

pubs.acs.org/jnp

Article

# Antiviral Rotenoids and Isoflavones Isolated from *Millettia oblata* ssp. *teitensis*

Ivan Kiganda, Jonathan Bogaerts, Lianne H. E. Wieske, Tsegaye Deyou, Yoseph Atilaw, Colores Uwamariya, Masum Miah, Joanna Said, Albert Ndakala, Hoseah M. Akala, Wouter Herrebout, Edward Trybala, Tomas Bergström,\* Abiy Yenesew,\* and Mate Erdelyi\*



This is the first report on the use of VCD to assign the absolute configuration of rotenoids. The crude leaves and root wood extracts displayed anti-RSV (human respiratory syncytial virus) activity with IC<sub>50</sub> values of 0.7 and 3.4  $\mu$ g/mL, respectively. Compounds **6**, **8**, **10**, **11**, and **14** showed anti-RSV activity with IC<sub>50</sub> values of 0.4–10  $\mu$ M, while compound **3** exhibited anti-HRV-2 (human rhinovirus 2) activity with an IC<sub>50</sub> of 4.2  $\mu$ M. Most of the compounds showed low cytotoxicity for laryngeal carcinoma (HEp-2) cells; however compounds **3**, **11**, and **14** exhibited low cytotoxicity also in primary lung fibroblasts. This is the first report on rotenoids showing antiviral activity against RSV and HRV viruses.

n contrast to the large variety of antimicrobials accessible lacksquare for the treatment of bacterial and fungal infections, antiviral drugs are so far only available for the treatment of ten human viruses, namely, the human immunodeficiency virus (HIV), the hepatitis B and C viruses, three herpesviruses, the respiratory syncytial virus (RSV), the influenza viruses, the human papillomavirus,<sup>1,2</sup> and recently for SARS coronavirus-2. The epidemiological control of most viral infections focuses on the isolation of cases, quarantine of contacts, personal protection, and mass vaccination since specific treatments are not available against most viral infections.<sup>3,4</sup> A variety of viral diseases have evolved to endemics and pandemics and lead to significantly increased mortality rates and social, economic, and political changes.<sup>1,5,6</sup> Antiviral vaccines and new drugs are thus required. Natural products have been sources of inspiration for the development of new antimicrobial drugs.<sup>7</sup> Some flavonoids<sup>8</sup> and specifically flavones,<sup>9</sup> for instance, exhibit antiviral properties by inhibiting enzymes and preventing virus penetration.<sup>10,11</sup>

and VCD data with those calculated for the possible stereoisomers.

Out of the over 260 *Millettia* species (Fabaceae/ Leguminosae) distributed over the tropics of Asia, Australia, and Africa, 139 are found in Africa.<sup>12–14</sup> Traditionally, some of these species have been applied in the treatment of diseases including paralysis, insect bites, snake bites, and dysmenorrhea.<sup>15,16</sup> The genus *Millettia* is known to be a rich source of various subclasses of flavonoids, especially isoflavones, chalcones, and rotenoids.<sup>17-19</sup> Some of these compounds presented cytotoxic, antiplasmodial, anti-inflammatory, antileishmanial, and antibacterial activities.<sup>12,15,20</sup> Isoflavones specifically affect various stages of the viral life cycle by targeting cellular components of importance for viral replication and, accordingly, have an effect on many viruses.<sup>21</sup> The isolation and biological applications of some rotenoids have recently been reviewed.<sup>22</sup> We previously isolated rotenoids and isoflavones from the leaves and stem bark of Millettia oblata ssp. teitensis.<sup>18,19</sup> Here, we report the investigation of the secondary metabolites of root wood and the reinvestigation of its leaves. We have characterized three new rotenoids from the leaves (1-3), a new isoflavone glycoside (4) from the root wood, and 14 additional known compounds from this plant. The antiviral activity against RSV and human rhinovirus 2 (HRV-2) and the cytotoxicity of the crude extracts and a selection of the isolated compounds were

Received:December 30, 2023Revised:March 24, 2024Accepted:March 24, 2024Published:April 5, 2024



Downloaded via UNIV OF ANTWERPEN on April 29, 2024 at 08:44:02 (UTC). See https://pubs.acs.org/sharingguidelines for options on how to legitimately share published articles.

1003

explored in human laryngeal epidermoid carcinoma (HEp-2) and cervical cancer (HeLa) cells and in human embryonic lung fibroblasts.

# RESULTS AND DISCUSSION

The  $CH_2Cl_2/MeOH$  (1:1) extract of the leaves of M. oblata ssp. teitensis was subjected to a silica gel column chromatography, followed by purification by Sephadex LH-20 and preparative thin layer chromatography (TLC) to afford the three new rotenoids oblarotenoid E(1), oblarotenoid F(2), and oblarotenoid G (3). Similar treatment of the root wood afforded the new isoflavone glycoside obloneside (4), the six known rotenoids oblarotenoid C (5)<sup>18</sup> oblarotenoid A (6)<sup>18</sup> oblarotenoid D (7),<sup>18</sup> 12a-hydroxymunduserone (8),<sup>18</sup> tephrosin (9),<sup>23</sup> and deguelin (10),<sup>17</sup> the seven known isoflavones ichthynone (11),<sup>24</sup> 7,2',5'-trimethoxy-3',4'-methylenedioxyiso-flavone (12),<sup>18</sup> isoerythrin-A-4'-prenyl ether (13),<sup>25</sup> 4'prenyloxyderrone (14),<sup>26</sup> aglycuneatin methyl ether (15),<sup>27</sup> calopogonium isoflavone B (16),<sup>28</sup> maximaisoflavone G (17),<sup>17</sup> and milldurone (18),<sup>27</sup> and the known chalcone isobavachromene (19).<sup>29</sup> The known compounds were identified by comparison of the maximum compounds were identified by comparison of the experimental spectroscopic data to the literature data (Supporting Information). The <sup>13</sup>C NMR assignment of compounds 1-4 was confirmed by CSEARCH.



Compound 1 was obtained as a white, amorphous solid that showed UV absorption maxima at 280 and 305 nm. Its molecular formula was established as  $C_{19}H_{16}O_7$  based on HRESIMS ( $[M + H]^+$  at m/z 357.0976, calcd for  $C_{19}H_{17}O_7$ 357.0974, Figures S1–S8, Supporting Information) and NMR data (Table 1). The <sup>1</sup>H NMR spectrum displayed four sets of

coupled aliphatic protons at  $\delta_{\rm H}$  4.21 (br d, *J* = 12.1 Hz, H-6), 4.65 (dd, J = 12.1, 3.0 Hz, H-6), 4.94 (m, H-6a), and  $\delta_{\rm H}$  3.81 (H-12a, overlapping with a OMe signal), which are consistent with the NMR signals expected for the B ring of a rotenoid skeleton.<sup>31</sup> The <sup>13</sup>C NMR spectrum displayed the corresponding signals at  $\delta_{\rm C}$  values of 66.8 (C-6), 72.7 (C-6a), 189.0 (C-12), and 45.0 (C-12a). Two methoxy groups ( $\delta_{\rm H}$  3.71 and 3.81;  $\delta_{\rm C}$  57.2 and 56.2) and a methylenedioxy group ( $\delta_{\rm H}$  5.91 and  $\delta_{\rm H}$  5.97;  $\delta_{\rm C}$  102.7) were also identified based on their characteristic chemical shifts. The three mutually coupled aromatic protons of ring D showed an AMX spin system at  $\delta_{\rm H}$ 7.85 (d, J = 8.9 Hz, H-11),  $\delta_{\rm H}$  6.59 (dd, J = 8.9, 2.3 Hz, H-10), and  $\delta_{\rm H}$  6.44 (d, J = 2.3 Hz, H-8) (Table 2 and Figures S1–S8, Supporting Information). The placement of the methoxy group at C-9 ( $\delta_{\rm C}$  167.0) was supported by the HMBC cross peaks from H-8 ( $\delta_{\rm H}$  6.44), H-10 ( $\delta_{\rm H}$  6.59), and H-11 ( $\delta_{\rm H}$ 7.85) to C-9 ( $\delta_{\rm C}$  167.0) and from OMe-9 ( $\delta_{\rm H}$  3.85) to C-9. The OMe-9 group further showed a NOE correlation to H-8 ( $\delta_{\rm H}$  6.44). This substitution pattern is consistent with the biogenetically expected oxygenation (OMe) at C-9.32 The remaining aromatic proton ( $\delta_{
m H}$  6.39) was assigned to H-1, based on its HMBC correlations with C-1a ( $\delta_{\rm C}$  109.4), C-4a  $(\delta_{\rm C} 133.2)$ , and C-12a  $(\delta_{\rm C} 45.0)$ . This places the second methoxy group and the methylenedioxy group at C-2, C-3, and C-4, giving rise to two alternative structures that have the methoxy group at C-2 or at C-4. The observed <sup>13</sup>C NMR chemical shift of the methoxy ( $\delta_{\rm C}$  57.2) better fits to oxygenation at C-2, as it would be expected to have a chemical shift above 59 ppm if it was at C-4, due to di-ortho substitution.<sup>33</sup> The placement of the methoxy group at C-2 ( $\delta_{\rm C}$ 138.9) was further supported by the NOE correlation between OMe-2 ( $\delta_{
m H}$  3.73) and H-1 ( $\delta_{
m H}$  6.31) and by the HMBC correlations of H-1 ( $\delta_{\rm H}$  6.31) with C-1a ( $\delta_{\rm C}$  109.4), C-2 ( $\delta_{\rm C}$ 138.9), C-3 ( $\delta_{\rm C}$  136.3), C-4a ( $\delta_{\rm C}$  133.2), and C-12a ( $\delta_{\rm C}$  45.0) as well as between OMe-2 ( $\delta_{
m H}$  3.73) and C-2 ( $\delta_{
m C}$  138.9) (Table 1). Based on the spectroscopic data, the gross structure of compound 1 was identified as 2,9-dimethoxy-3,4methylenedioxyrotenoid. It has an uncommon trisubstitution with a methoxy and a methylenedioxy group at its ring A, which may indicate it to be biosynthesized from 7,2',5'trimethoxy-3',4'-methylenedioxyisoflavone (12), a co-metabolite that has the same oxygenation pattern in its B ring as the A ring of 1. The chemical shift of H-1 ( $\delta_{\rm H}$  6.39) along with the small coupling constant (J = 2.9 Hz) between H-6 ( $\delta_{\rm H}$  4.65 and 4.21) and H-6a  $(\delta_{\rm H}$  4.94) suggested that H-6a is equatorially oriented, and compound 1 hence has a cis configuration at its B/C ring junction, which has been reported as the thermodynamically most favorable configuration.<sup>18</sup> The NOE correlation between H-6a ( $\delta_{\rm H}$  4.94) and H-12a ( $\delta_{\rm H}$  3.81) corroborates the cis configuration of the B/C ring junction. To determine the absolute configuration, the electronic circular dichroism (ECD) spectrum of 1 was recorded (Figure 1). According to Slade et al. (2005), dextrorotatory rotenoids with a *cis*-B/C ring junction, showing a positive Cotton effect (CE) at 300–330 nm ( $\pi$  to  $\pi^*$  transition) and negative CE at 348– 360 nm (for n to  $\pi^*$  transition), are associated with the (6aR,12aR) configuration. On the other hand, levorotatory rotenoids with a *cis*-B/C ring junction, showing a negative CE at 300-330 nm and a positive CE at 348-360 nm, are associated with a (6aS,12aS) configuration. The n to  $\pi^*$ transition at 348-360 nm is generally weak, and for many rotenoids, it has not been observed and thus may not be a reliable indicator for determination of rotenoids' absolute

