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An Analytical Framework for Estimating Drying Shrinkage Strain of OPC Based Hardened Cement Paste

Saeid Babaei^{*(1)a,b,c}, Suresh C. Seetharam^{(2)a}, Arnaud Dizier^{(3)c}, Gunther Steenackers^{(4)b,d} and Bart Craeye^{(5)b,e}

^a Engineered and Geosystems Analysis Unit, Institute for Environment, Health, and Safety, Belgian Nuclear Research Centre (SCK•CEN), Boeretang 200, B-2400 Mol, Belgium.

^b Faculty of Applied Engineering, University of Antwerp, EMIB Research Group, Groenenborgerlaan 171 - 2020 Antwerpen , Belgium

^c EIG, EURIDICE, Belgian Nuclear Research Centre (SCK•CEN), Boeretang 200, B-2400 Mol, Belgium

^d Faculty of Applied Engineering, University of Antwerp | Op3Mech Research Group Groenenborgerlaan 171 - 2020 Antwerpen

^e Odise University College, Industrial Services & Technologies, DUBIT Research Unit, Belgium

(1)* Tel: +32 14 333118, saeid.babaei@uantwerpen.be; saeid.babaei@sckcen.be

(2) Tel: +32 14 333208, suresh.seetharam@sckcen.be

(3) Tel: +32 14 332998, arnaud.dizier@euridice.be

(4) Tel: +32 00 000000, gunther.steenackers@uantwerpen.be

(5) Tel: +32 00 000000, bart.craeye@uantwerpen.be

ABSTRACT

A new analytical framework that relies on minimal inputs and combines a number of existing techniques to estimate reversible drying shrinkage strain of OPC-based materials is presented. This includes a multiscale framework for estimating water (de)sorption isotherm (WSI), an analytical homogenization technique to estimate bulk modulus, and a multi-mechanism based drying shrinkage formulation. The minimal inputs needed are the cement composition, microstructural information and mechanical properties of hydrated phases of hardened cement paste. A pore network model that forms the core module of the multiscale WSI provides a quantitative basis for the drying shrinkage formulation. The unique feature of the framework is that only two calibration parameters are involved: (i) a geometric parameter used in the pore network model, and (ii) a constant in the disjoining pressure relationship, which is set to unity mainly due to a lack of knowledge. Importantly, there is no need to calibrate these parameters for every experiment. **Results from the framework are compared against shrinkage data from literature that encompass both virgin materials (samples that have never been dried prior to the test) and non-virgin materials.** A reasonably good correspondence has been achieved with respect to the non-virgin materials, whereas, the results for the virgin materials are examined mainly to gain qualitative understanding of the role of the microstructure on irreversible deformation and thus to propose a phenomenological model.

KEYWORDS

Hardened cement paste, Drying shrinkage, Poroelasticity, Disjoining pressure, Surface free energy, Multi-mechanism shrinkage, Homogenization, Multiscale

2 1 INTRODUCTION

3 For massive civil engineering concrete structures, the drying shrinkage strain is usually
4 neglected because water exchange with the surrounding environment is very slow and mostly
5 its effect such as cracking is limited to a thin outer layer of the structure [1]. Moreover, in
6 massive structures, peak temperature due to heat of hydration remains only for a few days thus
7 limiting any adverse effect on drying rate [2]. However, this may not necessarily be the case in
8 applications related to massive non-reinforced concrete engineered barriers for deep geological
9 disposal of radioactive waste [3, 4]. In particular, the so-called Supercontainer concept currently
10 under consideration in Belgium, encapsulates within a concrete buffer, high-level radioactive
11 waste (HLW) materials that release decay heat over hundreds of years. Depending on the type
12 of waste, temperatures can reach 100 °C at the interface between the HLW canisters and
13 concrete buffet [5], with an increased tendency to initiate a severe drying front at the interface
14 and further into outer layers of the buffer. Therefore, the knowledge of drying shrinkage strain
15 in the entire relative humidity (h) range becomes necessary. This is in addition to the
16 contribution from other eigenstrains such as thermal and creep strains. In such applications,
17 optimum choice of cement formulation at the design phase is essential and hence *a priori*
18 knowledge of drying shrinkage strain would be a valuable input for numerical assessment of
19 thermo-hydro-mechanical (THM) behaviour of structural concrete. Therefore, an approach that
20 allows *a priori* estimation of drying shrinkage strain of hardened cement paste from the
21 knowledge of cement composition and microstructural characteristics of the material paves a
22 way forward for better understanding of the cracking potential of cementitious components or
23 structures. Such an exercise is not limited to the aforementioned application alone but to other
24 situations where drying shrinkage cracking is a problem, which underlines the necessity the
25 importance and renovation of this framework.

26 The development of predictive models for drying shrinkage strain has significantly advanced
27 in the last half a century (e.g. [6-10]). The basis for most of the advanced approaches rely on
28 the idea of multiple mechanisms operating at different pore scales (Powers [11], Brochard et al.
29 [12], Vandamme et al. [13], Pinson et al., [14], Luan and Ishida [15], Nguyen et al. [16]) and
30 importantly the approaches are relevant for reversible drying shrinkage strains only. The
31 commonly adopted multiple mechanisms approach was in fact originally put forward by Powers
32 [17], who presented a thermodynamic analysis of volumetric shrinkage strain of hardened
33 cement paste attributable to solid surface tension or surface free energy (Eq. 12 in [11]),
34 disjoining pressure (Eq. 17 [11]) and capillary pressure (Eq. 19 in [11]), but only included
35 qualitative examples of individual volumetric strains. Their thermodynamic analysis essentially
36 relates change in Gibb's free energy to water content in different pore classes via Kelvin's law
37 and involves only one unknown constant in the disjoining pressure equation. A fundamental
38 input is the water content in different pore classes: (de)sorption isotherm is the basis for this
39 type of analysis and all similar approaches discussed further. Furthermore, they argue that the
40 capillary pressure term represents the combined effect of both disjoining and capillary pressure
41 for capillary pore range, but capillary pressure is not applicable for lower humidity range (~
42 <0.45), where only disjoining pressure is dominant. In what follows, particular attention is paid
43 to the state of the art multi-mechanisms models for reversible drying shrinkage strain similar to
44 that of Powers [11].

45 Coussy [9] showed that the capillary pressure alone cannot capture observed total volumetric
46 strain of hardened cement paste and thus introduced an additional interfacial energy term,
47 whose value increases with decrease in h . However, they conclude that their macroscopic
48 approach of combining capillary pressure and interfacial energy fails to capture the macroscopic
49 strain below relative humidity of 50-40%. Luan and Ishida [15] and Rezvani [18] used a multi-
50 mechanism approach similar to Powers [11], in which they consider contribution of shrinkage

51 strains from capillary pressure and disjoining pressure only. In particular, Luan and Ishida [15]
52 argue that the effect of surface energy is only relevant at very low h and that the change in
53 disjoining pressure can be regarded as being equivalent to the change in surface energy at
54 complete desorption. They demonstrate excellent agreement with measured uniaxial shrinkage
55 strains for cement paste at two W/C ratios. Pinson et al.[14] also follow similar idea as Powers
56 [11] by proposing three mechanisms operating at three pore classes (capillary, gel and
57 interlayer) to capture total reversible shrinkage strain. Unlike Powers [11] who considers a
58 thermodynamic relationship for the shrinkage contribution due to disjoining pressure, Pinson
59 et al. [14] use a molecular approach plus a calibration factor to capture the shrinkage strain
60 contribution from the interlayer water. They also demonstrate a good agreement with desorption
61 experiment although their approach predicts a transitory swelling upon drying between about
62 30% and 20% RH. More recently, Nguyen et al. [16] proposed a multi-mechanism drying
63 shrinkage approach similar to Powers [11]. Starting from Gibb's free energy equation, they
64 derive a three-term equivalent pore pressure equation representing shrinkage contribution from
65 capillary pressure, surface free energy and disjoining pressure, which are then embedded within
66 a poroelastic theory to estimate the shrinkage strain. Note that their equivalent pore pressure is
67 not the same as Coussy [9], where only capillary and interfacial energy is considered. Two
68 calibration factors enter their drying shrinkage equation, one for the surface energy and the
69 other for disjoining pressure and it appears that they need to be calibrated for each material.
70 They show excellent correspondence with experimental results for Portland cement (CEM I)
71 cement for two different W/C ratios of 0.3 and 0.47. Finally, an interesting approach, which
72 does not belong to the afore mentioned multi-mechanism approaches, is that of Vlahinić et al.
73 [19] who proposes a constitutive model for drying of an elastic porous material based on the
74 Bishop [20] effective stress theory. In their approach, instead of pressure averaging, they
75 consider weakening of the solid as a function of drying (degree of saturation). They also show

76 an excellent agreement against a second cycle drying experiment on a 56-day-old cement paste
77 sample. However, their model is valid under conditions where solid surface energy does not
78 play an important role in deformation and where capillary pressure is dominant, in other words,
79 h values above about 50% for hardened cement paste.

80 In conclusion, the validity of the multi-mechanism approach and the importance of sorption
81 isotherm is sufficiently justified for drying shrinkage predictions. Keeping in mind the intended
82 objective, which is to estimate drying shrinkage behaviour from cement composition, the study
83 presented in this paper deviates from the aforementioned literatures in the following aspects:

- 84 i. A multiscale water (de)sorption isotherm framework (WSI) is invoked to estimate water
85 content in different pore classes [21] (Section 2.1).
- 86 ii. An analytical homogenization approach principally based on Christensen [22, 23] is
87 implemented to evaluate both the solid and bulk effective modulus of hardened cement
88 paste (Section 2.2).
- 89 iii. A reversible drying shrinkage formulation is adopted comprising the Biot-Bishop's
90 poroelasticity [20, 24], Bangham's relationship [14, 25] and Power's thermodynamic
91 relationship [11] (Section 0).
- 92 iv. The role of microstructure on irreversible shrinkage strain is explored resulting in a
93 phenomenological model that should be seen as a first approximation (Section 3.4.2).

94 The performance of the analytical framework is examined against a wide variety of drying
95 shrinkage tests from literature, where complete data are available.

96 2 ANALYTICAL FRAMEWORK

97 An analytical framework for estimating drying shrinkage strain of hardened cement paste is
98 implemented by combining existing approaches/models as follows (Figure 1):

- 99 i. An existing cement hydration kinetics model, Virtual Cement and Concrete Testing
100 Laboratory (VCCTL), is used to estimate degree of hydration and volume fractions of
101 Portlandite, C-S-H and capillary porosity based on the initial composition of the
102 material. With the resultant degree of hydration, the volume fractions of high density
103 (HD) and low density (LD) C-S-H is estimated via Jennings-Tennis's hydration model
104 [26]. Depending on the ratio of HD and LD C-S-H, the porosity of the gel pore space is
105 also derived ([21]). These volume fractions are used in estimating effective bulk
106 modulus of the material (step (iii) below).
- 107 ii. A recently developed multiscale framework for estimating water desorption isotherm
108 (WSI) [21] based on an integration of a number of models, which also includes step (i)
109 above. This is the fundamental input necessary for computing drying shrinkage strain
110 of the material for all the mechanisms considered.
- 111 iii. An existing analytical homogenization scheme is invoked to compute effective bulk
112 modulus of the material based on inputs from (i) above. This parameter is an essential
113 input for the unsaturated poromechanics theory to compute drying shrinkage strain due
114 to capillary forces.
- 115 iv. An existing approach to estimate drying shrinkage strain principally based on the multi-
116 mechanism approach proposed by Powers [11], which is based on thermodynamic
117 equilibrium. The basic premise is that the total drying shrinkage strain can be attributed
118 to a number of co-existing forces such as capillary, surface tension and disjoining
119 pressure that operate at different relative humidity ranges, which are directly associated
120 with the underlying pore size heterogeneity.

121 Of the above, only (iii) and (iv) are described in detail, whereas (i) and (ii) have already been
122 dealt with in [21] but briefly covered in Section 2.1.

123 **2.1 DESORPTION ISOTHERM FROM A MULTISCALE APPROACH**

124 Babaei et al. [21] presented a multiscale framework to estimate desorption isotherm via the
125 integration of the following models: (i) particle packing, (ii) cement hydration kinetics, and (iii)
126 pore network. The first two models provide inputs for constructing pore size distribution as well
127 as volume fractions of various pores, viz., gel (HD C-S-H, LD C-S-H) and capillary pores. The
128 pore network model uses Kelvin's equation to determine distribution of equilibrium water
129 content in the network for different increments of capillary pressure, P_c , in other words, the
130 desorption isotherm for a given cement paste. For the shrinkage strain due to capillary forces,
131 the desorption isotherm (i.e. P_c vs. S_w) provides direct input as required by Equation (7). For
132 the shrinkage strain due to surface tension, the pore network model not only provides
133 equilibrium volumetric water content (θ) as a function of P_c (or h), but also the volume of empty
134 pores with surface adsorbed water, which is needed to compute σ as surface area of emptied
135 pore per volume of porous material in Equation (10). For the disjoining pressure, the pore
136 network model provides equilibrium water content (weight), w_d in pores smaller than 2.75 nm
137 as a function of P_c (or h) (i.e. for $h < 0.45$) as required by Equation (12).

138 **2.2 EFFECTIVE BULK MODULUS FROM ANALYTICAL** 139 **HOMOGENIZATION**

140 The effective bulk modulus of cement paste, K_b , is estimated using an analytical
141 homogenization approach proposed by Christensen [22, 23] for two-phase material, which is
142 based on Hashin's [27] composite spheres assemblage (CSA) model. The above can be
143 generalized to a multiphase system as shown in Xi and Jennings [10]:

$$K_{s,eff} = K_{s,i} + \frac{S_{i-1,1}[(K_{s,eff})_{i-1} - K_{s,i}]}{1 + (1 - S_{i-1,1}) \frac{(K_{s,eff})_{i-1} - K_{s,i}}{K_{s,i} + \frac{4}{3} G_i}} \quad (1)$$

144 where $K_{s,i}$ and G_i are the bulk and shear modulus of different phases, respectively, and s is the
 145 volume fraction defined as:

$$s_{i-1,i} = \frac{\sum_{j=1}^{i-1} f_j}{\sum_{j=1}^i f_j} \quad \text{from } i = 2 \text{ to } i = N - 1 \quad (2)$$

$$s_{N-1,N} = 1 - f_N$$

146 f_i is the volume fraction of phase i such that:

$$\sum_{j=1}^N f_j = 1 \quad (3)$$

147 The homogenization sequence is illustrated in Figure 2. The first step computes effective bulk
 148 modulus of C-S-H gel by considering HD C-S-H and LD C-S-H as the two phases. The effect
 149 of gel pores in these phases are reflected in their stiffness values. The second step computes the
 150 effective bulk modulus of cement paste by considering a three-phase system: homogenized C-
 151 S-H gel obtained from the first step, Portlandite plus other crystalline hydration products, and
 152 the anhydrous cement grains.

153 The effective bulk modulus of solid skeleton is calculated using the abovementioned technique
 154 but to calculate the bulk modulus of porous structure, i.e. including capillary pores, Hashin and
 155 Shtrikman [28] found the effective bulk modulus for two-phase composite where voids are
 156 considered as a separate phase as follows:

$$K_b = K_{s,eff} \left(\frac{1 - \eta_c}{1 + \eta_c} \right) \quad (4)$$

157 Equation (4) was further modified as [29, 30]:

$$K_b = K_{s,eff}(1 - \eta_c)^2 \quad (5)$$

158 where η_c is the capillary porosity.

159 **2.3 DRYING SHRINKAGE**

160 Based on the proposal by Powers [11], the total shrinkage strain in pure OPC material may be
161 attributed to three main mechanisms[11, 17, 31] :

- 162 i. **Capillary forces:** Capillary water in pores are in a state of tension, which results in
163 compressive stress in the solid phase, thus causing shrinkage of the material (Powers
164 [31]). Powers [11] reasoned that capillary water cannot exist at h lower than
165 approximately 0.45 because at this humidity only pores roughly above 2.6 nm will be
166 de-saturated (or in equilibrium with $h=0.45$) on the basis of Kelvin-Laplace's equation.
167 However, pores below 2.6 nm will be under the influence of strong interfacial forces
168 (see point (iii) below) such that capillary menisci cannot be formed. Hence, the
169 capillary-condensation theory is not valid anymore to estimate the drying shrinkage
170 strain due to capillary forces. Therefore, shrinkage strain due to capillary forces (ϵ_{vc}) is
171 postulated to operate in the relative humidity range 0.45 to 1.
- 172 ii. **Solid surface tension:** Adsorption or desorption of water molecules on the surface of
173 hardened cement microstructure is accompanied by a change in surface tension or
174 equivalently surface free energy of the material. More specifically, there will be a
175 decrease in energy during adsorption and an increase in energy during desorption. It is
176 well documented that this change of energy is accompanied by volumetric strain (e.g.
177 [32-34]). It is possible to relate the change in surface free energy to the change in vapour
178 pressure by means of Gibb's equation ([25], [11] and [32]) and thus to the volumetric
179 strain. Shrinkage strain due to solid surface tension (ϵ_{vs}) is postulated to operate in the
180 entire relative humidity range of 0 to 1. This assumption is reasonable because at any

181 given humidity there will always be pores that will have adsorbed layer of water in a
182 given representative volume element. Note that both Feldman and Sereda [32] and
183 Pinson et al. [14] also consider it to be operative in the entire relative humidity range. It
184 is however unclear if Powers [11, 17] considered the contribution of surface tension to
185 the drying shrinkage strain above $h=0.45$.

