullet$  Configuration 4: from March 16<sup>th</sup> 2022 to March 16<sup>th</sup> 2023

More detailed information on the monitoring periods of the different configurations can be found in SM B. The points sampled and the analysed parameters varied according to sampling period: 245 Monitoring period 1 and 2 focused on Total N (TN) and P for all sample collection points, while 246 also Chemical Oxygen Demand (COD),  $NH_4-N$ ,  $NO_3-N$ ,  $NO_2-N$  were monitored in the NDN tank. A wider range of parameters were analysed during period 3 for all sampling locations including pH, EC, Dry Matter (DM), Suspended solids (SS), COD, Biological Oxygen Demand (BOD), TN, 249 NH<sub>4</sub>-N, NO<sub>3</sub>-N, P, K, Sulphur (S), Total Organic Carbon (TOC), Copper (Cu) and Zinc (Zn). The corresponding analysis methods are described in SM C.

251

252 2.3. Calculation of separation, recovery and removal efficiencies

253 Separation efficiency (Se) was determined for the centrifuge and indicates the amount of a 254 component in the solid and liquid fraction respectively compared to the total input of this element 255 (Eq. 1).

$$
Se = \frac{X * C_x}{Y * C_y}
$$
 Eq. 1

256 Where Se represents separation efficiency,  $X$  (kg) is the amount of outgoing fractions,  $Cx$  (g kg<sup>-1</sup>) 257 represents the concentration of a particular component (e.g. P) in the outgoing fractions,  $Y$  (kg) is 258 the quantity of ingoing raw manure and Cy (g kg<sup>-1</sup>) is the concentration of a a particular component 259 in the ingoing substrate.

264

261 Recovery efficiency (Re) was calculated for the NH<sup>3</sup> SS unit and P precipitation unit and stands 262 for the mass of component in the recovered fertiliser (i.e.  $NH<sub>4</sub>$ <sup>+</sup> salt solution or FePO<sub>4</sub> precipitate) 263 respectively as a proportion of the total input from the unit process (Eq. 2).

> $Re =$  $X * C_{x}$  $Y * C_y$ Eq. 2

265 The mass of the recovered fertiliser is represented by  $X$  (in kg), the concentration of the considered compound in the recovered fertiliser is Cx (in g  $kg^{-1}$ ), the mass of the influent is Y (in kg), and the 267 compound concentration is Cy (in  $g kg^{-1}$ ).

268

269 Removal efficiencies (Rm) for NDN and CW were determined by the difference between the 270 effluent and influent concentrations of the considered component (Eq. 3).

271

$$
Rm = \frac{C_y - C_z}{C_y} \tag{Eq. 3}
$$

272

273 Where Cz (mg  $kg^{-1}$ ) is the concentration of the compound under consideration in the effluent, while 274 Cy (mg  $kg^{-1}$ ) reports on the concentration of this compound in the influent.

275

276 2.4. Economic assessment

This study estimated the CAPEX and OPEX of the different configurations based on quotes from constructors received by the studied manure treatment facility and online market prices. The CAPEX costs are determined by the actual investment costs paid for each technology unit, while 280 the OPEX costs included electricity use, resource consumption (i.e. polymer,  $H_2SO_4$ ,  $HNO_3$ , Methanol and FeCl3), insurance, maintenance and labour. The maintenance of the PO tank is managed by the PO supplier, with associated costs included in the PO's selling price. Consequently, we did not separately account for additional maintenance and labour costs for the treatment facility beyond these considerations. Table 1 The total investment cost of the treatment plant and maintenance costs per technology unit





286  $^{\circ}$  <sup>a</sup>De Vrieze et al. (2019) <sup>b</sup>full-time equivalent

287

288 The capital cost of all unit processes was amortised according to Eq. 4 assuming the process unit 289 specific investment cost (C) (Table 1) with an interest rate (r) of 5% depreciation period (n) of 10 290 years (Møller, 2000).

$$
Q = C * \frac{r((1+r)^n)}{(1+r)^n - 1}
$$
 Eq. 4

291

The OPEX costs associated with the use of chemicals, labour, electricity, disposal of by-products and the estimated market value of the RENURE fertilizer products were gathered from online databases and through accounting data from the treatment facility under study (Table 2). As prices of these consumables are volatile, the cost of it was assessed by applying a probabilistic approach through Monte Carlo simulation with uniform distributions created by RAND() function in excel for the data provided in Table 2*.* The OPEX and CAPEX of each configuration was determined based on the total volume of manure treated and the associated electricity and resource consumption in a certain monitoring period.

300 Table 2 The considered cost price of the used chemicals, electricity, labour and by-products disposal and the estimated market value for the produced fertilisers.