position $\delta_c$ 1 106		1			2			3	
1 106	, type	$\delta_{\rm H}~(J~{ m in}~{ m Hz})$	HMBC <sup>b</sup>	$\delta_{\mathrm{C}}$ , type	$\delta_{\rm H}$ (J in Hz)	HMBC	$\delta_{ m C}$ , type	$\delta_{\rm H}~(J~{ m in}~{ m Hz})$	HMBC
	.6, CH	6.31 s	C-1a, C-2, C-3, C-4a, C- 12a	105.9, CH	6.54 s	C-1a, C-2, C-3, C-4a, C- 12a	107.1, CH	8.32 s	C-2, C-3, C-4a, C- 12a
la 109	.4, C			109.9, C			111.8, C		
2 138	.9, C			149.5, C			142.9, C		
3 136	.3, C			142.4, C			147.4, C		
4 136	.9, C			99.3, CH	6.46 s	C-1a, C-2, C-3	98.7, CH	6.53 s	C-1a, C-2, C-3
4a 133	.2, C			149.5, C			147.5, C		
6 66	.8, CH <sub>2</sub>	4.65 dd (12.5, 2.6)	C-4a, C-6a, C-12a	64.0, CH <sub>2</sub>	4.58 dd (13.5, 2.6)	C-4a, C-6a, C-12a	65.0, CH <sub>2</sub>	4.96 s	C-4a, C-6a, C-12a
		4.21 br d (12.5)	C-4a,C-6a		4.46 d (13.5, 2.6)	C-4a,C-6a			
6a 72	.7, CH	4.94 dd (2.6)	C-1a, C-4a, C-6, C-12, C- 12a	76.4, CH	4.56 br d (3.6)	C-1a, C-4a, C-6, C-12, C- 12a	156.8, C		
7a 163	.0, C			157.5, C			156.7, C		
8 100	.9, CH	6.44 d (2.3)	C-7a, C-9, C-10, C-11a	100.3, CH	6.40 s	C-7a, C-9, C-10, C-11a	100.3, CH	6.83 d (2.2)	C-7a, C-9, C-10, C- 11a
9 167	.0, C			157.3, C			164.0, C		
10 110	.9, CH	6.59 dd (8.9, 2.3)	C-8, C-9, C-11a	145.3, C		C-8, C-9, C-11a	114.7, CH	7.00 dd (8.9, 2.2)	C-8, C-11a
11 129	.6, CH	7.85 d (8.9)	C-7a, C-9, C-12	106.9, CH	7.25 s	C-7a, C-9, C-12	127.9, CH	8.20 d (8.9)	C-7a, C-9, C-12
11a 113	.1, C			109.5, C			118.7, C		
12 189	.0, CO			191.3, CO			174.2, CO		
12a 45	.0, CH	3.81 <sup>c</sup>	C-1a, C-4, C-6	67.9, C			112.3, C		
OCH <sub>2</sub> O 102	.7, CH <sub>2</sub>	5.97 d (1.4); 5.91 d (1.4)	C-3, C-4	101.5, CH <sub>2</sub>	5.83 d (1.4); 5.86 d (1.4)	C-2, C-3	101.5, CH <sub>2</sub>	5.96 s	C-2, C-3
OMe-2 57	.2, CH <sub>3</sub>	3.73 s	C-2						
OMe-9 56	.2, CH <sub>3</sub>	3.81 s	C-9	56.4, CH <sub>3</sub>	3.86 s	C-9	56.0, CH <sub>3</sub>	3.92 s	C-9
OMe-10				56.5, CH <sub>3</sub>	3.88 s	C-10			

Table 2. NMR Spectroscopic Data (<sup>1</sup>H 800 and <sup>13</sup>C 200 MHz, CD<sub>3</sub>OD) for Obloneside (4)

position	$\delta_{\rm C}$ , type	$\delta_{ m H}$ m (J in Hz)	HMBC
2	154.2, CH	8.17 s	C-1', C-3, C-4, C- 8a
3	126.9, C		
4	177.1, CO		
4a	115.2, C		
5	101.8, CH	7.12 s	C-4, C-4a, C-6, C- 7, C-8a
6	157.0, C		.,
7	142.1, C		
8	153.9, C		
8a	155.5, C		
1'	126.0, C		
2'-	110.8, CH	7.03 d (1.7)	C-3, C-3', C-4', C-6'
3′	148.9, C		
4′	148.9, C		
5'	109.0, C	6.84 d (8.0)	C-1', C-3', C-4'
6'	123.8, CH	6.97 dd (8.0, 1.7)	C-2', C-3, C-4'
OCH <sub>2</sub> O	102.5, CH <sub>2</sub>	5.98 d (1.2)	C-3', C-4'
OMe-7	62.8, OMe	3.92 s	C-7
OMe-8	62.3, OMe	3.91 s	C-8
$\beta$ -glycoside	moiety 1		
1″	101.6, CH	5.05 d (7.9)	C-3", C-5", C-6
2″	73.6, CH	3.78 dd (8.7, 7.9)	C-1", C-4"
3″	76.7, CH	3.70 dd (9.5, 7.9)	C-4", C-5", C-7"
4″	89.4, CH	3.61 dd (9.5, 9.1)	C-1"", C-2", C-3""
5″	70.2, CH	3.46 dd (9.3, 9.1, 7.3)	C-1", C-3", C-4", C-7"
7″	67.8, OCH <sub>2</sub>	4.08 d (9.3) 3.58 (7.3)	C-1"'', C-2", C-4", C-5"
$\beta$ -glycoside	moiety 2		
1‴	105.4, CH	4.53 d (7.8)	C-3", C-4", C-5"
2‴	75.2, CH	3.34 dd (8.9, 7.8)	C-1‴, C-5‴
3‴	76.6, CH	3.55 dd (9.7, 8.9)	C-2‴, C-4‴
4‴	71.7, CH	3.32 dd (9.7, 9.7)	C-3‴, C-5‴, C-7‴
5‴	77.6, CH	3.43 ddd (9.7, 9.2, 7.2)	C-1‴, C-3‴, C-4‴, C-7‴
7‴	68.5, OCH <sub>2</sub>	4.04 d (9.2) 3.58 d (7.2)	C-1‴″, C-2‴″, C- 4‴
α-rhamnosi	de moiety 3		
1‴''	102.1, CH	4.71 d (1.2)	C-2‴', C-3‴', C- 5‴', C-7″
2‴′	71.9, CH	3.90 dd (3.3, 1.2)	C-1 <sup>""'</sup> , C-3 <sup>""'</sup> , C- 4 <sup>""'</sup>
3‴′	72.2, CH	3.70 dd (9.4, 3.3)	C-4‴', C-5‴'
4‴′	74.1, CH	3.37 dd (9.5, 9.4)	C-2‴', C-3‴', C- 5‴'
5‴′′	70.0, CH	3.67 dd (9.5, 6.2)	C-1‴', C-3‴', C- 4‴''
7‴′	18.1, CH <sub>3</sub>	1.28 d (6.2)	C-4‴', C-5‴'
lpha-rhamnosi	de moiety 4		
1‴″	102.4, CH	4.74 d (1.4)	C-2‴″, C-3‴″, C- 5‴″, C-7‴
2‴″	72.0, CH	3.92 dd (3.4, 1.4)	C-1‴″, C-4‴″
3‴″	72.4, CH	3.80 dd (9.4, 3.4)	C-1‴″, C-2‴″, C- 4‴″
4‴″	74.0, CH	3.39 dd (9.5, 9.4)	C-2""", C-3""", C- 5"""
5‴″	69.9, CH	3.63 dd (9.5, 6.2)	C-1‴″, C-3‴″, C- 4‴″
7‴″	18.0, CH <sub>3</sub>	1.21 d (6.2)	C-4‴″, C-5‴″





Figure 1. Experimental ECD spectra of compounds 1 (top) and 2 (bottom).

configuration.<sup>18</sup> Compound 1 is dextrorotatory with a  $[\alpha]$  =  $^{20}_{D}$  +44.0 (c 0.2 CH<sub>3</sub>OH) and showed a negative CE at 330 nm for its  $\pi$  to  $\pi^*$  transition in its ECD spectrum. This contradicts the above literature proposal.<sup>34</sup> A similar contradiction has previously been reported for some rotenoids, which had either low positive or low negative specific rotation.<sup>18</sup> In contrast to prenylated rotenoids with a cis-B/C ring junction that show specific rotations of  $[\alpha]_{D}^{20} < -100$  or  $[\alpha]_{D}^{20} >$ +100, with the sign of the rotation reliably indicating the absolute configuration,  $^{35,36}$  nonprenylated rotenoids with a *cis*-B/C ring junction typically show specific rotations of -50 < $[\alpha]^{20}_{D}$  < +50, and the sign of the rotation is not reliable for determining their absolute configuration.<sup>18</sup> For compound 1, which has a positive specific rotation of  $[\alpha]_{D}^{20}$  +44.0 (c 0.2 CH<sub>3</sub>OH), we relied on the ECD spectrum and assigned its absolute configuration as (6aS,12aS)-1 upon comparison of its ECD spectrum with that reported for (6aS,12aS)-9-methoxy-2,3-methylenedioxyrotenoid, a related rotenoid whose absolute configuration was established by single-crystal X-ray crystallography.<sup>34</sup> The absolute configuration of compound 1 was confirmed by acquiring a vibrational circular dichroism (VCD) spectrum (Figure 2), which showed a good match with that simulated for the (6aS,12aS) stereoisomer of 1, whereas it showed no match for the predicted spectrum of the (6aS,12aR) stereoisomer. The vibrational IR pattern of 1 at 1300-1200 cm<sup>-1</sup> (Figure S135, Supporting Information) was further consistent with a *cis* configuration at the B/C ring junction.<sup>37,38</sup> This is the first report of the use of VCD spectroscopy to determine the relative and absolute configuration of rotenoids. Based on the spectroscopic data, the new compound (1) was identified as (6aS,12aS)-2,9-dimethoxy-3,4-methylenedioxyrotenoid and was given the trivial name oblarotenoid E. It is structurally closely related to the previously reported oblarotenoid  $C_{,1}^{18}$  differing in the substitution of its ring A.