186 iii. **Disjoining pressure:** In the specific case of overlapping interfacial regions such as a
187 thin layer of adsorbed water between two solid surfaces, the difference in the hydrostatic
188 pressure of the adsorbed water in the interlayer and contiguous bulk water from which
189 the adsorbed water phase was formed is referred to as the disjoining pressure [35, 36],
190 and it is a function of thickness of the interlayer, and RH and temperature of the
191 surrounding environment. For the disjoining pressure to be non-zero, the distance
192 between the two solid surfaces must be less than a certain threshold value. For the case
193 of hardened cement paste, Powers [11] estimated this value to be around 2.6 nm. He
194 also estimated the mean inter-particle distance for the gel pores to be roughly 1.3 nm,
195 which implies that the disjoining pressure can be active in majority of the gel pore space.
196 This also implies that in this pore space, van der Waals attractive forces dominate giving
197 rise to compressive forces between opposite surfaces, which are counter balanced by
198 the disjoining pressure and the compressive stress of the solid phase (Powers, 1968
199 [17]). Therefore, it is imperative that any loss of water in the pore space due to drying
200 is likely to result in shrinkage of the material. In light of the reasoning in point (i) above,
201 the volumetric shrinkage strain due to disjoining pressure (ϵ_{vd}) is postulated to operate
202 in the relative humidity range 0 to 0.45.

203 In the absence of external load and generally observed small strain (Pinson et al. [14]), the total
204 reversible volumetric drying shrinkage strain ($\epsilon_{v,r}$) can be mathematically expressed as:

$$\varepsilon_{v,r} = \varepsilon_{vc} + \varepsilon_{vs} + \varepsilon_{vd} \quad (6)$$

205

206 **2.3.1 SHRINKAGE STRAIN DUE TO CAPILLARY FORCES ($0.45 < h < 1$)**

207 Assuming pore air pressure (u_a) to be significantly smaller than pore water pressure (u_w), ε_{vc}
 208 can be derived from the Bishop's "single effective stress" constitutive equation [20, 37] :

$$\varepsilon_{vc} = \frac{\chi P_c \alpha_B}{K_b} \quad (7)$$

$$\alpha_B = \left(1 - \frac{K_b}{K_s}\right) \quad (8)$$

$$P_c = (u_a - u_w) = \frac{RT}{M v_w} \ln(h) \quad (9)$$

209 where χ is the Bishop's effective stress parameter taken as equal to the degree of water
 210 saturation (S_w), α_B is the Biot's coefficient, P_c is the capillary pressure (Pa), K_b is the bulk
 211 modulus of the skeleton (Pa) and K_s is the bulk modulus of the solid phase (C-S-H) (Pa), R is
 212 the gas constant (J/mol/K), T is the temperature (K), M is the molar mass of water (g/mol), v_w
 213 is the specific volume of water (m³/kg).

214 Especially, within the geomechanical/geotechnical community there are numerous discussions
 215 on χ as well as applicability of single effective stress approach, which is beyond the scope of
 216 this paper. Readers are referred to reviews by Jennings and Burland [38] and Nuth and Laloui
 217 [39] concerning the single effective stress approach for partially saturated soils and the
 218 difficulties in measuring a unique value of χ , and Vlahinic et al. [19] concerning the derivation
 219 and interpretation of χ from micro-poromechanics. Nevertheless, Eq. (7) has been successfully
 220 applied by Di Bella et al. [40] and appears to be fairly accurate for second cycle (or reversible
 221 part) of drying but only at $h > 0.5$.

222 **2.3.2 SHRINKAGE STRAIN DUE TO SOLID SURFACE TENSION ($0 < h < 1$)**

223 This study is similar to Pinson's [14] approach, which is essentially the Bangham equation [25]
224 that describes volumetric strain from change of surface tension (surface free energy), ε_{vs} :

$$\varepsilon_{vs} = \frac{\Delta(\sigma\gamma)}{K_b} \quad (10)$$

225 where σ is the surface area of emptied pores per volume of porous material, which unlike Pinson
226 [14], is directly obtained from the pore network model (Section 2.1). γ is the surface free energy
227 of solid that is equal to additional surface tension of pore wall to the adsorbed water [14, 34]
228 layer and it is computed via:

$$\gamma = \gamma_0 - \frac{RT}{M} \int_{h_0}^h \theta \frac{dh}{h} \quad (11)$$

229 where γ_0 is the surface tension at h_0 , θ is the volumetric water content of the surface adsorbed
230 water. $h=1$ is considered as the reference state with the corresponding surface tension set equal
231 to the surface tension of bulk water.

232 **2.3.3 SHRINKAGE STRAIN DUE TO DISJOINING PRESSURE ($0 < h < 0.45$)**

233 Based on a thermodynamic analysis, Powers [11] proposed an expression for the volumetric
234 strain due to the disjoining pressure (Eq. (12)):

$$\varepsilon_{vd} = \kappa\beta' \frac{RT}{M\nu_w} \int_{h_1}^{h_2} \frac{w_d}{V_s} d\ln(h) \quad (12)$$

235 where ν_w is the molar volume of water, β' is the coefficient of compressibility of the material
236 under sustained stress, which is taken as the inverse of bulk modulus of cement paste, K_b (Pa),
237 and k is a constant of proportionality, which is taken as unity as a first approximation and w_d
238 is water content in pores smaller than 2.75 nm. V_s is the volume of the adsorbent (m^3) defined
239 as:

$$V_s = V_p(1 - \eta_t) \quad (13)$$

240 where V_p is the volume of cement paste and η_t is the total porosity of the paste.

241 **2.3.4 OTHER MODELS FOR DRYING SHRINKAGE STRAIN**

242 This study is particularly focussed on estimating drying shrinkage strain based on multi-
243 mechanism approach (Section 0 to 2.3.3). However, there are other approaches, in particular,
244 the equivalent pore pressure approach of Coussy et al. [9] and effective bulk modulus approach
245 of Vlahinic et al. [19] that captures these mechanisms in a single framework. These are briefly
246 covered in Appendix-A as the performance of the multi-mechanism approach will be compared
247 with these single framework approaches.

248

249

250 **3 VALIDATION**

251 The analytical framework is validated against a number of available experimental data that
252 encompass total shrinkage strains with and without irreversible strains for various hardened
253 cement pastes [41-44]. The available experimental shrinkage strain is usually the ultimate
254 shrinkage strain, which is an asymptotic value of the hyperbolic shrinkage strain equation as
255 defined, for example, in ACI-209. Recall from Section 2-iv that the shrinkage strain equations
256 (Equations (7), (10) and (12)) are based on thermodynamic equilibrium, which implies that the
257 calculated strains are equilibrium values for a given RH , and hence can be directly compared
258 with the experimental ultimate shrinkage strain. The shrinkage data are available for two types
259 of materials: (i) non-virgin samples that were dried and rewetted to yield total shrinkage strains
260 without irreversible strain component (samples CP1 to CP3), and (ii) virgin samples that were
261 cured (Table 1) right after casting and kept saturated to yield total shrinkage strains, which
262 include irreversible strain component (samples CP4 to CP9). Desorption isotherms are also
263 available for these materials [41]. Recall that the drying shrinkage formulation (Section 0) is
264 only able to estimate reversible shrinkage strain, but not the total shrinkage strain that includes
265 irreversible strain. Nevertheless, the main purpose of comparing the predicted results with the
266 shrinkage experiments of virgin samples is to (i) explore the extent of deviation between the
267 predicted and measured values and (ii) to quantitatively evaluate the role of microstructure on
268 the irreversibility. The chemical composition, curing condition and experimental techniques of
269 the materials (CP1 to CP9) are presented in Table 1.

270 **3.1 CEMENT HYDRATION KINETICS**

271 The results obtained from the cement hydration kinetics model, VCCTL [45], are presented in
272 Table 2, which includes degree of hydration, volume fractions of Portlandite, C-S-H, and
273 capillary porosity at the end of the respective curing periods. Table 2 also includes the volume
274 fractions of HD and LD C-S-H based on Jennings-Tennis's hydration model. Note that the

275 results for the samples CP1-CP3 were already reported in Babaei et al. [21], but reproduced
276 here for immediate reference. As expected, the models predict higher volume fractions of LD
277 C-S-H, capillary porosity and final degree of hydration for compositions with higher water to
278 cement ratio, which are qualitatively consistent with the known behaviour of OPC [26, 46-48].

279 **3.2 WATER DESORPTION ISOTHERMS**

280 Based on the multiscale WSI framework of Babaei et al. [21], desorption isotherms for materials
281 CP1 to CP9 are estimated. Figure 3 and Figure 4 shows a comparison of predicted and
282 experimental results of desorption isotherms for CP1 to CP3 and CP4 to CP9, respectively.
283 Once again note that the results for CP1-CP3 were already discussed in Babaei et al. [21], but
284 reproduced here for immediate reference. For materials CP4 to CP9, it is seen that the predicted
285 results show reasonably good correlation with experimental results. The coefficient of
286 determination ranges from 0.88 to 0.95 for predicted isotherm desorption curves. This increases
287 confidence in the use of the multiscale WSI framework. In other words, with the available
288 knowledge of cement microstructure and the set of models used in the WSI framework, it is
289 possible to arrive at the desorption isotherm directly from cement composition.

290 **3.3 EFFECTIVE BULK MODULUS**

291 Based on the volume fractions of various hydration products (Table 2) and experimental data
292 on Young's modulus and Poisson's ratio of individual phases of the cement paste (Table 3), K_b
293 and K_s of the materials CP1 to CP9 are estimated as shown in Table 4. With the exception of
294 materials CP1, CP2 and CP8, the homogenization technique captures experimental K_b results
295 well. The deviations in the case of CP1, CP2 and CP8 are attributable to the differences between
296 the actual material and the microstructural model results, for instance, with respect to the
297 volume fractions of various phases and ratio of LD and HD C-S-H.

298 3.4 DRYING SHRINKAGE STRAIN

299 3.4.1 NON-VIRGIN MATERIAL - REVERSIBLE STRAIN

300 Figure 5(a)-(c) show a comparison of ultimate drying shrinkage strain of non-virgin materials
301 (CP1 to CP3) as a function of degree of saturation. Note that for CP3, the experimental drying
302 range is above $RH=0.45$ (corresponding $S_w=0.47$), where the disjoining pressure is postulated
303 to be inactive, hence the shrinkage strain attributable to the disjoining pressure is zero. Overall,
304 the predicted values show good correspondence with experimental data with coefficient of
305 determination of 0.98, 0.91 and 0.99 respectively for CP1, CP2 and CP3, although with a slight
306 overestimation for CP1 and CP2 at very low degree of saturation. Even though the WRC for
307 CP1 and CP3 are slightly less accurate, the drying shrinkage strains are reasonably well
308 predicted. However, data concerning experimental uncertainty are not available to completely
309 confirm the degree of accuracy. In relative terms, CP2 shows less overall accuracy based on the
310 coefficient of determination (0.91). Note that CP2 has also the most unconventional
311 composition i.e. $w/c = 0.8$ and is a blended cement. The fundamental input for the construction
312 of pore network originates from the cement hydration kinetics model, which provides volume
313 fractions of various type of pores and hydration products; the latter also linked to the estimation
314 of bulk modulus of the material. The accuracy of the microstructural model for such a blend
315 relies on the extent of calibration (with isothermal calorimetric data) that have been performed
316 with this unconventional material type, which could be one source of uncertainty. The
317 consequence is that the predicted WRC is slightly less accurate in the entire range of degree of
318 saturation. In addition, since CP2 has a lower strength compared to CP1 and CP3, there is a
319 possibility that CP2 has higher microcrack density, which is not captured by the multi-
320 mechanism model.

321 Figure 5(a)-(c) also shows contributions from the three shrinkage mechanisms. The general
322 trend is that the contribution of surface free energy to the shrinkage strain is relatively less than

323 the disjoining and capillary forces, but is still quantitatively important. The exception is
324 however for CP2, where the contribution from surface free energy is more than the capillary
325 forces. The exception is because the total porosity of CP2 is very high 0.47 (W/C=0.8), which
326 is directly accounted for in the σ term in the surface free energy (Equation (10)). Whereas, for
327 the capillary force, the porosity is reflected in two properties: (i) K_b (Equation (5)), and (ii)
328 desorption isotherm. Firstly, although K_b is important, it does not explain the difference even
329 if the predicted K_b is replaced with experimental K_b (Table 4). Secondly, the high porosity
330 results in a desorption isotherm that is characterized by lower capillary pressure for a given
331 degree of saturation, compared to the materials with lower W/C ratios (CP1 and CP3). This
332 results in a lower contribution from the capillary forces to the total shrinkage strain. However,
333 the validity of the assumption $\chi = S_w$ remains questionable.

334 **Figure 6** presents a comparison of results from the analytical framework that includes multi-
335 mechanisms, Coussy et al. [9] that includes interface energy (Appendix A1) and Vlahinic et al.
336 [19] that includes effective bulk modulus (Appendix A2) for CP1-CP3. The coefficient of
337 determination of the predicted results varies from 0.91 to 0.99, 0.21 to 0.97 and 0.49 to 0.79 for
338 the analytical framework, Coussy et al. [9] and Vlahinic et al. [19] respectively, thus offering
339 an improved confidence in the capability of the analytical framework. **Recall that Coussy's**
340 **model (Equation (16)) mainly relies on the WRC ($S_w P_c$) to capture interfacial energies, and is**
341 **also stated to be reliable up to $RH=0.4-0.5$ according to Coussy et al. [9]. It is noted that as long**
342 **as the capillary forces ($S_w P_c$) dominate (Figure 5a and 5c), Coussy's model shows reasonable**
343 **correspondence with experimental data, which is the case with CP1 and CP3, although the**
344 **deviation is much more with the latter. However, for CP2, which has a relatively high W/C=0.8,**
345 **it is shown that the calculated surface forces (Equation (10)) and disjoining pressure (Equation**
346 **(12)) are dominant compared to the capillary forces (Equation (7)) (Figure 5b). Therefore,**
347 **Coussy's model shows considerable deviation, which implies that their interfacial energy term**

348 does not fully compensate for the surface forces and disjoining pressure predicted by Equation
349 (10) and Equation (12), respectively, specifically for high W/C.

350 3.4.2 VIRGIN MATERIAL - TOTAL STRAIN

351 Figure 7 (a)-(f) show a comparison of ultimate drying shrinkage strain of virgin materials (CP4
352 to CP9) as a function of degree of saturation. The predicted values generally show poor
353 correspondence with experimental data except in the higher saturation range ($S_w > 0.8$). The
354 coefficient of determination for the multi-mechanism model ranges from 0.37 to 0.80 with a
355 mean value of 0.64, for Coussy's model it ranges from 0.27 to 0.70 with a mean value of 0.57
356 and for Vlahinic's model it ranges from 0.2 to 0.6 with a mean value of 0.36. This is to be
357 expected because during the first drying permanent deformation occurs (irreversible strain),
358 which accounts for 29% to 40% of the total ultimate shrinkage strain (Table 2). Irreversible
359 shrinkage may include processes such as densification of LD C-S-H [48-51] and/or formation
360 of microcracks [47, 51, 52] that are not captured by the multi-mechanism approach. In
361 particular, it is clear that the strain due to capillary forces, surface free energy and disjoining
362 pressure relies on two fundamental parameters, which are (i) desorption isotherm and (ii) bulk
363 modulus. Firstly, a constant bulk modulus is considered for all the mechanisms and hence no
364 microstructural changes are reflected. While it is possible to consider the variation of the bulk
365 modulus as a function of degree of saturation as in the effective bulk modulus concept
366 (Appendix A1) of Vlahinic et al. [19], it still cannot compensate for the difference between the
367 total strain and reversible strain, for example, as shown in Figure 7 (f) for CP9. Secondly,
368 desorption isotherm is not significantly sensitive to small microstructural changes (Section 3.1
369 in [21]) and thus even though desorption isotherm may be determined on virgin samples, it will
370 still not quantitatively reflect the microstructural changes.