<b>Chemicals, products or energy</b>	Standard price $(\epsilon t^{-1})$	Range in price ( $\epsilon t^{-1}$ )
Polymer <sup>a</sup>	1,700	$1,000 - 4,000$
$H2SO4a$	120	$100 - 300$
$HNO3$ <sup>a</sup>	200	$150 - 400$
FeCl <sub>3</sub> <sup>a</sup>	100	$80 - 250$
$O_2^a$	1.3	$0.7 - 3.3$
$NH_4(SO_4)_2$ solution <sup>b</sup>	79	$0 - 260$
$NH4NO3$ solution <sup>b</sup>	210	$0 - 400$
SF disposal <sup>c</sup>	20	$15 - 25$
NDN effluent disposal <sup>c</sup>	$\overline{4}$	$3 - 5$



Sources: <sup>a</sup> Price quote of retailer <sup>b</sup>Vingerhoets et al. 2023a., 2023 °Derden and Dijkmans, 2020 <sup>d</sup>De Vrieze et al. 2019

## 303

## 304 **3. RESULTS AND DISCUSSION**

- 305 3.1. Overall mass balance of the different treatment configurations
- 306 On average a higher processing capacity was achieved after the implementation of the changes to
- 307 intensify the recovery process. The daily processing capacity for configuration 1, 2 and 3 was on
- 308 average 91, 102 and 116 tonne raw manure, respectively (Figure 3A).





310

*Figure 3* Mass balance for A) overall mass B) nitrogen and phosphorus expressed in t d<sup>-1</sup> of solid and liquid products within the process flows of the manure treatment plan

The centrifuge unit first separated raw manure into LF and SF, recovering the majority of P in the SF with a separation efficiency above 85% for all configurations. A review of 11 studies evaluating mechanical separation technologies used in manure treatment by Lyons et al. (2021) found that the P separation efficiencies for a decanter centrifuge without chemical addition varied between 30– 91%, while screw presses recorded a P separation efficiency of only 4–34%. Polymer addition (cationic polyacrylamide), inducing coagulation between solids, along with the most efficient swine slurry separation technology (i.e. centrifugation), resulted in elevated separation efficiencies compared to the literature findings in our study (Hjorth et al., 2010).The majority of N was found in the LF as separation efficiencies for N did not exceed 35%.

321

322 NDN treatment achieved similar removal efficiencies for LF in configurations 1 and 2, but the 323 utilization of PO as the aeration agent resulted in an increased capacity for LF treatment (76 m<sup>3</sup> LF 324 d<sup>-1</sup> m<sup>-3</sup> vs 86 m<sup>3</sup> LF d<sup>-1</sup> m<sup>-</sup>3) with a higher daily N load (0.135 kg N m<sup>3</sup>d<sup>-1</sup> vs 0.154 kg N m<sup>3</sup>d<sup>-1</sup>) as 325 compared to configuration 1. The suspension was further processed using a settling tank to obtain 326 3-6 t d<sup>-1</sup> of activated sludge and 73-80 t d<sup>-1</sup> of effluent depending on the configuration. The effluent 327 of the settler was sold to neighbouring farmers to bring it to the crop land.

328 Configuration 3 involved a more complex LF treatment, processing an average of 29 t  $d^{-1}$  of LF in  $329$  the NH<sub>3</sub> SS unit, where a counter-airflow captures the NH<sub>3</sub> of the liquid phase. The recirculation  $330$  gas - rich in NH<sub>3</sub> - was sent over a NH<sub>3</sub> absorber containing 0.4 t HNO<sub>3</sub> diluted in water to produce 331 0.7 t AN solution (15% of N) in configuration 3a or 0.2 t  $H_2SO_4$  diluted in water to produce 0.8 t 332 d<sup>-1</sup> of AS solution (7% of N) in configuration 3b/4. Between 32 and 36% of TN contained in the 333 SS influent was recovered as fertilizer suspension depending on the counter acid used, 334 corresponding to an NH<sub>4</sub>–N recovery efficiency of 56 and 57 %. The N-poor stripped LF is then 335 mixed with 69 t d<sup>-1</sup> non-stripped LF and biologically treated through an NDN system. The NDN 336 unit was fed with 508 kg N d<sup>-1</sup> of TN, of which 59 % was present in the form of NH<sub>4</sub>–N and 41 % 337 in the form of Org-N, and achieved a N removal efficiency of 92% resulting in an effluent 338 containing 43 kg N d<sup>-1</sup> and 31 kg P d<sup>-1</sup>. The NDN effluent proceeded to the settling tank where we  $\frac{339}{100}$  recovered 4 m<sup>-1</sup> d<sup>-1</sup> of activated sludge from the effluent containing 16 kg N d<sup>-1</sup> and 3 kg P d<sup>-1</sup>. In 340 configuration 3a/b, the effluent was applied on land, whereas the effluent was further polished in 341 configuration 4. In the polishing step, P precipitation is induced in the effluent by adding  $4.5 \text{ H}_2\text{SO}_4$ 342 (55%) and 3.5 l of FeCl<sub>3</sub> (40%) per tonne effluent before being fed to the CW. P was mostly

recovered in the P sludge (26 kg P d<sup>-1</sup>), while TN, mainly in the form of NH<sub>4</sub>–N and NO<sub>3</sub>-N, was 344 associated with effluent proceeding to the CW (25 kg N  $d^{-1}$ ). The CW further removed N, P, BOD 345 and COD by plant uptake, microbiological degradation, and sedimentation, resulting in the effluent 346 meeting surface water discharge criteria (15 mg  $l^{-1}$  for N, 1 mg  $l^{-1}$  for P, for 250 mg  $l^{-1}$  COD, 25 347 mg  $l^{-1}$  for BOD and 35 mg  $l^{-1}$  for SS) (VCM, 2021).

