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1 **Suspect screening analysis in house dust from Belgium using high resolution**
2 **mass spectrometry; prioritization list and newly identified chemicals**

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11 **Abstract**

12 In recent years, several changes have been made to the composition of various products which are
13 used indoors. Plenty of new chemical additives have been incorporated to materials to comply with
14 current legislation and safety rules. Consequently, the emission profiles of contaminants detected
15 indoors may change over time, requiring continuous monitoring. In this study, dust samples were
16 collected from 25 homes located in the Flemish region of Belgium during different seasons (winter and
17 summer). Our aim was the development of a suspect screening workflow for the identification of new
18 chemicals which might have been applied to indoor goods, released into the indoor environment, and
19 accumulated in dust. An *in-house* suspect list was curated including selected groups of compounds,
20 namely “phthalates”, “phosphates”, “terephthalates”, “citrates”, “trimellitates”, (di-, tri-, tetra-)
21 “carboxylic acids”, “adipates”, “azelates”, “sebacates”, (di-)“benzoates”, and “succinates”. 63
22 chemicals were prioritized based on their level of identification and detection frequency in samples.
23 Seasonal comparison was tested, indicating that higher temperatures of summer might facilitate the
24 release of few chemicals from the products into the indoor environment. Seven chemicals, to the best
25 of our knowledge not previously reported, were selected out of the 63 listed and identified for
26 structure confirmation using high-resolution mass spectrometry. Tributyl trimellitate (TBTM), bis
27 (3,5,5-trimethylhexyl) phosphate (Bis-3,5,5-TMHP), iso-octyl 2-phenoxy ethyl terephthalate (IOPhET),
28 dimethyl azelate (DMA), dimethyl sebacate (DMS), dipropylene glycol dibenzoate (DiPGDB) and 3,5-di-
29 tert-butyl-4-hydroxybenzaldehyde (BHT-CHO) were detected at frequencies ranging from 8-52% in
30 winter and 4-56% in summer dust.

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32 **Keywords:** suspect screening analysis, indoor dust, new chemicals, prioritization, LC-QTOF-MS

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34 1. Introduction

35 During the last decades, indoor environment attracted scientific attention due to the need of constant
36 monitoring of its quality (Rostkowski et al., 2019). The modern living conditions include modified
37 materials for buildings and products and lower rates of renewed air in favor of the energy consumption
38 reduction. In combination with higher time spent indoors, the indoor environment has been
39 characterized as one of the major contributors to health problems due to the presence of numerous
40 chemicals (Laborie et al., 2016). In particular, indoor dust has been identified as a significant route of
41 human exposure to semi volatile organic compounds (SVOCs), including wide range of priority
42 pollutants such as flame retardants (FRs), polycyclic aromatic hydrocarbons (PAHs), plasticizers and
43 pesticides (Bui et al., 2016; Little et al., 2012; Harrad et al., 2010; Weschler & Nazaroff, 2008). The
44 main pathways of transfer for SVOCs from materials to indoor dust are (i) volatilization and subsequent
45 partitioning, (ii) transfer of particles or fibers from products to dust or (iii) direct contact between
46 surfaces and dust (Rauert et al., 2016). As a consequence, dust acts as a sink for SVOCs released from
47 indoor sources and through dust inhalation, ingestion and dermal contact, humans are exposed to a
48 wide range of chemicals.(Rauert et al., 2016; Kanazawa et al., 2010; Araki et al., 2014).

49 Recent evidence from biomonitoring studies in human blood and urine indeed indicates remarkable
50 human exposure to more than a hundred SVOCs (Calafat et al., 2017). Such exposure may lead to
51 increased health risks for sensitive subpopulations, such as infants and pregnant women (Calafat et
52 al., 2017;Zheng et al., 2015). In particular, children under 6 years of age tend to be more exposed to
53 indoor dust, due to their natural behavior of crawling and hand-to-mouth contact (Mercier et al.,
54 2011). Strong indications of harmful health effects have led to the restriction or ban of several SVOCs
55 (e.g. polybrominated diphenyl ethers-PBDEs, hexabromocyclododecanes-HBCDs, polychlorinated
56 biphenyls-PCBs, legacy plasticizers- bis(2-ethylhexyl) phthalate (DEHP), di-n-butyl phthalate (DBP),
57 benzyl butyl phthalate (BBP), di-iso-butyl phthalate (DIBP) (Yadav et al., 2017;Reche et al., 2019;Calafat
58 et al., 2017). Such restrictions/bans for "legacy" chemicals have resulted in their replacement by new
59 chemicals with SVOC properties which were frequently reported as "new", "alternative", "emerging",
60 "replacements", "chemicals of new generation" or "chemicals of emerging concern". For those ones
61 toxicological information is scarce and knowledge on potential exposure and adverse health effects in
62 humans from residential exposure remains still limited. (Subedi et al., 2017;Liu et al., 2012;Covaci et
63 al., 2011).

64 So far, several studies focusing on target analyses have been published reporting the levels of legacy
65 compounds in dust (Laborie et al., 2016;Little et al., 2012 ;Niu et al., 2019;Yadav et al., 2019;Dodson
66 et al., 2012;Christia et al., 2019;Eede et al., 2016;Larsson et al., 2017). Their main advantage of high

sensitivity is confronting the limited coverage of chemicals especially for a complex material like dust which is hosting a wide range of them (Dong et al., 2019). Since replacements started being introduced into global chemical market, the need to investigate the occurrence and fate of emerging contaminants in indoor environment is urgent and could be achieved by alternative approaches (Ouyang et al., 2017; Moschet et al., 2018; Rostkowski et al., 2019). Suspect screening analysis (SSA) is an analytical tool for providing a comprehensive outlook in dust matrix. SSA represents a group of analytical workflows developed for high-resolution mass spectrometry (HRMS). Molecular features of samples, which are compound-specific data acquired by LC-QTOF-MS instrument, accompanied by accurate mass, retention time and mass spectral information, are compared with information from lists of chemicals or databases with the aim to identify potential matches (Oberacher et al., 2020; Phillips et al., 2018). Thousands of expected compounds belonging to various groups of interest (e.g. alternative plasticizers, film forming agents, emollients, additives in food contact materials, etc.) can be congregated in suspect lists. The application of databases combined with the reliable results that HRMS can provide, reassures better evidence of their presence indoors. Since multiple candidates from the suspect list can match positively with the same feature, Schymanski et al., 2014 developed a system of identification levels or levels of confidence (LoC) that is widely adopted to communicate outcomes derived from SSA approach.

Since the knowledge on new chemicals that may affect the quality of the indoor environment and consequently human health is still at an early stage, SSA applications are required to investigate the profile of alternative chemicals indoors. Ensuring that purpose, a sampling campaign was designed and, in total 25 individual floor dust samples were collected from homes all over the Flemish region of Belgium during winter and summer season of 2019 (total $n=50$ paired samples). The main aims were; (i) to generate innovative information linked to the presence of alternative chemicals in residential dust, (ii) to develop an in-house suspect list with alternative chemical groups and suitable workflow of data analysis (iii) to build up prioritization lists of chemicals based on the obtained LoC, the relevant indoor applications, and the calculated theoretical bio-accessibilities.