371 To further explore the role of microstructure, irreversible shrinkage strains are extracted from
372 the experimental data for CP4-CP9 by subtracting the total shrinkage strain obtained from the

373 drying and wetting branch of the experimental isotherms at $RH=1$. Table 2 presents the
374 maximum irreversible shrinkage strains for materials CP4-CP9 (column 11). A first observation
375 is that the irreversible shrinkage strain is proportional to the extent of drying. For example,
376 sample CP4 is subjected to more drying ($S_w \approx 0.2$) compared to CP7 ($S_w \approx 0.35$), and accordingly
377 the irreversible shrinkage strain is slightly higher in the case of CP4. It is also seen that the
378 higher the amount of LD C-S-H, the higher is the irreversibility (Figure 8(a) and Figure 8(b)).
379 Jennings [46, 48] argued in his C-S-H conceptual model that drying densifies the low density
380 C-S-H. Thus the experimental results confirm Jennings [46, 48] model. Furthermore, the only
381 shrinking phase in the hardened cement matrix is C-S-H, therefore, it is evident that the volume
382 fraction of C-S-H gel is proportional to the total shrinkage strain. Accordingly, Figure 8(c)
383 shows the irreversible shrinkage strain as a function of product of volume fraction of LD C-S-
384 H and total C-S-H. It is also observed that the surface area of the material has a noticeable
385 influence on shrinkage [14, 32, 53]. Since surface energy is the only force active throughout
386 the whole drying range (Figure 7), it can be a valid candidate for estimating irreversible
387 component of shrinkage. Other factors affecting the irreversibility are the solid bulk modulus
388 and porous bulk modulus, which are functions of volume fraction of various hydration products
389 but mostly C-S-H and porosity. These factors provide a basis to propose a phenomenological
390 approach to account for the volumetric irreversible shrinkage strain, $\varepsilon_{v,irr}$. One proposal could
391 take the form:

$$\varepsilon_{v,irr} = (\varepsilon_{vs} V_{C-S-H} V_{LD\ C-S-H}) / \eta_t \quad (14)$$

392 Where ε_{vs} is shrinkage due to surface free energy, V_{C-S-H} is volume fraction of C-S-H,
393 $V_{LD\ C-S-H}$ is volume fraction of LD C-S-H. which is observed to be at least valid for the six
394 datasets presented in this paper (Figure 8(d)). Adding $\varepsilon_{v,irr}$ with $\varepsilon_{v,r}$ will yield the total drying
395 shrinkage strain as shown in Figure 7 (legend: multi-mechanism). It is seen that the multi-

396 mechanism model results now correspond well with the experimental data for virgin materials,
397 especially for CP6-CP9 with coefficient of determination of 0.99 for all the four. However, the
398 level of accuracy is less satisfactory for CP4 and CP5 whose coefficient of determination are
399 0.86 and 0.91, respectively indicating that the multi-mechanism approach may still be missing
400 some important mechanisms or it is possible that there are some experimental uncertainties. It
401 is important to note that Equation (14) is merely a phenomenological model, which happens to
402 work on these materials and no further conclusion can be made given such small number of
403 data points.

404

405 4 CONCLUSIONS

406 A new analytical framework to estimate drying shrinkage strain for OPC-based materials is
407 presented. As a starting point, the framework principally requires cement composition,
408 microstructural information and mechanical properties of hydrated phases. There are only two
409 calibration parameters: (i) a geometric parameter used in the pore network model, and (ii) a
410 constant in the disjoining pressure relationship, which is set to unity because of a lack of
411 knowledge (hence strictly no calibration). Importantly, there is no need to calibrate these
412 parameters for every experiment. The following specific conclusions are reached:

- 413 i. Predicted desorption isotherms are in good correspondence with **wide ranging**
414 experimental data **from literature**. In this study, six isotherms have been validated,
415 which is in addition to the eleven isotherms already validated by the authors in their
416 previous work Babaei et al. [21, 54], thus offering further confidence in the pore
417 network model that forms the core module of the multiscale WSI framework.
- 418 ii. With some exceptions, the predicted bulk modulus of hardened cement paste is in
419 good agreement w.r.t. the experimental data **from literature**. The deviations are
420 attributed to the uncertainty in the results of the hydration model.
- 421 iii. The chosen drying shrinkage formulation has offered reasonably good results and
422 offers insights into the active mechanisms during drying. In particular, the general
423 trend is that the contribution of surface free energy to the shrinkage strain is relatively
424 less than the disjoining and capillary forces, but is still quantitatively important for
425 accuracy. Moreover, this trend depends on the W/C ratio. The formulation **performs**
426 **generally well** compared to the equivalent pore pressure and effective bulk modulus
427 concepts.
- 428 iv. It is not surprising that the drying shrinkage formulation does not offer satisfactory
429 results w.r.t. experiments on virgin materials, which are subject to first drying cycle.

430 Examining the experimental results vis-à-vis hydration kinetics model suggest that
431 the higher the amount of LD C-S-H, the higher is the irreversibility. A
432 phenomenological model is proposed that quantitatively captures the irreversible
433 shrinkage strain.

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439 paper for their constructive criticism, which improved the quality of this paper.

440 **APPENDIX A**

441 **A1. EQUIVALENT PORE PRESSURE CONCEPT – COUSSY**

442 Coussy et al. [9] used equivalent pore pressure concept to compute drying shrinkage strain. In
443 their model, interface energy, U , was defined as the sum of energy of all the interfaces
444 including, liquid-gas, solid-liquid and solid-gas:

$$U = \int_{S_w}^1 P_c(s) ds \quad (15)$$

445 Equivalent pore pressure, π , is defined via:

$$\pi = P^* - U \quad (16)$$

446 where P^* is the average pore pressure ($S_w P_c$). The drying shrinkage strain is then calculated
447 via:

$$\varepsilon = \frac{\alpha_B \pi}{K_b} \quad (17)$$

448 **A2. EFFECTIVE BULK MODULUS CONCEPT – VLAHINIC ET AL.**

449 Vlahinic et al. [19, 55] proposed a constitutive model, which considers loss of stiffness of the
450 material as the main parameter that dictates the volumetric deformation, which is attributed to
451 microstructural evolution during drying. Their approach deviates from Bishop [20], which
452 considers average pore pressure as the main parameter that dictates the volumetric deformation.
453 The loss of stiffness is thus expressed in the form of reduction of K_s with decrease in degree of
454 saturation, $\bar{K}(S_w)$, which is an experimentally aided estimate as defined in Eq. (19).

$$\varepsilon_{vc} = P_c \left(\frac{1}{K_b} - \frac{1}{\bar{K}(S_w)} \right) \quad (18)$$

$$\bar{K}(S_w) \approx K_s - \frac{K_s - K_b}{\varphi_0} \varphi(S_w) \quad (19)$$

$$\varphi(S_w) = \frac{(1 - S_w)\varphi_0}{1 - S_w\varphi_0} \quad (20)$$

455 where φ is the porosity of the effective solid, φ_0 is the initial porosity and S_w is the degree of
456 water saturation.

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Table 1. Chemical composition of the samples (% mass).

Material code	Material	W/C	C3S	C2S	C3A	C4AF	Curing method	Experimental method	Extent of drying of samples (RH)	Reference
CP1*	CEM II	0.50	0.21	0.53	0.10	0.15	Endogenous curing conditions for 1 year	Drying controlled by saturated salt solutions, T=20°C	0.25	[43]
CP2*	CEM II	0.80	0.21	0.53	0.10	0.15	Endogenous curing conditions for 1 year	Drying controlled by saturated salt solutions, T=20°C	0.30	[43]
CP3*	CEM I	0.45	0.56	0.18	0.06	0.11	Immersion in limewater for 56 days then dried for 270 days and rewetted for 28 days	Drying progressively for 270 days using ASTM C157, T=25 ± 0.2	0.45	[44]
CP4	CEM I	0.55	0.62	0.19	0.07	0.10	Saturated conditions for 91 days (100% RH)	Climate chamber with <i>h</i> control using sodium hydrate solution. T=20°C	0.2	[41, 42]
CP5	CEM I	0.40	0.62	0.19	0.07	0.10	same	same	0.2	[41, 42]
CP6	CEM I	0.55	0.42	0.38	0.04	0.12	same	same	0.2	[41, 42]
CP7	CEM I	0.40	0.42	0.38	0.04	0.12	same	same	0.2	[41, 42]
CP8	CEM I	0.55	0.24	0.62	0.02	0.08	same	same	0.2	[41, 42]
CP9	CEM I	0.40	0.24	0.62	0.02	0.08	same	same	0.2	[41, 42]

* Babaei et al. [21]

Table 2. Results from the cement hydration kinetics model at the end of respective curing periods, including experimental data of shrinkage strains.

Material code	W/C	Volume fraction LD-CSH	Volume fraction HD-CSH	Total C-S-H	Capillary porosity	Total porosity	DOH	Portlandite	Unhydrated Clinker	Other products	Limestone	Experimental		
												Ultimate shrinkage (m ³ /m ³)	Irreversible shrinkage (m ³ /m ³)	Irreversible/ultimate shrinkage
CP1*	0.50	0.28	0.10	0.38	0.21	0.31	0.85	0.11	0.06	0.04	0.10	-0.003721	-	-
CP2*	0.80	0.32	0.05	0.37	0.33	0.42	0.95	0.10	0.02	0.02	0.07	-0.005144	-	-
CP3*	0.45	0.27	0.23	0.50	0.17	0.28	0.82	0.12	0.06	0.05	-	-0.002960	-	-
CP4	0.55	0.39	0.09	0.48	0.21	0.32	0.88	0.12	0.04	0.04	-	-0.004906	-0.001709	0.40
CP5	0.40	0.25	0.27	0.52	0.14	0.25	0.78	0.12	0.08	0.03	-	-0.004106	-0.001219	0.35
CP6	0.55	0.41	0.10	0.51	0.21	0.34	0.88	0.09	0.04	0.02	-	-0.005210	-0.001804	0.36
CP7	0.40	0.25	0.27	0.52	0.15	0.27	0.78	0.10	0.08	0.03	-	-0.004255	-0.001189	0.29
CP8	0.55	0.39	0.10	0.49	0.20	0.33	0.88	0.08	0.07	0.03	-	-0.006282	-0.002673	0.40
CP9	0.40	0.23	0.28	0.51	0.14	0.28	0.78	0.09	0.09	0.03	-	-0.004366	-0.001312	0.29

* Babaei et al. [21]

Table 3. Mechanical properties of hardened cement paste constituents [56-59].

	E (GPa)	ν (-)
C-S-H Gel		
HD C-S-H	29.4±2.4	0.24
LD C-S-H	21.7±2.2	0.24
Cement Paste		
C ₃ S	135	0.3
C ₂ S	140	0.3
C ₃ A	145	0.3
C ₄ AF	125	0.3
CH	38	0.305
Other products	52	0.32

Table 4. Calculated bulk modulus vs experimental data.

Material code	Experimental bulk modulus (K_b) (GPa)	Calculated Bulk modulus (K_b) using proposed model (GPa)	Calculated solid bulk modulus (K_s) using proposed model
CP1	10.5	11.52	18.5
CP2	6.0	7.50	16.6
CP3	12	12.20	17.4
CP4	11.0	11.85	18.9
CP5	13.8	13.67	18.48
CP6	11.0	11.27	18.05
CP7	12.4	12.61	17.45
CP8	9.27	10.34	16.25
CP9	12.8	12.02	16.10

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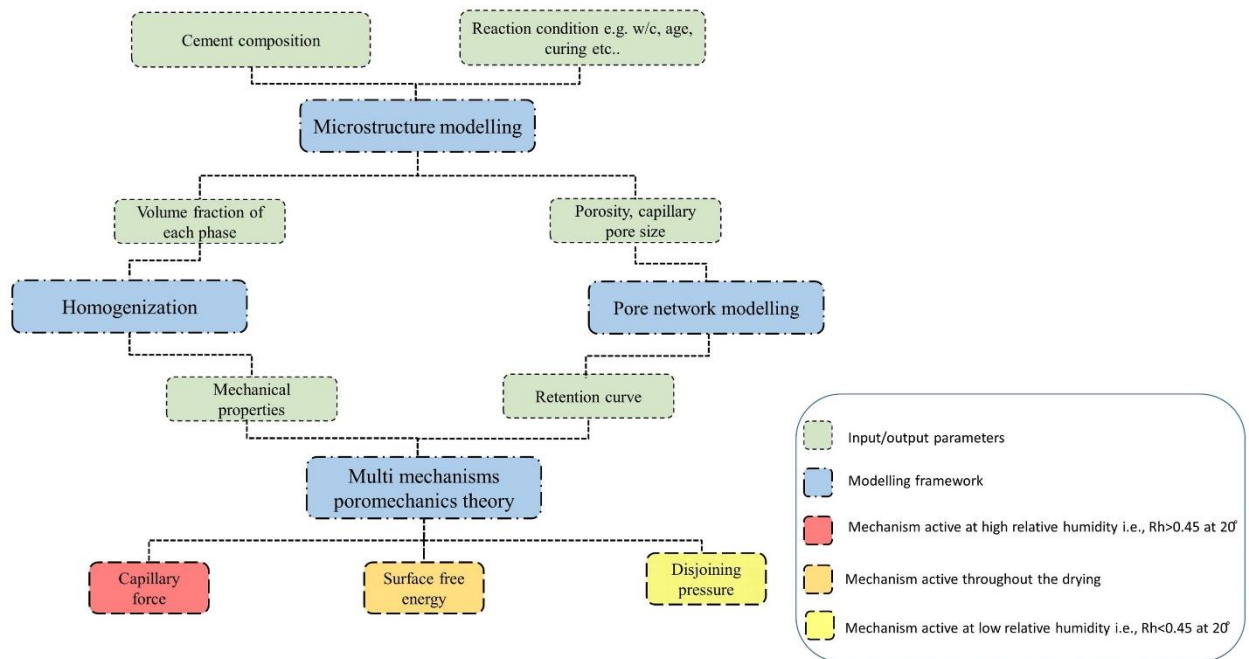


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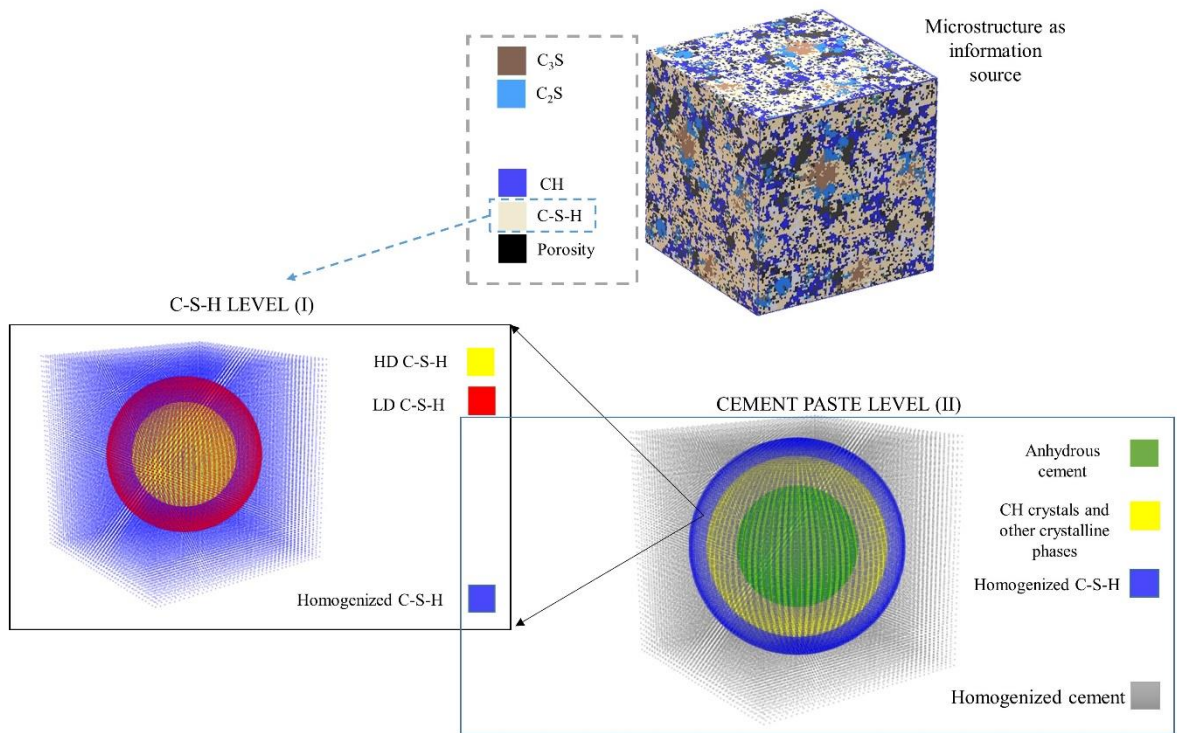


Figure 2. Generalized Homogenization method with its different levels.