Compound 2 was isolated as a white, amorphous solid. Its molecular formula was established as  $C_{19}H_{16}O_8$  based on MS ( $[M + H]^+$  at m/z 373.2) and HRESIMS data ( $[M + H - H_2O]^+$  at m/z 355.0354), consistent with the formula  $C_{19}H_{15}O_7$  (calcd 355.0818) and NMR data (Table 1). The UV absorbance at  $\lambda_{max}$  285 and 305 nm as well as the <sup>1</sup>H NMR [ $\delta_H$  4.58 dd, J = 13.5, 2.6 Hz (H-6), 4.46, dd J = 13.5, 2.6 Hz (H-6), and 4.56, d, J = 3.6 Hz (H-6a)] and <sup>13</sup>C NMR [( $\delta_C$  64.6 (C-6), 76.4 (C-6a), 191.3 (C-12), and 67.9 (C-12a)] data



**Figure 2.** VCD spectra observed for **1** (black) and those predicted for its (6a*S*,12a*S*)-**1** (blue) and (6a*S*,12a*R*)-**1** (red) stereoisomers. The numbers 1–9 indicate band assignment of key importance for AC assignment and are given to facilitate comparison of the experimental and simulated spectra.

(Table 1, Figures S10-S17, Supporting Information) were consistent with a 12a-hydroxyrotenoid skeleton.<sup>18</sup> The NMR spectra indicated a methylenedioxy ( $\delta_{\rm H}$  5.83 and 5.86;  $\delta_{\rm C}$ 101.5) and two methoxy ( $\delta_{\rm H}$  3.86 and 3.88;  $\delta_{\rm C}$  56.4 and 56.5) functionalities. The four singlet aromatic protons H-1 ( $\delta_{\rm H}$ 6.54), H-4 ( $\delta_{\rm H}$  6.46), H-8 ( $\delta_{\rm H}$  6.40), and H-11 ( $\delta_{\rm H}$  7.25) were consistent with C-2, C-3, C-9, and C-10 oxygenation of the 12a-hydroxyrotenoid skeleton. The NOE correlation of H-11  $(\delta_{\rm H}\,7.25)$  with OMe-10  $(\delta_{\rm H}\,3.88)$  and the NOE of OMe-9  $(\delta_{\rm H}$ 3.86) with H-8 ( $\delta_{\rm H}$  6.40) along with the HMBC of H-8 ( $\delta_{\rm H}$ 6.40) to C-7a ( $\delta_{\rm C}$  157.5), C-9 ( $\delta_{\rm C}$  157.3), C-10 ( $\delta_{\rm C}$  149.5), and C-11a ( $\delta_{\rm C}$  109.5) as well as of H-10 ( $\delta_{\rm H}$  6.59) to C-8 ( $\delta_{\rm C}$ 100.3), C-9 ( $\delta_{\rm C}$  157.3), and C-11a ( $\delta_{\rm C}$  109.5) placed the two OMe groups of ring D at C-9 ( $\delta_{\rm C}$  145.3) and C-10 ( $\delta_{\rm C}$  157.3), respectively. The methylenedioxy moiety was placed at C-2/C-3 based on the HMBC correlations of H-1 ( $\delta_{\rm H}$  6.54) to C-1a  $(\delta_{\rm C} 109.9)$ , C-2  $(\delta_{\rm C} 109.9)$ , C-3  $(\delta_{\rm C} 109.9)$ , C-4a  $(\delta_{\rm C} 149.5)$ , and C-12a ( $\delta_{\rm C}$  67.9), as well as of H-4 ( $\delta_{\rm H}$  6.46) to C-1a ( $\delta_{\rm C}$ 109.9), C-2 ( $\delta_{\rm C}$  109.9), and C-3 ( $\delta_{\rm C}$  109.9), leading to the planar structure 9,10-dimethoxy-2,3-methylenedioxy-12a-hydroxyrotenoid for compound 2. The chemical shift of H-1  $(\delta_{\rm H} 6.54)$  along with the small coupling constants,  $J \leq 3.3$  Hz, between H-6a ( $\delta_{\rm H}$  4.59) and the CH<sub>2</sub>-6 protons ( $\delta_{\rm H}$  4.58 and 4.46) indicated a *cis* B/C ring junction,<sup>34</sup> whereas the negative Cotton effect at ca. 340 nm suggested a (6aR,12aR)-2 absolute configuration.<sup>18</sup> Similar to compound 1, the positive specific rotation  $[\alpha]_{D}^{20}$  +42 (c 0.3 CH<sub>3</sub>OH) was not found to be reliable in assigning the configuration of 2. Instead, its (6aR, 12aR)-2 absolute configuration was determined by the comparison of its experimental VCD (Figure 3) with calculated spectra. This configuration was corroborated by the IR band pattern between 1350 and 1000 cm<sup>-1</sup> (Figure S136, Supporting Information), which indicated a cis configuration.<sup>37,38</sup> Based on the spectroscopic data, the new compound 2 was identified as (6aR,12aR)-9,10-dimethoxy-2,3methylenedioxy-12a-hydroxyrotenoid and was given the trivial



Figure 3. Experimental VCD spectrum of 2 (black) and the spectra predicted for (6aR,12aR)-2 (blue) and (6aR,12aS)-2 (red). The numbers 1–11 indicate band assignment of key importance for AC assignment and are given to facilitate comparison of the experimental and simulated spectra.

name oblarotenoid F. It is structurally closely related to the previously reported oblarotenoid A,<sup>18</sup> differing in the substitution of its ring D. It should be emphasized that despite structural similarities, the specific rotation of oblarotenoid F,  $[\alpha]^{20}_{D}$  +42, has an opposite sign as compared to those of oblarotenoid A,  $[\alpha]^{20}_{D}$  -38.3, and of (-)-(6aR,12aR)-12a-hydroxy- $\alpha$ -toxicarol,  $[\alpha]^{20}_{D}$  -108, recently reported from *Millettia brandisiana*.<sup>39</sup> The ECD and the specific rotation data of rotenoids may not be complementary, which has been previously pointed out.<sup>18,39</sup>

Compound 3 was isolated as a light yellow solid. Its molecular formula, C18H12O6, was established based on HRESIMS ( $[M + H]^+$  m/z 325.0714, calcd 325.0712) and NMR data (Table 1, Figures S18-S24 Supporting Information). UV absorptions at  $\lambda_{\rm max}$  237, 278, and 310 nm along with the NMR data, especially the chemical shifts of the oxygenated methylene protons H-6 ( $\delta_{\rm H}$  4.96) and of C-6a ( $\delta_{\rm C}$  156.8), suggested 3 to have a 6a,12a-dehydrorotenoid skeleton. Furthermore, typical for 6a,12a-dehydrorotenoids, the deshielded chemical shift of the singlet signal at  $\delta_{\rm H}$  8.32 was assigned to H-1 of ring A.40 This is due to the magnetic anisotropy of the nearby carbonyl C-12 ( $\delta_{\rm C}$  174.2) and the flat geometry of the molecule [(sp<sup>2</sup>-hybridized C-6a ( $\delta_{\rm C}$  156.8) and C-12a ( $\delta_{\rm C}$  112.3)]. The NMR data further indicated the presence of methoxy ( $\delta_{\rm H}$  3.92,  $\delta_{\rm C}$  56.0) and methylenedioxy ( $\delta_{\rm H}$  5.96,  $\delta_{\rm C}$  101.5) moieties. Here, the appearance of the two singlets H-1 ( $\delta_{\rm H}$  8.32) and H-4 ( $\delta_{\rm H}$  6.53) was indicative of the placement of the methylenedioxy moiety at C-2/C-3 of ring A, which was further supported by the HMBCs between the methylenedioxy protons ( $\delta_{\rm H}$  5.96) and C-2 ( $\delta_{\rm C}$  142.9) and C-3  $(\delta_{\rm C}$  147.4). The placement of the methoxy group at C-9 was established upon the presence of three mutually coupled protons, H-11 ( $\delta_{\rm H}$  8.20, d, J = 8.9 Hz), H-10 ( $\delta_{\rm H}$  7.0, dd, J = 2.2, 8.9 Hz), and H-8 ( $\delta_{\rm H}$  6.83, d, J = 2.2 Hz), in ring D. This was further supported by the HMBCs of the OMe-9 ( $\delta_{\rm H}$  3.92) to C-9 ( $\delta_{\rm C}$  164.0), of H-11 ( $\delta_{\rm H}$  8.20) to C-9 ( $\delta_{\rm C}$  164.1) and C-

12 ( $\delta_{\rm C}$ 174.2), and of H-8 ( $\delta_{\rm H}$  6.83) to C-7a ( $\delta_{\rm C}$  156.7) and C-11a ( $\delta_{\rm C}$  118.7) as well as by the NOE between OMe-9 ( $\delta_{\rm H}$ 3.92) and H-8 ( $\delta_{\rm H}$  6.83). Therefore, the new compound 3 was identified as 9-methoxy-2,3-methylenedioxy-6a-2a-dehydrorotenoid and was given the trivial name oblarotenoid G. It has previously been reported as a synthetic product.<sup>41</sup> Here, we report it as a natural product and provide its complete NMR data for the first time. It is structurally close to the previously reported oblarotenoid A<sup>18</sup> and may be obtained by its dehydration. Oblarotenoid G has been observed in the crude extract, by TLC profiling, confirming that it is not an isolation artifact.