2. Materials and methods

2.1 Chemicals and reagents

Organic solvents of LC-grade were used for the analyses. LC-grade ultrapure water (H_2O) was obtained from a PURELAB Flexsystem (18.2 M Ω cm, Milli-Q, Millipore). Methanol (MeOH) was purchased from Fischer Scientific (Loughborough Leics, United Kingdom) ($\geq 99.9\%$). The solvents of toluene, acetone, dichloromethane (DCM), ethyl acetate (EtAc) were purchased from Merck (Darmstadt, Germany) and *n*-Hexane (*n*-Hex) was purchased from Acros Organics (Geel, Belgium). The SPE cartridges (Florisil®

ENVI 500 mg, 3 mL) were purchased from Supelco (Bellefonte, PA, USA). Centrifugal filters (modified nylon membrane) 0.45 μm pore size were purchased from VWR™ (North America). Eight labeled internal standards (ISs), native standards (Ns) (≥ 98) were used during the development of quality assurance and quality control (QA/QC) actions. Newly identified standards were purchased from AccuStandards (New Heaven, CT, USA), Sigma Aldrich, Chiron AS (Trondheim, Norway) or were custom synthesized (Dr. Vladimir Belov, Max Planck Institute for Biophysical Chemistry, Göttingen, Germany) and were used for structure confirmation (Table S1).

2.2 Dust Collection

Two sampling campaigns were organized with 25 homes located in the Flemish region of Belgium during winter and summer 2019 (Figure S1). The two periods were selected for testing possible seasonal variations in the chemical profile of indoor dust. The different cities/provinces were selected to cover sufficiently the area of Flanders including urban and rural areas, the size of families ranged from 3 to 5 people and the age of the houses and the indoor equipment varied along the cohort for obtaining a better understanding of profiles. Relevant questionnaires were completed by the participants during the sample collections. In total, 50 dust samples were collected during winter (W) ($n=25$; January-March) and summer (S) ($n=25$; June-July). A regular vacuum cleaner equipped with nylon socks (25 μm pore size, Allied Filter Fabrics Ltd, Berkeley Vale, Australia) was employed to collect floor dust from living rooms under a previously applied protocol (Christia et al., 2019). Briefly, 4 m^2 bare floor was vacuumed for 4 min or 1 m^2 carpeted surfaces was vacuumed for 1 min. To prevent cross-contamination, one nylon sock was used per each sample, and all involved parts of the cleaner were thoroughly rinsed with ethanol before and after each sampling. Additionally, 3 field blanks (FBLKs) (i.e. pre-cleaned Na_2SO_4 ; n-Hex washed using ultrasonication for 30 min followed by baking at 400°C overnight) were used per each sampling campaign. Briefly, 5 g of the pre-cleaned Na_2SO_4 were sprinkled in aluminum foil and vacuum cleaned following the dust collection protocol. All socks were firmly closed, placed into zip lock plastic bags and transferred to the laboratory. Each sample was sieved using a 500 μm mesh sieve, each FBLK was divided in 2 aliquots (~ 20 mg) and all samples were stored at room temperature until analysis.

2.3 Extraction and clean up

The sample preparation protocol was based on existing protocol of our laboratory after the necessary modifications (Christia et al., 2019). It is a time and cost effective method (up to 24 samples can be analyzed in parallel) and of relative low consumption of organic solvents. Briefly, dust aliquot of approximately 20 mg was weighted in pre-cleaned glass tubes and spiked with 50 μL of labeled ISs for positive and negative electrospray ionization mode (ESI+, ESI-) in concentrations ranging from 1-10

ng/ μ L (Table S1). The extraction cycle was performed by using 2.5 mL of a prepared mix of *n*-Hex:Acetone (1:1 v/v) and 0.5 mL toluene, combined with 1 min vortexing and 5 min of ultrasonication. The extracts were centrifuged at 3000 rpm for 3 min. The supernatants were transferred into pre-cleaned glass tubes. The extraction cycle was repeated one more time and both supernatants were pooled, evaporated near dryness under a gentle nitrogen stream, re-dissolved in 1 mL of *n*-Hex and vortexed for 1 min. Florisil® ENVI cartridges (500 mg, 3 mL) were prewashed with 3 mL of MeOH, 3 mL of EtAc and 3 mL of *n*-Hex. The extract was transferred into the cartridges, which were washed up with 8 mL of *n*-Hex: DCM (1:1 v/v) and compounds were eluted with 10 mL of EtAc:MeOH (1:1 v/v). The elution fraction was evaporated to near dryness under a gentle nitrogen stream. After evaporation, was re-dissolved in 100 μ L of MeOH:Water (1:1 v/v) and filtered using filters of 0.45 μ m in micro-centrifuge at 8000 rpm for 5 min. Then each aliquot was transferred to injection vial for further liquid chromatography combined with quadrupole time-of-flight mass spectrometry (LC-QTOF-MS) analysis.

2.4 LC-QTOF-MS analysis

An Agilent 1290 Infinity UPLC coupled to an Agilent 6530 QTOF (Agilent, Santa Clara, CA, US) was employed for the analysis of the dust extracts. For chromatographic separation, an Alltima HP C18-EPS column (100 mm x 2.1 mm; 3 μ m particle size, Grace, Belgium) was applied and the mobile phases of Water with 0.1% v/v formic acid (A) and MeOH/Water (80:20 v/v) with 0.1% formic acid (B) were used for ESI+. Water with 0.1% v/v acetic acid (A) and MeOH/Water (80:20 v/v) with 0.1% acetic (B) were used for ESI-. Flow rate was set at 0.4 mL/min in both modes. The injection volume was set at 5 μ L and the column temperature at 30 °C. All samples were analyzed in both ESI+ and ESI-. Details of the applied linear gradients are presented in Table S2. The QTOF-MS was operated in 2GHz, extended dynamic range mode. Agilent Jet-Stream electrospray ionization (AJS-ESI) was applied for the ionization of the compounds and the same parameters were applied for ESI+ and ESI-. Briefly, drying gas temperature was set at 300 °C and the flow at 10 mL/min. The sheath gas was set at 350°C and 12 mL/min flow rate. The nebulizer pressure was fixed at 45 psi, and the voltages of capillary, nozzle and fragmentor were set at 3500 V, 500 V, and 120 V respectively. The mode of acquisition was selected in Auto MS/MS operation in which the precursors ions were automatically selected by the software based on the relative abundance. Target MS/MS option was applied only when the compound of interest was not automatically selected for MS/MS spectra generation. By choosing this option, the instrument is able to select specific precursors that will be forced to pass through the quadrupole and produce the MS/MS spectra. The acquired mass range was 80-1100 m/z, the MS scan rate was 4 spectra/sec and the MS/MS scan rate 7 spectra/scan. Two levels of collision-induced dissociations (CID) were used at 10 and 30 V. The active exclusion mode was enabled to prevent the repetitive acquisition of MS/MS

spectra for the same ion and was set at 0.1 min. The generated data files were stored in centroid mode before exported for further data analysis.