- 348
- 349 3.1.1. NH<sup>3</sup> SS unit



Figure 4 Mass balance for the SS unit using different counter acids; a) sulphuric acid b) nitric acid

The physicochemical characteristics of the influent, effluent, AS and AN can be found in Table 3. Caustic soda or lime were not added during the SS process of LF manure, however, there was an observed increase in pH from 8.1-8.2 to 8.5-8.6 through partially venting the recirculation air which 353 induces  $CO_2$  stripping (Palakodeti et al., 2022). This increase can be attributed to the transfer of the carbonate buffer from LF into the ventilation flow, which occurs at a higher rate than NH<sub>3</sub> due 355 to the volatile nature of  $CO_2$ . Vingerhoets et al. (2023a) found that performing  $CO_2$  stripping to increase the pH and thus the NH3/NH4 ratio in the LF manure during the SS process is economically 357 more desirable than adding caustic acid. When  $H_2SO_4$  was used as scrubbing agent, slightly higher NH<sub>4</sub><sup>+</sup> removal efficiencies could be achieved as compared to HNO<sub>3</sub>. Therefore, the decrease in NH<sub>4</sub><sup>+</sup> content and NH<sub>4</sub>-N:TN ratio was less significant for the configuration with HNO<sub>3</sub> as 360 scrubbing agent. Because of the lower density of  $HNO<sub>3</sub>$  compared to  $H<sub>2</sub>SO<sub>4</sub>$ , the rate of ventilation flow had to be reduced to prevent HNO3 from being carried with the ventilation air from the

scrubbing unit to the stripping unit which would acidify the LF of manure and reduce stripping efficiency. However, the reduced air flow rate hampers the stripping efficiency because it has a strong impact on mass transfer coefficient, mixing, and gas-liquid interfacial area, which are important parameters determining gas-liquid transfer rate of NH3. For example, Liu et al. (2015) 366 showed that increasing the air flow rate from 60 m<sup>3</sup> h<sup>-1</sup> m<sup>-3</sup> to 840 m<sup>3</sup> h<sup>-1</sup> m<sup>-3</sup> enhances the hourly stripping efficiency from 8.6 to 86.4% when performing a stripping experiment on pig urine at a temperature of 50°C and an increased pH of 10. However, further increasing the air flow rate above 369 600 m<sup>3</sup> h<sup>-1</sup> m<sup>3</sup> showed reduced benefits on NH<sub>3</sub> removal rates.

370

371 Table 3 Composition (mean  $\pm$  standard deviation) of influent NH<sub>3</sub> SS (IS), effluent NH<sub>3</sub> SS (ES), ammonium nitrate (AN) solution, and ammonium sulphate (AS) solution for configuration 3a and 3b solution, and ammonium sulphate (AS) solution for configuration 3a and 3b

	Unit	Configuration 3a			Configuration 3b		
		<b>IS</b>	<b>ES</b>	AN	<b>IS</b>	ES	AS
pH		$8.2 \pm 0.1$	$8.6 \pm 0.1$	$6.0 \pm 0.6$	$8.1 \pm 0.2$	$8.5 \pm 0.1$	$5.7 \pm 1.1$
EC	$\text{mS cm}^{-1}$	$33.2 \pm 1.5$	$31.1 \pm 1.3$	$178 \pm 13$	$31.3 \pm 2.3$	$30.1 \pm 2.3$	$242 \pm 27$
<b>TOC</b>	$g kg^{-1}$	n.a.	n.a.	$0.12 \pm 0.03$	n.a.	n.a.	$0.82 \pm 0.11$
<b>COD</b>	$g kg^{-1}$	$48.6 \pm 6.1$	$44.8 \pm 4.7$	$0.21 \pm 0.04$	$55.4 \pm 2.8$	$52.1 \pm 2.2$	$0.32 \pm 0.09$
TN	$g kg^{-1}$	$5.91 \pm 0.26$	$3.97 \pm 0.45$	$153.1 \pm 26.4$	$5.82 \pm 0.4$	$3.73 \pm 0.51$	$74.4 \pm 8.4$
$NH_4-N$	$g kg^{-1}$	$3.34 \pm 0.35$	$1.46 \pm 0.21$	$76.2 \pm 18.1$	$3.73 \pm 0.3$	$1.61 \pm 0.38$	$74.3 \pm 8.1$
$NO3-N$	$g kg^{-1}$	n.a.	n.a.	$77.4 \pm 15.0$	n.a.	n.a.	n.a.
P	$g \text{ kg}^{-1}$	$0.54 \pm 0.09$	$0.52 \pm 0.06$	$0.06 \pm 0.01$	$0.31 \pm 0.11$	$0.32 \pm 0.09$	$0.05 \pm 0.01$
Cu	$mg \, kg^{-1}$	n.a.	n.a.	$1.2 \pm 0.7$	n.a.	n.a.	$2.2 \pm 0.9$
Zn	$mg \, kg^{-1}$	n.a.	n.a.	$3.4 \pm 1.3$	n.a.	n.a.	$5.5 \pm 2.4$