Compound 4 was isolated as a white amorphous solid and was assigned the molecular formula C45H54O25 based on HRESIMS ( $[M + H]^+ m/z$  959.3043) and NMR data (Table 2, Figures S25-S32, Supporting Information). The NMR spectra showed signals at  $\delta_{\rm H}$  8.17 (s, H-2),  $\delta_{\rm C}$  154.2 (C-2), 126.9 (C-3), and 177.1 (C-4) characteristic for an isoflavone substituted with a methylenedioxy ( $\delta_{\rm H}$  5.98,  $\delta_{\rm C}$  102.5), two methoxy ( $\delta_{\rm H}$  3.91;  $\delta_{\rm C}$  62.3 and  $\delta_{\rm H}$  3.92;  $\delta_{\rm C}$  62.8), and sugar moieties. Based on the HMBC of H-5 ( $\delta_{
m H}$  7.12) to C-4 ( $\delta_{
m C}$ 177.1), C-8a ( $\delta_{\rm C}$  155.5), and C-7 ( $\delta_{\rm C}$  142.1), the single aromatic proton of the trisubstituted ring A was placed at C-5  $(\delta_{\rm C}$  101.8). Its positioning was confirmed by the HMBC of H-5 ( $\delta_{\rm H}$  7.1) to C-4 ( $\delta_{\rm C}$  177.1). The chemical shift values  $\delta_{\rm C}$  62.8 and  $\delta_{\rm C}$  62.3 of the two OMe groups indicated that both are diortho-disubstituted.<sup>33</sup> They were placed at C-7 and C-8 based on the HMBC of 7-OMe ( $\delta_{\rm H}$  3.92) and C-7 ( $\delta_{\rm C}$  142.1) as well as between 8-OMe ( $\delta_{\rm H}$  3.91) and C-8 ( $\delta_{\rm C}$  153.9). The methylenedioxy moiety was placed at C-3'/C-4' ( $\delta_{\rm C}$  148.9) of ring B, which has an AMX-spin system with protons at  $\delta_{\rm H}$  7.03  $(d, J = 1.7 \text{ Hz}, \text{H-2'}), \delta_{\text{H}} 6.97 \text{ (dd}, J = 1.6, 8.0 \text{ Hz}, \text{H-6'}) \text{ and}$  $\delta_{\rm H}$  6.84 (d, *J* = 8.0 Hz, H-5'). The isoflavone is also substituted with four glycosides as indicated by the characteristic signals of four anomeric protons at  $\delta_{\rm H}$  5.08, 4.77, 4.73, and 4.55. Two of the sugars that possessed methyl groups at  $\delta_{\rm H}/\delta_{\rm C}$  1.14/17.9 and  $\delta_{\rm H}/\delta_{\rm C}$  1.09/17.8, respectively, were assigned as rhamnosyl, whereas the other two were assigned as glucosyl. The <sup>1</sup>H and <sup>13</sup>C NMR signals corresponding to each sugar (Table 2) were identified with the help of COSY, TOCSY, HSQC, HMBC, and NOESY spectra (Figures S26-S30, Supporting Information). Carbon C-6 of the isoflavone is the site of Oglycosidation, based on the NOE of H-5 ( $\delta_{
m H}$  7.13) and OMe-7 ( $\delta_{\rm H}$  3.80) with the anomeric proton H-1" ( $\delta_{\rm H}$  5.05) of the first glucose as well as on the HMBC of H-1" and C-6 ( $\delta_{\rm C}$ 157.0). The large coupling constant,  ${}^{3}J_{H1''H2''} = 7.9$  Hz of the anomeric proton H-1" ( $\delta_{\rm H}$  5.05) to H-2" ( $\delta_{\rm H}$  3.78) suggested the  $\beta$ -linkage of the sugar to the isoflavone.<sup>42</sup> The large diaxial coupling constants of the protons of this sugar moiety (Table 2) suggested it to be a  $\beta$ -glucose. Both H-1<sup>'''</sup> ( $\delta_{\rm H}$  4.53), the anomeric proton of the second sugar moiety, and H-2"  $(\delta_{
m H}$ 3.80) showed HMBC with C-4" ( $\delta_{\rm C}$  89.4), which together with the NOE of H-4" ( $\delta_{\rm H}$  3.61) and H-1" ( $\delta_{\rm H}$  4.53) and with the  ${}^{3}J_{\text{H1}''\text{H2}''} = 7.8$  Hz identified the  $\beta$ -glucopyranosyl- $(4'' \rightarrow$ 1<sup>'''</sup>)- $\beta$ -glucopyranosyl linkage between the two glycoside moieties. The characteristic large diaxial coupling constants (Table 2) and the NOEs of H-1<sup>'''</sup> ( $\delta_{\rm H}$  4.53) with 1<sup>''''</sup> ( $\delta_{\rm H}$  4.74), H-3<sup>''''</sup> ( $\delta_{\rm H}$  3.70), and H-3<sup>''''</sup> ( $\delta_{\rm H}$  3.80) indicated the second sugar moiety to also be a  $\beta$ -glucopyranosyl. The anomeric protons of the two rhamnosyl moieties, H-1  $^{\prime\prime\prime\prime}$  ( $\delta_{
m H}$ 4.71) and H-1""" ( $\delta_{\rm H}$  4.74), showed HMBC with C-7" ( $\delta_{\rm C}$ 67.8) and C-7<sup>'''</sup> ( $\delta_{\rm C}$  68.6), respectively. Furthermore, H-7<sup>''</sup> ( $\delta_{\rm H}$ 4.08 and 3.58) and H-7<sup>'''</sup> ( $\delta_{\rm H}$  4.04 and 3.55) showed HMBC

to the respective anomeric carbons of the rhamnosyl groups, C-1<sup>'''</sup> ( $\delta_{\rm C}$  102.1) and C-1<sup>'''''</sup> ( $\delta_{\rm C}$  102.4). These HMBCs along with the NOEs of H-7" ( $\delta_{\rm H}$  4.11 and 3.60) and H-7" ( $\delta_{\rm H}$  4.06 and 3.58) to H-1"" ( $\delta_{\rm H}$  4.71) and H-1"" ( $\delta_{\rm H}$  4.74), respectively, established the  $\alpha$ -rhamnosyl- $(1''' \rightarrow 7'')$ - $\beta$ -glucosyl and the  $\alpha$ -rhamnosyl- $(1'''' \rightarrow 7'')$ - $\beta$ -glucosyl linkages. The equatorial orientation of H-1"'' ( $\delta_{\rm H}$  4.71) and H-1"''' ( $\delta_{\rm H}$ 4.74) was assigned upon the observation of characteristic small, diequatorial  ${}^{3}J_{H1'''H2'''} = 1.2$  Hz and  ${}^{3}J_{H1'''H2'''} = 1.4$  Hz coupling constants and of the absence of NOEs between H- $1''''''(\delta_{\rm H} 4.74)$  and H- $3''''''(\delta_{\rm H} 3.80)$  and H- $1'''''(\delta_{\rm H} 4.74)$  and H-3""" ( $\delta_{\rm H}$  4.80). The identity of the aglycone was further confirmed by comparison of its NMR data with that of its literature-known 6-methoxy derivative (Figures S33-S39).43 The absolute configuration of compound 4 has not been elucidated. However, natural sugars are known to almost always be D-glucose and L-rhamnose.<sup>44,45</sup> Based on the above spectroscopic data, obloneside (4) was characterized as isoplatycarpanetin-6-O- $\beta$ -glucosyl- $(7'' \rightarrow 1'''')$ - $\alpha$ -rhamnosyl- $(4'' \rightarrow 1''') - \beta$ -glucosyl- $(7''' \rightarrow 1'''') - \alpha$ -rhamnoside.

The extracts and the isolated compounds were tested for antiviral activity against RSV by the viral plaque reduction assay in cultures of HEp-2 cells (Table 3, Supporting

Table 3. Anti-RSV Activity, Cytotoxicity for HEp-2 Cells, and Selectivity Indices (SI) of Compounds Isolated from *Milletia oblata* 

compound/ extract         IC <sub>50</sub> ( $\mu$ M)         CC <sub>50</sub> ( $\mu$ M)         cytostatic activity <sup>a</sup> (f)           1         >100.0         >100.0         1           2         >100.0         >100.0         1           3         22.0         90.0         4           4         >100.0         >100.0         1           5         53         >100.0         1           6         1.4         50.0         PCS (20.0 $\mu$ M)         3           7         1.8         6.7         3         3           8         1.5         42.0         PCS (20.0 $\mu$ M)         2           9         0.8         12.5         PCS (0.8 and 4.0 $\mu$ M)         10	SI <sup>b</sup>
1       >100.0       >100.0       1         2       >100.0       >100.0       1         3       22.0       90.0       4         4       >100.0       >100.0       1         5       53       >100.0       1         6       1.4       50.0       PCS (20.0 $\mu$ M)       3         7       1.8       6.7       3         8       1.5       42.0       PCS (20.0 $\mu$ M)       2         9       0.8       12.5       PCS (0.8 and 4.0 $\mu$ M)       1         10       0.4       26.5       PCS (40 $\mu$ M)       6	CC <sub>50</sub> / (C <sub>50</sub> )
2       >100.0       >100.0       1         3       22.0       90.0       4         4       >100.0       >100.0       1         5       53       >100.0       5         6       1.4       50.0       PCS (20.0 $\mu$ M)       3         7       1.8       6.7       3         8       1.5       42.0       PCS (20.0 $\mu$ M)       2         9       0.8       12.5       PCS (0.8 and 4.0 $\mu$ M)       1         10       0.4       26.5       PCS (40.0 M)       1	.0
3       22.0       90.0       4         4       >100.0       >100.0       1         5       53       >100.0       5         6       1.4       50.0       PCS (20.0 $\mu$ M)       3         7       1.8       6.7       3         8       1.5       42.0       PCS (20.0 $\mu$ M)       3         9       0.8       12.5       PCS (0.8 and 4.0 $\mu$ M)       1         10       0.4       26.5       PCS (4.0 $\mu$ M)       6	.0
4       >100.0       >100.0       1         5       53       >100.0       >         6       1.4       50.0       PCS (20.0 $\mu$ M)       3         7       1.8       6.7       3         8       1.5       42.0       PCS (20.0 $\mu$ M)       3         9       0.8       12.5       PCS (0.8 and 4.0 $\mu$ M)       1         10       0.4       26.5       PCS (4.0 $\mu$ M)       1	1.1
5       53       >100.0       >         6       1.4       50.0       PCS (20.0 $\mu$ M)       33         7       1.8       6.7       33         8       1.5       42.0       PCS (20.0 $\mu$ M)       33         9       0.8       12.5       PCS (0.8 and 4.0 $\mu$ M)       14         10       0.4       26.5       PCS (4.0 $\mu$ M)       14	.0
6       1.4       50.0       PCS (20.0 μM)       3         7       1.8       6.7       3         8       1.5       42.0       PCS (20.0 μM)       2         9       0.8       12.5       PCS (0.8 and 4.0 μM)       1         10       0.4       26.5       PCS (4.0 μM)       6	>1.9
7       1.8       6.7       3         8       1.5       42.0       PCS ( $20.0 \mu$ M)       2         9       0.8       12.5       PCS ( $0.8 \text{ and } 4.0 \mu$ M)       1         10       0.4       26.5       PCS ( $40 \mu$ M)       6	35.7
8         1.5         42.0         PCS (20.0 μM)         2           9         0.8         12.5         PCS (0.8 and 4.0 μM)         1           10         0.4         26.5         PCS (4.0 μM)         6	3.8
<b>9</b> 0.8 12.5 PCS (0.8 and 4.0 $\mu$ M) 1 <b>10</b> 0.4 26.5 PCS (4.0 $\mu$ M) 6	28.0
10 0.4 265 PCS $(40 \mu M)$ 6	5.6
	61.6
11 8.0 91.0 PCS (20.0 μM) 1	1.4
12 >100.0 >100 I	.0
13         2.1         >100.0         PCS (4.0, 20.0, and 100.0 μM)         >	×47.6
14 10.0 >100.0 PCS (100.0 $\mu$ M) >	>10
15 10.0 65.0 PCS (20.0 μM) 6	ó.5
<b>16</b> >100.0 >100.0	.0
17 >100.0 >100.0	.0
18 $ND^c$ $ND^c$ $ND^c$ $ND^c$	ND <sup>c</sup>
19 >0.8 2.0	<2.5
ribavirin 10.6 >100 PCP (100 μM)	»9.4
DMSO >100 >100 1	