2.5 Quality assurance and quality control

The quality assurance and quality control of the analysis included several steps. The analysis of a total of 6 FBLKs and 6 procedural blanks (BLKs) along with the 50 dust samples (Ds) was applied to control the background contamination. All glassware was pre-cleaned using *n*-Hex and baked at 400°C overnight. To reassure the high quality of the instrumental performance, several steps were applied. The QTOF instrument was automatically recalibrated during analyses by continuously infused into the source the Agilent calibration solution via a calibration delivery system two characteristic ions, 121.05058 and 922.0098 for ESI+, 112.9856 and 966.0007 for ESI-, were monitored. Samples were spiked with ISs suitable for both ESI+ and ESI- (Table S1) before the extraction step to estimate the ppm tolerance in terms of the mass accuracy (± 5 ppm), the retention time shifting (retention time delta) (± 0.2 min) and the isotopic pattern match score ($\geq 85\%$). These parameters were further used for setting the criteria in the suspect screening analysis- algorithm "Find by Formula" against a database/ library and test the inter batch repeatability of the produced results. A quality control solution (QC) including all ISs was injected multiple times along the sequence and the results were evaluated for the instrument performance on the same analytes. Solvent blanks were applied every 6th injection for checking any impurities or additives in MeOH and Water. Both MeOH and water produced reproducible signals of low intensity.

2.6 Data analysis workflow

An *in-house* suspect list was formed called "HECHIER" using Agilent MassHunter Quantitative Library Editor software (version 10.0). Relevant keywords were used for retrieving compounds belonging to the groups of interest such as "*phthalates*", "*phosphates*", "*terephthalates*", "*citrates*", "*trimellitates*", (di-, tri-, tetra-) "*carboxylic acids*", "*adipates*", "*azelates*" "*sebacates*", (di-) "*benzoates*", "*succinates*", from open access databases (e.g. NORMAN, HBM4EU, PubChem, US EPA dashboard) and international chemical agencies (e.g. ECHA, US EPA, EFSA, ANSES). A list of 2108 compounds was formed including the information of compound names, compound formulas, monoisotopic masses, chemical classifications, CAS numbers, InChIKeys, and canonical smiles (Calafat et al., 2017).

A new workflow for data analysis was developed based on a suspect screening approach for both W and S samples analysis (Figure 1). Four sets of data files, two for ESI+, two for ESI-, were generated and used for feature extraction using Agilent MassHunter Qualitative software (version B.07.00). Feature extraction was performed by applying the algorithm "Find by Formula" combined with "HECHIER" suspect list. Alignment and peak selection was done according to criteria of mass tolerance for parent

ions at ± 10 ppm for ESI+ and ± 7 ppm for ESI-, mass tolerance for product ions at ± 25 ppm for both ionization modes, selected matching score above 75%, and the isotope abundance score set at 80 % between the measured and the predicted one, based on the ISs results as described previously (2.5 *Quality control and quality assurance*). Mass Profiler Professional Software (version 15.0) was used for statistical analysis. All features below a fold change of 5 against the FBLKs were excluded (ANOVA test for significance of fold change, $p < 0.05$) reducing the number of the features for manual identification. Four final lists of compounds were produced and used for one-by-one tentative identification starting from match of formula and monoisotopic mass (LoC 5) and match scores between the measured isotopic patterns and the theoretical ones predicted by Agilent Isotope Distribution Calculator (version 7.0.7024.0) (LoC 4). Further identification was achieved based on the available MS/MS spectra. Product ions that were formed at collision induced dissociation (CID) 10 and 30 V were identified by applying Agilent formula Calculator tool (version 7.0.7024.0) and/or checking compound structures (LoC 3). On-line spectra libraries (Mass Bank of North America, MoNA and m/z cloud) were used to cross check the MS/MS spectra (where available) (LoC 2). The LoCs were provided for each step following Schymanski et al., 2014 methodology.

2.7 Seasonal Variation

To assess the seasonal difference in dust composition for winter and summer, compounds reported as native standards in Table S1, were identified (LoC 1) following the developed workflow. The selected compounds were belonging to the groups of organophosphate flame retardants (PFRs), phthalates (PHs) and alternative plasticizers (APs) which are frequently detected in indoor dust worldwide and thus were expected to be present in Flemish dust (Larsson et al., 2017; EPA Comptox; Shoeib et al., 2012; Dodson et al., 2017; Bu et al., 2019; Abb et al., 2009). Additionally, the availability of analytical standards provided the highest LoC. To enrich the seasonal estimation, we include compounds belonging to five chemical groups: “azelates”, “benzoates”, “citrate”, “sebacates” and “trimellitates”, which were identified at LoC 3. Since our SSA developed workflow is qualitative, the detection frequencies and the abundances of the identified compounds were employed to test disparities between winter and summer (t-test: two sample assuming unequal variances). Regarding the abundance, a response factor was calculated and used based on the following equation:

$$\text{Abundance Response Factor (x)} = \frac{\text{Abundance of compound (x)}}{\text{Mean Abundance of DBzP-d4 (x)}} \quad (1)$$

where the abundance of the compound was the instrumental abundance (counts) derived in sample (x) and mean abundance of DBzP-d4 was the mean instrumental abundance of an IS derived from the fortified samples. The selection of DBzP-d4 was due to its higher abundance and to generate abundance response factors based on the same parameter for all the compounds.

2.8 LogK_{ow} and theoretical bio-accessibility

Theoretical bio-accessibility, that refers to the fraction of total amount of a substance that is potentially available for absorption, was estimated in this study based on the LogK_{ow} values of the tentative identified compounds. The logK_{ow} values were predicted by EPIWEB software (version 4.1) and theoretical bio-accessibility was calculated based on methodology reported by Dong et al., 2019 as described in the following equation:

$$B_a = a + \frac{(b-a) \cdot (8 - \log K_{ow})}{8 - 5} \quad (2)$$

where B_a is the theoretical bio-accessibility when logK_{ow} value is between 5 and 8, a and b are the constants assumed as 0.2 and 0.8 respectively. In case of LogK_{ow} < 5, B_a assumed to be 0.8 and in case of LogK_{ow} > 8, B_a assumed to be 0.2 [25],[17].

3. Results and discussion

3.1 Quality control and quality assurance results

Mass accuracy was performed by recalibrating the mass axis for every injection by monitoring the 4 ions; 121.0508, 922.0098 (ESI+) and 112.9856, 966.0007 (ESI-) Reference mass solution was introduced into the system through the calibrant delivery system during the whole acquisition. ISs were identified with LoC 1 in all fortified samples in terms of mass accuracy, Rt consistency, isotopic match > 90% and identification of fragments (CID 10 & 30V). The results of QC vials for the same ISs were equivalent for mass accuracy, match of isotopic pattern and fragment identification. Total ion chromatograms (TICs) for both cases were reproducible along the instrumental run. The selected compounds were identified at LoC 1 with DF% ranging from 8 to 100% for PFRs, 16 to 92% for PHs and 28 to 88% for APs (Table S1). Background contamination was controlled by the analysis of FBLKs combined with BLKs. Molecular features detected in dust samples were cross checked in FBLKs and BLKs. The majority of them were not detected (n.d.) in blank-type samples. For those who were detected, abundance response factors were calculated and compared with the dust ratios (as *Equation 1*) indicating lower presence compared to dust. The quality of the suggested workflow was assured following exclusion of the false positives and false negatives during Mass Profiler Professional data analysis, and by applying the algorithm “Find by Formula” using specific formulas which were tested for all samples.