373

374 Figure 4 shows the mass balance of N in the  $NH<sub>3</sub>$  SS unit using 60% HNO<sub>3</sub> solution and 98% H<sub>2</sub>SO<sub>4</sub> 375 as an absorption agent. When treating 1 t of LF, 24.5 kg of AN (15.3  $\%$  N) is produced wherein 376 1.9 kg of NH4-N is recovered which only amount for half of the TN content in the produced AN 377 solution as the addition of 14.3 kg of  $60\%$  HNO<sub>3</sub> to scrubbing solution contributes the remaining 378 50% in the form of NO<sub>3</sub>-N. The treatment of 1 t LF with scrubbing solution containing 7.1 kg of  $379$  H<sub>2</sub>SO<sub>4</sub> resulted in the production 28.4 kg of 7.4 % N AS solution, which contains 2.1 kg of 380 recovered NH<sub>4</sub><sup>+</sup>-N. The use of HNO<sub>3</sub> instead of H<sub>2</sub>SO<sub>4</sub> results in a slightly lower recovery ratio of 381 TN from the influent in the form of AS and AN solution. However, the organic N content remained 382 the same as only mineral N is recovered during the SS process. It was found that higher N 383 concentrations could be achieved in the recovered  $NH_4NO_3$  solution as compared to the  $(NH_4)_2SO_4$ 

- solution because of the low solubility of (NH4)2SO4 resulting in crystallisation. A higher N concentration is more favourable as it has agronomical benefits and reduces costs of transportation and land spreading (Sigurnjak et al., 2019).
- A considerable number of experiments to assess the efficacy of the implementation of NH<sup>3</sup> SS technology to recover N from (digested) LF of pig manure has been performed at laboratory, pilot and full-scale over the last decades. Brienza et al. (2023) recovered on average 22% of N in the 390 form of AN (81 g  $kg^{-1}$  TN) when stripping LF of digestate at ambient temperature and low pH (8.0) in a pilot-scale installation for 2 hours. A range of temperatures (55-65°C) were tested in series of trials by Pintucci et al. (2017) for a low pH (7.8) and found NH3 removal rates ranging 393 from 28 to 46%. Baldi et al. (2018) achieved an high removal efficiency of  $62\%$  NH<sub>4</sub><sup>+</sup> in a stripping experiment on digestate by subjecting it to a temperature of 48°C and pH of 9.5 over a period of 2 hours, while Bolzonella et al. (2018) recovered 22% of TN in the form of (NH4)2SO4 solution (26 396 g kg<sup>-1</sup> TN) when stripping a mixture of digested swine and cow manure at pilot scale. As compared 397 to other studies, the efficiency found in our study  $(32 - 36\%$  of TN and  $56 - 67\%$  of NH<sub>4</sub><sup>+</sup>-N) was 398 at the higher end of the spectrum as it was performed at both high pH  $(8.5 - 8.6)$  and temperature 399 (50 °C). Also, high N concentrations in the NH<sub>4</sub><sup>+</sup> solutions were achieved in our study compared to literature results.

Assuming the quality criteria for manure derived RENURE products and liquid inorganic macronutrient fertilisers set by European FPR are met, the net costs of the proposed N recovery pathway were calculated by considering the sale of the produced AS and AN solution at current N 404 fertilisation prices. We found that the concentration of TOC was below the prescribed 1% while complying with the minimum required TN content, as shown in Table 4. Moreover, the AS and AN solutions satisfied the maximal TOC:TN and the mineral N:TN ratio requirements. The 407 concentration of hazardous elements Cu and Zn in the  $NH_4^+$  solutions also remained well below the RENURE and FPR requirements. Therefore, it can be concluded that the AS and AN solution obtained during the SS process can be regarded as a substitute for mineral N, which confirms the pricing assumption.

Table 4. Composition requirements for the different fertilisers products defined by the Fertilising Product

Regulation (EU) 1009/2019 and Joint Research Centre (JRC) RENURE products (Huygens et al., 2020).



415 Despite the NH $_4$ <sup>+</sup> solutions recovered less than 1% of the COD embedded in the influent LF, a 416 reduction in of 8% in the COD content of LF was found during the SS process. This trend was also 417 described by Brienza et al. (2023) who found a loss in COD content of 13% when HNO<sub>3</sub> was used 418 as absorbent, and [Bonmatı and Flotats \(2003\)](https://www.sciencedirect.com/science/article/pii/S0925857423000289#bb0030) who recorded COD losses above 5% when using 419 H<sub>2</sub>SO<sub>4</sub> as scrubber acid. Therefore, it can be assumed that  $HNO<sub>3</sub>$  nor  $H<sub>2</sub>SO<sub>4</sub>$  are not able to absorb 420 volatile organics which stress the need for additional air treatment to reduce impact on the 421 environment.