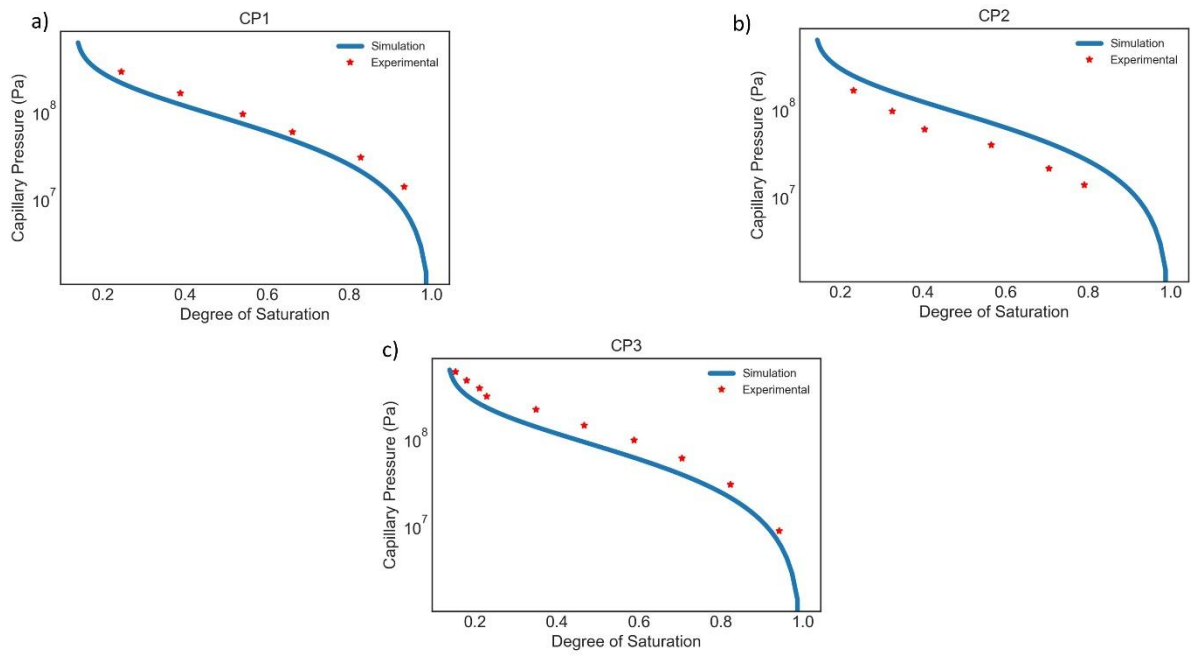


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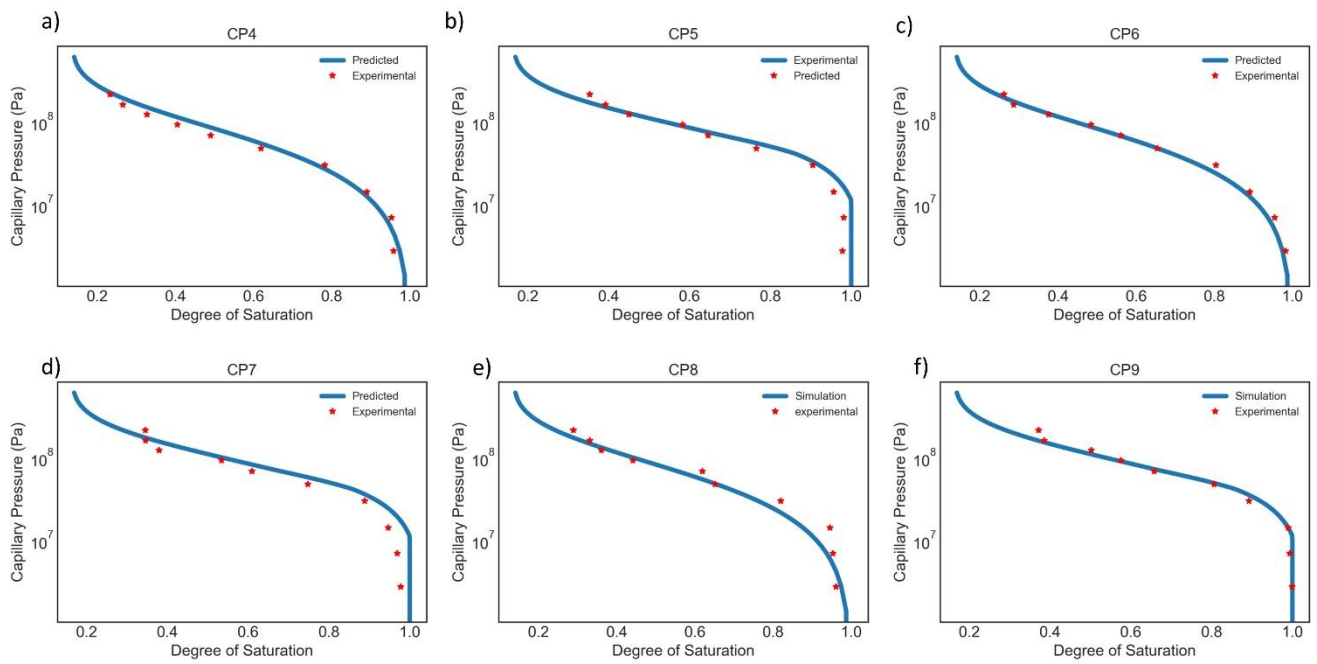


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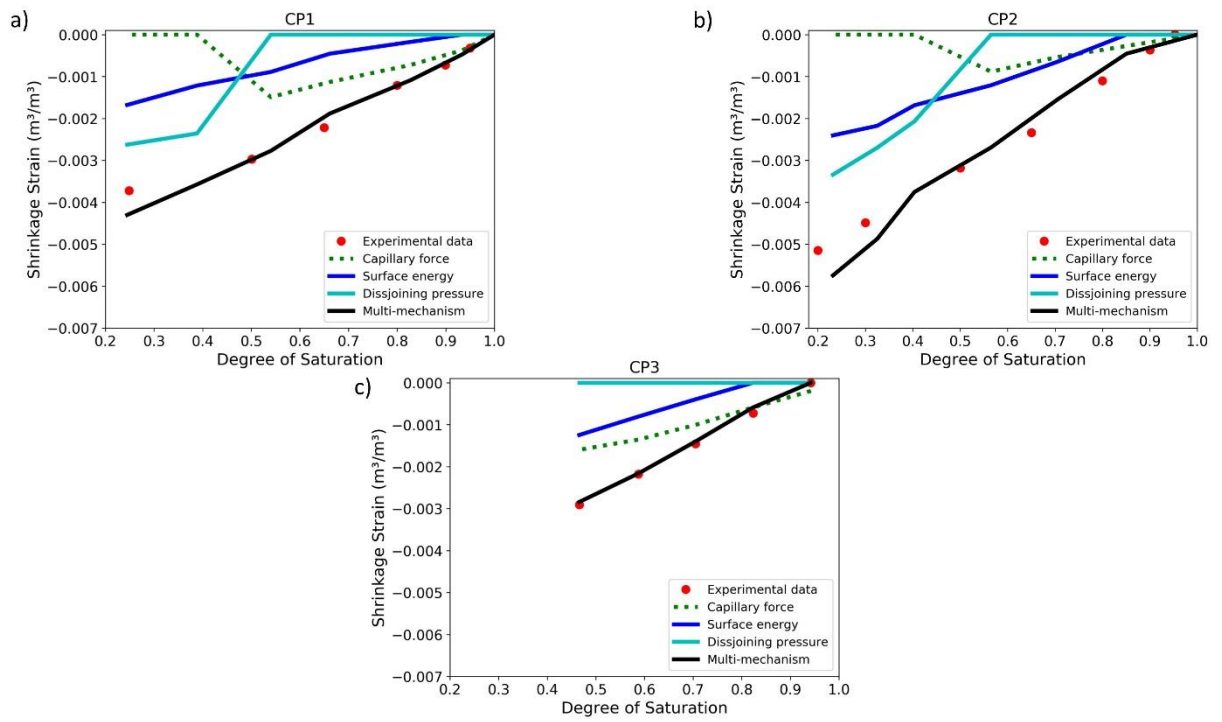


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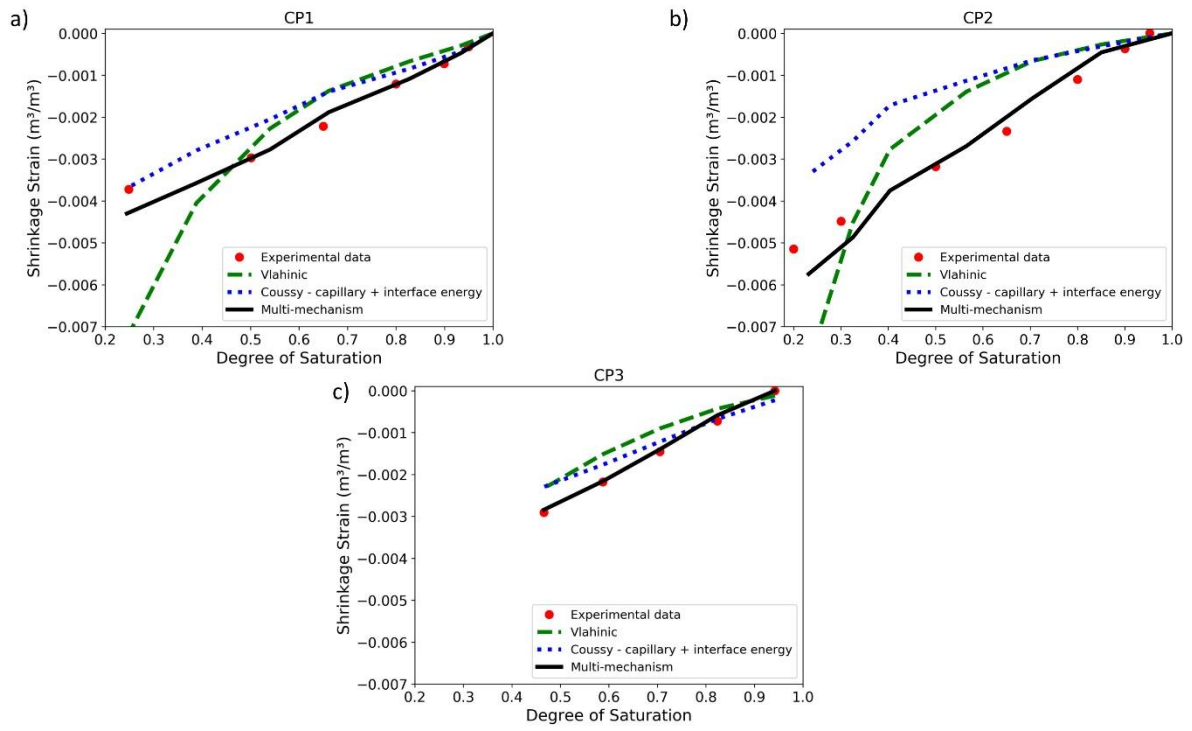


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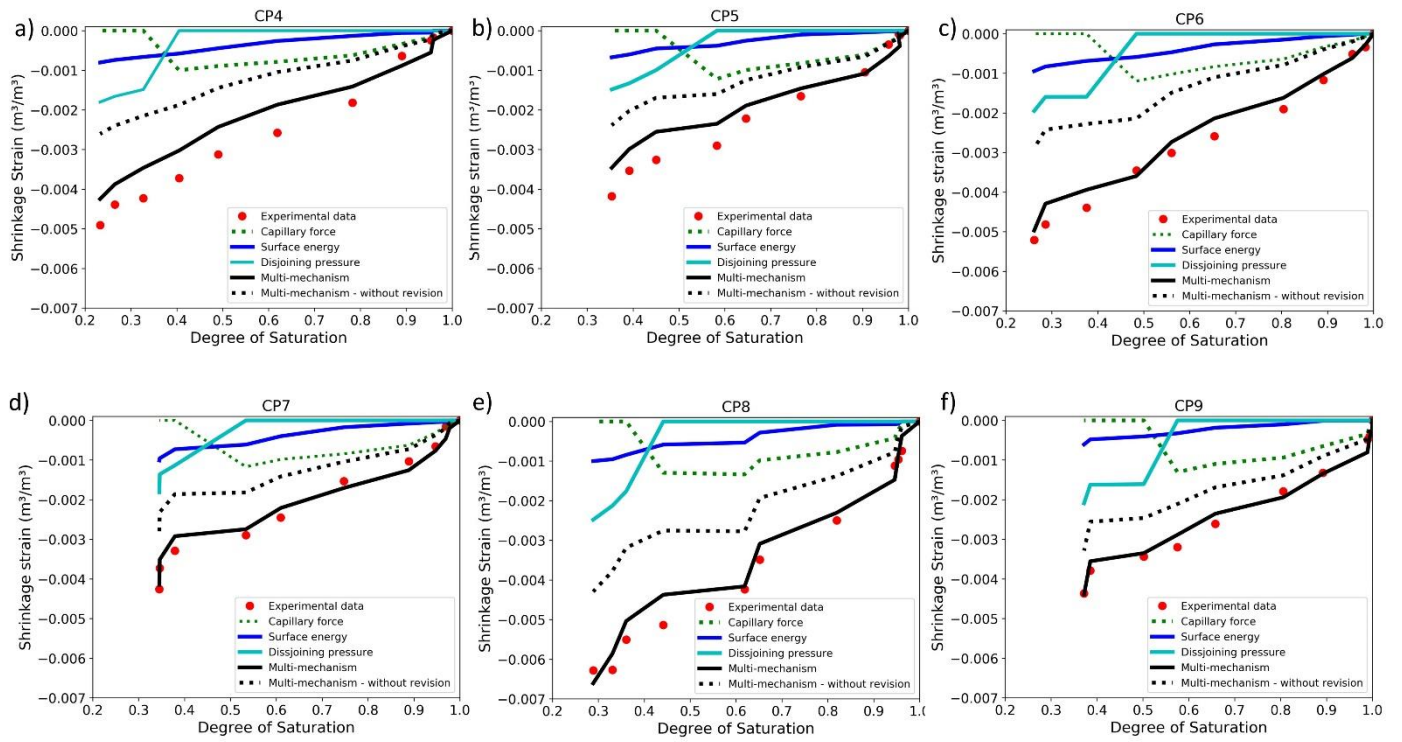


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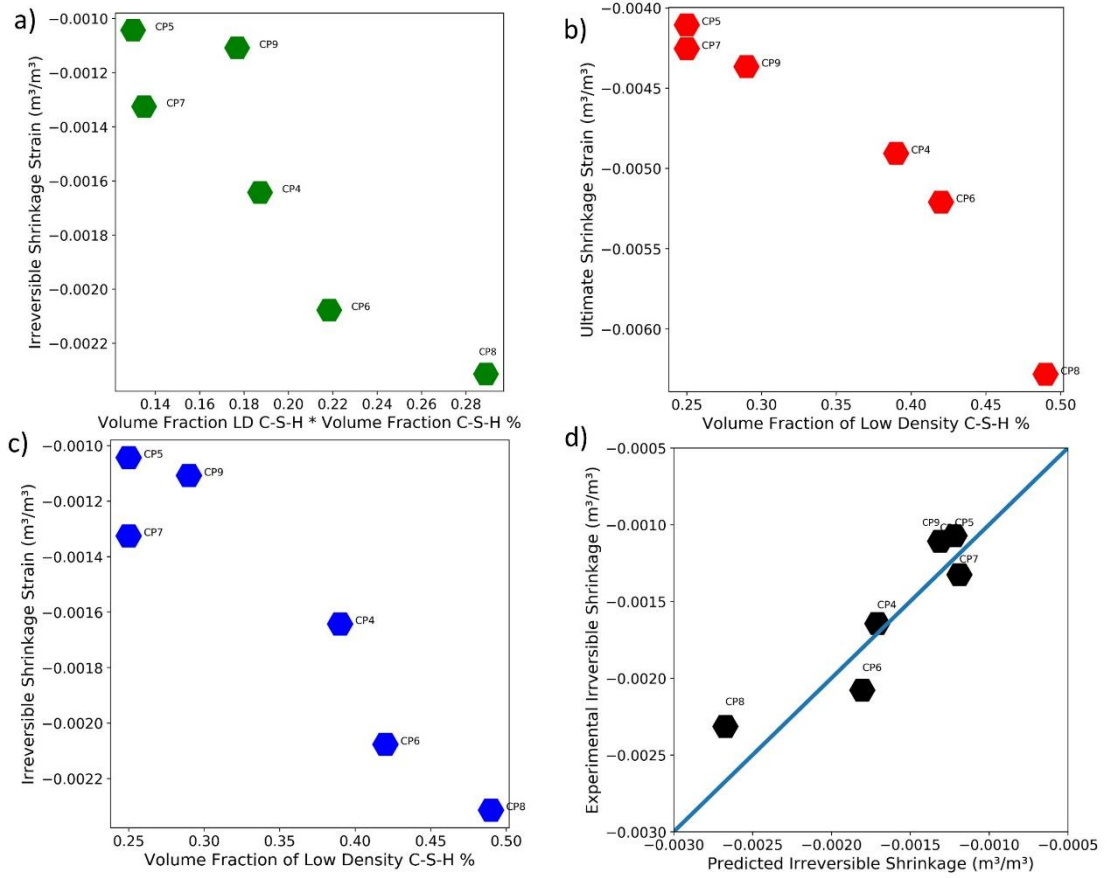


Figure 8. Irreversible shrinkage and its correlation with microstructural information.