Twelve features were detected only in FBLKs indicating the dust sampling equipment as a possible source, two compounds were detected only in BLKs indicating low contamination from the laboratory environment and 13 in both types of blank samples. However, the comparison of the abundance response factors between dust samples and both types of blanks, was higher from 4 up to 67 times for

the majority of the compounds, indicating stable and reproducible results in terms of instrumental performance. The fold change (FC) that was applied was ≥ 5 (Table S5).

3.2 Flemish indoor dust profile

Two lists of the identified compounds in Flemish dust are reported in Tables S3 and S4. In total, 44 and 12 compounds were identified for ESI+ and ESI-, respectively. Information of identifiers (Name, Formula, ICHI Key, Canonical Smiles), sampling season, detection frequency in dust samples (DF%), LoC, chemical category, relevant exposure sources, estimated $\log K_{ow}$, and theoretical B_a are provided in both tables. The compounds were prioritized based on their $LoC \geq 3$ due to MS/MS spectra availability and product ion confirmation. Additionally, DF% in dust samples were reported but no threshold was set for this parameter. The possible exposure sources and theoretical B_a were taken into consideration for the final selection of standards purchased for the final confirmation. In the ESI+ list, 9% of the compounds were identified at LoC 2 after the successful match of their MS/MS spectra with the equivalent ones of the online libraries of MoNA and mzCloud. The rest of the compounds reached only LoC 3 due to lack of available online spectra. In the ESI- list, all identified compounds reached LoC 3 and were classified by chemical groups for comprehension of the new substitutes. The group of (di-) carboxylic acids was the dominant family for both ionization modes, followed by phthalates for ESI+ and phosphates for the ESI-. In ESI+, a larger diversity of chemical groups was noticed than ESI-, including citrates, terephthalates, acrylates, hexanoid, cinnamic and salicylic acids, benzoates, and sebacates. This might indicate ESI+ as a suitable ionization mode for these chemicals. On the other hand, the group of butoxides were identified only in ESI- (Figure 2). The indoor sources possibly responsible for the presence of these chemicals were included in priority lists, aiming to a better understanding of the home environment. Main applications were additives in food packaging materials and children toys, food flavors or food residues, additives in personal care products (PCPs), emulsifiers, plasticizers, lubricants, adhesives and sealants, UV filters, flame retardants, emollients, paints, heat stabilizers, fragrances, automotive, and pet care products. It is evident that all reported applications can be directly related with goods and materials that are used indoors. The wide range of applications and the multiple functionality of many listed compounds complicate the relation of compound-specific indoor source.

The next introduced parameter was the theoretical bio-accessibility (B_a) and was taken into consideration for the final selection of compounds used in standard confirmation (LoC 1). Based on $\log K_{ow}$ (see Section 2.7), a theoretical $B_a \geq 0.8$ was calculated for a considerable number of the listed suggesting a potential risk increase of human intake (Dong et al., 2019). In the ESI+ list, 53% of the compounds could be considered as major contributors to human exposure, as their estimated B_a was

> 0.8. Lower percentages of 17% and 23% were calculated for compounds with Ba ranging between 0.8 to 0.2 and < 0.2, respectively. The latter indicate that these compounds are less likely to contribute to human exposure (Figure 3). In the ESI- list, 67% of the compounds could be considered as higher contributors with Ba > 0.8, 16% as medium contributors with Ba between 0.8 and 0.2, and 17% as lower contributors to human exposure with Ba < 0.2. However, our approach is theoretical and not experimental based. The bio-accessible fraction of a chemical is a complex factor which requires a particular study setup focused on *in vitro* digestion models by using synthetic bio-fluids (Collins et al., 2015).

3.3 Newly identified compounds in Flemish dust

Seven compounds belonging to different chemical groups and applications related to indoor environment, were selected based on their LoC (≥ 3) and DF% were reported for Ds. Analytical standards were used in order to confirm a LoC 1 (Table 1). The theoretical Ba of the compounds and the availability of the analytical standards was considered for the selection. Individual solutions of the analytical standards of tributyl trimellitate (TBTM), bis (3,5,5-trimethylhexyl) phosphate (Bis-3,5,5-TMHP), isooctyl 2-phenoxy ethyl terephthalate (IOPhET), dimethyl azelate (DMA), dimethyl sebacate (DMS), dipropylene glycol dibenzoate (DiPGDB) and 3,4-di-tert-butyl-4-hydroxy benzaldehyde (BHT-CHO) at concentration level 5 ng/ μ L (in MeOH) were analyzed by LC-QTOF-MS to generate MS/MS spectra. Figure S2 (a) to (g) presents the MS/MS spectra in CID 10 and 30 V together with the structures of the precursor and the product ions. The confirmed structure was achieved by matching the MS/MS spectra of the chemical in the analytical standard and the dust sample. Retention times were confirmed, the mass accuracy was <5 ppm, match of isotopic patterns were > 90%, and characteristic products ions for both levels of CID were confirmed (Table 1).

TBTM is a trimellitate that is usually found in market with the commercial name MorflexTM 540 or DIPLAST TM 4 (PubChem, ECHA). It is used as a plasticizer and additive in adhesives, printing inks, nail coatings or nail coating removers (CompTox Chemical Dashboard; EPA; Pubchem). The DF was calculated at 8% and 24% for W and S samples respectively. Bis-3,5,5-TMHP is a diester phosphate with DF 32% in W samples and 8% in S samples. It is applied as a chemical blowing agent for polyurethane foams (Pubchem). The term “blowing agent” is referred to chemicals which have the ability to produce a cellular structure through a foaming process, during hardening or phase transitions processes of materials such as polymers, plastics, and metals.

IOPhET is the next chemical from the group of terephthalates that was structurally confirmed. It is an additive in adhesives and was detected in 52% and 4% between seasons.

DMA, classified in the group of azelates, was found in dust samples with DF 28% and 56% in W and S and it is used as an additive in food packaging materials. It can also be present in pesticide mixtures as an active ingredient (CompTox Chemical Dashboard, EPA; Pubchem).

DMS is a sebacate applied as additive in paints, adhesives, fillers, and pesticides. Its DF in dust was 12% and 20% for W and S seasons, respectively.

DiPGDB is a classified benzoate that it is commercially known as Finsolv® PG-22, VELSIFLEX® 328, or Velsiflex 342 (PubChem, ECHA). It showed rather ubiquitous presence in dust with DFs of 40% and 48% for both seasons. Main applications are considered adhesives, building construction materials, paints and personal care products. Additionally, it can be used for the manufacture of furnishing equipment (CompTox Chemical Dashboard, EPA; Pubchem).

Finally, BHT-CHO, detected in ESI-, belongs to the chemical group of aldehydes. The dust DF was 8% and 28% for the two seasons. It is an additive in toys and drug products, or it is related with the manufacture of the drugs (CompTox Chemical Dashboard).

In literature, there is scarce toxicological information regarding these compounds with the majority of them to be derived by toxicological prediction models. Only for TBTM, *in vivo* studies in rats and rabbits are reported by ECHA, but no adverse effects were observed to these animals. All available information is reported in Table S6. Nevertheless, there is a lack of *in vivo* studies testing the carcinogenicity and endocrine disrupting behavior of the newly identified chemicals.