422

### 423 3.1.2. NDN system and settling tank

424 Figure SM D shows the long-term measurement data of the NDN tank, while Figure 5 presents the 425 average physicochemical characteristics of the effluent for the different configurations. The N 426 removal efficiency in the NDN tank was on average 91% of N for configuration 1, 94% of N for 427 configuration 2 and 92% of N for configuration 3. Despite the similar N removal efficiencies 428 observed for all configurations, the N effluent concentration varied depending on the configuration. 429 The NH<sub>4</sub><sup>+</sup> concentration in the NDN tank during configurations 2 and 3 were considerably lower 430 as compared to configuration 1. For all configurations,  $NH<sub>4</sub><sup>+</sup>$  concentrations in the effluent 431 increased during the months August-February as the land spreading limitation increased the 432 loading rate during these months. The high loading rates are causing elevated  $NH<sub>4</sub>$ <sup>+</sup> concentration 433 which eventually led to the free ammonia (FA) toxicity effects inhibiting  $NH_3$  and  $NO_2^-$  oxidation 434 (Nehmtow et al., 2016; Tian et al., 2013). The inhibition of this process causes the accumulation 435 of NO<sub>2</sub><sup>-</sup> during these months. When NH<sub>4</sub><sup>+</sup> concentration exceeds 160 mg/l, the negative feedback 436 loop was triggered reducing the LF input into the NDN tank which enables the process to restore 437 its concentration of NH<sub>4</sub><sup>+</sup> into healthy ranges. This negative feedback loop limits the process 438 capacity of configuration 1 to 76 m<sup>3</sup> LF d<sup>-1</sup> m<sup>-3</sup> (0.135 kg N m<sup>3</sup>d<sup>-1</sup>), while a capacity of 86 m<sup>3</sup> LF 439 d<sup>-1</sup> m<sup>-3</sup> (0.156 kg N m<sup>3</sup>d<sup>-1</sup>) could be reached after the implementation of PO aeration. Despite

increasing the N loading rates in configuration 2, the number of occasions where  $NO<sub>2</sub>$ 440 441 accumulation occurred diminished compared to configuration 1 (9 vs 20 per year) indicating on an 442 elevated nitrification rate in the NDN tank through the usage of PO. A similar N load was treated 443 in configuration 3 (0.157 kg N m<sup>3</sup>d<sup>-1</sup>) compared to configuration 2 (0.156 kg N m<sup>3</sup>d<sup>-1</sup>), but an 444 increased quantity (96 m<sup>3</sup> LF d<sup>-1</sup> m<sup>-3</sup>) could be treated because of the reduced N content in the 445 stripped LF.



446

447 Figure 5 Average concentration of N, P and COD compounds in the NDN tank during the monitoring configurations of the different configurations. configurations.

 Corbala Robles et al. (2018) reported on a typical wastewater treatment plant treating the LF of pig 450 manure after centrifugation in Flanders with a tank volume of  $2846 \text{ m}^3$  which could process on 451 average 58 m<sup>3</sup> LF manure per day with an associated N loading rate of 0.09 kg N m<sup>3</sup> d<sup>-1</sup>. Smet et al. (2003) evaluated 14 different biological manure treatment installations according to the Trevi 453 concept with a total treatment capacity of  $300,000$  m<sup>3</sup> y<sup>-1</sup> and found that the operation of the 454 biological treatment installations allows a loading rate of 0.13 kg N  $m^3 d^{-1}$ . The average loading 455 rate sustained by the NDN tank during configuration 1 in our study equalled 0.14 kg N  $m^3d^{-1}$  and is thus similar as the values found in literature. A review of 25 studies by Skouteris et al. (2020) showed that the replacement of ambient air by PO as aeration agent could enhance the treatment capacity by increasing the oxygen transfer rate through an elevated partial pressure of oxygen, especially for high strength wastewaters with high Mixed Liquor Suspended Solids (MLSS). Similar conclusions were yielded by Rodríguez et al. (2012) who compared the use of PO and air on the nitrification rate in a pilot-scale MBR system used for wastewater treatment and found that

462 aeration by PO could enhance the nitrification rate with 8-13%. As the nitrification rate increases, 463 it effectively diminishes the  $NH_4^+$  concentration within the NDN tank, preventing the accumulation of NH<sup>4</sup> + 464 and thereby mitigating the inhibitory effects of FA. This inhibition generally impedes the 465 entirety of the nitrification process when FA concentrations exceed 150 mg N  $1<sup>-1</sup>$  (Elawwad, A. 466 2018). Conversely, the inhibition of  $NO<sub>2</sub>$  oxidation commences at FA levels surpassing 2.8 mg N  $1<sup>-1</sup>$  (Jubany et al. 2008). Hawkins et al. (2010) conducted a comprehensive review of 15 studies, 468 revealing that the FA concentration threshold at which  $NO<sub>2</sub>$  oxidation begins to be inhibited varies 469 significantly across studies, yet predominantly initiates at FA concentrations of 2-3 mg N  $1^{-1}$ . This 470 finding is consistent with observations of approximately 160 mg  $1^{\text{-}1}$  of total NH<sub>4</sub>-N present in the 471 NDN tank under standard operational conditions.