<sup>a</sup>The CC<sub>50</sub> assay was complemented by microscopic recording of possible cytostatic activity of test compounds manifested as poor cell proliferation (PCP), altered cell shape (ACS), or poor cell staining (PCS). <sup>b</sup>SI of 1 indicates lack of antiviral activity, i.e., lack of cell protection against RSV; SI of 2–10 suggests that antiviral activity may at least in some compounds be unspecific due to cytostatic activity (poor cell proliferation/altered cell shape) that occurs at concentrations just below the CC<sub>50</sub> and may not be detectable by the cell toxicity assay used; SI > 10 suggests specific antiviral activity for most compounds tested. <sup>c</sup>ND, not determined

Information). A tetrazolium-based cytotoxicity assay on HEp-2 cells was also performed to exclude that the anti-RSV activity was a result of general cytotoxicity. The cytotoxicity assay was complemented by microscopy observation of the compoundand extract-treated cells to detect possible cytostatic activity, such as poor cell growth and/or altered cell shape that may affect antiviral activity. The leaf extract exhibited anti-RSV activity with an IC<sub>50</sub> of 0.7  $\mu$ g/mL and comparably low toxicity, CC<sub>50</sub>, of 50.0  $\mu$ g/mL, for HEp-2 cells (Table S1 and Figures S129-S130, Supporting Information), whereas the root wood extract showed anti-RSV activity with an  $\mathrm{IC}_{50}$  of 3.4  $\mu$ g/mL. Both crude extracts decreased the cell viability also at concentrations less than their  $CC_{50}$  values (Figure S130, Supporting Information), suggesting that they are rich in cytotoxic/cytostatic compounds that may interfere with their anti-RSV activities (Figure S129, Supporting Information). Out of the 19 isolated compounds tested, 6 (IC<sub>50</sub> 1.4  $\mu$ M), 8  $(IC_{50} 1.5 \ \mu M)$ , 10  $(IC_{50} 0.4 \ \mu M)$ , 11  $(IC_{50} 8.0 \ \mu M)$ , and 14  $(IC_{50} \ 10.0 \ \mu M)$  exhibited substantial anti-RSV activity and showed cytotoxicity (26.5  $\mu$ M to >100.0  $\mu$ M) and cytostatic activity only at much higher concentrations than their antiviral IC<sub>50</sub> values (Table 3 and Figures S131–S132, Supporting Information). Compounds 9 (IC<sub>50</sub> 0.8  $\mu$ M) and 13 (IC<sub>50</sub> 2.1  $\mu$ M) showed substantial anti-RSV activity with selectivity indices (SI) of 15.6 and >47.6, respectively; however, both exhibited cytostatic activity at concentrations close to their  $IC_{50}$  values (Table 3). At this stage, it is difficult to assess whether the cytostatic activity of these specific compounds contributed to the protection of cells against virus infection, and accordingly the specificity of anti-RSV action of 9 and 13 requires further investigation. An attempt was made to address this issue by testing cytotoxic activities of the most promising compounds in human embryonic lung fibroblasts (HELFs), i.e., primary cells derived from lung interstitium. In these cells, the CC<sub>50</sub> values were >100  $\mu$ M for compounds 3, 11, and 14 and 47.1  $\mu$ M for compound 15, thus confirming the low cytotoxic activity of these compounds found in HEp-2 cells (Table 3), an observation that further strengthens their antiviral potential. In contrast the other active compounds showed CC<sub>50</sub> values at low concentrations, i.e., 1.3  $\mu$ M for 6, 4.8  $\mu$ M for 8, 0.3  $\mu$ M for 10, and 9.0  $\mu$ M for 13, indicating that these compounds were more toxic for HELF cells than for HEp-2 cells, and the specificity of their anti-RSV activity requires further studies in HELF cells. The addition of the same volume of pure DMSO as the volume of the solution of the studied compounds was used as a negative control in this study, while ribavirin, the only antiviral drug approved in the form of aerosol formulations for the treatment of RSV disease, was used as a positive control. In our hands, ribavirin inhibited RSV infection of HEp-2 cells with an IC<sub>50</sub> of 10.6  $\mu$ M and was not toxic for cells at concentrations up to 100.0  $\mu$ M (CC<sub>50</sub> > 100.0  $\mu$ M), but showed cytostatic activity at 100.0  $\mu$ M (Table 3 and Figures S131 and S132, Supporting Information).

We also tested the crude extracts and the isolated compounds for their capability to inhibit HRV-2 infection of adenocarcinoma cells of the uterine cervix (HeLa). Compound **3** displayed anti-HRV-2 activity by protecting the cells at an  $IC_{50}$  of 4.2  $\mu$ M while showing toxicity for HeLa cells first at a  $CC_{50}$  of 48.0  $\mu$ M (SI 11.4, Figure S133, Supporting Information). Compound **3** also showed anti-RSV activity ( $IC_{50}$  22.0  $\mu$ M, Table 3 and Figures S131 and S132, Supporting Information). Our observations emphasize the antiviral potential of rotenoids, as some of these compounds

were reported to inhibit the Newcastle disease virus and the herpes simplex virus (HSV),<sup>46</sup> and 12a-hydroxyrotenoids showed modest anti-HSV activity.<sup>47</sup> Isoflavonoids are believed to be produced by plants for protection against microbes, and they can therefore be expected to show activity against viruses.<sup>48</sup>

In conclusion, 19 natural products, including three new roteinoids (1-3) and a new glycosylated isoflavone (4), were isolated from the CH<sub>2</sub>Cl<sub>2</sub>/MeOH (1:1) extract of the leaves and the root wood of M. oblata ssp. teitensis. The absolute configuration of rotenoids was for the first time determined by VCD spectroscopy. In addition to the crude extracts ( $IC_{50}$  0.7 and 3.4  $\mu$ g/mL), the isolated compounds oblarotenoid A (6), 12a-hydroxymunduserone (8), deguelin (10), ichthynone (11), and 4'-prenyloxyderrone (14) showed anti-RSV activity  $(IC_{50} 0.4-10.0 \ \mu M)$  with low cytotoxicities in HEp-2 cells (>26.5  $\mu$ M). Tephrosin (9) and isoerythrin-A-4'-prenyl ether (13) showed substantial anti-RSV activity (IC<sub>50</sub> 0.8 and 2.1  $\mu$ M, respectively), yet also exhibited cytostatic activity (CC<sub>50</sub> 12.5 and >100.0  $\mu$ M). Besides its anti-RSV activity (IC<sub>50</sub> 22.0  $\mu$ M), 9-methoxy-2,3-methylenedioxy-6a-2a-dehydrorotenoid (3) protected cells from HRV-2 infection (IC<sub>50</sub> of 4.2  $\mu$ M) whereas showing low cytotoxicity (CC<sub>50</sub> of 48.0  $\mu$ M).

## EXPERIMENTAL SECTION

General Experimental Procedures. Optical rotations were measured on a PerkinElmer 341-LC instrument. NMR spectra were acquired on an Agilent MR-400-DD2 spectrometer equipped with a 5 mm OneNMR probe and on a Bruker Avance NEO 500 MHz or Bruker Avance NEO 800 MHz instrument equipped with a 5 mm cryogenic TXO probe. The spectra were processed using MestReNova 14.1 software and were referenced to the residual solvent peak. LC-ESIMS data were acquired on a Waters Micromass ZQ multimode ionization electrospray ionization (ESI) instrument connected to an Agilent 1100 series gradient pump system and a C8 column (Gemini), using a Milli-Q H<sub>2</sub>O/MeCN (5:95 to 95:5, with 1% HCO<sub>2</sub>H over 4 min) eluent mixture. HRESIMS spectra were acquired with a Q-TOF LC/MS spectrometer with a lock mass-ESI source (Stenhagen Analysis Lab AB, Gothenburg, Sweden), using a 2.1  $\times$  30 mm, 1.1  $\mu$ m reverse phase<sup>49</sup>-C<sub>18</sub> column and a H<sub>2</sub>O/MeCN gradient (5:95 to 95:5, with 0.2% HCO2H). TLC analyses were carried out on Merck precoated silica gel 60 F<sub>254</sub> plates. Preparative TLCs were performed on glass plates of  $20 \times 20$  cm dimension, precoated with silica gel  $60F_{254}$  having 0.25 to 1 mm thickness. Column chromatography was performed on silica gel (40-63  $\mu$ m mesh), and gel filtration was performed on Sephadex LH-20.

**Plant Materials.** The root wood and leaves of *Millettia oblata* ssp. *teitensis* were collected in July 2014 from Ngaongao forest, Taita Hill, Taita County, Kenya, and were assigned voucher number TD-04/2014. The plant was identified by Mr. Patrick Chalo Mutiso of the Department of Biology, University of Nairobi, Kenya, where the specimen was deposited.

**Extraction and Isolation.** The dried and ground leaves of *M.* oblata (795 g) were extracted with CH<sub>2</sub>Cl<sub>2</sub>/MeOH (1:1) (4 × 1.5 l) for 24 h by cold percolation. The extract was filtered, and the supernatant was concentrated under reduced pressure to obtain the extract (35 g). A portion of this crude extract (27 g) was subjected to column chromatography on silica gel (300 g) using *iso*-hexane (a mixture of isomeric branched chain hexanes) containing increasing amounts of EtOAc to give a total of 100 fractions. The first 30 fractions eluted with 0–3% EtOAc in *iso*-hexane were not further investigated. Fractions 31–40 eluted with 6% EtOAc in *iso*-hexane were combined, based on their TLC profile, and were subjected to column chromatography on Sephadex LH-20 (CH<sub>2</sub>Cl<sub>2</sub>/MeOH, 1:1) to give oblarotenoid C (5, 8 mg) and 9-methoxy-2,3-methylendioxy-6a-12a-dehyrorotenoid (3, 10 mg). Fractions 41–48 eluted at 7% EtOAc in *iso*-hexane afforded a white precipitate, which was washed with MeOH to afford oblarotenoid A (6, 11 mg). Fractions 49-57 eluted with 8% EtOAc in iso-hexane showed the presence of five compounds by TLC analysis and were separated by column chromatography on silica gel (eluent: iso-hexane/CH<sub>2</sub>Cl<sub>2</sub>, 1:1) to yield an additional amount of oblarotenoid A (6, 2 mg) together with 12a-hydroxymunduserone (8, 4 mg), tephrosin (9, 7 mg), deguelin (10, 10 mg), and a mixture, which was further separated by column chromatography over silica gel (eluent; iso-hexane/CH2Cl2/EtOAc, 60:35:10) to afford deguelin (10, 2 mg) and (6aS,12aS)-4,9dimethoxy-2,3-methylenedioxyrotenoid (1, 3 mg). The fractions eluted with 10-14% EtOAc gave a mixture of four compounds, which was further purified on Sephadex LH-20 eluting with CH<sub>2</sub>Cl<sub>2</sub>/ MeOH (1:1) to yield 7,2',5'-trimethoxy-3',4'-methylenedioxyisoflavone (12, 10 mg), oblarotenoid D (7, 8 mg), and a fraction containing three compounds. These three compounds were separated by preparative TLC developed in a mixture of iso-hexane/  $CH_2Cl_2/EtOAc$  (13:5:2) to yield ichthynone (11, 4 mg), oblarotenoid D (7, 1 mg), and (6aR,12aR)-9,10-dimethoxy-2,3methylenedioxyrotenoid (2, 4 mg).

The root wood (800 g) was extracted to give a light brown crude extract (30 g) and was subjected to column chromatography over silica (300 g), eluting it as described above. The fractions eluted with 4–8% EtOAc in *iso*-hexane afforded isoerythrin-A-4' (13, 7 mg), 4'-prenyloxyderrone (14, 15 mg), and a mixture of two compounds, which were separated on Sephadex LH-20 (CH<sub>2</sub>Cl<sub>2</sub>/MeOH, 1:1) to give cuneatin methyl ether (15, 6 mg) and calopogonium isoflavone B (16, 10 mg). The fractions eluted with 10% EtOAc were further purified on Sephadex LH20 to yield isobutyl alcohol (19, 11 mg). Fractions eluted with 12–18% EtOAc in *iso*-hexane afforded a mixture of two compounds, which upon purification by column chromatography over silica gel (*iso*-hexane/CH<sub>2</sub>Cl<sub>2</sub>, 1:1) yielded maximaiso-flavone G (17, 9 mg) and milldurone (18, 8 mg). The fractions eluted with 100% EtOAc were further purified by Sephadex LH-20 (100% MeOH) chromatography and yielded obloneside (4, 15 mg).

(6aS,12aS)-2,9-Dimethoxy-3,4-methylenedioxyrotenoid (1). White amorphous solid;  $[\alpha]^{20}_{D}$  +44.0 (*c* 0.2 CH<sub>3</sub>OH); UV  $\lambda_{max}$  (log  $\epsilon$ ) 280 nm (4.2) and 305 nm (4.0); <sup>1</sup>H and <sup>13</sup>C NMR, see Table 1 and Figures S1–S9, Supporting Information; HRESIMS *m*/*z* 357.0976 [M + H]<sup>+</sup> (calcd 357.0974 for C<sub>19</sub>H<sub>17</sub>O<sub>7</sub>).

(6aR,12aR)-9,10-Dimethoxy-2,3-methylenedioxyrotenoid (2). White amorphous solid,  $[\alpha]^{20}_{D}$  +42.0 (*c* 0.3 CH<sub>3</sub>OH); UV  $\lambda_{max}$  (log  $\epsilon$ ) 285 nm (4.3) and 305 nm (3.9); <sup>1</sup>H and <sup>13</sup>C NMR, see Table 1 and Figures S10–S17, Supporting Information; HRESIMS *m*/*z* 355.0354 [M + H – H<sub>2</sub>O]<sup>+</sup> (calcd for C<sub>19</sub>H<sub>15</sub>O<sub>7</sub> 355.0815), LC-ESIMS *m*/*z* 373.2 [M + 1]<sup>+</sup> (calcd 373.1 for C<sub>19</sub>H<sub>18</sub>O<sub>9</sub>).

9-Methoxy-2,3-methylenedioxy-6a,2a-dehydrorotenoid (3). White amorphous solid; UV  $\lambda_{max}$  (log  $\epsilon$ ) 237 nm (4.5), 278 nm (2.3), and 310 nm (4.0); <sup>1</sup>H and <sup>13</sup>C NMR, see Table 1 and Figures S18–S24, Supporting Information; HRESIMS m/z 325.074 [M + H]<sup>+</sup> (calcd 325.0712 for C<sub>18</sub>H<sub>13</sub>O<sub>6</sub>).

Obloneside (4). Light yellow amorphous solid; UV  $\lambda_{max}$  (log  $\epsilon$ ) 290 nm (4.3); <sup>1</sup>H and <sup>13</sup>C NMR, see Table 2 and Figures S25–S32, Supporting Information; HRESIMS m/z 959.3043 [M + H]<sup>+</sup> (calcd 959.3033 for C<sub>42</sub>H<sub>55</sub>O<sub>25</sub>).

**Cells and Viruses.** HEp-2 cells were cultivated in Dulbecco's modified Eagle's medium (DMEM) supplemented with 8% fetal calf serum and 292  $\mu$ g/mL of L-glutamine. Human adenocarcinoma cells of uterine cervix (HeLa) were grown in Eagle's minimum essential medium supplemented with 5% fetal calf serum and antibiotics. HELFs were grown in Eagle's minimum essential medium supplemented with 5% fetal calf serum, L-glutamine, and antibiotics. The A2 strain of RSV (ATCC VR-1540) was used. The RSV stock was prepared and stored as described previously by Lundin et al.<sup>50</sup> In some experiments, strain HGP of HRV-2 (ATCC, VR-482) was used.

**Antiviral Assays.** The RSV plaque reduction assay was performed as described previously<sup>27,51</sup> with some modifications. Briefly, the test extracts and compounds were serially diluted at a range of 1.6–200  $\mu$ g/mL and 0.16–100  $\mu$ M, respectively, in DMEM supplemented with 2% heat-inactivated fetal calf serum, 60  $\mu$ g/mL of penicillin, 100  $\mu$ g/mL of streptomycin, and 292  $\mu$ g/mL of L-glutamine (DMEM-M)

and then added to one-day-old cultures of HEp-2 cell in 24-well culture plates. Following 15 min of incubation at 37 °C in a humidified atmosphere comprising 5% CO<sub>2</sub> (incubator), the cells were inoculated with 50–100 plaque-forming units of RSV A2. After incubation of the virus–compound mixture with cells for 2.5 h in the CO<sub>2</sub> incubator, the inoculum was removed and 750  $\mu$ L of 1% methylcellulose solution in DMEM-M comprising the same concentration of test compound/extract was added and left for a further 3 days in the CO<sub>2</sub> incubator. The cells were stained with a 1% solution of crystal violet, and the viral plaques were counted under a microscope. Ribavirin, a drug approved for the treatment of RSV disease, was used as a positive control, whereas DMSO at concentrations corresponding to those present in the test compounds was used as solvent control.

The anti-HRV-2 activity of extracts and compounds was assessed by a tetrazolium-based colorimetric assay<sup>52</sup> with some modifications. Briefly, serial 5-fold dilutions of test samples (0.16-100  $\mu$ M) in EMEM supplemented with 2% fetal calf serum, 1% pest stock, 1% Lglutamine stock, 30 mM MgCl<sub>2</sub>, and 20 mM HEPES (pH 7.1) (EMEM-M) were added to one-day-old cultures of HeLa cells seeded in 96-well culture plates. Following the incubation of cells with test samples for 3 h in a 34 °C CO2 incubator, 100 tissue culture infectious doses (TCID<sub>50</sub>) in EMEM-M were added. In some wells addition of the test samples and the virus was omitted to serve as the virus control and uninfected cell control, respectively. After incubation of the test plates in the CO<sub>2</sub> incubator for 3 days, the CellTiter 96 Aqueous One solution reagent (Promega, Madison, WI, USA) was added, and following further incubation of plates in the CO<sub>2</sub> incubator for 1 h, the absorbance was recorded at 490 nm. The % of the test sample protection of cells against HRV-2 infection was calculated as (absorbance of the test sample – virus control)  $\times 100/$ absorbance of the cell control – virus control.

**The Cell Toxicity Assay.** The tetrazolium-based cytotoxicity assay using a CellTiter 96 Aqueous One solution reagent (Promega) was performed for the test extracts and isolated compounds in HEp-2, HELFs, and HeLa cells as described previously.<sup>50</sup>

**Optical Spectroscopy.** UV absorbance and ECD spectra were collected simultaneously on a 0.02 mg/mL sample in MeOH using a path length of 10 mm on a ChiraScan-Plus instrument (Applied Photophysics) with a scan speed of 30 nm/min under a continuous N<sub>2</sub> flow. The solvent spectra were recorded under identical conditions to remove solvent bands in the UV spectra and to baseline correct the ECD spectra. IR and VCD spectra were recorded simultaneously on a ChiraIIR-2X (Biotools) equipped with a dual PEM system running at a resolution of 4 cm<sup>-1</sup> and the central PEM frequency set to 1400 cm<sup>-1</sup>. Samples were dissolved in CDCl<sub>3</sub>, in a cell equipped with BaF<sub>2</sub> windows and a path length of 100  $\mu$ m, and a total of 96 000 scans were recorded (32 h). The final experimental VCD spectra were obtained by subtracting the solvent VCD spectra recorded under identical conditions.

Calculations. A low-energy conformation library of postulated compounds 1 and 2 in both the cis and trans configuration was created using PCModel<sup>53</sup> with the incorporated MMFF94 force field. All conformers within a cutoff of 5 kcal/mol from the lowest energy conformer were retained and subjected to DFT optimization and spectral calculations at the B3LYP/6-311++G(d,p) level of theory. DFT calculations were performed on conformers exhibiting a Boltzmann weight higher than 1% (based on the enthalpy). The solvent was implicitly taken into account using the IEFPCM model, as implemented in the Gaussian suite. All DFT level calculations were performed using the Gaussian 16 software package<sup>54</sup> with tight convergence criteria and ultrafine integration grids. For each conformer, IR absorbance and VCD spectra were created by applying a Lorentzian broadening with full width at half-maximum of 10 cm<sup>-1</sup> and subsequent Boltzmann-weighted based on their enthalpies. To compare the calculated spectra with the experimental one, a spectra scaling factor of 0.98 was applied on the calculated frequencies.

# ASSOCIATED CONTENT

#### **Data Availability Statement**

The original FIDs and MestreNova files for all compounds, NMReDATA<sup>55,56</sup> files, and CSEARCH<sup>30</sup> results for the new compounds 1-4, original UV and IR spectra, and DFT-computed conformers for compounds 1 and 2 are freely available on Zenodo (DOI: 10.5281/zenodo.10846050).

#### **Supporting Information**

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.jnatprod.3c01288.

NMR, MS, and optical spectroscopy data for the isolated compounds and antiviral activity and cytotoxicity data (PDF)

## AUTHOR INFORMATION

#### **Corresponding Authors**

Tomas Bergström – Department of Infectious Diseases/ Virology, Institute of Biomedicine, Sahlgrenska Academy, University of Gothenburg, S-413 46 Gothenburg, Sweden; Email: tomas.bergstrom@microbio.gu.se

Abiy Yenesew – Department of Chemistry, University of Nairobi, 00100 Nairobi, Kenya; orcid.org/0000-0002-1123-3200; Email: ayenesew@uonbi.ac.ke

Mate Erdelyi – Department of Chemistry – BMC, Uppsala University, SE-751 23 Uppsala, Sweden; orcid.org/0000-0003-0359-5970; Email: mate.erdelyi@kemi.uu.se

#### Authors

Ivan Kiganda – Department of Chemistry, University of Nairobi, 00100 Nairobi, Kenya; Department of Chemistry – BMC, Uppsala University, SE-751 23 Uppsala, Sweden

Jonathan Bogaerts – Department of Chemistry, University of Antwerp, 2020 Antwerp, Belgium; Orcid.org/0000-0001-8089-7759

Lianne H. E. Wieske – Department of Chemistry – BMC, Uppsala University, SE-751 23 Uppsala, Sweden; orcid.org/0000-0003-4617-7605

**Tsegaye Deyou** – Department of Chemistry, Salale University, QPVQ+6C7 Fitche, Ethiopia

Yoseph Atilaw – Department of Chemistry – BMC, Uppsala University, SE-751 23 Uppsala, Sweden

**Colores Uwamariya** – Department of Infectious Diseases/ Virology, Institute of Biomedicine, Sahlgrenska Academy, University of Gothenburg, S-413 46 Gothenburg, Sweden

Masum Miah – Department of Infectious Diseases/Virology, Institute of Biomedicine, Sahlgrenska Academy, University of Gothenburg, S-413 46 Gothenburg, Sweden

Joanna Said – Department of Infectious Diseases/Virology, Institute of Biomedicine, Sahlgrenska Academy, University of Gothenburg, S-413 46 Gothenburg, Sweden

Albert Ndakala – Department of Chemistry, University of Nairobi, 00100 Nairobi, Kenya

Hoseah M. Akala – Walter Reed Army Institute of Research -Africa (WRAIR-A), Kenya Medical Research Institute (KEMRI), 40100 Kisumu, Kenya

Wouter Herrebout – Department of Chemistry, University of Antwerp, 2020 Antwerp, Belgium; Occid.org/0000-0002-3167-8944

Edward Trybala – Department of Infectious Diseases/ Virology, Institute of Biomedicine, Sahlgrenska Academy, University of Gothenburg, S-413 46 Gothenburg, Sweden

Complete contact information is available at:

## https://pubs.acs.org/10.1021/acs.jnatprod.3c01288

## Notes

The authors declare no competing financial interest.

# ACKNOWLEDGMENTS

We thank the Swedish Research Council (No. 2019-03715 to M.E. and Nos. 2021-06386 and 2018-03918 to T.B.) for financial support. We are grateful to Mr. P. C. Mutiso of Herbarium, Botany Department, University of Nairobi, for the identification of the plant species for the study, and to Dawson McCall and Yasin Katwere for their support of I.K. at the initial stage of this work.

# REFERENCES

(1) Christy, M. P.; Uekusa, Y.; Gerwick, L.; Gerwick, W. H. J. Nat. Prod. 2021, 2021, 161–182.

(2) De Clercq, E.; Li, G. Clin. Microbiol. Rev. 2016, 29, 695-747.

(3) Boldogh, I.; Albrecht, T.; Porter, D. D. Persistent Viral Infections. In *Medical Microbiology*, 4th ed.; Baron, S., Ed.; University of Texas Medical Branch: Galveston, TX, 1996.

(4) Cheng, V. C.-C.; Chan, J. F-W; Hung, I. F. N.; Yiuen, K-Y Viral Infections, an Overview with a Focus on Prevention of Transmission; Elsevier, 2016.

(5) Ghoran, S. H.; Mohamed, E.-S.; Nazim, S.; Anake, K. *Molecules* **2021**, *26*, 3–28.

(6) Sampath, S.; Khedr, A.; Qamar, S.; Tekin, A.; Singh, R.; Green, R.; Kashyap, R. *Cureus* **2021**, *13*, No. e18136.

(7) Prasansuklab, A.; Theerasri, A.; Rangsinth, P.; Sillapachaiyaporn, C.; Siriporn, C.; Tencomnao, T. *J. Tradit. Complement. Med.* **2021**, *11*, 144–157.

(8) Zakaryan, H.; Arabyan, E.; Adrian, O.; Keivan, Z. Arch. Virol. 2017, 162, 2539-2551.

(9) Wang, L.; Song, J.; Liu, A.; Xiao, B.; Li, S.; Wen, Z.; Lu, Y.; Du, G. Nat. Prod. Bioprospect **2020**, 10, 271–283.

(10) Jo, S.; Kim, S.; Shin, D. H.; Kim, M. S. J. Enzyme Inhib. Med. Chem. 2020, 35, 145–151.

(11) Juca, M. M.; Cysne Filho, F. M. S.; de Almeida, J. C.; Mesquita,

D. D. S.; Barriga, J. R. M.; Dias, K. C. F.; Barbosa, T. M.; Vasconcelos, L. C.; Leal, L.; Ribeiro, J. E.; Vasconcelos, S. M. M. *Nat. Prod. Res.* **2020**, *34*, 692–705.

(12) Jena, R.; Rath, D.; Rout, S. S.; Kar, D. M. Saudi Pharm. J. 2020, 28, 1686–1703.

(13) Gomes, G. d. S.; Silva, G. S.; Domingos, L. d. S. S.; Regiglaucia, R. d. O.; Gonçalo, M. d. C. Asian J. Environ. Ecology 2018, 4, 1–10.
(14) Chen, K.; Tang, H.; Zheng, L.; Wang, L.; Xue, L.; Peng, A.-h.; Tang, M.-h.; Long, C.; Chen, X.; Ye, H.-y.; Chen, L.-j. Phytochem. Lett. 2018, 25, 60–64.

(15) Raksat, A.; Maneerat, W.; Andersen, R. J.; Pyne, S. G.; Laphookhieo, S. J. Nat. Prod. 2018, 81, 1835–1840.

(16) Banzouzi, T. J.; Prost, A.; Rajemiarimirabo, M.; Ongoka, P. *Int. J. Botany* **2008**, *4*, 406–420.

(17) Yenesew, A.; Jacob, O. M.; Peter, G. W. Pergamon 1998, 9, 295–300.

(18) Deyou, T.; Marco, M.; Heydereich, M.; Pan, F.; Gruhonjic, A.; Fitzpatrick, P. A.; Koch, A.; Derese, S.; Pelletier, J.; Rissanen, K.; Yenesew, A.; Erdelyi, M. J. Nat. Prod. **2017**, 80, 2060–2066.

(19) Derese, S.; Barasa, L.; Akala, H. M.; Yusuf, A. O.; Kamau, E.; Heydenreich, M.; Yenesew, A. *Phytochem. Lett.* **2014**, *8*, 69–72.

(20) Zingue, S.; Tchoumtchoua, J.; Ntsa, D. M.; Sandjo, L. P.; Cisilotto, J.; Nde, C. B. M.; Winter, E.; Awounfack, C. F.; Ndinteh, D. T.; Clyne, C.; Njamen, D.; Halabalaki, M.; Creczynski-Pasa, T. B. *BMC Complement Altern Med.* **2016**, *16*, 421.

(21) Andres, A.; Donovan, S. M.; Kuhlenschmidt, M. S. J. Nutr Biochem 2009, 20, 563-569.

(22) Kumar, P. K. P.; Priyadharshini, A.; Muthukumaran, S. Mini-Rev. Med. Chem. 2021, 21, 1734–1746. (23) Carlson, D. G.; Weislede, D.; Tallent, W. H. Tetrahedron 1973, 29, 2731–2741.

(24) Ren, Y.; Benatrehina, P. A.; Munoz Acuna, U.; Yuan, C.; Chai, H. B.; Ninh, T. N.; Carcache de Blanco, E. J.; Soejarto, D. D.; Kinghorn, A. D. *Planta Med.* **2016**, *82*, 1096–1104.

(25) Buyinza, D.; Yang, L. J.; Derese, S.; Ndakala, A.; Coghi, P.; Heydenreich, M.; Wong, V. K. W.; Moller, H. M.; Yenesew, A. *Nat. Prod. Res.* **2021**, *35*, 2744–2747.

(26) Panchareon, O.; Athipornchai, A.; Panthong, A.; Taylor, W. C. Chem. Pharm. Bull. 2008, 56, 835–838.

(27) Veitch, N. C.; Sutton, P. S. E.; Kite, G. C.; Ireland, H. E. J. Nat. Prod. 2003, 66, 210–216.

(28) Marco, M.; Deyoua, T.; Gruhonjicb, A.; Hollerand, J.; Duffyd, S.; Heydenreiche, M.; Firtzpatrickc, P. A.; Landbergc, G.; Koche, A.; Deresea, S.; Pelletierf, J.; Averyd, V. M.; Erdélyi, M.; Yenesew, A. *Phytochem. Lett.* **2017**, *21*, 216–220.

(29) Ma, S.; Huang, Y.; Zhao, Y.; Du, G.; Feng, L.; Huang, C.; Li, Y.; Guo, F. *Phytochem. Lett.* **2016**, *16*, 213–218.

(30) Robien, W. Molecules 2021, 26, 3413.

(31) Yenesew, A.; Derese, S.; Midiwo, J. O.; Oketch-Rabah, H. A.; Lisgarten, J.; Palmer, R.; Heydenreich, M.; Peter, M. G.; Akala, H.; Wangui, J.; Liyala, P.; Waters, N. C. *Phytochemistry* **2003**, *64*, 773–

779. (22) Photosic D. Complete L. Sonders M. Militian D. A. Char

(32) Bhandari, P.; Crombie, L.; Sanders, M.; Whiting, D. A. Chem. Commun. 1988, 1085–1086.

(33) Panichpol, K.; Waterman, P. G. Phytochemistry 1978, 17, 1363–1367.

(34) Slade, D.; Ferreira, D.; Marais, J. P. Phytochemistry 2005, 66, 2177–215.

(35) Unai, T.; Yamamoto, I. Agr Biol. Chem. Tokyo **1973**, 37, 897–901.

(36) Starks, C. M.; Williams, R. B.; Norman, V. L.; Lawrence, J. A.; Goering, M. G.; O'Neil-Johnson, M.; Hu, J. F.; Rice, S. M.; Eldridge, G. R. *Phytochem. Lett.* **2011**, *4*, 72–74.

(37) Kostova, I.; Ognyanov, I. Monatsh. Chem. 1986, 117, 689–693.
(38) Vasconcelos, J. N. E.; Santiago, G. M. P.; Lima, J. Q.; Mafezoli, J.; de Lemos, T. L. G.; da Silva, F. R. L.; Lima, M. A. S.; Pimenta, A.

J.; de Lentos, T. L. G.; da Silva, F. K. L.; Lind, M. A. S.; Finlend, A. T. A.; Braz-Filho, R.; Arriaga, A. M. C.; Cesarin-Sobrinho, D. Quim. Nova **2012**, 35, 1097–1100.

(39) Meesakul, P.; Suthiphasilp, V.; Teerapongpisan, P.; Rujanapun, N.; Chaiyosang, B.; Tontapha, S.; Phukhatmuen, P.; Maneerat, T.; Charoensup, R.; Duangyod, T.; Patrick, B. O.; Andersen, R. J.; Laphookhieo, S. *Phytochemistry* **2022**, *204*, No. 113440.

(40) Silva, P. B.; Bernardo, R. R.; Parente, P. J. *Phytochemistry* **1998**, 49, 1787–1789.

(41) Miyano, M. J. Org. Chem. 1970, 35, 246-249.

(42) Szeja, W.; Grynkiewicz, G.; Rusin, A. Curr. Org. Chem. 2016, 21, 218–235.

(43) Torrance, S. J.; Wiedhopf, R. M.; Hoffmann, J. J.; Cole, J. R. *Phytochemistry* **1979**, *18*, 366–368.

(44) Szarek, W. A.; Hay, G. W.; Vyas, D. M.; Ison, E. R.; Hronowski, L. J. J. Can. J. Chem. **1984**, 62, 671–674.

(45) Pikuta, E. V.; Hoover, R. B.; Klyce, B.; Davies, P. C. W.; Davies, P. Instruments, Methods, and Missions for Astrobiology IX; SPIE, 2006; Vol. 6309.

(46) Takatsuki, A.; Nakatani, N.; Morimoto, M.; Tamura, G.; Matsui, M.; Arima, K.; Yamaguchi, I.; Misato, T. *Appl. Microbiol.* **2002**, *18*, 660–667.

(47) Phrutivorapongkul, A.; Lipipun, V.; Ruangrungsi, N.; Watanabe, T.; Ishikawa, T. *Chem. Pharm. Bull.* **2002**, *50*, 534–537.

(48) Andres, A.; Donovan, S. M.; Kuhlenschmidt, M. S. J. Nutr. Biochem. 2009, 20, 563-9.

(49) Azani, N.; Babineau, M.; Bailey, C. D.; Banks, H.; Barbosa, A.; Pinto, R. B.; Boatwright, J.; Borges, L.; Brown, G.; Bruneau, A.; Candido, E.; Cardoso, D.; Chung, K.-F.; Clark, R.; Conceição, A. d.; Crisp, M.; Cubas, P.; Delgado-Salinas, A.; Dexter, K.; Doyle, J.; Duminil, J.; Egan, A.; De La Estrella, M.; Falcão, M.; Filatov, D.; Fortuna-Perez, A. P.; Fortunato, R.; Gagnon, E.; Gasson, P.; Rando, J. G.; Azevedo Tozzi, A. M. G. d.; Gunn, B.; Harris, D.; Haston, E.; Hawkins, J.; Herendeen, P.; Hughes, C.; Iganci, J. V.; Javadi, F.; Kanu, S. A.; Kazempour-Osaloo, S.; Kite, G.; Klitgaard, B.; Kochanovski, F.; Koenen, E. M.; Kovar, L.; Lavin, M.; Roux, M. l.; Lewis, G.; de Lima, H.; López-Roberts, M. C.; Mackinder, B.; Maia, V. H.; Malécot, V.; Mansano, V.; Marazzi, B.; Mattapha, S.; Miller, J.; Mitsuyuki, C.; Moura, T.; Murphy, D.; Nageswara-Rao, M.; Nevado, B.; Neves, D.; Ojeda, D.; Pennington, R. T.; Prado, D.; Prenner, G.; de Queiroz, L. P.; Ramos, G.; Ranzato Filardi, F.; Ribeiro, P.; Rico-Arce, M. d. L.; Sanderson, M.; Santos-Silva, J.; São-Mateus, W. B.; Silva, M. S.; Simon, M.; Sinou, C.; Snak, C.; de Souza, É.; Sprent, J.; Steele, K.; Steier, J.; Steeves, R.; Stirton, C.; Tagane, S.; Torke, B.; Toyama, H.; Cruz, D. T. d.; Vatanparast, M.; Wieringa, J.; Wink, M.; Wojciechowski, M.; Yahara, T.; Yi, T.; Zimmerman, E. *Taxon* 2017, 66, 44–77.

(50) Lundin, A.; Bergström, T.; Trybala, E. Methods Mol. Biol. 2013, 2013, 345–363.

(51) Lundin, A.; Bergström, T.; Bendrioua, L.; Kann, N.; Adamiak, B.; Trybala, E. *Antiviral Res.* **2010**, *88*, 317–324.

(52) Pauwels, R.; Balzarini, J.; Baba, M.; Snoeck, R.; Schols, D.; Herdewijn, P.; Desmyter, J.; De Clercq, E. *J. Virol. Meth.* **1988**, *20*, 309–321.

(53) Aerts, R.; Vanhove, J.; Herrebout, W.; Johannessen, C. Chem. Sci. 2021, 12, 5952–5964.

(54) Frisch, M. J. T. G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Petersson, G. A.; Nakatsuji, H.; Li, X.; Caricato, M.; Marenich, A. V.; Bloino, J.; Janesko, B. G.; Gomperts, R.; Mennucci, B.; Hratchian, H. P.; Ortiz, J. V.; Izmaylov, A. F.; Sonnenberg, J. L.; Williams-Young, D.; Ding, F.; Lipparini, F.; Egidi, F.; Goings, J.; Peng, B.; Petrone, A.; Henderson, T.; Ranasinghe, D.; Zakrzewski, V. G.; Gao, J.; Rega, N.; Zheng, G.; Liang, W.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Throssell, K.; Montgomery, J. A., Jr.; Peralta, J. E.; Ogliaro, F.; Bearpark, M. J.; Heyd, J. J.; Brothers, E. N.; Kudin, K. N.; Staroverov, V. N.; Keith, T. A.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A. P.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Millam, J. M.; Klene, M.; Adamo, C.; Cammi, R.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Farkas, O.; Foresman, J. B.; Fox, D. J. Gaussian 16, Revision C.01; Gaussian, Inc.: Wallingford, CT, 2022.

(55) Kuhn, S.; Wieske, L. H. E.; Trevorrow, P.; Schober, D.; Schlorer, N. E.; Nuzillard, J. M.; Kessler, P.; Junker, J.; Herraez, A.; Fares, C.; Erdelyi, M.; Jeannerat, D. *Magn. Reson. Chem.* **2021**, *59*, 792–803.

(56) Pupier, M.; Nuzillard, J. M.; Wist, J.; Schlorer, N. E.; Kuhn, S.; Erdelyi, M.; Steinbeck, C.; Williams, A. J.; Butts, C.; Claridge, T. D. W.; Mikhova, B.; Robien, W.; Dashti, H.; Eghbalnia, H. R.; Fares, C.; Adam, C.; Kessler, P.; Moriaud, F.; Elyashberg, M.; Argyropoulos, D.; Perez, M.; Giraudeau, P.; Gil, R. R.; Trevorrow, P.; Jeannerat, D. *Magn. Reson. Chem.* **2018**, *56*, 703–715.