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An Analytical Framework for Estimating Drying Shrinkage Strain of OPC Based Hardened Cement Paste

Saeid Babaei^{*(1)a,b,c}, Suresh C. Seetharam^{(2)a}, Arnaud Dizier^{(3)c}, Gunther Steenackers^{(4)b,d} and Bart Craeye^{(5)b,e}

^a Engineered and Geosystems Analysis Unit, Institute for Environment, Health, and Safety, Belgian Nuclear Research Centre (SCK•CEN), Boeretang 200, B-2400 Mol, Belgium.

^b Faculty of Applied Engineering, University of Antwerp, EMIB Research Group, Groenenborgerlaan 171 - 2020 Antwerpen , Belgium

^c EIG, EURIDICE, Belgian Nuclear Research Centre (SCK•CEN), Boeretang 200, B-2400 Mol, Belgium

^d Faculty of Applied Engineering, University of Antwerp | Op3Mech Research Group Groenenborgerlaan 171 - 2020 Antwerpen

^e Odise University College, Industrial Services & Technologies, DUBIT Research Unit, Belgium

(1)* Tel: +32 14 333118, saeid.babaei@uantwerpen.be; saeid.babaei@sckcen.be

(2) Tel: +32 14 333208, suresh.seetharam@sckcen.be

(3) Tel: +32 14 332998, arnaud.dizier@euridice.be

(4) Tel: +32 00 000000, gunther.steenackers@uantwerpen.be

(5) Tel: +32 00 000000, bart.craeye@uantwerpen.be

ABSTRACT

A new analytical framework that relies on minimal inputs and combines a number of existing techniques to estimate reversible drying shrinkage strain of OPC-based materials is presented. This includes a multiscale framework for estimating water (de)sorption isotherm (WSI), an analytical homogenization technique to estimate bulk modulus, and a multi-mechanism based drying shrinkage formulation. The minimal inputs needed are the cement composition, microstructural information and mechanical properties of hydrated phases of hardened cement paste. A pore network model that forms the core module of the multiscale WSI provides a quantitative basis for the drying shrinkage formulation. The unique feature of the framework is that only two calibration parameters are involved: (i) a geometric parameter used in the pore network model, and (ii) a constant in the disjoining pressure relationship, which is set to unity mainly due to a lack of knowledge. Importantly, there is no need to calibrate these parameters for every experiment. Results from the framework are compared against shrinkage data from literature that encompass both virgin materials (samples that have never been dried prior to the test) and non-virgin materials. A reasonably good correspondence has been achieved with respect to the non-virgin materials, whereas, the results for the virgin materials are examined mainly to gain qualitative understanding of the role of the microstructure on irreversible deformation and thus to propose a phenomenological model.

KEYWORDS

Hardened cement paste, Drying shrinkage, Poroelasticity, Disjoining pressure, Surface free energy, Multi-mechanism shrinkage, Homogenization, Multiscale

2 1 INTRODUCTION

3 For massive civil engineering concrete structures, the drying shrinkage strain is usually
4 neglected because water exchange with the surrounding environment is very slow and mostly
5 its effect such as cracking is limited to a thin outer layer of the structure [1]. Moreover, in
6 massive structures, peak temperature due to heat of hydration remains only for a few days thus
7 limiting any adverse effect on drying rate [2]. However, this may not necessarily be the case in
8 applications related to massive non-reinforced concrete engineered barriers for deep geological
9 disposal of radioactive waste [3, 4]. In particular, the so-called Supercontainer concept currently
10 under consideration in Belgium, encapsulates within a concrete buffer, high-level radioactive
11 waste (HLW) materials that release decay heat over hundreds of years. Depending on the type
12 of waste, temperatures can reach 100 °C at the interface between the HLW canisters and
13 concrete buffet [5], with an increased tendency to initiate a severe drying front at the interface
14 and further into outer layers of the buffer. Therefore, the knowledge of drying shrinkage strain
15 in the entire relative humidity (h) range becomes necessary. This is in addition to the
16 contribution from other eigenstrains such as thermal and creep strains. In such applications,
17 optimum choice of cement formulation at the design phase is essential and hence *a priori*
18 knowledge of drying shrinkage strain would be a valuable input for numerical assessment of
19 thermo-hydro-mechanical (THM) behaviour of structural concrete. Therefore, an approach that
20 allows *a priori* estimation of drying shrinkage strain of hardened cement paste from the
21 knowledge of cement composition and microstructural characteristics of the material paves a
22 way forward for better understanding of the cracking potential of cementitious components or
23 structures. Such an exercise is not limited to the aforementioned application alone but to other
24 situations where drying shrinkage cracking is a problem, which underlines the necessity the
25 importance and renovation of this framework.

26 The development of predictive models for drying shrinkage strain has significantly advanced
27 in the last half a century (e.g. [6-10]). The basis for most of the advanced approaches rely on
28 the idea of multiple mechanisms operating at different pore scales (Powers [11], Brochard et al.
29 [12], Vandamme et al. [13], Pinson et al., [14], Luan and Ishida [15], Nguyen et al. [16]) and
30 importantly the approaches are relevant for reversible drying shrinkage strains only. The
31 commonly adopted multiple mechanisms approach was in fact originally put forward by Powers
32 [17], who presented a thermodynamic analysis of volumetric shrinkage strain of hardened
33 cement paste attributable to solid surface tension or surface free energy (Eq. 12 in [11]),
34 disjoining pressure (Eq. 17 [11]) and capillary pressure (Eq. 19 in [11]), but only included
35 qualitative examples of individual volumetric strains. Their thermodynamic analysis essentially
36 relates change in Gibb's free energy to water content in different pore classes via Kelvin's law
37 and involves only one unknown constant in the disjoining pressure equation. A fundamental
38 input is the water content in different pore classes: (de)sorption isotherm is the basis for this
39 type of analysis and all similar approaches discussed further. Furthermore, they argue that the
40 capillary pressure term represents the combined effect of both disjoining and capillary pressure
41 for capillary pore range, but capillary pressure is not applicable for lower humidity range (~
42 <0.45), where only disjoining pressure is dominant. In what follows, particular attention is paid
43 to the state of the art multi-mechanisms models for reversible drying shrinkage strain similar to
44 that of Powers [11].

45 Coussy [9] showed that the capillary pressure alone cannot capture observed total volumetric
46 strain of hardened cement paste and thus introduced an additional interfacial energy term,
47 whose value increases with decrease in h . However, they conclude that their macroscopic
48 approach of combining capillary pressure and interfacial energy fails to capture the macroscopic
49 strain below relative humidity of 50-40%. Luan and Ishida [15] and Rezvani [18] used a multi-
50 mechanism approach similar to Powers [11], in which they consider contribution of shrinkage

51 strains from capillary pressure and disjoining pressure only. In particular, Luan and Ishida [15]
52 argue that the effect of surface energy is only relevant at very low h and that the change in
53 disjoining pressure can be regarded as being equivalent to the change in surface energy at
54 complete desorption. They demonstrate excellent agreement with measured uniaxial shrinkage
55 strains for cement paste at two W/C ratios. Pinson et al.[14] also follow similar idea as Powers
56 [11] by proposing three mechanisms operating at three pore classes (capillary, gel and
57 interlayer) to capture total reversible shrinkage strain. Unlike Powers [11] who considers a
58 thermodynamic relationship for the shrinkage contribution due to disjoining pressure, Pinson
59 et al. [14] use a molecular approach plus a calibration factor to capture the shrinkage strain
60 contribution from the interlayer water. They also demonstrate a good agreement with desorption
61 experiment although their approach predicts a transitory swelling upon drying between about
62 30% and 20% RH. More recently, Nguyen et al. [16] proposed a multi-mechanism drying
63 shrinkage approach similar to Powers [11]. Starting from Gibb's free energy equation, they
64 derive a three-term equivalent pore pressure equation representing shrinkage contribution from
65 capillary pressure, surface free energy and disjoining pressure, which are then embedded within
66 a poroelastic theory to estimate the shrinkage strain. Note that their equivalent pore pressure is
67 not the same as Coussy [9], where only capillary and interfacial energy is considered. Two
68 calibration factors enter their drying shrinkage equation, one for the surface energy and the
69 other for disjoining pressure and it appears that they need to be calibrated for each material.
70 They show excellent correspondence with experimental results for Portland cement (CEM I)
71 cement for two different W/C ratios of 0.3 and 0.47. Finally, an interesting approach, which
72 does not belong to the afore mentioned multi-mechanism approaches, is that of Vlahinić et al.
73 [19] who proposes a constitutive model for drying of an elastic porous material based on the
74 Bishop [20] effective stress theory. In their approach, instead of pressure averaging, they
75 consider weakening of the solid as a function of drying (degree of saturation). They also show

76 an excellent agreement against a second cycle drying experiment on a 56-day-old cement paste
77 sample. However, their model is valid under conditions where solid surface energy does not
78 play an important role in deformation and where capillary pressure is dominant, in other words,
79 h values above about 50% for hardened cement paste.

80 In conclusion, the validity of the multi-mechanism approach and the importance of sorption
81 isotherm is sufficiently justified for drying shrinkage predictions. Keeping in mind the intended
82 objective, which is to estimate drying shrinkage behaviour from cement composition, the study
83 presented in this paper deviates from the aforementioned literatures in the following aspects:

84 i. A multiscale water (de)sorption isotherm framework (WSI) is invoked to estimate water
85 content in different pore classes [21] (Section 2.1).

86 ii. An analytical homogenization approach principally based on Christensen [22, 23] is
87 implemented to evaluate both the solid and bulk effective modulus of hardened cement
88 paste (Section 2.2).

89 iii. A reversible drying shrinkage formulation is adopted comprising the Biot-Bishop's
90 poroelasticity [20, 24], Bangham's relationship [14, 25] and Power's thermodynamic
91 relationship [11] (Section 0).

92 iv. The role of microstructure on irreversible shrinkage strain is explored resulting in a
93 phenomenological model that should be seen as a first approximation (Section 3.4.2).

94 The performance of the analytical framework is examined against a wide variety of drying
95 shrinkage tests from literature, where complete data are available.

96 **2 ANALYTICAL FRAMEWORK**

97 An analytical framework for estimating drying shrinkage strain of hardened cement paste is
98 implemented by combining existing approaches/models as follows (Figure 1):

- 99 i. An existing cement hydration kinetics model, Virtual Cement and Concrete Testing
100 Laboratory (VCCTL), is used to estimate degree of hydration and volume fractions of
101 Portlandite, C-S-H and capillary porosity based on the initial composition of the
102 material. With the resultant degree of hydration, the volume fractions of high density
103 (HD) and low density (LD) C-S-H is estimated via Jennings-Tennis's hydration model
104 [26]. Depending on the ratio of HD and LD C-S-H, the porosity of the gel pore space is
105 also derived ([21]). These volume fractions are used in estimating effective bulk
106 modulus of the material (step (iii) below).
- 107 ii. A recently developed multiscale framework for estimating water desorption isotherm
108 (WSI) [21] based on an integration of a number of models, which also includes step (i)
109 above. This is the fundamental input necessary for computing drying shrinkage strain
110 of the material for all the mechanisms considered.
- 111 iii. An existing analytical homogenization scheme is invoked to compute effective bulk
112 modulus of the material based on inputs from (i) above. This parameter is an essential
113 input for the unsaturated poromechanics theory to compute drying shrinkage strain due
114 to capillary forces.
- 115 iv. An existing approach to estimate drying shrinkage strain principally based on the multi-
116 mechanism approach proposed by Powers [11], which is based on thermodynamic
117 equilibrium. The basic premise is that the total drying shrinkage strain can be attributed
118 to a number of co-existing forces such as capillary, surface tension and disjoining
119 pressure that operate at different relative humidity ranges, which are directly associated
120 with the underlying pore size heterogeneity.

121 Of the above, only (iii) and (iv) are described in detail, whereas (i) and (ii) have already been
122 dealt with in [21] but briefly covered in Section 2.1.

123 **2.1 DESORPTION ISOTHERM FROM A MULTISCALE APPROACH**

124 Babaei et al. [21] presented a multiscale framework to estimate desorption isotherm via the
125 integration of the following models: (i) particle packing, (ii) cement hydration kinetics, and (iii)
126 pore network. The first two models provide inputs for constructing pore size distribution as well
127 as volume fractions of various pores, viz., gel (HD C-S-H, LD C-S-H) and capillary pores. The
128 pore network model uses Kelvin's equation to determine distribution of equilibrium water
129 content in the network for different increments of capillary pressure, P_c , in other words, the
130 desorption isotherm for a given cement paste. For the shrinkage strain due to capillary forces,
131 the desorption isotherm (i.e. P_c vs. S_w) provides direct input as required by Equation (7). For
132 the shrinkage strain due to surface tension, the pore network model not only provides
133 equilibrium volumetric water content (θ) as a function of P_c (or h), but also the volume of empty
134 pores with surface adsorbed water, which is needed to compute σ as surface area of emptied
135 pore per volume of porous material in Equation (10). For the disjoining pressure, the pore
136 network model provides equilibrium water content (weight), w_d in pores smaller than 2.75 nm
137 as a function of P_c (or h) (i.e. for $h < 0.45$) as required by Equation (12).

138 **2.2 EFFECTIVE BULK MODULUS FROM ANALYTICAL** 139 **HOMOGENIZATION**

140 The effective bulk modulus of cement paste, K_b , is estimated using an analytical
141 homogenization approach proposed by Christensen [22, 23] for two-phase material, which is
142 based on Hashin's [27] composite spheres assemblage (CSA) model. The above can be
143 generalized to a multiphase system as shown in Xi and Jennings [10]:

$$K_{s,eff} = K_{s,i} + \frac{S_{i-1,1}[(K_{s,eff})_{i-1} - K_{s,i}]}{1 + (1 - S_{i-1,1}) \frac{(K_{s,eff})_{i-1} - K_{s,i}}{K_{s,i} + \frac{4}{3} G_i}} \quad (1)$$

144 where $K_{s,i}$ and G_i are the bulk and shear modulus of different phases, respectively, and s is the
 145 volume fraction defined as:

$$s_{i-1,i} = \frac{\sum_{j=1}^{i-1} f_j}{\sum_{j=1}^i f_j} \quad \text{from } i = 2 \text{ to } i = N - 1 \quad (2)$$

$$s_{N-1,N} = 1 - f_N$$

146 f_i is the volume fraction of phase i such that:

$$\sum_{j=1}^N f_j = 1 \quad (3)$$

147 The homogenization sequence is illustrated in Figure 2. The first step computes effective bulk
 148 modulus of C-S-H gel by considering HD C-S-H and LD C-S-H as the two phases. The effect
 149 of gel pores in these phases are reflected in their stiffness values. The second step computes the
 150 effective bulk modulus of cement paste by considering a three-phase system: homogenized C-
 151 S-H gel obtained from the first step, Portlandite plus other crystalline hydration products, and
 152 the anhydrous cement grains.

153 The effective bulk modulus of solid skeleton is calculated using the abovementioned technique
 154 but to calculate the bulk modulus of porous structure, i.e. including capillary pores, Hashin and
 155 Shtrikman [28] found the effective bulk modulus for two-phase composite where voids are
 156 considered as a separate phase as follows:

$$K_b = K_{s,eff} \left(\frac{1 - \eta_c}{1 + \eta_c} \right) \quad (4)$$

157 Equation (4) was further modified as [29, 30]:

$$K_b = K_{s,eff}(1 - \eta_c)^2 \quad (5)$$

158 where η_c is the capillary porosity.

159 **2.3 DRYING SHRINKAGE**

160 Based on the proposal by Powers [11], the total shrinkage strain in pure OPC material may be
161 attributed to three main mechanisms[11, 17, 31] :

- 162 i. **Capillary forces:** Capillary water in pores are in a state of tension, which results in
163 compressive stress in the solid phase, thus causing shrinkage of the material (Powers
164 [31]). Powers [11] reasoned that capillary water cannot exist at h lower than
165 approximately 0.45 because at this humidity only pores roughly above 2.6 nm will be
166 de-saturated (or in equilibrium with $h=0.45$) on the basis of Kelvin-Laplace's equation.
167 However, pores below 2.6 nm will be under the influence of strong interfacial forces
168 (see point (iii) below) such that capillary menisci cannot be formed. Hence, the
169 capillary-condensation theory is not valid anymore to estimate the drying shrinkage
170 strain due to capillary forces. Therefore, shrinkage strain due to capillary forces (ϵ_{vc}) is
171 postulated to operate in the relative humidity range 0.45 to 1.
- 172 ii. **Solid surface tension:** Adsorption or desorption of water molecules on the surface of
173 hardened cement microstructure is accompanied by a change in surface tension or
174 equivalently surface free energy of the material. More specifically, there will be a
175 decrease in energy during adsorption and an increase in energy during desorption. It is
176 well documented that this change of energy is accompanied by volumetric strain (e.g.
177 [32-34]). It is possible to relate the change in surface free energy to the change in vapour
178 pressure by means of Gibb's equation ([25], [11] and [32]) and thus to the volumetric
179 strain. Shrinkage strain due to solid surface tension (ϵ_{vs}) is postulated to operate in the
180 entire relative humidity range of 0 to 1. This assumption is reasonable because at any

181 given humidity there will always be pores that will have adsorbed layer of water in a
182 given representative volume element. Note that both Feldman and Sereda [32] and
183 Pinson et al. [14] also consider it to be operative in the entire relative humidity range. It
184 is however unclear if Powers [11, 17] considered the contribution of surface tension to
185 the drying shrinkage strain above $h=0.45$.