472 These findings support the increased processing capacity found for configuration 2. When the LF 473 was partially pretreated by NH<sup>3</sup> SS, an increased volume of LF could be processed by the NDN 474 tank. However, only a slight difference was found in processed N load between configuration 2 475 and 3, which could be due to the more favorable COD:N ratio (Phanwilai et al., 2020). Lower COD 476 concentrations were found for configurations 1 and 2 as compared to configuration 3 due to the 477 higher COD loading rates of configuration 3. In addition to augmenting the treatment capacity of 478 the NDN system, N recovery by SS in configuration 3 has the potential to decrease N<sub>2</sub>O emissions 479 per m<sup>3</sup> of LF treated. Specifically, the N<sub>2</sub>O emissions are estimated to be 0.057 kg N<sub>2</sub>O-N in 480 configuration 3, compared to 0.063 kg N<sub>2</sub>O-N per m<sup>3</sup> of LF treated in configuration 1, assuming 481 N2O emissions account for 1.1% of the N load in the NDN (de Haas and Andrews, 2022), The 482 conventional N2O emission factors are applicable on concentrated piggery wastewaters (Ravi et 483 al., 2023). However, the reduction in  $NO<sub>2</sub>$  accumulation in configuration 3 is expected to 484 contribute to an even more pronounced decrease in  $N_2O$  emissions, given the strong association 485 between  $NO_2^-$  accumulation and  $N_2O$  emission rates (Van Hulle et al., 2011). The NDN effluent 486 was further processed by a clarifier, which resulted in the production of  $3-6$  t d<sup>-1</sup> activated sludge, 487 containing 9-16 kg N d<sup>-1</sup> and 2-6 kg N d<sup>-1</sup>, and 73-94 t d<sup>-1</sup> clarified effluent, containing 23-26 kg 488 N d<sup>-1</sup> and 6-28 kg P d<sup>-1</sup>. As the settled activated sludge contains a considerable amount of water, it 489 is looped to the centrifuge for further processing.

490

491 3.1.3. P precipitation and wetland

492 Table 5 presents the average influent and effluent concentrations of the P precipitation unit and 493 effluent of the wetland. The results illustrate that the P precipitation unit achieves a high P recovery 494 efficiency of 91.5 % by inducing P precipitation as FePO<sub>4</sub> though the addition of 4.5 H<sub>2</sub>SO<sub>4</sub> (55%) 495 and 3.5 l of FeCl3 (40%). In the sedimentation tank, also the DM content was partly recovered, 19 496 and 7 % respectively. On average 0.01 t of sludge, recovering almost 10% of N and 92% of P, and 497 0.99 t P poor effluent, containing 0.26 kg of N, 0.03 kg of P and 10.9 kg of COD, were produced 498 during the treatment of 1 tonne effluent. Meers et al. (2006) achieved a removal efficiency of 39, 499 88 and 95% of P when applying 1, 3 and 5 l of FeCl<sub>3</sub> to LF of pig manure treated by NDN with a 500 P content of 332 mg P l<sup>-1</sup>. Decreasing pH below 8 by the addition of  $H_2SO_4$  increases the P removal 501 rates for similar FeCl<sub>3</sub> dosage. Through the acidification of the NDN effluent in our study, high P 502 removal efficiencies could be achieved for low FeCl<sub>3</sub> dosages which is crucial to meet the stringent 503 local Cl discharge limits of  $1000 \text{ mg Cl}^{-1}$  (VCM, 2021).

504

505 Table 5 Composition (mean  $\pm$  standard deviation) of influent P precipitation unit (I-PU), effluent P precipitation unit (E-PU), and effluent constructed wetland (E-CW) effluent constructed wetland (E-CW)

		I-PU	E-PU	E-CW
pH		$8.1 \pm 0.1$	$7.9 \pm 0.2$	$7.9 \pm 0.1$
EC	$\text{mS cm}^{-1}$	$33 \pm 1.5$	$31 \pm 1.3$	$9.8 \pm 1.8$
DM	$g kg^{-1}$	$14 \pm 1.7$	$11 \pm 0.8$	$0.12 \pm 0.03$
SS	$g kg^{-1}$	$0.56 \pm 0.14$	$0.52 \pm 0.21$	$0.01 \pm 0.00$
<b>COD</b>	$g kg^{-1}$	$12 \pm 2.2$	$8.7 \pm 1.8$	$0.12 \pm 0.07$
<b>BOD</b>	$g kg^{-1}$	$5.9 \pm 1.7$	$5.8 \pm 2.1$	$0.01 \pm 0.00$
TN	$mg \, kg^{-1}$	$283 \pm 153$	$258 \pm 45$	$13 \pm 5.9$
$NH_4-N$	$mg \, kg^{-1}$	$81 \pm 100$	$75 \pm 33$	$1.5 \pm 0.2$
$NO3-N$	$mg \, kg^{-1}$	$401 \pm 70$	$37 \pm 10$	$5.7 \pm 1.4$
$\mathbf P$	$mg \, kg^{-1}$	$304 \pm 45$	$28 \pm 8.9$	$0.32 \pm 0.09$

507

508 After precipitation, the effluent flows into the CW. The average influent and effluent so oncentrations of the CW can be found in Table 5. As on average 4.6 t  $d^{-1}$  of effluent were treated, 510 the N and P loading rates equalled 0.94 g N m<sup>-2</sup> d<sup>-1</sup> and 0.11 g N m<sup>-2</sup> d<sup>-1</sup>, wherefore a removal 511 efficiency of 95% of N, 99% of P and 99% of COD is obtained. Meers et al. (2006) conducted 512 bench experiment with a corresponding loading rate of 0.23 g N m<sup>-2</sup> d<sup>-1</sup>, 0.27 g P m<sup>-2</sup> d<sup>-1</sup> and 3.2 513 g COD m<sup>-2</sup> d<sup>-1</sup> and achieved removal efficiencies between 73%–83% for N, 71% - 98% for P and 514 64 – 75% for COD, while Meers et al. (2008) reported on a CW of 4500 ha that could sustain a