3.4 Seasonal variation in Flemish dust

Analytical standards were used to investigate seasonal differences of indoor dust profile between winter and summer where the temperatures and the type/frequency of ventilation varied. Since results are qualitative, calculated abundance response factors (*equation 1*) were compared (Figure 4, Table 2). The 5th, median and 95th percentiles were calculated for 15 compounds for W and S. The order of dominant compounds in winter according to their calculated mean abundance response factors was DEHP/DEHT (isomers) (2.28) > DIBP/DNBP (0.92) > DIDP = DINP = DINCH (0.86) > TBOEP (0.72) > ATBC (0.63) > BBP (0.30) > THTM (0.23) > TCIPP (0.21) > TIBP/TNBP (0.13) > DMP (0.11) > TEHP (0.09) > DEP (0.08) > TPHP (0.03) > TDCIPP (0.01). The order for summer was DIBP/DNBP (1.67) > DEHP/DEHT (1.06) > DINCH (0.86) > ATBC (0.79) > TBOEP (0.73) > DMP (0.63) > BBP (0.60) > THTM (0.45) > TCIPP (0.43) > DINP (0.42) > TIBP/TNBP (0.40) > DEP (0.38) > TEHP (0.15) > TPHP (0.11) > TDCIPP (0.13), respectively. Mean abundance ratios of TPHP, TCIPP, DMP and DEP were significantly higher in samples which were collected during S (p<0.05). That might indicate that elevated temperatures of the season combined with relatively higher vapor pressures of the compounds may enhance their migration from materials

to the indoor environment (Table 1). Only DINP showed higher abundance ratio during W compared to S ($p < 0.05$). For the rest of the compounds, no significant difference was observed between the two seasons ($p > 0.05$).

Ten compounds belonging to chemical families of azelates ($n=1$), benzoates ($n=1$), citrates ($n=3$), sebacates ($n=3$), terephthalates ($n=1$) and trimellitates ($n=1$) and which were identified at LoC 3, were used for season comparison (Figure 5, Table 2). The order of dominant compounds for the W season was 2,3-bis(acetyloxy)propyl 2-methyl tetradecanoate (0.97) > 2-hydroxyethyl 3-hydroxypropyl benzene-1,4-dicarboxylate (0.80) > trihexyl trimellitate (0.41) > 11-oxoundecyl benzoate = dimethyl azelate (0.23) > 3-[(2-ethylhexyl)oxy]-2-hydroxypropyl 2,2-dimethyl octanoate (0.21) > diethyl 2-(1-hydroxybutyl)propanedioate (0.19) > diethyl decanedioate (0.17) > dimethyl decanedioate (0.15) > diethyl undecanedioate (0.08). The order for the S season was dimethyl azelate (0.47) > dimethyl decanedioate (0.40) > diethyl 2-(1-hydroxybutyl) propanedioate = diethyl undecanedioate (0.30) > diethyl decanedioate = 3-[(2-ethylhexyl)oxy]-2-hydroxypropyl 2,2-dimethyl octanoate (0.20) > 2-hydroxyethyl 3-hydroxypropyl benzene-1,4-dicarboxylate (0.17) > 11-oxoundecyl benzoate (0.14) > trihexyl-trimellitate (0.11). Statistical season variations were proved only for 2 of the 10 reported compounds. 11-oxoundecyl benzoate found to be higher in W samples whereas diethyl decanedioate found to be higher in S season, according to their abundance response factors ($p < 0.05$). This could be attributed to the difference in the vapor pressures as explained previously for the analytical standards. For the rest of the compounds, there was no seasonal statistical difference ($p > 0.05$).

Dimethyl azelate was presented in 28% of W and in 56% of the S samples. 11-oxoundecyl benzoate ($C_{18}H_{26}O_3$) belongs to chemical group of benzoates and was detected in 64% of winter samples and 36% of the summer ones. For the group of citrates, three compounds were identified; 2,3-bis(acetyloxy) propyl 2-methyl tetradecanoate ($C_{22}H_{40}O_6$) in 40% only in W samples, 3-[(2-ethylhexyl)oxy]-2-hydroxypropyl 2,2-dimethyl octanoate ($C_{21}H_{42}O_4$) in 72% and 80% in winter and S samples respectively and diethyl 2-(1-hydroxybutyl) propanedioate ($C_{11}H_{20}O_5$) in 24% and 32% in W and S samples. No further information regarding their applications was available up to our knowledge. Three compounds were identified for the group of sebacates; diethyl sebacate ($C_{14}H_{26}O_4$) in 16% W and 20% S, dimethyl sebacate ($C_{12}H_{22}O_4$) in 12% of W and 20% of S samples and diethyl undecanedioate ($C_{15}H_{28}O_4$) in 16% of W and 20% of S samples. Relevant information for the application was available only for diethyl sebacate which is reported as a side ingredient of drug products and dimethyl sebacate, which is classified as additive in paints, pesticides, fillers and adhesives (CompTox Chemical Dashboard, EPA, PubChem). 2-Hydroxyethyl 3-hydroxypropyl benzene-1,4-dicarboxylate ($C_{13}H_{16}O_6$) and trihexyl trimellitate ($C_{27}H_{42}O_6$) from the chemical groups of terephthalates and trimellitates respectively were

found in dust in 4% W and 32% S and 8% W & S samples. No further information was found regarding their uses.

3.5. Limitations

An important limitation is the lack of information in on-line libraries preventing the achievement of LoCs 3 and 2, along with the low availability of commercial analytical standards which prompts to costly and time consuming solution of the custom synthesis. The lack of fragmentation (MS/MS spectra generation) is an ambiguous characteristic of suspect screening analyses, which can be solved by applying target MS/MS analytical mode, but in the same time cannot guarantee the accurate results and requires time for sample batch re-analysis.

4. Conclusions

A suspect screening analysis workflow was developed and applied successfully in indoor dust with regards to efficiency, repeatability, acceptable control of background contamination and appropriate elimination of false positives/negatives. The available generated lists of compound prioritization indicate the new profile of indoor dust in the Flemish area of Belgium and are useful for identification of emerging contaminants. The application of the lists can be relevant to other matrices of interest (eg. environmental matrices). An important outcome of our study is the reporting of seven new compounds which up to our knowledge, have not been previously reported in dust. These compounds, TBTM, Bis-3,5,5-TMHP, IOPhET, DMA, DMS, DiPGDB, and BHT-CHO, have relevant applications in domestic products or materials, thus would be of high importance their further investigation in indoor environments. The season appeared to influence the indoor chemical profile of some compounds. Compounds with higher vapor pressure values showed higher abundance ratios during summer, supporting the hypothesis that higher temperatures favored their increased migration from products to the indoor environment and dust.

Conflicts of interest

The authors declare that they have no conflict of interest.

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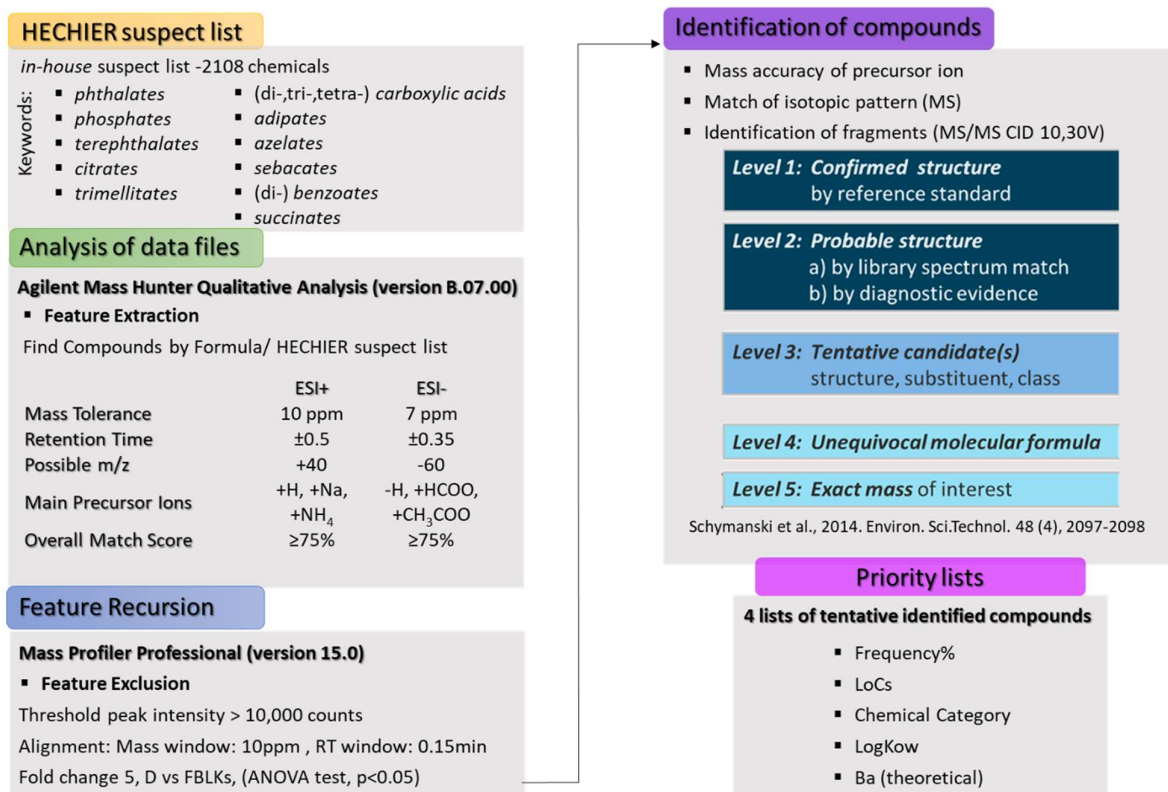


Figure 1. Workflow of the data analysis. Possible m/z; parameter of the expansion of values for chromatogram extraction. The values were set based on the predicted adducts. ESI; electrospray ionization, RT; retention time, D; dust samples, FBLKs; field blanks, CID; collision-induced dissociation, LoC; level of confidence, LogKow; logarithmic octanol-water partition coefficient, Ba; bioaccessibility.

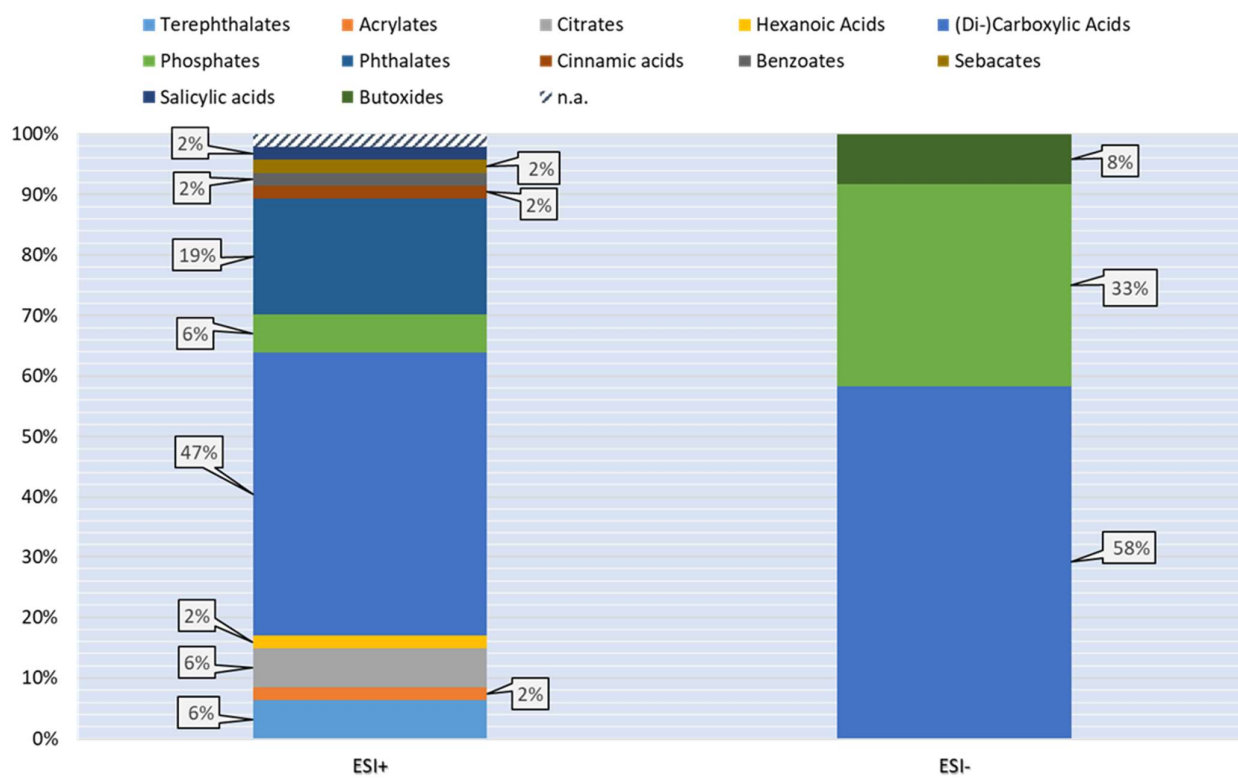
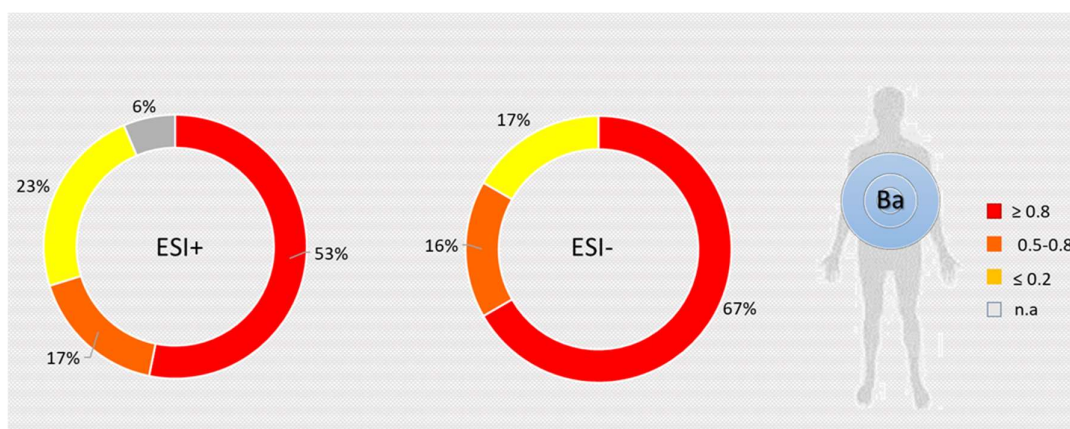


Figure 2. Main chemical groups in priority lists for ESI+ and ESI-.

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553

554 **Figure 3.** Theoretical bio-accessibility (Ba) and compound classification in priority lists.

Chemical Name	CAS No.	Molecular Weight	Chemical Structure	Molar Mass	Density	Boiling Point	Melting Point	Log P	Log S	Log D	Mass Spectrometry		Elemental Analysis			Physical Properties	Safety Data	Applications
											CID 10V		CID 30V					
Diethyl phthalate (23-4)	TBTM	DIPLAST TM 4, Morflex TM 540	C ₂₁ H ₃₀ O ₆	+	18.7	378.2041	379.2116	[M-H] ⁺	-0.13	99.95	305.135, 249.074, 193.012, 57.070	249.0742, 193.0124, 57.0704	8/24	5.92	0.61		Plasticizer, additive in adhesives or printing inks, nail coatings, nail coatings removers	
Bis(2-ethylhexyl) phosphate (98-2)	Bis-3,5,5-TMHPH	n.a.	C ₁₈ H ₃₉ PO ₄	+	14.8	350.2585	351.2623	[M-H] ⁺	0.52	94.08	98.9839, 71.0856, 57.0699	98.9833, 71.0850, 57.0698	32/8	6.8	0.44		Chemical blowing agent in polyurethane foams	
Diethyl phthalate (2-75-5)	IOPhET	n.a.	C ₂₄ H ₃₀ O ₅	+	18.6	398.2094	399.21161	[M-H] ⁺	1.19	95.23	305.1750, 261.1468, 149.0205	305.1778, 261.1507, 149.0255, 121.0288, 65.0414	52/4	6.76	0.44		Additive in adhesives	
Diethyl phthalate (10-1)	DMA	n.a.	C ₁₁ H ₂₀ O ₄	+	12.2	216.1361	217.1436	[M-H] ⁺	-0.2	99.73	125.0950, 97.1005, 83.0860, 55.0545	55.0546, 125.0927	28/56	2.86	0.8		Additive in materials of food packaging, active ingredient in pesticide products	
Diethyl phthalate (79-6)	DMS	n.a.	C ₁₂ H ₂₂ O ₄	+	13.8	230.1518	231.1596	[M-H] ⁺	-2.29	99.95	199.1334, 171.1365, 139.1114, 69.0705, 55.0502	69.0703, 55.0547	12/20	3.35	0.8		Additive in paints, fillers, adhesives, pesticides and ingredient used in automotive industry	
Diethyl phthalate (8-31-4)	DiPGDB	VELSIFLEX® 328, Velsiflex 342, Finsolv® PG-22	C ₂₀ H ₂₂ O ₅	+	14.9	342.1467	343.1544	[M-H] ⁺	-1.26	99.88	163.0755, 105.0338, 77.0395	163.0753, 105.0340, 77.0393	40/48	3.9	0.8		Additive in adhesives, building construction materials, paints, manufacture of furniture, personal care products	
Di-tert-butyl 4-hydroxybenzoate (98-0)	BHT-CHO	n.a.	C ₁₅ H ₂₂ O ₂	-	14.3	234.162	233.1547	[M-H] ⁻	-1.25	99.8	217.1226 , 59.0147	217.1221, 59.0147, 133.0656	8/28	4.2	0.8		Additive chemical in toys, drug product or related with the manufacturing of drugs	

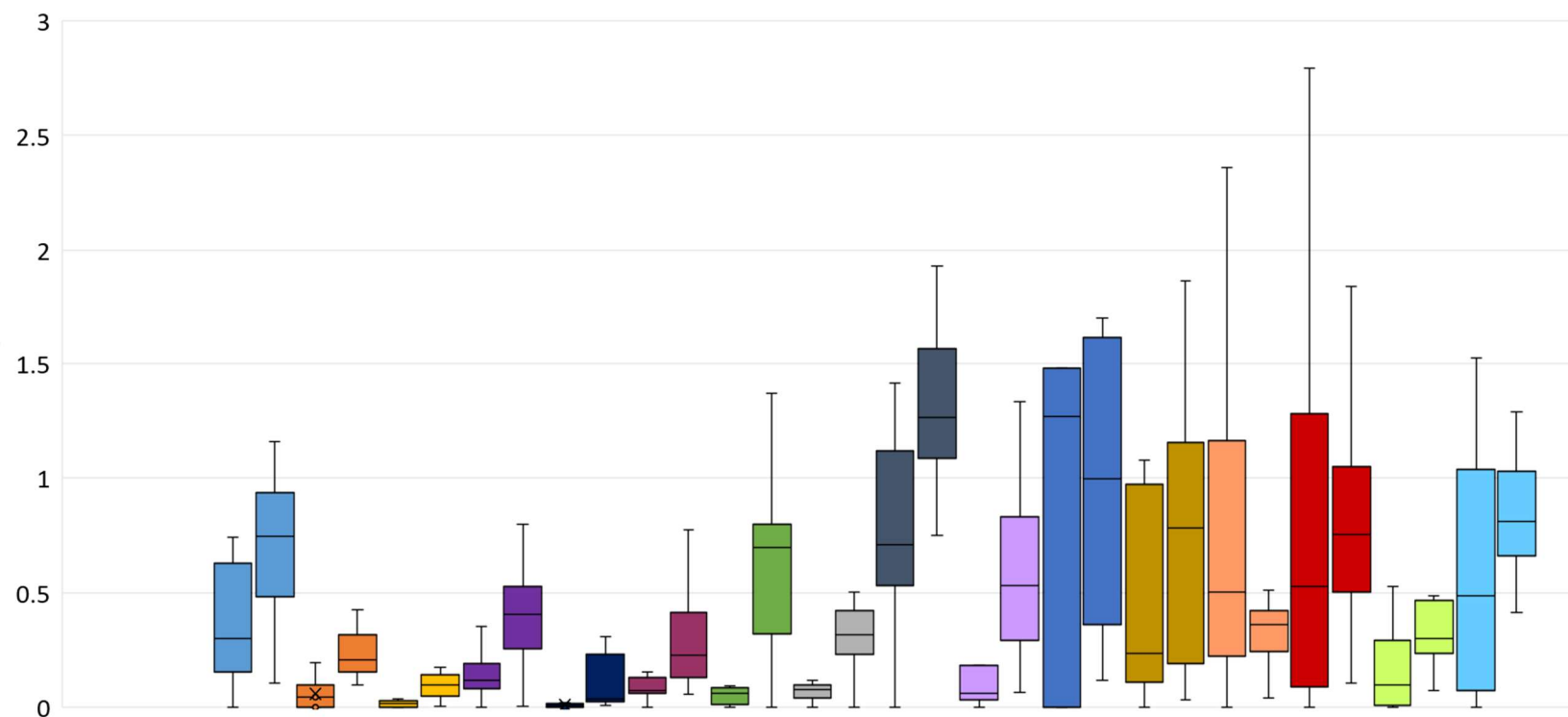
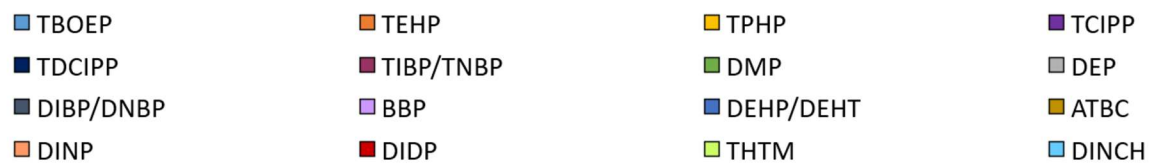


Figure 4. Abundance ratios of 15 native compounds identified in dust at LoC 1 (color-coupled error bars winter/summer).

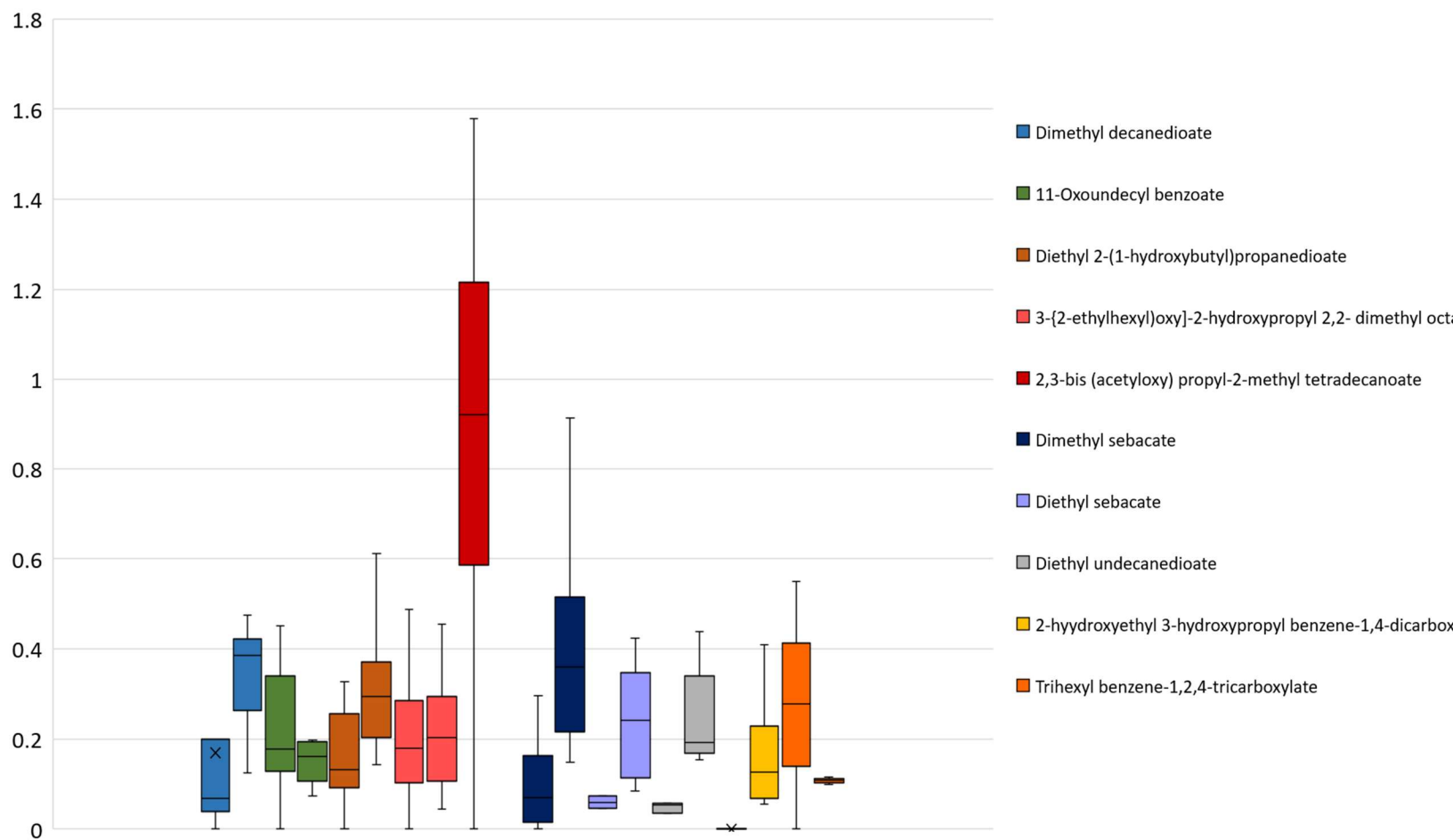


Figure 5. Abundance ratios of 10 alternative compounds identified in dust at LoC 3 (color-coupled error bars winter/summer).

Figure 2. Comparison of representative compounds between winter and summer season.

Compound	Winter					Summer				
	DF %	Mean Abundance Response Factor	5 th perc.	95 th perc.	DF%	Mean Abundance Response Factor	5 th perc.	95 th perc.	t-test (p-value))	Vapor Pressure (mmHg)
IS										
DEP	84	0.72	0.11	2.7	80	0.73	0.16	1.7	0.960	5.51E-06
BP	16	0.09	0.04	0.18	8	0.15	0.10	0.20	0.426	8.25 E-08
P	8	0.03	0.03	0.03	32	0.11	0.04	0.17	0.002	2.39E-05
PP	96	0.21	0.03	0.64	100	0.43	0.13	0.92	0.003	4.44E-03
IPP	12	0.01	0.01	0.02	20	0.08	0.01	0.25	0.295	2.60E-05
BP/TIBP	60	0.13	0.05	0.37	32	0.40	0.13	0.94	0.055	1.13E-03
Is & APs										
P	16	0.11	0.05	0.20	92	0.63	0.21	0.95	6.07E-07	3.08 E-03
P	60	0.08	0.03	0.12	88	0.38	0.17	1.0	6.51E-06	2.10E-03
P/DNBP	64	0.92	0.32	1.7	72	1.7	0.83	2.8	0.06	2.23E-05
P	44	0.30	0.02	1.1	88	0.60	0.15	1.1	0.06	8.25E-06
HP/DEHT	12	2.28	1.3	3.8	28	1.1	0.19	1.7	0.33	1.42E-07
PC	48	0.63	0.08	1.7	44	0.79	0.07	1.8	0.59	7.97E-07
P	76	0.86	0.17	2.1	56	0.42	0.09	0.96	0.04	5.40E-07
P	100	0.86	0.06	2.6	88	0.80	0.36	1.3	0.77	5.28E-07
PM	28	0.23	0.01	0.47	40	0.45	0.21	0.97	0.07	2.57E-08
CH	80	0.86	0.05	2.8	72	0.83	0.46	1.2	0.92	1.85E-07