186 iii. **Disjoining pressure:** In the specific case of overlapping interfacial regions such as a
187 thin layer of adsorbed water between two solid surfaces, the difference in the hydrostatic
188 pressure of the adsorbed water in the interlayer and contiguous bulk water from which
189 the adsorbed water phase was formed is referred to as the disjoining pressure [35, 36],
190 and it is a function of thickness of the interlayer, and RH and temperature of the
191 surrounding environment. For the disjoining pressure to be non-zero, the distance
192 between the two solid surfaces must be less than a certain threshold value. For the case
193 of hardened cement paste, Powers [11] estimated this value to be around 2.6 nm. He
194 also estimated the mean inter-particle distance for the gel pores to be roughly 1.3 nm,
195 which implies that the disjoining pressure can be active in majority of the gel pore space.
196 This also implies that in this pore space, van der Waals attractive forces dominate giving
197 rise to compressive forces between opposite surfaces, which are counter balanced by
198 the disjoining pressure and the compressive stress of the solid phase (Powers, 1968
199 [17]). Therefore, it is imperative that any loss of water in the pore space due to drying
200 is likely to result in shrinkage of the material. In light of the reasoning in point (i) above,
201 the volumetric shrinkage strain due to disjoining pressure (ϵ_{vd}) is postulated to operate
202 in the relative humidity range 0 to 0.45.

203 In the absence of external load and generally observed small strain (Pinson et al. [14]), the total
204 reversible volumetric drying shrinkage strain ($\epsilon_{v,r}$) can be mathematically expressed as:

$$\varepsilon_{v,r} = \varepsilon_{vc} + \varepsilon_{vs} + \varepsilon_{vd} \quad (6)$$

205

206 **2.3.1 SHRINKAGE STRAIN DUE TO CAPILLARY FORCES ($0.45 < h < 1$)**

207 Assuming pore air pressure (u_a) to be significantly smaller than pore water pressure (u_w), ε_{vc}
 208 can be derived from the Bishop's "single effective stress" constitutive equation [20, 37] :

$$\varepsilon_{vc} = \frac{\chi P_c \alpha_B}{K_b} \quad (7)$$

$$\alpha_B = \left(1 - \frac{K_b}{K_s}\right) \quad (8)$$

$$P_c = (u_a - u_w) = \frac{RT}{M v_w} \ln(h) \quad (9)$$

209 where χ is the Bishop's effective stress parameter taken as equal to the degree of water
 210 saturation (S_w), α_B is the Biot's coefficient, P_c is the capillary pressure (Pa), K_b is the bulk
 211 modulus of the skeleton (Pa) and K_s is the bulk modulus of the solid phase (C-S-H) (Pa), R is
 212 the gas constant (J/mol/K), T is the temperature (K), M is the molar mass of water (g/mol), v_w
 213 is the specific volume of water (m^3/kg).

214 Especially, within the geomechanical/geotechnical community there are numerous discussions
 215 on χ as well as applicability of single effective stress approach, which is beyond the scope of
 216 this paper. Readers are referred to reviews by Jennings and Burland [38] and Nuth and Laloui
 217 [39] concerning the single effective stress approach for partially saturated soils and the
 218 difficulties in measuring a unique value of χ , and Vlahinic et al. [19] concerning the derivation
 219 and interpretation of χ from micro-poromechanics. Nevertheless, Eq. (7) has been successfully
 220 applied by Di Bella et al. [40] and appears to be fairly accurate for second cycle (or reversible
 221 part) of drying but only at $h > 0.5$.

222 **2.3.2 SHRINKAGE STRAIN DUE TO SOLID SURFACE TENSION ($0 < h < 1$)**

223 This study is similar to Pinson's [14] approach, which is essentially the Bangham equation [25]
224 that describes volumetric strain from change of surface tension (surface free energy), ε_{vs} :

$$\varepsilon_{vs} = \frac{\Delta(\sigma\gamma)}{K_b} \quad (10)$$

225 where σ is the surface area of emptied pores per volume of porous material, which unlike Pinson
226 [14], is directly obtained from the pore network model (Section 2.1). γ is the surface free energy
227 of solid that is equal to additional surface tension of pore wall to the adsorbed water [14, 34]
228 layer and it is computed via:

$$\gamma = \gamma_0 - \frac{RT}{M} \int_{h_0}^h \theta \frac{dh}{h} \quad (11)$$

229 where γ_0 is the surface tension at h_0 , θ is the volumetric water content of the surface adsorbed
230 water. $h=1$ is considered as the reference state with the corresponding surface tension set equal
231 to the surface tension of bulk water.

232 **2.3.3 SHRINKAGE STRAIN DUE TO DISJOINING PRESSURE ($0 < h < 0.45$)**

233 Based on a thermodynamic analysis, Powers [11] proposed an expression for the volumetric
234 strain due to the disjoining pressure (Eq. (12)):

$$\varepsilon_{vd} = \kappa\beta' \frac{RT}{Mv_w} \int_{h_1}^{h_2} \frac{w_d}{V_s} d\ln(h) \quad (12)$$

235 where v_w is the molar volume of water, β' is the coefficient of compressibility of the material
236 under sustained stress, which is taken as the inverse of bulk modulus of cement paste, K_b (Pa),
237 and k is a constant of proportionality, which is taken as unity as a first approximation and w_d
238 is water content in pores smaller than 2.75 nm. V_s is the volume of the adsorbent (m^3) defined
239 as:

$$V_s = V_p(1 - \eta_t) \quad (13)$$

240 where V_p is the volume of cement paste and η_t is the total porosity of the paste.

241 **2.3.4 OTHER MODELS FOR DRYING SHRINKAGE STRAIN**

242 This study is particularly focussed on estimating drying shrinkage strain based on multi-
243 mechanism approach (Section 0 to 2.3.3). However, there are other approaches, in particular,
244 the equivalent pore pressure approach of Coussy et al. [9] and effective bulk modulus approach
245 of Vlahinic et al. [19] that captures these mechanisms in a single framework. These are briefly
246 covered in Appendix-A as the performance of the multi-mechanism approach will be compared
247 with these single framework approaches.

248

249

250 **3 VALIDATION**

251 The analytical framework is validated against a number of available experimental data that
252 encompass total shrinkage strains with and without irreversible strains for various hardened
253 cement pastes [41-44]. The available experimental shrinkage strain is usually the ultimate
254 shrinkage strain, which is an asymptotic value of the hyperbolic shrinkage strain equation as
255 defined, for example, in ACI-209. Recall from Section 2-iv that the shrinkage strain equations
256 (Equations (7), (10) and (12)) are based on thermodynamic equilibrium, which implies that the
257 calculated strains are equilibrium values for a given RH , and hence can be directly compared
258 with the experimental ultimate shrinkage strain. The shrinkage data are available for two types
259 of materials: (i) non-virgin samples that were dried and rewetted to yield total shrinkage strains
260 without irreversible strain component (samples CP1 to CP3), and (ii) virgin samples that were
261 cured (Table 1) right after casting and kept saturated to yield total shrinkage strains, which
262 include irreversible strain component (samples CP4 to CP9). Desorption isotherms are also
263 available for these materials [41]. Recall that the drying shrinkage formulation (Section 0) is
264 only able to estimate reversible shrinkage strain, but not the total shrinkage strain that includes
265 irreversible strain. Nevertheless, the main purpose of comparing the predicted results with the
266 shrinkage experiments of virgin samples is to (i) explore the extent of deviation between the
267 predicted and measured values and (ii) to quantitatively evaluate the role of microstructure on
268 the irreversibility. The chemical composition, curing condition and experimental techniques of
269 the materials (CP1 to CP9) are presented in Table 1.

270 **3.1 CEMENT HYDRATION KINETICS**

271 The results obtained from the cement hydration kinetics model, VCCTL [45], are presented in
272 Table 2, which includes degree of hydration, volume fractions of Portlandite, C-S-H, and
273 capillary porosity at the end of the respective curing periods. Table 2 also includes the volume
274 fractions of HD and LD C-S-H based on Jennings-Tennis's hydration model. Note that the

275 results for the samples CP1-CP3 were already reported in Babaei et al. [21], but reproduced
276 here for immediate reference. As expected, the models predict higher volume fractions of LD
277 C-S-H, capillary porosity and final degree of hydration for compositions with higher water to
278 cement ratio, which are qualitatively consistent with the known behaviour of OPC [26, 46-48].

279 **3.2 WATER DESORPTION ISOTHERMS**

280 Based on the multiscale WSI framework of Babaei et al. [21], desorption isotherms for materials
281 CP1 to CP9 are estimated. Figure 3 and Figure 4 shows a comparison of predicted and
282 experimental results of desorption isotherms for CP1 to CP3 and CP4 to CP9, respectively.
283 Once again note that the results for CP1-CP3 were already discussed in Babaei et al. [21], but
284 reproduced here for immediate reference. For materials CP4 to CP9, it is seen that the predicted
285 results show reasonably good correlation with experimental results. The coefficient of
286 determination ranges from 0.88 to 0.95 for predicted isotherm desorption curves. This increases
287 confidence in the use of the multiscale WSI framework. In other words, with the available
288 knowledge of cement microstructure and the set of models used in the WSI framework, it is
289 possible to arrive at the desorption isotherm directly from cement composition.

290 **3.3 EFFECTIVE BULK MODULUS**

291 Based on the volume fractions of various hydration products (Table 2) and experimental data
292 on Young's modulus and Poisson's ratio of individual phases of the cement paste (Table 3), K_b
293 and K_s of the materials CP1 to CP9 are estimated as shown in Table 4. With the exception of
294 materials CP1, CP2 and CP8, the homogenization technique captures experimental K_b results
295 well. The deviations in the case of CP1, CP2 and CP8 are attributable to the differences between
296 the actual material and the microstructural model results, for instance, with respect to the
297 volume fractions of various phases and ratio of LD and HD C-S-H.

298 3.4 DRYING SHRINKAGE STRAIN

299 3.4.1 NON-VIRGIN MATERIAL - REVERSIBLE STRAIN

300 Figure 5(a)-(c) show a comparison of ultimate drying shrinkage strain of non-virgin materials
301 (CP1 to CP3) as a function of degree of saturation. Note that for CP3, the experimental drying
302 range is above $RH=0.45$ (corresponding $S_w=0.47$), where the disjoining pressure is postulated
303 to be inactive, hence the shrinkage strain attributable to the disjoining pressure is zero. Overall,
304 the predicted values show good correspondence with experimental data with coefficient of
305 determination of 0.98, 0.91 and 0.99 respectively for CP1, CP2 and CP3, although with a slight
306 overestimation for CP1 and CP2 at very low degree of saturation. Even though the WRC for
307 CP1 and CP3 are slightly less accurate, the drying shrinkage strains are reasonably well
308 predicted. However, data concerning experimental uncertainty are not available to completely
309 confirm the degree of accuracy. In relative terms, CP2 shows less overall accuracy based on the
310 coefficient of determination (0.91). Note that CP2 has also the most unconventional
311 composition i.e. $w/c = 0.8$ and is a blended cement. The fundamental input for the construction
312 of pore network originates from the cement hydration kinetics model, which provides volume
313 fractions of various type of pores and hydration products; the latter also linked to the estimation
314 of bulk modulus of the material. The accuracy of the microstructural model for such a blend
315 relies on the extent of calibration (with isothermal calorimetric data) that have been performed
316 with this unconventional material type, which could be one source of uncertainty. The
317 consequence is that the predicted WRC is slightly less accurate in the entire range of degree of
318 saturation. In addition, since CP2 has a lower strength compared to CP1 and CP3, there is a
319 possibility that CP2 has higher microcrack density, which is not captured by the multi-
320 mechanism model.

321 Figure 5(a)-(c) also shows contributions from the three shrinkage mechanisms. The general
322 trend is that the contribution of surface free energy to the shrinkage strain is relatively less than

323 the disjoining and capillary forces, but is still quantitatively important. The exception is
324 however for CP2, where the contribution from surface free energy is more than the capillary
325 forces. The exception is because the total porosity of CP2 is very high 0.47 (W/C=0.8), which
326 is directly accounted for in the σ term in the surface free energy (Equation (10)). Whereas, for
327 the capillary force, the porosity is reflected in two properties: (i) K_b (Equation (5)), and (ii)
328 desorption isotherm. Firstly, although K_b is important, it does not explain the difference even
329 if the predicted K_b is replaced with experimental K_b (Table 4). Secondly, the high porosity
330 results in a desorption isotherm that is characterized by lower capillary pressure for a given
331 degree of saturation, compared to the materials with lower W/C ratios (CP1 and CP3). This
332 results in a lower contribution from the capillary forces to the total shrinkage strain. However,
333 the validity of the assumption $\chi = S_w$ remains questionable.

334 Figure 6 presents a comparison of results from the analytical framework that includes multi-
335 mechanisms, Coussy et al. [9] that includes interface energy (Appendix A1) and Vlahinic et al.
336 [19] that includes effective bulk modulus (Appendix A2) for CP1-CP3. The coefficient of
337 determination of the predicted results varies from 0.91 to 0.99, 0.21 to 0.97 and 0.49 to 0.79 for
338 the analytical framework, Coussy et al. [9] and Vlahinic et al. [19] respectively, thus offering
339 an improved confidence in the capability of the analytical framework. Recall that Coussy's
340 model (Equation (16)) mainly relies on the WRC ($S_w P_c$) to capture interfacial energies, and is
341 also stated to be reliable up to $RH=0.4-0.5$ according to Coussy et al. [9]. It is noted that as long
342 as the capillary forces ($S_w P_c$) dominate (Figure 5a and 5c), Coussy's model shows reasonable
343 correspondence with experimental data, which is the case with CP1 and CP3, although the
344 deviation is much more with the latter. However, for CP2, which has a relatively high W/C=0.8,
345 it is shown that the calculated surface forces (Equation (10)) and disjoining pressure (Equation
346 (12)) are dominant compared to the capillary forces (Equation (7)) (Figure 5b). Therefore,
347 Coussy's model shows considerable deviation, which implies that their interfacial energy term

348 does not fully compensate for the surface forces and disjoining pressure predicted by Equation
349 (10) and Equation (12), respectively, specifically for high W/C.

350 **3.4.2 VIRGIN MATERIAL - TOTAL STRAIN**

351 Figure 7 (a)-(f) show a comparison of ultimate drying shrinkage strain of virgin materials (CP4
352 to CP9) as a function of degree of saturation. The predicted values generally show poor
353 correspondence with experimental data except in the higher saturation range ($S_w > 0.8$). The
354 coefficient of determination for the multi-mechanism model ranges from 0.37 to 0.80 with a
355 mean value of 0.64, for Coussy's model it ranges from 0.27 to 0.70 with a mean value of 0.57
356 and for Vlahinic's model it ranges from 0.2 to 0.6 with a mean value of 0.36. This is to be
357 expected because during the first drying permanent deformation occurs (irreversible strain),
358 which accounts for 29% to 40% of the total ultimate shrinkage strain (Table 2). Irreversible
359 shrinkage may include processes such as densification of LD C-S-H [48-51] and/or formation
360 of microcracks [47, 51, 52] that are not captured by the multi-mechanism approach. In
361 particular, it is clear that the strain due to capillary forces, surface free energy and disjoining
362 pressure relies on two fundamental parameters, which are (i) desorption isotherm and (ii) bulk
363 modulus. Firstly, a constant bulk modulus is considered for all the mechanisms and hence no
364 microstructural changes are reflected. While it is possible to consider the variation of the bulk
365 modulus as a function of degree of saturation as in the effective bulk modulus concept
366 (Appendix A1) of Vlahinic et al. [19], it still cannot compensate for the difference between the
367 total strain and reversible strain, for example, as shown in Figure 7 (f) for CP9. Secondly,
368 desorption isotherm is not significantly sensitive to small microstructural changes (Section 3.1
369 in [21]) and thus even though desorption isotherm may be determined on virgin samples, it will
370 still not quantitatively reflect the microstructural changes.