515 nutrient load of 0.75-1.22 g N m<sup>-2</sup> d<sup>-1</sup> and 0.04 g P m<sup>-2</sup> d<sup>-1</sup> with a removal efficiency of 96% and 516 99%, respectively. Lee et al. (2014) loaded 0.84 g N m<sup>-2</sup> d<sup>-1</sup> into a CW (4492 ha) in the form of piggery effluent and achieved a removal efficiency of 55% N. The plant uptake was estimated by 518 the crop cut method (Sapkota et al. 2016) to account for 17 % of N and 26% of P removal, which is considerably lower than the values found by Meers et al. (2006), Meers et al. (2008) and Lee et al. (2014) as the biomass yield remained under expectations.

- However, COD, BOD and N were mainly removed by microbiological degradation and 522 denitrification processes resulting in C and N losses to the atmosphere as  $CO<sub>2</sub>$  and N<sub>2</sub>, while the remainder of the P removal is attributed to sorption by substrate and sedimentation. Due to the finite capacity of these P removal mechanisms, sorption and sedimentation will not sustain long-term P removal (Meers et al., 2008). Therefore, it is crucial to establish a correlation between the loading rates of P and its uptake by the biomass that can be harvested to maintain the long-term effectiveness of a system in removing P. By reducing the P load through the introduction of a P precipitation before the effluent enters the CW, an elevated amount of effluent can be treated without compromising the sustainable operation of the CW.
- The regional discharge criteria limit the effluent concentration to 15 mg  $l^{-1}$  for N, 1 mg  $l^{-1}$  for P, 531 for 250 mg l<sup>-1</sup> COD, 25 mg l<sup>-1</sup> for BOD and 35 mg l<sup>-1</sup> for SS (VCM, 2021). Figure SM E illustrates the effluent concentration of the tertiary treatment system over time. The CW effectively removed P, COD, BOD and SS during the entire year as the effluent concentration were continuously below discharge limits, while N was above the discharge limits during the winter period. Therefore, the results of this study indicated that N is the limiting component determining the CW operational capacity while being in correspondence with the legal discharge limits.

3.2. Energy balance

The energy balance was computed for each configuration based on the total consumed energy and amount treated during the period the respective configuration was tested. Table 6 shows the average consumed energy per tonne of raw manure treated. The electricity required by the manure 541 processing ranged between 14.8-16.4 kWh  $t^{-1}$  raw manure depending on the used configuration. For conventional processing (configuration 1), NDN treatment was the most energy-intensive step, followed by the centrifugation. Replacing air by PO in the oxygenation step resulted in a significant reduction of energy requirement for the NDN process. When installing a SS unit before the NDN treatment, the costs by oxidation requirement even decreased further until 11.4 kWh  $t^{-1}$  raw manure.





566 Table 6 Energy consumption (kWh  $t^{-1}$  raw manure) per process technology unit

567

568 3.3. Economic assessment of intensified manure processing

569 The different configurations were economically evaluated to find the effect of intensification of the 570 processing configuration on the total costs associated with raw manure treatment. The cost benefit 571 analysis for the different configurations is illustrated in Figure 6. CAPEX compromised capital 572 costs, including technology cost and land usage, for centrifuge, NDN tank and settling tank in all 573 configurations, while amortisation costs associated with the installation of a pure liquid oxygen 574 oxygenation tank,  $NH<sub>3</sub>$  SS unit and CW were considered depending on the configuration. The 575 amortisation costs associated with the centrifuge unit varied between 0.7 and 1.0  $\epsilon$ , while the 576 amortisation costs of the NDN treatment including the settling tank, depending on the treatment 577 capacity of the configuration, amounted between 3.6 - 4.7  $\epsilon t^{-1}$  with an additional cost of 0.09 - 0.1 578  $\epsilon$  t<sup>-1</sup> for the PO tank. Implementing the NH<sub>3</sub> SS unit involved a cost of  $\epsilon$  0.23 M which resulted in 579 a cost of 1.8  $\epsilon t^{-1}$  LF manure or 0.7  $\epsilon t^{-1}$  raw manure treated. Considering the found treatment 580 capacity of the CW, the tertiary treatment via P precipitation and CW amounted for 2.2  $\epsilon t^{-1}$ LF or 581 1.8  $\epsilon$  t<sup>-1</sup> raw manure.

582



584 Figure 6 Economic results for the different configurations, calculated for the treatment of 1 tonne of raw manure. (C= Centrifuge, 585 S= SS unit, NDN= Nitrification denitrification unit, A= Aeration tank and CW= Const 585 S= SS unit, NDN= Nitrification denitrification unit, A= Aeration tank and CW= Constructed wetland) (C1=Configuration 1, C2<br>586 = Configuration 2, C3a=Configuration 3a, C3b=Configuration 3b, C4=Configuration 4)  $=$ Configuration 2, C3a=Configuration 3a, C3b=Configuration 3b, C4=Configuration 4)

The OPEX comprised electrical energy requirements, chemicals, disposal, insurance, maintenance, and labour costs. The insurance costs for the separation, NDN and settling unit decreased per tonne raw manure as the capacity of these units increased while these costs remained the same. However, the alternative configuration implicated additional costs associated with insurance, maintenance, and labour for the PO tank, NH3 SS unit, P precipitation unit and CW depending on the configuration. Because of the lower aeration requirements of configuration 2 compared to 593 configuration 1, a lower cost for energy consumption was obtained. The incorporation of the NH<sub>3</sub>

SS step in configuration 3 resulted in a further decrease in energy consumption during NDN processing, however, this was not enough to compensate additional energy requirements of the NH<sup>3</sup> SS unit itself resulting in an increased cost for energy in configuration 3. To recover N in 597 configuration 3a and 3b, 6 kg of HNO<sub>3</sub> and 2.3 kg of H<sub>2</sub>SO<sub>4</sub> was used amounting in a cost of 1.1 598 and  $0.2 \text{ } \in \text{ } t^1$ , respectively. The N recovery via NH<sub>3</sub> SS unit provided an economic benefit of 0.8 – 599 1.6  $\epsilon t^{-1}$  depending on the choice of counter acid as it can be sold as replacement for synthetic mineral N fertilisers, while the implementation of a CW saves farmers the costs associated with effluent disposal.

602 The total costs for pig manure treatment ranged between 18.9 - 20.7  $\epsilon t^{-1}$ . The gradual implementation of additional treatment units could reduce the cost; PO oxygenation led to a cost 604 reduction of almost 4%, the implementation of a  $NH<sub>3</sub>$  SS could further reduce the costs with 1 to 2% depending on counter acid used, while the expansion of the treatment line with a CW accounted for the lowest total cost (-9% compared to reference situation of using configuration 1).

 Spiller et al. (2022) reported in review study on conflicting results regarding the economic feasibility of a direct N recovery via technologies such as NH<sup>3</sup> SS and membrane filtration. This is because the traditional "indirect" reuse loop, which involves reactive N removal by NDN in waste 610 streams as  $N_2$  and refixation via the Haber-Bosch process, appears to be the more financially viable in many cases. Derden and Dijkmans (2020) estimated the overall treatment costs involving 612 separation and NDN for raw manure at 19 € t<sup>-1</sup>, while the implementation of N recovery via NH<sub>3</sub> 613 SS is expected to result in a cost increment of  $10 \in \mathfrak{t}^{-1}$ . A similar conclusion was made by De Vrieze et al. (2019) who found a cost increase of 51% when involving NH3 SS as compared to the conventional treatment configuration. However, both studies did not consider the synergetic effects between NH3 SS and NDN processing such as capacity increase and reduced oxygen requirement. 

The Monte Carlo simulations demonstrated a different impact of uncertainty in resource pricing for the different configurations (Figure 7). For configuration 1 and 2, an increase of 42 and 43% in costs was observed at 80% probability as these configurations have a lower dependence on the external resources (including energy, counter acid, with varying prices and uncertain revenue from 622 the produced NH<sub>4</sub><sup>+</sup> salts). Due to the strong dependency of configuration 3a-c on resource pricing, the increase in overall costs at 80% probability was more pronounced (45-46%). The uncertainty 624 in the market price of counter acid and market value of NH<sub>4</sub><sup>+</sup> salt solution mainly impacted the

total net cost of configurations 3. The impact of the NHO3 price and market value of AN solution 626 was stronger compared to the impact of  $H_2SO_4$  and AS pricing, due to the higher acid consumption, 627 revenue, and price range. The additional consumption of  $H_2SO_4$  and  $FeCl_3$  in configuration 4 compared to configuration 3b was reflected in the fact that the elevation of total net cost was higher at 80% probability (+ 0.7 %). Although the higher electricity consumption depicted in configuration 3 had a noticeable impact on the overall net cost, its effect was significantly less than that of the market price of chemicals. The sensitivity analysis of configuration 3a-c illustrates that process intensification can be highly economically beneficial compared to the conventional 633 situation when high market values for the recovered NH<sub>4</sub><sup>+</sup> salts prevail, but can also turn out more expensive when resource costs for the use of chemical and energy are high and the market values of the produced products are not favourable.



637 Figure 7 Monte Carlo probability distributions for the sensitivity analysis for the different manure processing configuration 638 (C1=Configuration 1, C2 = Configuration 2, C3a=Configuration 3a, C3b=Configuration 3b, C (C1=Configuration 1, C2 =Configuration 2, C3a=Configuration 3a, C3b=Configuration 3b, C4=Configuration 4)

#### **4. CONCLUSION**

This study presents a comparative techno-economic assessment of different configurations studying the effect of the implementation of process intensifying measures along the swine manure 643 treatment line, including PO aeration, pretreatment by NH<sub>3</sub> SS and tertiary treatment by a CW. The techno-economic assessment was performed based steady-state mass balances derived from long-term monitoring campaigns at a full-scale pig manure treatment facility focusing on total mass, N and P. Inserting additional treatment steps to the treatment train was found to have a beneficial effect on the overall treatment costs as it enhances the processing capacity of the conventional configuration, reduces costs of by-product disposal and generates addition revenue by producing

- high quality fertilising products. The usage of PO oxygenation reduces the total with almost 4%,
- the addition of a SS unit further reduces the costs with 1 to 2% depending on counter acid used and
- the implementation of a CW even reduces the total costs further with 4%.

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