371 To further explore the role of microstructure, irreversible shrinkage strains are extracted from
372 the experimental data for CP4-CP9 by subtracting the total shrinkage strain obtained from the

373 drying and wetting branch of the experimental isotherms at $RH=1$. Table 2 presents the
374 maximum irreversible shrinkage strains for materials CP4-CP9 (column 11). A first observation
375 is that the irreversible shrinkage strain is proportional to the extent of drying. For example,
376 sample CP4 is subjected to more drying ($S_w \approx 0.2$) compared to CP7 ($S_w \approx 0.35$), and accordingly
377 the irreversible shrinkage strain is slightly higher in the case of CP4. It is also seen that the
378 higher the amount of LD C-S-H, the higher is the irreversibility (Figure 8(a) and Figure 8(b)).
379 Jennings [46, 48] argued in his C-S-H conceptual model that drying densifies the low density
380 C-S-H. Thus the experimental results confirm Jennings [46, 48] model. Furthermore, the only
381 shrinking phase in the hardened cement matrix is C-S-H, therefore, it is evident that the volume
382 fraction of C-S-H gel is proportional to the total shrinkage strain. Accordingly, Figure 8(c)
383 shows the irreversible shrinkage strain as a function of product of volume fraction of LD C-S-
384 H and total C-S-H. It is also observed that the surface area of the material has a noticeable
385 influence on shrinkage [14, 32, 53]. Since surface energy is the only force active throughout
386 the whole drying range (Figure 7), it can be a valid candidate for estimating irreversible
387 component of shrinkage. Other factors affecting the irreversibility are the solid bulk modulus
388 and porous bulk modulus, which are functions of volume fraction of various hydration products
389 but mostly C-S-H and porosity. These factors provide a basis to propose a phenomenological
390 approach to account for the volumetric irreversible shrinkage strain, $\varepsilon_{v,irr}$. One proposal could
391 take the form:

$$\varepsilon_{v,irr} = (\varepsilon_{vs} V_{C-S-H} V_{LD C-S-H}) / \eta_t \quad (14)$$

392 Where ε_{vs} is shrinkage due to surface free energy, V_{C-S-H} is volume fraction of C-S-H,
393 $V_{LD C-S-H}$ is volume fraction of LD C-S-H. which is observed to be at least valid for the six
394 datasets presented in this paper (Figure 8(d)). Adding $\varepsilon_{v,irr}$ with $\varepsilon_{v,r}$ will yield the total drying
395 shrinkage strain as shown in Figure 7 (*legend: multi-mechanism*). It is seen that the multi-

396 mechanism model results now correspond well with the experimental data for virgin materials,
397 especially for CP6-CP9 with coefficient of determination of 0.99 for all the four. However, the
398 level of accuracy is less satisfactory for CP4 and CP5 whose coefficient of determination are
399 0.86 and 0.91, respectively indicating that the multi-mechanism approach may still be missing
400 some important mechanisms or it is possible that there are some experimental uncertainties. It
401 is important to note that Equation (14) is merely a phenomenological model, which happens to
402 work on these materials and no further conclusion can be made given such small number of
403 data points.

404

405 **4 CONCLUSIONS**

406 A new analytical framework to estimate drying shrinkage strain for OPC-based materials is
407 presented. As a starting point, the framework principally requires cement composition,
408 microstructural information and mechanical properties of hydrated phases. There are only two
409 calibration parameters: (i) a geometric parameter used in the pore network model, and (ii) a
410 constant in the disjoining pressure relationship, which is set to unity because of a lack of
411 knowledge (hence strictly no calibration). Importantly, there is no need to calibrate these
412 parameters for every experiment. The following specific conclusions are reached:

- 413 i. Predicted desorption isotherms are in good correspondence with wide ranging
414 experimental data from literature. In this study, six isotherms have been validated,
415 which is in addition to the eleven isotherms already validated by the authors in their
416 previous work Babaei et al. [21, 54], thus offering further confidence in the pore
417 network model that forms the core module of the multiscale WSI framework.
- 418 ii. With some exceptions, the predicted bulk modulus of hardened cement paste is in
419 good agreement w.r.t. the experimental data from literature. The deviations are
420 attributed to the uncertainty in the results of the hydration model.
- 421 iii. The chosen drying shrinkage formulation has offered reasonably good results and
422 offers insights into the active mechanisms during drying. In particular, the general
423 trend is that the contribution of surface free energy to the shrinkage strain is relatively
424 less than the disjoining and capillary forces, but is still quantitatively important for
425 accuracy. Moreover, this trend depends on the W/C ratio. The formulation performs
426 generally well compared to the equivalent pore pressure and effective bulk modulus
427 concepts.
- 428 iv. It is not surprising that the drying shrinkage formulation does not offer satisfactory
429 results w.r.t. experiments on virgin materials, which are subject to first drying cycle.

430 Examining the experimental results vis-à-vis hydration kinetics model suggest that
431 the higher the amount of LD C-S-H, the higher is the irreversibility. A
432 phenomenological model is proposed that quantitatively captures the irreversible
433 shrinkage strain.

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439 paper for their constructive criticism, which improved the quality of this paper.

440 **APPENDIX A**

441 **A1. EQUIVALENT PORE PRESSURE CONCEPT – COUSSY**

442 Coussy et al. [9] used equivalent pore pressure concept to compute drying shrinkage strain. In
443 their model, interface energy, U , was defined as the sum of energy of all the interfaces
444 including, liquid-gas, solid-liquid and solid-gas:

$$U = \int_{S_w}^1 P_c(s) ds \quad (15)$$

445 Equivalent pore pressure, π , is defined via:

$$\pi = P^* - U \quad (16)$$

446 where P^* is the average pore pressure ($S_w P_c$). The drying shrinkage strain is then calculated
447 via:

$$\varepsilon = \frac{\alpha_B \pi}{K_b} \quad (17)$$

448 **A2. EFFECTIVE BULK MODULUS CONCEPT – VLAHINIC ET AL.**

449 Vlahinic et al. [19, 55] proposed a constitutive model, which considers loss of stiffness of the
450 material as the main parameter that dictates the volumetric deformation, which is attributed to
451 microstructural evolution during drying. Their approach deviates from Bishop [20], which
452 considers average pore pressure as the main parameter that dictates the volumetric deformation.
453 The loss of stiffness is thus expressed in the form of reduction of K_s with decrease in degree of
454 saturation, $\bar{K}(S_w)$, which is an experimentally aided estimate as defined in Eq. (19).

$$\varepsilon_{vc} = P_c \left(\frac{1}{K_b} - \frac{1}{\bar{K}(S_w)} \right) \quad (18)$$

$$\bar{K}(S_w) \approx K_s - \frac{K_s - K_b}{\varphi_0} \varphi(S_w) \quad (19)$$

$$\varphi(S_w) = \frac{(1 - S_w)\varphi_0}{1 - S_w\varphi_0} \quad (20)$$

455 where φ is the porosity of the effective solid, φ_0 is the initial porosity and S_w is the degree of
456 water saturation.

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Table 1. Chemical composition of the samples (% mass).

Material code	Material	W/C	C3S	C2S	C3A	C4AF	Curing method	Experimental method	Extent of drying of samples (RH)	Reference
CP1*	CEM II	0.50	0.21	0.53	0.10	0.15	Endogenous curing conditions for 1 year	Drying controlled by saturated salt solutions, T=20°C	0.25	[43]
CP2*	CEM II	0.80	0.21	0.53	0.10	0.15	Endogenous curing conditions for 1 year	Drying controlled by saturated salt solutions, T=20°C	0.30	[43]
CP3*	CEM I	0.45	0.56	0.18	0.06	0.11	Immersion in limewater for 56 days then dried for 270 days and rewetted for 28 days	Drying progressively for 270 days using ASTM C157, T=25 ± 0.2	0.45	[44]
CP4	CEM I	0.55	0.62	0.19	0.07	0.10	Saturated conditions for 91 days (100% RH)	Climate chamber with <i>h</i> control using sodium hydrate solution. T=20°C	0.2	[41, 42]
CP5	CEM I	0.40	0.62	0.19	0.07	0.10	same	same	0.2	[41, 42]
CP6	CEM I	0.55	0.42	0.38	0.04	0.12	same	same	0.2	[41, 42]
CP7	CEM I	0.40	0.42	0.38	0.04	0.12	same	same	0.2	[41, 42]
CP8	CEM I	0.55	0.24	0.62	0.02	0.08	same	same	0.2	[41, 42]
CP9	CEM I	0.40	0.24	0.62	0.02	0.08	same	same	0.2	[41, 42]

* Babaei et al. [21]

Table 2. Results from the cement hydration kinetics model at the end of respective curing periods, including experimental data of shrinkage strains.

Material code	W/C	Volume fraction LD-CSH	Volume fraction HD-CSH	Total C-S-H	Capillary porosity	Total porosity	DOH	Portlandite	Unhydrated Clinker	Other products	Limestone	Experimental		
												Ultimate shrinkage (m ³ /m ³)	Irreversible shrinkage (m ³ /m ³)	Irreversible/ultimate shrinkage
CP1*	0.50	0.28	0.10	0.38	0.21	0.31	0.85	0.11	0.06	0.04	0.10	-0.003721	-	-
CP2*	0.80	0.32	0.05	0.37	0.33	0.42	0.95	0.10	0.02	0.02	0.07	-0.005144	-	-
CP3*	0.45	0.27	0.23	0.50	0.17	0.28	0.82	0.12	0.06	0.05	-	-0.002960	-	-
CP4	0.55	0.39	0.09	0.48	0.21	0.32	0.88	0.12	0.04	0.04	-	-0.004906	-0.001709	0.40
CP5	0.40	0.25	0.27	0.52	0.14	0.25	0.78	0.12	0.08	0.03	-	-0.004106	-0.001219	0.35
CP6	0.55	0.41	0.10	0.51	0.21	0.34	0.88	0.09	0.04	0.02	-	-0.005210	-0.001804	0.36
CP7	0.40	0.25	0.27	0.52	0.15	0.27	0.78	0.10	0.08	0.03	-	-0.004255	-0.001189	0.29
CP8	0.55	0.39	0.10	0.49	0.20	0.33	0.88	0.08	0.07	0.03	-	-0.006282	-0.002673	0.40
CP9	0.40	0.23	0.28	0.51	0.14	0.28	0.78	0.09	0.09	0.03	-	-0.004366	-0.001312	0.29

* Babaei et al. [21]

Table 3. Mechanical properties of hardened cement paste constituents [56-59].

	<i>E</i> (GPa)	<i>v</i> (-)
C-S-H Gel		
HD C-S-H	29.4±2.4	0.24
LD C-S-H	21.7±2.2	0.24
Cement Paste		
C ₃ S	135	0.3
C ₂ S	140	0.3
C ₃ A	145	0.3
C ₄ AF	125	0.3
CH	38	0.305
Other products	52	0.32

Table 4. Calculated bulk modulus vs experimental data.

Material code	Experimental bulk modulus (K_b) (GPa)	Calculated Bulk modulus (K_b) using proposed model (GPa)	Calculated solid bulk modulus (K_s) using proposed model
CP1	10.5	11.52	18.5
CP2	6.0	7.50	16.6
CP3	12	12.20	17.4
CP4	11.0	11.85	18.9
CP5	13.8	13.67	18.48
CP6	11.0	11.27	18.05
CP7	12.4	12.61	17.45
CP8	9.27	10.34	16.25
CP9	12.8	12.02	16.10

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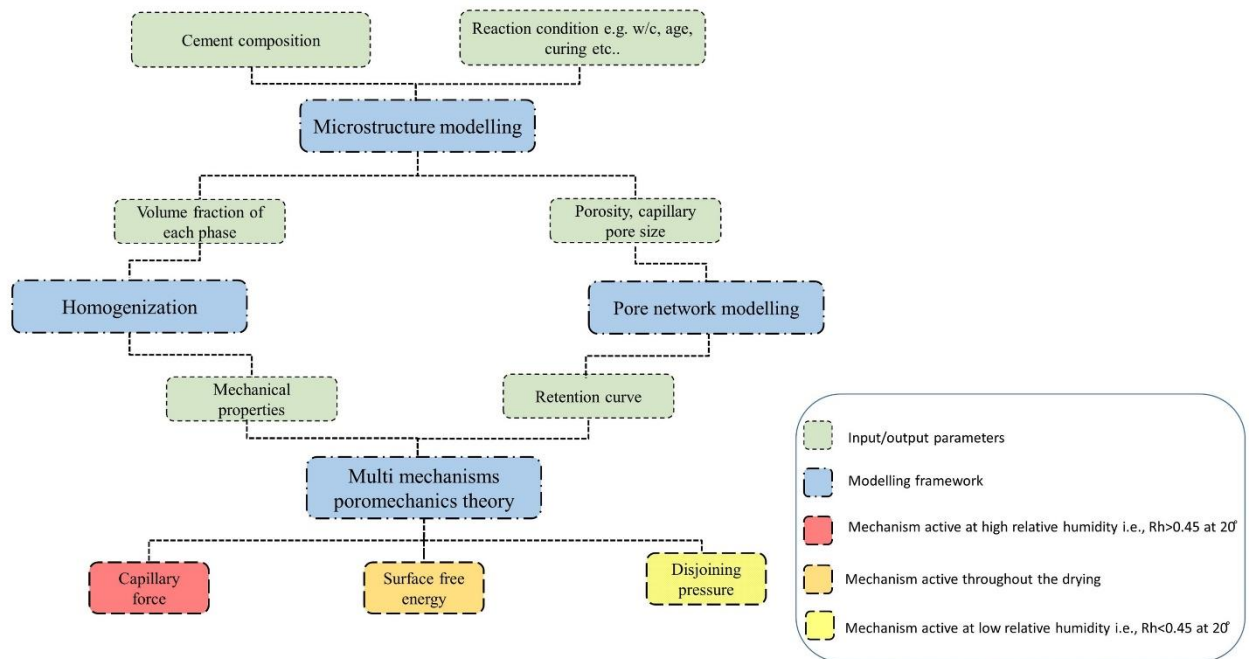


Figure 1. Analytical framework for estimating drying shrinkage strain.

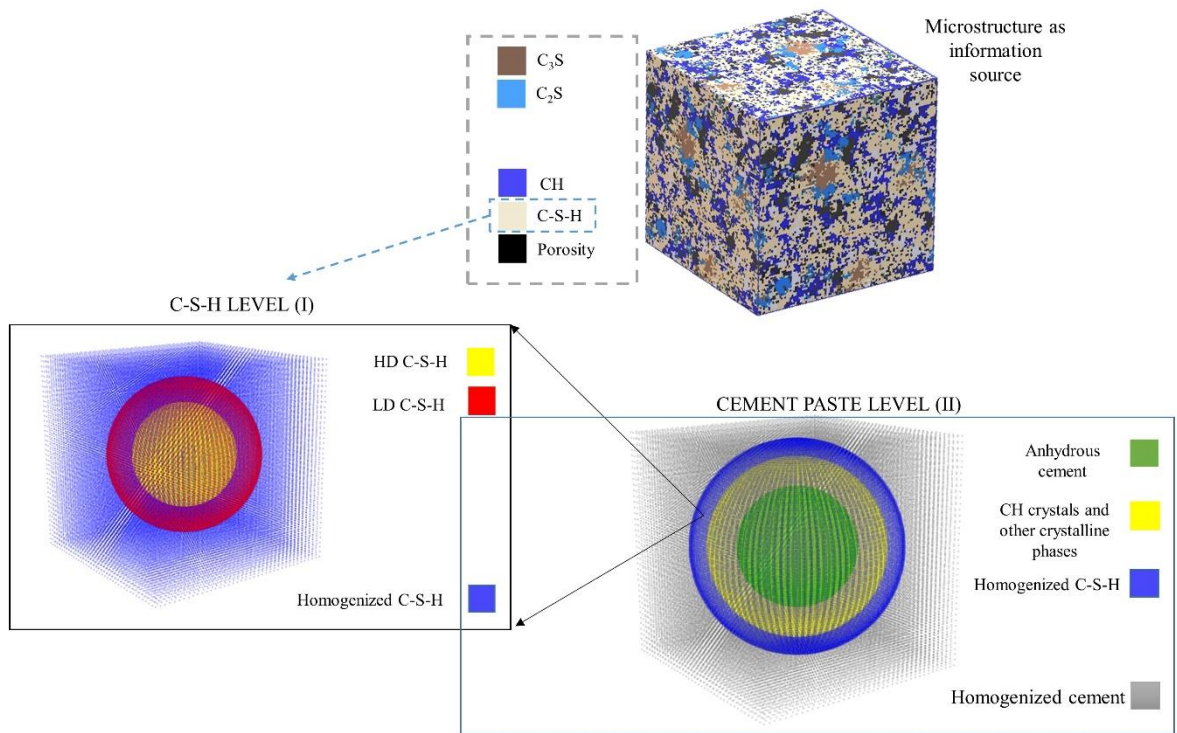


Figure 2. Generalized Homogenization method with its different levels.

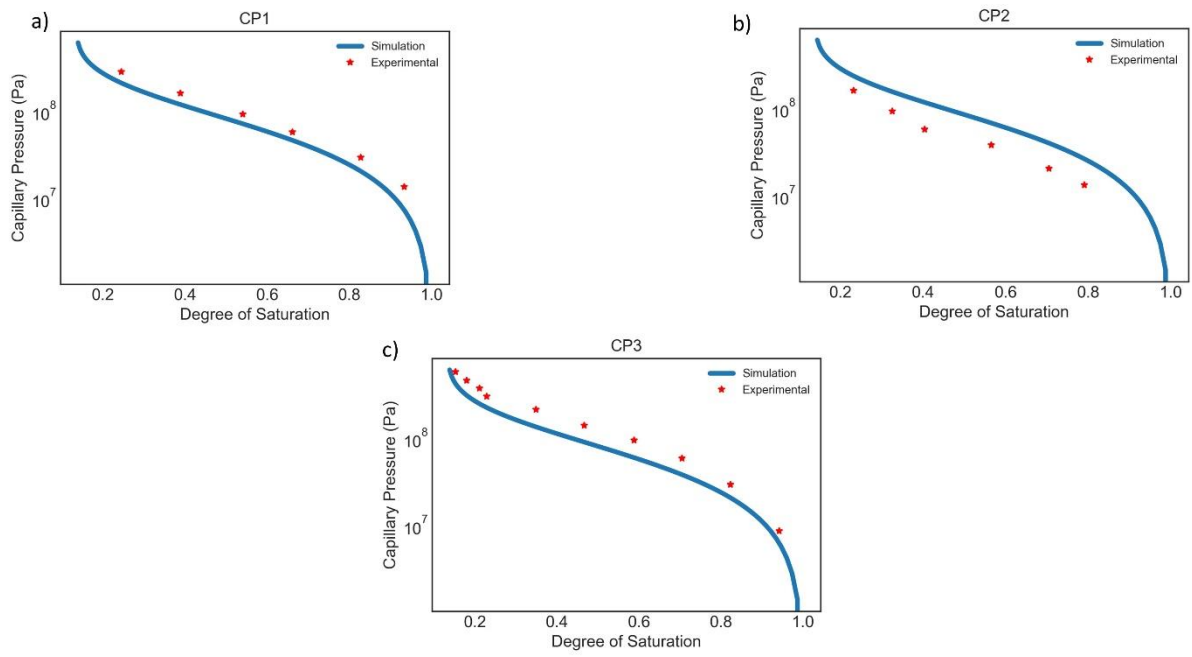


Figure 3. Estimated desorption isotherms for materials CP1-CP3 (previously reported in Babaei et al. [21]).

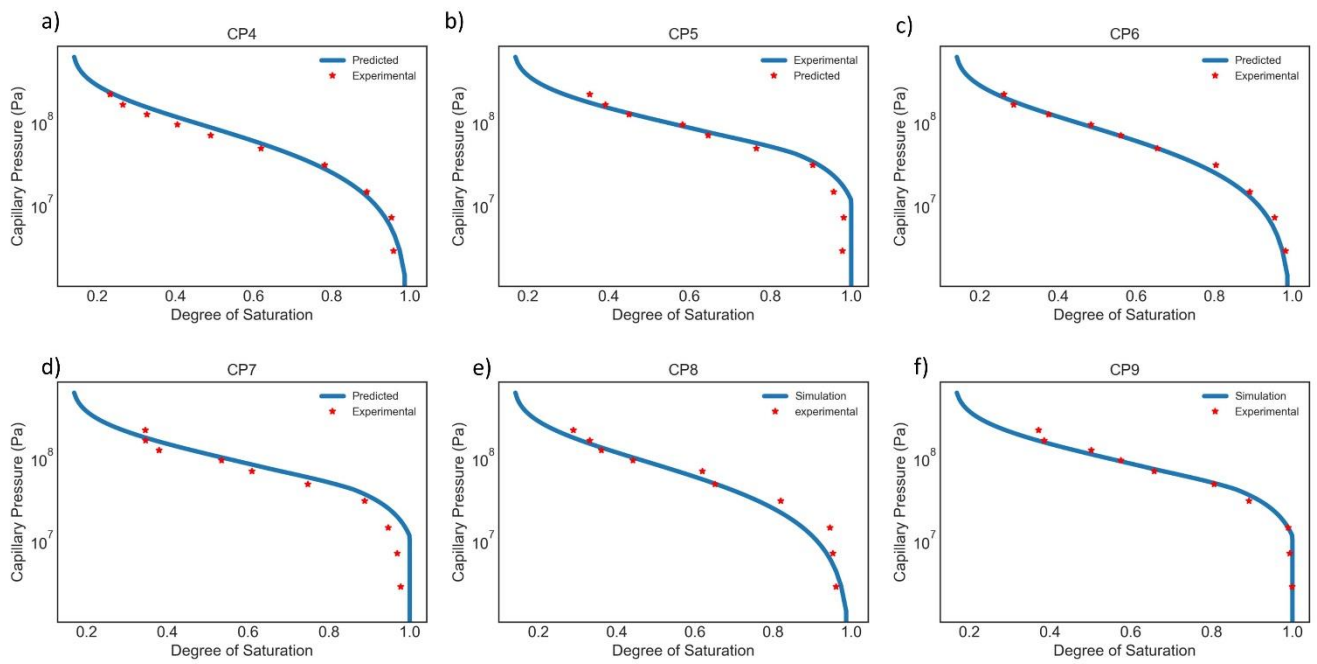


Figure 4. Estimated desorption isotherms for samples CP4-CP9 using Babaei et al. [21] approach vs experimental data.

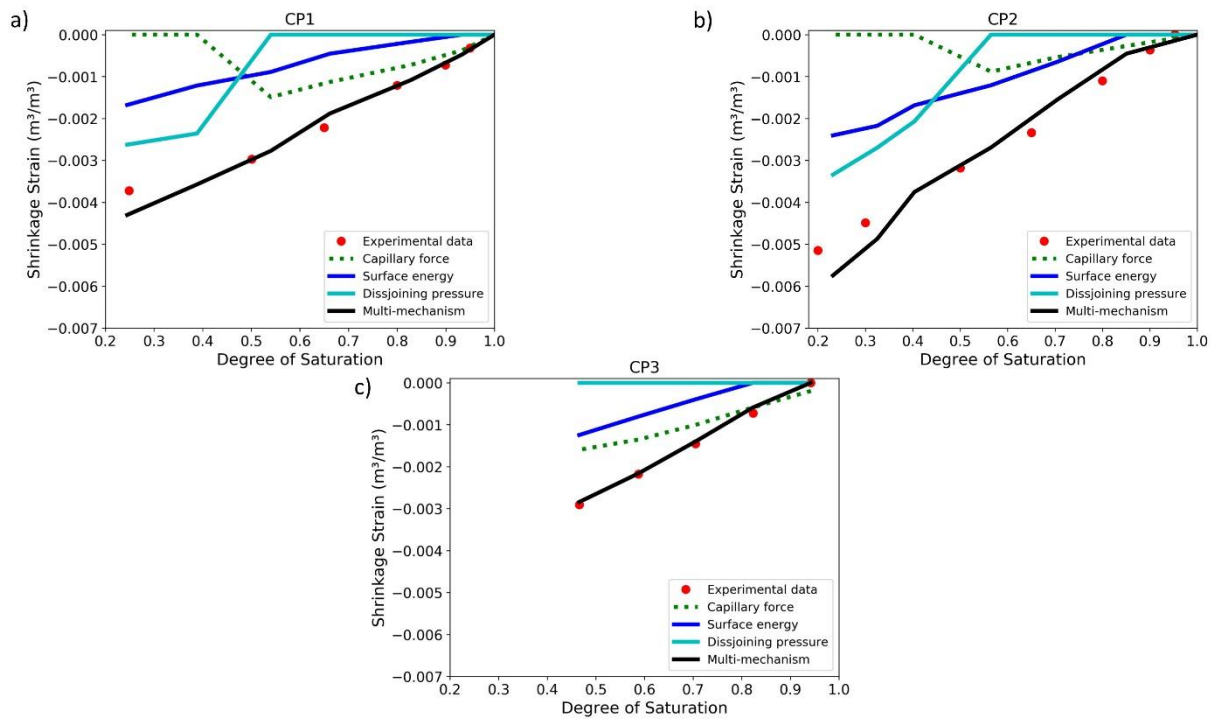


Figure 5. Contribution of each mechanism on predicted ultimate drying shrinkage vs. experimental data for non-virgin materials.

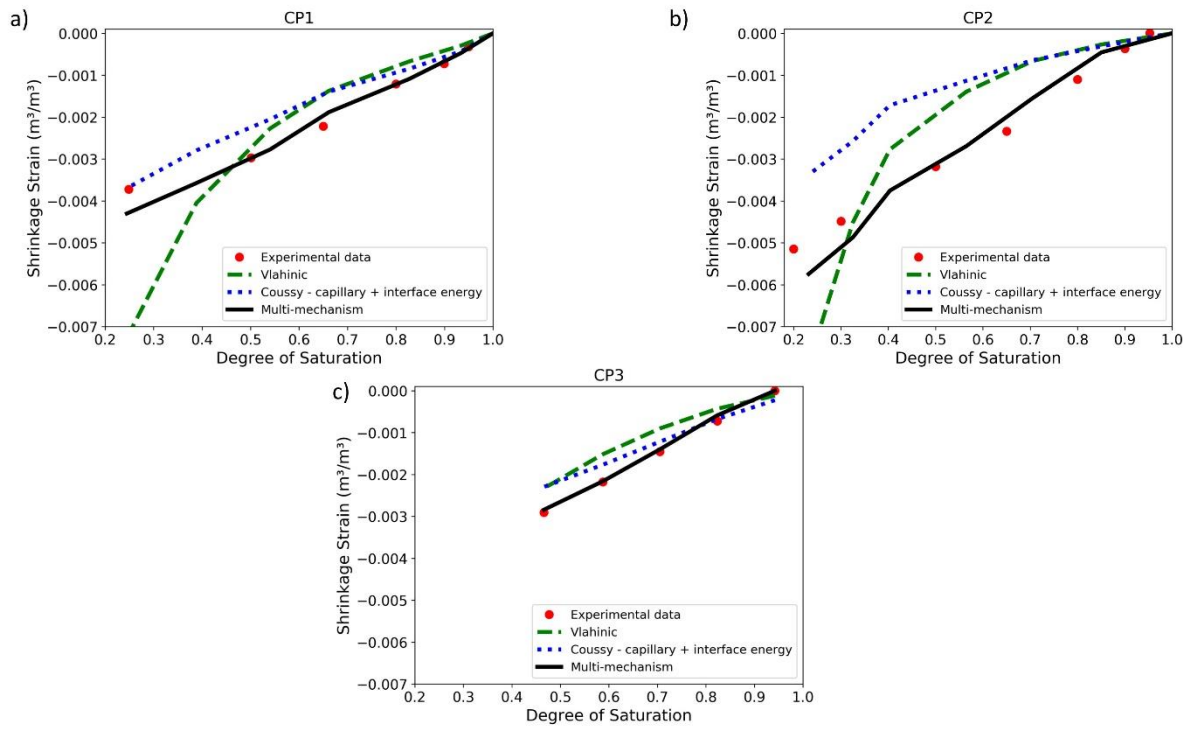


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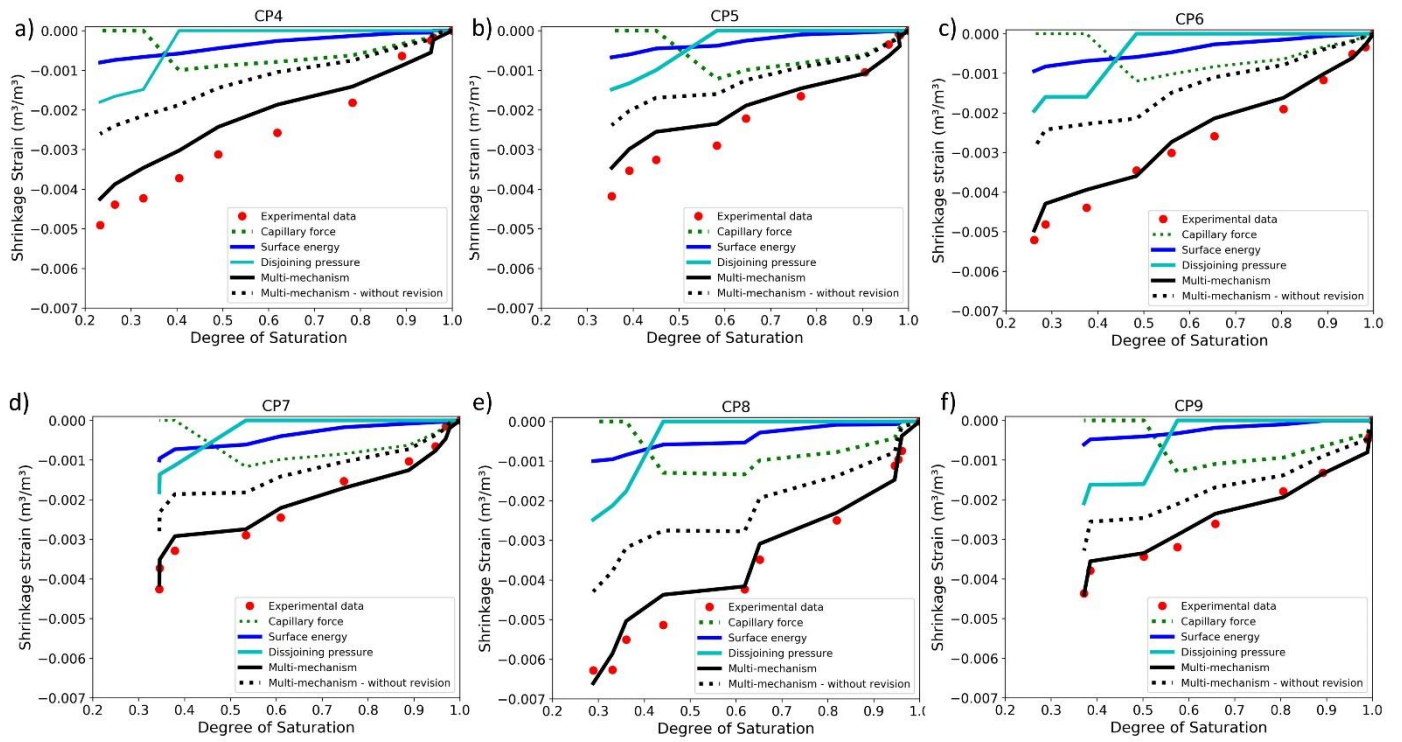


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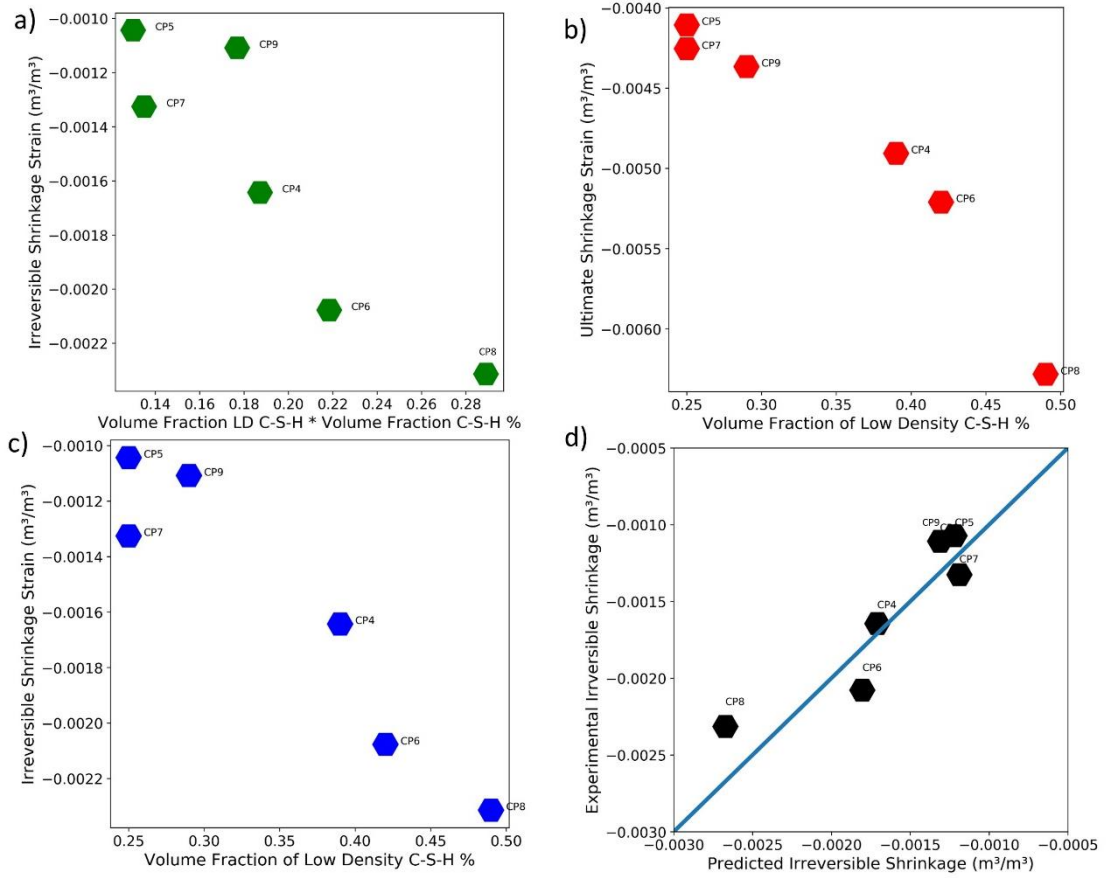


Figure 8. Irreversible shrinkage and its correlation with microstructural information.

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Declaration of interests

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests: