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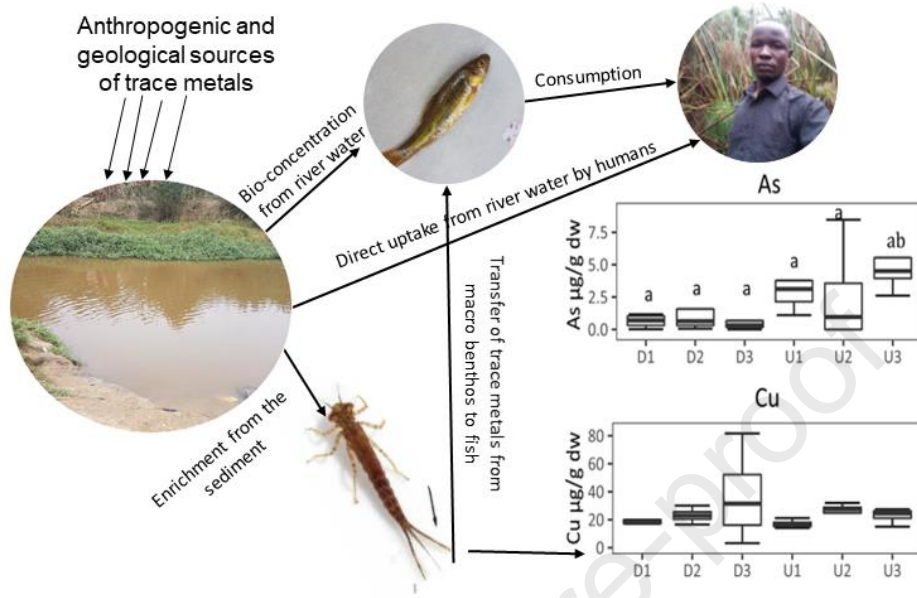
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1 **Trace metal concentrations in the abiotic and biotic components of River Rwizi ecosystem in**
2 **Western Uganda, and the risks to human health.**

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13

14 **HIGHLIGHTS**

15 • Surface waters at Katenga and Kayanja were contaminated with gold and mercury

16 • Concerning metals the surface water was generally safe for human consumption at

17 most sites

18 • The sediment trace metal levels posed no ecological risks to the benthic biota ~~such~~

19 • Arsenic and mercury concentrations in *Brycinus sadleri* muscle posed a potential

20 human health risk.

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22 **Abstract**

23 The distribution of metals in the Rwizi River ecosystem was investigated and human
24 health risks were assessed. Samples of water, sediment, damselfly larvae (*Ceriatagrion*
25 *glabrum*) and fish species (*Brycinus sadleri* and *Barbus altianalis*), were collected at six
26 sites. In all samples the trace elements As, Al, Au, Cd, Co, Cu, Fe, Hg, Mn, Pb, Zn, were
27 quantified. Sediment samples near the gold mine had significantly higher
28 concentrations of Hg, Fe and Al although all the concentrations were below the probable
29 effect concentrations (PEC). The dissolved concentrations of trace metals were within
30 the European standards and WHO drinking water guidelines. However, Fe and Mn
31 concentrations exceeded the standards at three sites. The damselfly larvae were good
32 indicators of local metal pollution. The fish species accumulated metal levels in the order
33 gills>liver>muscle for most metals except for Hg. Multiple regressions between
34 accumulated metals in damselfly with environmental metal levels showed only for Au
35 and Cd significant positive relationships. Relating environmental metal levels and
36 physicochemical characteristics to the levels in the invertebrates, only for Cu and Pb
37 significant relationships were found. With respect to the measured metals, the fish were
38 safe for human consumption in most cases although *Brycinus sadleri* posed a potential
39 health risk due to a As hazard quotient (HQ) of 2.2 that exceeded the critical value of 1.
40 Similarly, the maximum edible risk-free quantity (Q) for As in *Brycinus sadleri* was 1.5 g
41 (95 % CI), less than the minimum risk free quantity of 31.5 g. In conclusion, the river
42 water was safe for drinking but the extraction of gold using Hg should be replaced with
43 an environmentally friendly method or an effective wastewater treatment should be
44 instituted. People should be cautioned from consuming *Brycinus sadleri* to avoid
45 potential health hazards.

46 **Keywords:** Hazard quotient, trace metals, River Rwizi, *Ceriatodon glabrum*, *Barbus*

47 *altianalis*, *Brycinus sadleri*

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49 **1 Introduction**

50 Trace metal pollution in aquatic ecosystems has become a global threat (Onyari and
51 Wandiga, 1989), caused by increased urbanization, industrialization, burning fossil fuels
52 and mining activities releasing metals into the environment. For example, the global
53 number of artisanal and small scale gold miners have increased from 6 million in 1993
54 to 40.5 million in 2017 (IGF, 2017), resulting in an augmented annual release of 1400
55 tons of mercury into the environment (IGF, 2017). Metals are increasingly used in
56 industrial, domestic, technological, medical, and agricultural fields (Tchounwou et al.,
57 2012; Rehman et al., 2018), thus, mining them is indispensable. Trace metals are
58 naturally present in the aquatic environment in low soluble quantities (Duffus, 2002).
59 However, the increase of metals in the environment is mainly anthropogenically
60 influenced. If trace metals are deposited into aquatic ecosystems, they dissolve or
61 adsorb to suspended particles or sediment (Mataba et al., 2016), rendering the benthic
62 habitats a sink and source for secondary contamination in the ecosystem (Onyari and
63 Wandiga, 1989). From the sediment or water, metals may be taken up by biota such as
64 fish, plants or invertebrates (Goodyear and Mcneill, 1999; Nabulo et al., 2008), and
65 potentially posing human health risks via the consumption of contaminated water and
66 food (Rice et al., 2014).

67 Trace metals are mostly transferred by rivers from contamination points to other aquatic
68 ecosystems such as seas, lakes, oceans, estuaries and wetlands. For example, the River
69 Nyamwamba-Rukoki in Uganda which originates from the Rwenzori Mountain
70 (Mwesigye and Tumwebaze 2017) deposits 30 tons of Cu and 13 kg of Cd annually into
71 Lake George from Kilembe mines (Hartwig et al., 2005). Rwizi River in southwestern
72 Uganda, which has been cited to be polluted with trace metals (Egor et al., 2014; Ojok

73 et al., 2017; Semwanga et al., 2020), drains into lakes Mburo, Kachera, Nakivale, and
74 Victoria.

75 In Africa, trace metal pollution in the environment has reached unprecedented levels
76 (Yabe et al., 2010; Fasinu and Orisakwe, 2013) although most governments have put less
77 effort to combat the menace and have prioritized malnutrition, infant and maternal
78 mortalities, and communicable diseases (Fasinu and Orisakwe, 2013). In Uganda,
79 presence of trace metals in sediment and invertebrates are usually not addressed,
80 hence leaving a knowledge gap. Because of bioavailability and toxicity mechanisms of
81 the trace metals in the aquatic ecosystem, measurements in the sediment,
82 invertebrates, fish and surface water should be included for a complete understanding
83 of the system status. Sediment inhabits biota and may act as a secondary source of
84 nutrients as well as bring metals into the aquatic ecosystem through resuspension
85 (Mataba et al., 2016). Further, most benthic invertebrate species are sedentary, and
86 therefore appropriate biological tools to monitor bioavailability of local pollutants
87 (Wakwabi et al., 2006).

88 Very little information is available on the distribution of metals in aquatic ecosystems in rivers
89 in Uganda and information on metals in edible fish from Uganda is scarce (Omara et al. 2019)
90 and non-existing for the study area.

91 Therefore the central aim of this study was to investigate the distribution of trace metals
92 in the River Rwizi ecosystem in the vicinity of a gold mine. More specifically the
93 objectives were (1) to evaluate the extent of metal pollution in water, sediment and
94 biota in the catchment of River Rwizi, (2) to establish relationships between trace metal
95 levels in the environment and the biota, taking into account water and sediment

96 characteristics and (3) to assess the human health risks of drinking water and consuming
97 fish contaminated with trace metals.

98

99 **2 Materials and methods**

100 **2.1 Study area**

101 The Rwizi River originates in the hills of Buhweju hills in Western Uganda and flows for 55
102 km through Lake Mburo, Nakivali, Kachera, Kijanebalola to Lake Victoria (Songa et al.,
103 2015; Semwanga et al., 2020). Despite being shallow with a maximum depth of about 2
104 to 3 m, it is the principal source of water for domestic, agricultural and industrial
105 activities in southwestern Uganda (NEMA, 2009). Ecologically, it supports a network of
106 highly biodiverse ecosystems such as the Lake Mburo National Park and two Ramsar
107 sites, namely Lake Mburo-Nakivali Wetland System (Songa et al., 2015) and the Sango
108 Bay Wetland Forests (Ojok et al., 2017). The river catchment area of 2,282 km² contains
109 natural resources such as gold, sand and fertile soil, which have attracted human
110 exploitation (Mugonola, 2013; Mugonola et al., 2015). Herewith, the wetlands and river
111 banks were reclaimed for cultivation, human settlement, gold and sand mining (NEMA,
112 2009; Mugonola, 2013).

113 Six study sites were selected and marked by GPS 12XL- Garmin (figure 1). Katenga (U1)
114 was situated upstream of the river near the gold mining field followed by Kayanja (U2).
115 Sheema (U3) was located downstream from the mining field and was only affected by
116 farm inputs. Rwebikona (D1) received industrial wastes and Katete (D2) received
117 municipal wastewater. Finally, Buleba (D3) was considered a control site with less human
118 interference.

119

120 **2.2 Water characteristics**

121 At each site, water pH, conductivity and temperature were measured in quadruple in-
122 situ at about 3 cm depth with a multi-probe Hanna meter (HI9828, USA).

123

124 **2.3 Collection, laboratory preparation, and analysis of samples**

125 **2.3.1 Field sample collection and preservation**

126 At each site, invertebrates, sediment and surface water samples were randomly
127 collected, stored in polypropylene tubes and preserved in a cooler box in the field. Four
128 replicate water samples were taken from each of the six sites (table 1) at a depth of ca.
129 5 cm using acid- washed 50 ml polypropylene (PP) tubes, which were closed airtight.
130 Four replicate sediment samples were collected with a 2.5-inch diameter hand-held inox
131 corer from the same locations where water samples were collected, and stored in 50 ml
132 PP tubes (table 1) Upon arrival in the lab water samples were filtered over a cellulose
133 acetate filter (0.2 μ m), acidified by adding 150 μ l HNO₃ (69% ultrapure) and transferred
134 to 14 ml PP tubes. All samples were frozen at -20°C before shipment to Antwerp
135 University for trace metal analysis.

136 For sediment characterization from each site, a subsample of 5 g was taken from each
137 replicate sample and pooled. Subsamples were obtained per site for the measurement
138 of Total Organic Carbon (TOC) and clay content. For TOC analysis about 10 g per site was
139 transferred to pre-weighed 50 ml PP tubes and dried at 60°C for 24 h to obtain the dry
140 weight (DW₆₀). The content was transferred to crucibles and placed in a muffle oven
141 that was slowly heated up to 550°C for one hour, hereafter this temperature was kept
142 during another four hours. The samples were cooled and weighed again (DW₅₅₀). The
143 TOC was calculated from the loss of ignition method (LOI) as described by Heiri et al.(

144 2001) according to the equation: $TOC (\%) = ((DW_{60} - DW_{550}) / DW_{60} * 100) / 1.742$. With
145 DW_{60} and DW_{550} the dry weight after heating at 60 and 550°C respectively. The
146 correction factor 1.742 was applied since it is assumed that 57 % of the organic matter
147 is carbon (Nelson & Sommers 1996) .

148 For the determination of the clay content another subsample of about 1 g was pre-
149 treated with 40ml of H_2O_2 (33%) and 9ml of HCl (30%) to digest organic material and
150 iron conglomerates. The percentage of clay (fraction < 2 μ m) in the sediment samples
151 was measured by laser diffraction (Malvern Mastersizer 2000).

152 Following the USEPA (1995) and Barbour et al. (1998), invertebrates samples were
153 randomly taken at every location using a 1-mm mesh sweep net and washed off through
154 a 500- μ m mesh benthic net. At each site four larvae of common waxtail damselfly
155 (*Ceragrion glabrum*) of comparable size were sorted out and preserved (table 1).

156 At each site an attempt was made to collect fish by a hook and line. Only at three out of
157 the six sites fish could be collected (table 1). In total, nineteen individual fish were
158 collected. Nine Rippon barbel (*Barbus altianalis*), of a total length (TL) of 8.9-25.9 cm
159 were captured from the river urban area (Katete, D2), and five of TL =17.4-28.7 cm from
160 the remote area (Rwebikona, D1) (table 1). Rippon barbel is an omnivorous freshwater
161 cyprinid inhabiting the inland waters in preferably sand and gravel substrate of most
162 water bodies in East Africa, and migrates upstream to spawn in marginal and vegetation
163 covers (Lévêque, 1997; FishBase team RMCA & Geelhand, 2016). Its stocks are
164 threatened by climate change and overfishing (Balirwa, 1979; Rutaisire et al. 2015). Five
165 Sadler's robber (*Brycinus sadleri*) of TL=11.2-14.2 cm were captured at Kayanja (U2)
166 which was near a gold mining site U1 (table 1). Sadler's robber is a freshwater characid,
167 which mostly lives in open waters, wetlands, marginal vegetation covers, and migrates

168 upstream to spawn (Ojuok 2008; Olowo et. al 2004; Fishbase 2016). Under drastic
169 conditions such as severe pollution, over-fishing and habitat degradation it becomes a
170 generalist, feeding on plants during day and insects at night (Wanink et al., 2007). Both
171 species were selected because of their high abundance during the study period. Before
172 dissection, fish were washed with deionized water. Gills, muscles and liver were
173 collected and stored at -20°C.

174

175 **2.3.2 Laboratory preparation of samples.**

176 Invertebrate, sediment and fish tissue samples were transferred into pre-weighed PP
177 tubes and the total weights were assessed (Mettler Toledo AT261 Delta Range).
178 Procedure blanks and standard reference material including mussel tissue (NIST-2976;
179 National Institute of Standards and Technology, USA)) and estuarine sediment (BCR-
180 277R) were included. The samples were freeze-dried for four days in a Thermo Scientific,
181 Heto PowerDry LL3000 freeze dryer, which was connected to a Thermo Savant VL P80
182 Valu Vacuum Pump. The samples were removed and stored until digestion. They were
183 grouped into two categories according to dry weight and appropriate acid volumes were
184 added.

185 Samples were transferred to 10 ml glass tubes, and a mixture of HNO₃ (69%) and HCl
186 (37%) (1:3; Aqua Regia) was added to keep mercury in a stable solution, hereafter they
187 were left at room temperature for 24h. Subsequently, 200 µl of H₂O₂ was added to digest
188 the fat tissue. The samples were digested in two steps at 120°C and 160°C respectively
189 in a pressurized microwave system (CEM Discover SP-D; CEM Corporation, Matthews,
190 NC 28106, USA). After digestion, the samples were diluted with Milli-Q water. For water

191 samples, a mixture of HNO₃ and HCl (1:3) was added and stored at -20°C until analysis
192 (Mataba et al., 2016).

193

194 **2.3.3 Trace metal analysis**

195 Ten trace metals: aluminium (Al), cadmium (Cd), cobalt (Co), copper (Cu), gold (Au), iron
196 (Fe), lead (Pb), mercury (Hg), manganese (Mn), zinc (Zn) and the metalloid arsenic (As)
197 were analysed using a High Resolution Inductively Coupled Plasma Mass Spectrometry
198 (HR-ICP-MS; Element XR, Thermo Scientific, Bremen, Germany). The instrument
199 detection limits varied from 0.001 µg/L for Cd, Au, Pb, Co, and As to 0.01µg/L for Hg, Al,
200 Mn, Fe, Cu, and Zn. The data was recorded in µg/L, and the recoveries for the metals
201 ranged from 80 to 120% except for Hg which was 150% and a correction factor of x/1.5
202 was applied on Hg concentrations except for water samples (Thompson et al., 1996). For
203 sediment and biota the concentrations were expressed as µg/g dry weight (µg/g dw)
204 except for the risk for human consumption where concentration in muscle tissues were
205 expressed on a wet weight basis (µg/g ww).

206

207 **2.3.4 Heavy metal evaluation index (HEI) for surface water**

208 The heavy metal evaluation index (HEI) by Edet & Offiong (2002) was used to evaluate
209 the pollution status of Rwizi River surface water (equation 1)

$$210 \text{ HEI} = \sum_{n=1}^n \frac{H_c}{H_{mac}} \quad (1)$$

211 With H_c: mean concentration of each metal in the surface water at each site, and H_{mac}:
212 maximum permissible levels in potable water by UNBS (2014). HEI values were
213 categorized according to the mean value of the HEI for the overall study area (Prasanna
214 et al., 2012; Biswas et al., 2017). The pollution status for each site was accordingly

215 categorized as low (HEI: < 400), medium (HEI: 400-800), and high (HEI: > 800) (Edet &
216 Offiong, 2002).

217

218 **2.4 Human health risk assessment: maximum edible risk-free quantity of fish**
219 **consumed without posing deleterious health risks to a person of 70kg average**
220 **body weight**

221 If an individual is exposed to trace metals through consuming contaminated fish meat,
222 a quantity at which no health risks are posed can be calculated. For Hg it is assumed that
223 80-100 % of the total mercury in the fish muscle is present as methylmercury (EFSA,
224 2012), and thus a conversion factor of 1 was applied (EFSA, 2012). The maximum edible
225 risk-free quantity (Q) was determined following Mataba et al. (2016) (equation 2):

226
$$Q = \frac{(W \times M \times 1000)}{C} \quad (2)$$

227 Where Q: Maximum quantity (g) of edible fish meat to be consumed without posing
228 health risks to an individual of 70kg average body weight;

229 W: Weight (kg) of an average person;

230 M: Minimum Risk Levels (mg/kg/day) for the trace metals as determined by ATSDR
231 (2018); and

232 C: the 50th and 95th percentile concentration ($\mu\text{g/g ww}$) in the fish meat for the two
233 species.

234 A Hazard Quotient (HQ) was determined as a ratio of the daily fish meat consumed per
235 person in Uganda (31.5 g, as noted by Kasozi et al. (2017)) to the maximum risk-free
236 quantity (Q). A HQ >1 signified potential human health risks posed with consumption of
237 contaminated fish meat with a particular metal and HQ < 1 showed no potential health
238 risks.

239

240 **2.5 Statistical analysis**

241 Data analysis was performed in R 4.0.2 (R Core Team, 2020). Shapiro-Wilk and Levene's
242 test were performed to test for data normality and homogeneity of variances
243 respectively before parametric tests were conducted. Kruskal Wallis or One-way ANOVA
244 was used to test for differences in the trace metal concentrations in water, sediment
245 ,invertebrates and fish among the different sites. Tukey (HSD) or Pair-wise Wilcoxon
246 tests were used as post hoc tests. Kruskal-Wallis was used to compare variations
247 between trace metal levels in fish gills and muscle. Spearman ranks correlation test was
248 performed to determine the relationship between the trace metal levels in water,
249 sediment and invertebrates. Multiple regressions were performed to determine the
250 relationship between the metal levels in damselfly with environmental trace metal
251 concentrations taking into account the physicochemical characteristics (pH,
252 conductivity, clay content, and TOC). No statistical tests were performed if 50% of the
253 measurements were below the limit of detection (Custer et al., 2000). Outliers were
254 examined by applying the Grubbs test (<http://www.graphpad.com/quickcalcs/>) and
255 removed if found statistically significant.

256 **3 Results**

257 **3.1 Water and sediment characteristics**

258 Temperature, pH and conductivity increased from upstream (U) to downstream (D) of
259 the river (Supplementary Information, SI table S1). Conductivity ranged from 28 to 100
260 $\mu\text{S}/\text{cm}$ with the highest value recorded at D3 and the lowest at U1 and this was
261 significantly different among sites ($\chi^2_{(1,5)} = 21.7, p = 0.0005$). Temperature ranged from
262 16.1 to 22.7°C, with the lowest measured at D2 and the highest at D1 (table 2). However,

263 the water temperature was not exactly measured at the same moment of the day at all
264 sites and thus the differences could not be statistically tested. For pH, there was a
265 significant difference between upstream and downstream ($\chi^2_{(1)} = 156.4, p = 0.0001$);
266 however, there were no statistically significant differences among sampled sites
267 ($p > 0.05$).

268 Table S2 show the mean % composition of clay content and Total Organic Content (TOC)
269 in the sediment. High clay content percentage was recorded at U3 (6.27 ± 1.26), followed
270 by U2 (5.64 ± 2.4), and least at D2 (1.24 ± 0.00). The sites did not significantly differ in the
271 % clay content ($\chi^2_{(1, 5)} = 4, p = 0.42$). High percentage TOC was recorded at U2
272 (0.51 ± 0.19), followed by D3 (0.28 ± 0.00), and lowest at D1 and U3 (0.01 ± 0.00). Similar
273 to clay content, the sites did not significantly differ in their TOC concentration in the
274 sediment ($\chi^2_{(1, 5)} = 5, p = 0.42$).

275

276 **3.2 Environmental trace metal levels in the sampled sites of Rwizi River (surface** 277 **water and sediment)**

278 Trace metals concentrations measured in surface water for the six sampled sites from
279 Rwizi River are presented in table 2. Except for As, the metal concentrations were
280 significantly different among sampled sites. Gold, As and Hg were below the detection
281 limit at most sampled sites except at U1 (gold mining site) where Au was above
282 detection limit. For other metals significant differences were recorded at particular sites.
283 For example, at U1 the levels of Al, Hg and Pb were significantly different from other
284 sites (table 2). Zinc, Mn and Co were significantly higher at U3 (Sheema) and comparable
285 levels were measured for Cd between U2 and U3. Similarly, Cu concentrations were

286 comparable between D3 and U3, and for Fe, significantly low levels were measured
287 between U2 and U3.

288 In Figure 3 and table S3 of the supplementary information, the trace metal
289 concentrations in the sediments are presented. Only Pb, Cd, Co, Cu and Hg levels did
290 not differ between the six sampled sites ($p>0.05$). Gold, Al, As, Fe, Mn and Zn
291 concentrations were significantly different among sites (Figure 3). Pair-wise Wilcoxon
292 comparisons among sites showed that Al and Au were highest at U1, As was lowest at
293 D1, Mn was highest at U2 while Fe and Zn were lowest at U2. Trace metal ranges in the
294 sediment samples were determined for comparison with the standard sediment quality
295 guidelines. Only trace metals with known probable effect concentrations and severe
296 effect concentrations were evaluated.

297 Correlations between concentrations in water and sediment resulted in significant R-values ($p<$
298 0.001) for Cd, Hg and Zn only with $R=0.99$; 0.98 and 0.92 respectively.

299

300 **3.3 Surface water pollution status of River Rwizi using Heavy metal evaluation index** 301 **(HEI)**

302 Table S4 presents the results of the Heavy metal Evaluation Index HEI for the six sites.
303 Two outliers were found in the concentrations of Fe at U1 and removed before
304 calculating the HEI index. HEI values decreased from upstream (U) to downstream (D)
305 and ranged from 1.3 to 8.7. The downstream sites values ranged from 1.3 to 1.8, and
306 the upstream from 2.4 - 8.7 with highest HEI at Sheema. The average HEI value for all
307 sites was 3.3 and HEI was categorized as low (0-3), medium (3 to 6) and high (>6) (Table
308 S4).

309

310 **3.4 Trace element concentrations in damselfly larvae**

311 Trace metal concentrations in damselfly larvae are presented in Figure 4 and table S5.
312 All trace elements in the larvae were above the detection limits in all the sampled sites.
313 Only the concentrations of Fe, Cu, Mn and Zn in the damselfly larvae did not statistically
314 differ among the sampled sites (Figure 3). Multiple pair-wise Wilcoxon comparisons
315 shows highest Au and Pb levels were observed at U1 while As, Cd, and Co concentrations
316 differed at U3 compared to other sites. Higher Al levels were recorded at U3 and lower
317 at U2.

318

319 **3.5 Trace metal concentrations in *Barbus altianalis* and *Brycinus sadleri***

320 Metal concentrations in gills from *Barbus altianalis* at D1 (Rwebikona) and D2 (Katete)
321 and from *Brycinus sadleri* at U2 (Kayanja) are presented in figure 4 and table S6. Only
322 Mn and As concentrations did not differ among the sampled sites. However, *Brycinus*
323 *sadleri* had significantly higher levels of Al Co, Cd, Hg and Zn compared to *Barbus*
324 *altianalis*. The levels of Pb were significantly lower at D1 compared to U2 and D2.
325 Concentrations in muscle from the same species at the three sites are presented in
326 figure 5 and table S6. Except for Cd and Cu at U2, for none of the measured trace metals
327 significant differences were found.

328 Liver could only be collected from *Barbus altianalis* from Rwebikona (D1) and Katete
329 (D2). No significant difference between the sites could be found for none of the
330 measured metals (table S6).

331 For *Barbus altianalis* the order of magnitude of metal concentration in tissues was
332 liver>gill>muscle for Au, Cd, Co, Cu, Fe and Zn and gill>liver>muscle for Al, Mn and lead.

333 For As no significant differences were found among the tissues and for Hg

334 concentrations in muscle equalled the ones in liver but were higher than in gills. For
335 *Brycinus sadleri* we could only compare between gills and muscle. For Cd, Co, Cu, Fe and
336 Mn ($p < 0.05$) significant higher concentrations were measured in gills. No significant
337 differences were found for Al, As, Co, Hg, Pb and Zn ($p > 0.05$).

338

339 **3.6 Relationship between environmental and accumulated concentrations**

340 Since fish could not be captured at the same sites as where the sediments and the
341 surface water were sampled, we could only relate the environmental concentrations to
342 the concentrations in the invertebrates. Significant correlations ($p < 0.01$) between water
343 and sediment concentrations were only found for Cd and Zn (R-values of respectively
344 0.99 and 0.92). For Hg correlation could not be investigated since only at one site the
345 dissolved concentrations was above the detection limit. Multiple regressions were
346 constructed to relate concentrations in the environment with accumulated
347 concentrations in invertebrates, taking into account water (pH, conductivity) or
348 sediment (TOC, clay content) characteristics. Only for Au, Cd, Cu, Fe, Hg and Pb
349 significant relationships were found between concentrations in water or sediment and
350 invertebrates. In a few cases, water or sediment characteristics contributed significantly
351 to the described variation in accumulated metals. This was the case for TOC with Au,
352 conductivity for Cd, pH for Fe and TOC for Hg and Pb (Table 3). Also, multiple regressions
353 with both concentrations in the water and sediment with invertebrates showed that
354 significant relationships were only observed for Au ($R^2=0.99$, $p < 0.001$), Cd ($R^2=0.958$, p
355 < 0.05) and Pb ($R^2=0.85$, $p < 0.01$) (Table 3).

356

357 **3.7 Human health risks assessment**

358 The maximum edible risk-free quantity (Q) consumed without potential health risks and
359 the hazard quotients (HQ) were calculated (table 4). The Qs for the two fish species at
360 both median and upper limit was greater than the daily consumption rate per person
361 (31.5 g) in Uganda except for As (14 g) in *B. sadleri* at upper limit bound (95% CI).
362 Similarly, the HQs for the assessed metals in both fish species were <1. However, the HQ
363 for As in *B. sadleri* was 2.2 at the 95th percentile (table.6). The HQs for Fe, Mn and Au
364 were not determined since no minimum risk levels (MRL) were available.

365

366

367 4 Discussion

368 4.1 Physicochemical characterisation of water and sediment

369 The surface water electrical conductivity (EC) significantly increased downstream of the
370 Rwizi River. However, pH and EC were within the Uganda natural water quality standards
371 (UNBS, 2014). Electrical conductivity is governed by natural river geology and hydrology
372 but the high levels in the urban sites could be due to influx of ions from the catchment
373 and transportation from the upstream sites of the river (Van Butsel et al., 2017).

374 Surface water pH of the downstream sites was significantly lower compared to the
375 upstream sites. Earlier studies conducted on the Rwizi River in Uganda had similar pH
376 ranges from 6 to 7 in the downstream areas (Egor et al., 2014; Semwanga et al., 2020).

377 The low pH upstream could be attributed to leaching of hydrogen ions from the acidic
378 soils of the riverbed and catchment (Banga, 2014). Studies on Sondu-Miriu Rivers in
379 Kenya and Okpokiri River in Nigeria showed the variations in water pH were attributed
380 to demineralisation from the riverbed (Vuai et al., 2012; Evbuomwan & Obinuchi, 2018).

381 The pH range from 5.7 to 7.1 in the downstream part was perhaps influenced by influx
382 of wastewaters from the urban catchment. The pH ranges were similar to Egor et al.,
383 (2014) and Semwanga et al. (2020) although sampling were conducted in different sites.

384 In the present study, the TOC and clay content were not significantly different among
385 sampled sites. However, both Katenga (U1) and Katete (D2), which received high silt and
386 organic matter from the goldmines and urban areas, respectively, had relatively high
387 clay and TOC levels in the sediment which were higher than the levels in Thigithe River
388 in Tanzania (Mataba et al., 2016). The sediment properties such as organic content and
389 clay or silt content may have an important effect on the concentration of trace metals
390 in the sediment (Mason, 2013). Total organic carbon increases binding capacity of some

391 trace metals in the sediment and determines their release into the water or
392 bioavailability to the biota such as invertebrates and fish (Allen, 1993). For example,
393 metal bioavailability to chironomids decreased with the increase in TOC in the sediment
394 (Bervoets et al., 1997, 1998). Further, the clay content will determine the rate of ion
395 exchange over the sediment surface (Allen, 1993). For example, clay types such as
396 smectite and vermiculite are natural cation agents in the soils and sediment (Allen,
397 1993).

398

399 **4.2 Trace metal concentrations in the sediment and water**

400 In the present study, Au, As, Al, Mn, Zn and Fe were significantly different in the
401 sediment among the sampled sites. The level of Au, was significantly higher at the
402 goldmine which implied an anthropogenic enrichment due to wastewater disposal. In
403 the surface water, Hg was only above the detection limit at Katenga and Kayanja, sites
404 within and near the goldmining area respectively. The insignificant levels of Hg in
405 sediment samples among sites but significantly higher concentrations in water at
406 Katenga could perhaps be attributed to low sedimentation rate. Mataba et al. (2016)
407 also observed significantly higher Hg levels near the goldmine in the sediment of
408 Thigithe River in Tanzania. The significantly higher concentration of Al at Rwebikona
409 (D1), and As at Katete (D2) were possibly due to enrichment from the catchment since
410 these sites were located in the urban areas. Similar studies on trace metal levels in the
411 sediment, for example, on lakes Kwania, Nakuwa, Opeta and Kyoga in Uganda measured
412 0-50 $\mu\text{g/g dw}$ of Cu, 0-133 $\mu\text{g/g dw}$ Zn and 10140-98930 $\mu\text{g/g dw}$ Fe in sediments (Ocaya,
413 2010). These exceeded the levels of Cu and Fe measured in River Rwizi in the present
414 study although the lakes were not affected by anthropogenic pollution sources (Ocaya

415 2010). Thus, trace metal enrichment in the sediment could be determined by natural
416 geological sources (Jagus et al., 2013; Machowski et al., 2019). Bugenyi (1982) identified
417 an influx of trace metals from the Kilembe copper mine into lakes George and Edward
418 in Uganda. In the present study, a similar trend was observed for Hg, Al and Fe
419 concentrations at U1 which was in close proximity to the Katenga gold mine. Previous
420 studies which measured trace metals in River Rwizi ecosystem did not quantify the
421 enrichment in sediment and hence the present study can be used a baseline for the
422 river.

423 In comparison with numerical sediment quality guidelines (MacDonald et al. 2000), the
424 assessed metals were all below the Severe Effect Levels (SEL). However, both Zn and Hg
425 levels exceeded the Probable Effect Concentrations (PEC) (Zn: 315 $\mu\text{g/g}$ and Hg: 0.486
426 $\mu\text{g/g}$). Thus, Zn and Hg are potentially toxic for the benthic biota such as the damselfly.
427 Mataba et al. (2016) reported similar findings on Thigithe River although severe effects
428 levels were exceeded for As in the sediment. The present study findings were not
429 comparable with Sekabira et al. (2010) on the metal loads in the sediment from highly
430 polluted urban streams due to industrial discharge. For example, Zn levels of 177-442
431 $\mu\text{g/g dw}$ and 341-1968 $\mu\text{g/g dw}$ were measured in Nakivubo stream and industrial waste
432 sediment in Kampala, Uganda. These levels exceeded both the SEL (820 $\mu\text{g/g}$) and PEC
433 (315 $\mu\text{g/g}$) (MacDonald et al., 2000). Therefore, River Rwizi sediment can be considered
434 as moderately polluted by trace metals from anthropogenic sources.

435 The trace metal levels in the surface water were similar to findings by Semwanga et al.
436 (2020) but contrasted Egor et al. (2014) on Rwizi River although in all studies sampling
437 was conducted in different sites. Egor et al. (2014) sampled in the downstream part of
438 the river, and therefore less impact of gold mine wastewaters was expected. Since Hg

439 and Au were not detected in the downstream sites but only in the sites (U1 and U2) near
440 the goldmining site, it can be concluded that surface water was polluted
441 anthropogenically from Katenga gold mines. Also, significantly higher Fe and Al
442 concentrations at Katenga (U1) could suggest a possible influx from the mine wastes.
443 The higher Pb levels at Katete (D2) could perhaps be attributed to contamination from
444 the wastewaters from both domestic and industrial activities (Tchounwou et al., 2012).
445 Egor et al. (2014) reported similar trends of Pb levels increase in the urban sites although
446 the levels of 0.75 ug/L were higher than in the present study. Further, Katete (D2) was
447 near a road bridge and local people washed motorcycles in the river causing direct
448 dumping of oil and fuel wastes into the water which were potentially the sources of Pb
449 contamination at the site. At sites U2, U3 and D3 the maximal measured As
450 concentrations exceeded the European water quality standards (European Commission
451 2013) whereas this was only the case for Co and Zn at site U3 (table 2). This may pose
452 a potential risk to the aquatic communities.

453 At Buleba (D3), the significantly higher levels of Cu and Zn perhaps showed natural
454 enrichment since there were less human activities. However, Egor et al. (2014) reported
455 significantly higher levels of Zn 2.54 ug/L in 2011 and 2.53 ug/L in 2010 in the urban
456 sites which was attributed to influx of runoff from zinc corrugated iron sheets from the
457 urban area and perhaps it was carried downstream. The present study contrasted the
458 findings of Mataba et al. (2016) on Thigithe River where most metals in surface water
459 were below the detection limit. However, Omara et al. (2019) observed similar Hg
460 concentrations. For example, 0.15 µg/L was measured near the gold mine in
461 Namukombe stream in Uganda compared to 0.175 µg/L in Rwizi River near the Katenga
462 gold mine. Similar trends of Hg levels decreasing downstream from the gold mine have

463 been observed in Mataba et al. (2016). This trend suggested that the goldmines were
464 the major source of the Hg in surface water. Mercury is mostly used in amalgamating Au
465 from the mud slurry, however, contamination into the environment was attributed to
466 inadequate treatment of wastewater.

467

468 In terms of human drinking water assessment, the metal loads in the surface water were
469 within the Uganda national standards for potable natural water (WHO 2017) except for
470 Fe and Mn mean levels at U1 and U2 where standards were exceeded (table 2). Similar
471 findings on the same river were reported by Semwanga et al. (2020) and Egor et al.
472 (2014). However, although the surface water was safe for humans to drink at most of
473 the sites, this was based on the measured metals only at that particular time of
474 sampling. Therefore, it was not exclusive for other contaminants or pollutants such as
475 total coliforms, and turbidity, smell, taste and other compounds such as total nitrogen,
476 phosphorus, soluble reactive silica and organic micro pollutants.

477

478 **4.3 Surface water trace metal status using trace metal evaluation index**

479 In the present study, Katenga (U1) had trace metal hotspots for iron based on the
480 outliers which were detected at that site. However, the surface water had generally low
481 trace metal loads as shown by the mean HEI values. Rwebikona (D1), Katete (D2), Buleba
482 (D3) and Katenga (U1) had low surface water metal pollution. However, Kayanja (U2)
483 and Sheema (U3) had medium and high metal pollution, respectively. However, the
484 index was highly sensitive to outliers because the mean values were used to compute
485 the index (Edet and Offiong, 2002). The outliers of Fe concentration at Katenga were
486 attributed to the replicates obtained from stagnant water that possibly created a trace

487 metal hotspot. Possibly, the high metal pollution at Sheema could be associated with
488 natural elevated metal concentrations and to a lesser extent from point sources such as
489 the influx from the nearby farms. Probably, the low metal pollution in the river was
490 caused by a low influx from the catchment since the sampling was conducted in a dry
491 season. Although Egor et al. (2014) and Semwanga et al. (2020) conducted the sampling
492 in the wet seasons, the metal concentrations were not significantly different from the
493 present study. Incidentally, the samples were collected from different sites but the
494 studies indicated that natural enrichment was an important factor that determined the
495 metal concentrations of the Rwizi River.

496

497 **4.4 Trace metals in the biota: damselfly and fish.**

498 In the present study, Au and Pb were significantly higher in the damselfly larvae from
499 Katenga (U1) and Kayanja (U2). High accumulation of Au was observed at Katenga (U1),
500 which was in close proximity to the goldmines. A study of Bervoets et al. (1997) on the
501 impact of mining activities on Bolivian rivers, showed that chironomid larvae
502 accumulated higher levels of Zn, Pb, Cu and Cd compared to measured concentrations
503 in the damselfly from Rwizi River. Chironomid larvae are highly pollution tolerant
504 (Bazzanti, 2000) because they possess a red blood pigment for extraction of oxygen at
505 low levels (Sriariyanuwath et al., 2015). Simon et al. (2017)'s study on Tisza and Szamos
506 rivers in Turkey affected by mining activities showed that *Gomphus flavipes* larvae
507 accumulated Al, Zn, Mn, Fe, Pb and Cu concentrations similar to damselfly larvae from
508 Rwizi River. In the Olifant River Basin (ORB), Verhaert et al. (2019) measured total Hg
509 concentrations of 0.06 to 0.29 $\mu\text{g/g dw}$ in the snail *Tarebia granifera* and 0.08 to 0.69
510 $\mu\text{g/g dw}$ in odonata larvae, which were comparable to the 0.05 to 0.27 $\mu\text{g/g dw}$ in the

511 damselfly larvae in the present study. The trend of trace metal accumulation in the
512 damselfly larvae was similar to findings of Erasmus et al. (2020), which was higher in the
513 macroinvertebrate families near or within platinum mining sites. In contrast, the
514 annelids, molluscs, *Mugil sp.*, *Solea sp.*, and *Tilapia sp.* from Lake Qarun in Egypt
515 accumulated less Zn, Mn, Cu, Pb, Cd, Co and Fe (Mohamed and Mohamed, 2005), yet
516 the lake was polluted by industrial and agricultural waste (Hussein et al., 2008). Similar
517 to other studies, our study has indicated that macroinvertebrates significantly
518 accumulated higher trace metal levels near mining sites and urban areas, indicating they
519 are good bio-indicators of environmental pollution.

520 For most trace metals measured in *Barbus altianalis* gills and liver accumulated higher
521 levels than muscle. A similar trace metal accumulation pattern was reported by Bervoets
522 et al. (2001), Bervoets & Blust, (2003); Szarek-Gwiazda & Amirowicz (2006); Mataba et
523 al. (2016); Semwanga et al. (2020). However, these studies have been conducted on
524 different waterbodies and fish species. Copper, Fe, Zn, Pb, Cd and Mn concentrations in
525 the liver and muscle of African catfish were lower than in the present study except for
526 Al that was higher with 545 µg/g dw in the liver and with 69.2 µg/g dw in the muscle
527 (Semwanga et al., 2020). This was probably due to the higher Al levels in water (225
528 µg/L) in addition to species specific differences (Van Ael et al., 2017). Similar to Teunen
529 et al. (2017), Mataba et al. (2016), Gilbert et al. (2017) the order of Hg concentration
530 was muscle>liver>gills. In *Oreochromis niloticus* captured 30 m downstream of the
531 goldmine in Namukombe stream in Uganda, no Hg could be detected in the muscle
532 (Omara et al., 2019). However, Hg levels of 0.11 µg/g dw were measured in muscle of
533 *O. niloticus* captured within the goldmine (Omara et al., 2019). Similarly, low Hg levels

534 (0.08 $\mu\text{g/g dw}$) were detected in *Brycinus sadleri* muscle captured 200 m downstream
535 of Katenga gold mine in the present study.

536 Physiologically the gills are used for gaseous exchange and thus exposed continuously
537 to the polluted river water. Definitely, this makes them the main entry of dissolved
538 metal ions and the target organ of metal toxicity (Olsson et al., 1998). Structurally, gills
539 have a thin epithelium which provides a large surface area for uptake and storage of the
540 metals. Similar to gills, the liver is constantly exposed to trace metals during
541 detoxification and metabolism (Olsson et al., 1998). After detoxification, the metals are
542 usually stored in the liver in a non-toxic form (Olsson et al., 1998). The edible muscle are
543 neither in direct contact with polluted water nor physiologically active such as liver and
544 gills, but they accumulated higher Hg levels. Mercury was not detected in water in the
545 downstream sites but detected in the fish organs, which was a clear indication that the
546 main route of Hg exposure was through food rather than water. The general low edible
547 muscle metal loads might be attributed to the even distribution of metals over a larger
548 mass compared to other organs (Vinodhini & Narayanan, 2008). The non-significant
549 differences in the accumulated concentrations in fish tissues of most metals among sites
550 suggests similar pollution levels exist along the river.

551

552 **4.5 Relationship between environmental and biotic trace metals concentrations**

553 The concentrations of trace metals in water, sediment and their physiochemical
554 characteristics such as conductivity, pH for water and TOC, clay content and grain size
555 may affect the bioavailability of trace metals to biota such as macroinvertebrates
556 (Luoma, 1989; Allen, 1993; Mason, 2013). The invertebrates are vital in the food chain
557 as they are food for fish, which may be consumed by humans, and thus understanding

558 the metal accumulation in macroinvertebrates in aquatic ecosystem is necessary. Both
559 studied species feed on aquatic invertebrates (fishbase.org). In the present study, weak
560 correlations between metal levels in water and sediment were found except for Cd, Zn,
561 and Hg. Further, only Au, Cd, Cu, Fe, Pb and Hg had significant relationships between
562 concentrations in the invertebrates and concentrations in water or sediment. For Au, Hg
563 and Pb TOC contributed significantly to the relationship with the concentration in the
564 sediment. As expected and also found by Bervoets et al. (1997), for Pb, TOC contributed
565 negatively to the relationship. For Au and Hg, however, higher TOC were related to
566 higher accumulated levels in the damselfly larvae. This might be an indication that these
567 metals are rather taken up via the food with higher exposure due to metals bound to
568 the organic matter.

569 Clay content did not contribute to the relationships between metals in sediment and in
570 biota. This is probably due to the relative low concentration of clay at all sites. With
571 higher clay concentrations one would expect a significant negative effect of clay content
572 on the relationship given the high adsorption capacity of fine material (Allen, 1993;
573 Luoma ,1989).

574

575 When related to the water concentration, conductivity and pH contributed significantly
576 negative to the accumulated concentrations for respectively Cd and Fe. Higher water
577 conductivity is determined by higher amount of free ions and higher pH with lower
578 amount of hydrogen ions (Canli and Canli, 2015) which will affect the bioavailability of
579 metals to biota (e.g. Bervoets et al., 1996 and Bervoets & Blust, 2000).

580 Only for three metals significant relationships were found between concentrations in
581 water and sediment. This is probably due to the differences in sediment characteristics
582 between the sites (Bervoets & Blust, 2000).

583

584 **4.6 Human health risk assessment**

585 Except for As in *Brycinus sadleri* at the upper limit bound, the measured trace metals
586 posed no human health risk through fish consumption because Q values were higher
587 than the average fish consumption in Uganda. Also, the hazard quotient (HQ) was lower
588 than 1 (critical value). Mataba et al. (2016) found a Q of 45 g and a HQ of 0.38 for As in
589 Ningu from Thigithe River at upper limit bound. However, no immediate health risk were
590 expected since the safe consumption of 45 g was higher than the national Tanzanian
591 daily consumption of 17g (Mataba et al., 2016). Van Ael et al. (2017) reported similar
592 potential human health risks due to As levels in muscle of fish collected in the Scheldt
593 estuary.

594 Also, Hg accumulated in *Brycinus sadleri* and the Q value at upper limit bound was 85 g
595 comparable to 65 g reported by Mataba et al. (2016) in *Labeo victorianus* from Thigithe
596 River. About 80-100 % of total Hg in fish muscle is present as methylmercury (EFSA,
597 2012) which is highly neurotoxic in humans (Li et al., 2010; Pandey et al., 2012). Since
598 the fish were continuously exposed to Hg loaded wastewater from the goldmine, health
599 risks are prevalent. Similar studies by Ezemonye et al. (2019) on *Brycinus longipinus*
600 from Benin River in Nigeria reported a Pb hazard quotient of 1.9. In Uganda, Omara et
601 al. (2019) reported the HQ of 0.94 for Hg in *Oreochromis niloticus* suggesting potential
602 toxicity to humans. The present study is cognisant of the differences in the size, age and
603 location of the fish species assessed. Therefore, the study findings of *Brycinus sadleri*

604 were mostly applicable for sites near the gold mine and could not be generalised along
605 the river, and similarly, the *Brycinus altianalis* assessment was applicable for the urban
606 catchment and nearby areas. However, both species indicated the general river nature
607 in terms of health risk assessment.

608

609 **5 Conclusions**

610 From the present study it was shown that for most measured variables in the water no
611 risk for human consumption is expected. Concerning the environmental quality possible
612 risks may occur at sites U2 and U3. In addition the river sediment was anthropogenically
613 polluted near the goldmining site although natural metal loading may affect the
614 concentration in the sediment. As a consequence more research is needed to assess
615 possible effects on the aquatic communities. Concerning the risk for human
616 consumption, only in the case of fish captured near the gold mine, arsenic in muscle
617 posed a possible risk for human consumption. However, more edible species should be
618 measured in the future to assess possible health risks.

619 Because no trace metal concentrations have been measured in the invertebrates and
620 sediment from the study river before, the present study can be used as a baseline.

621

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627

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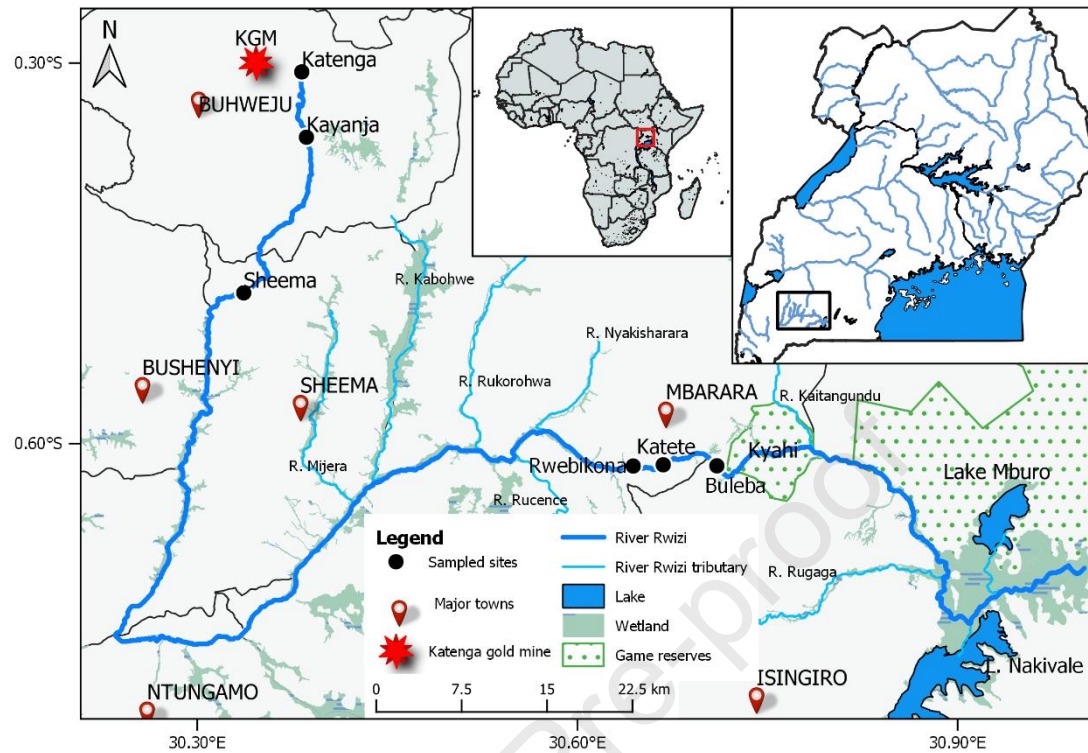
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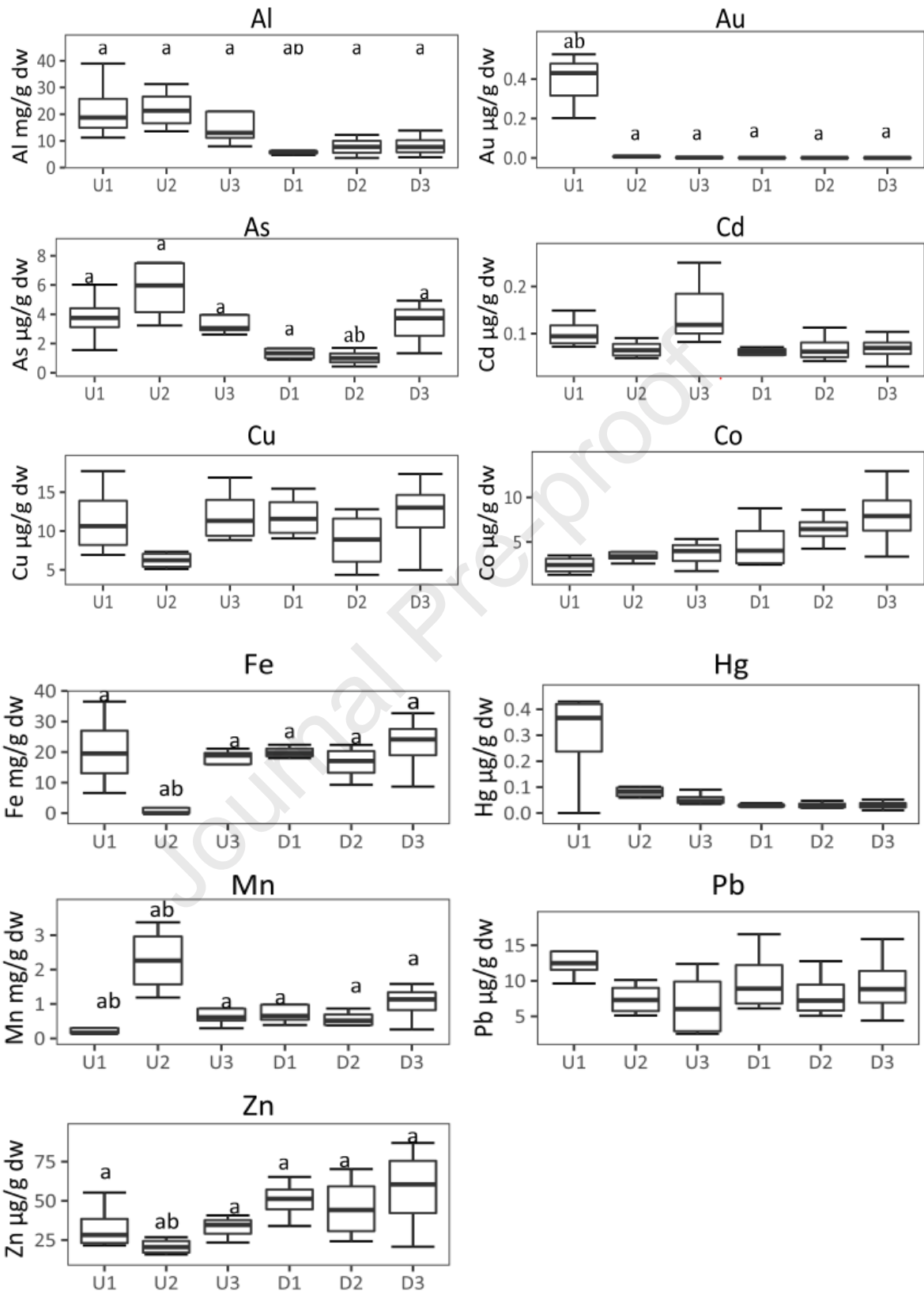
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886 Figure 1. Study sites; upstream sites; Katanga (U1) near the gold mine; Kayanja (U2) and
 887 Sheema (U3); downstream sites; Rwebikona (D1); Katete (D2) and Buleba (D3)

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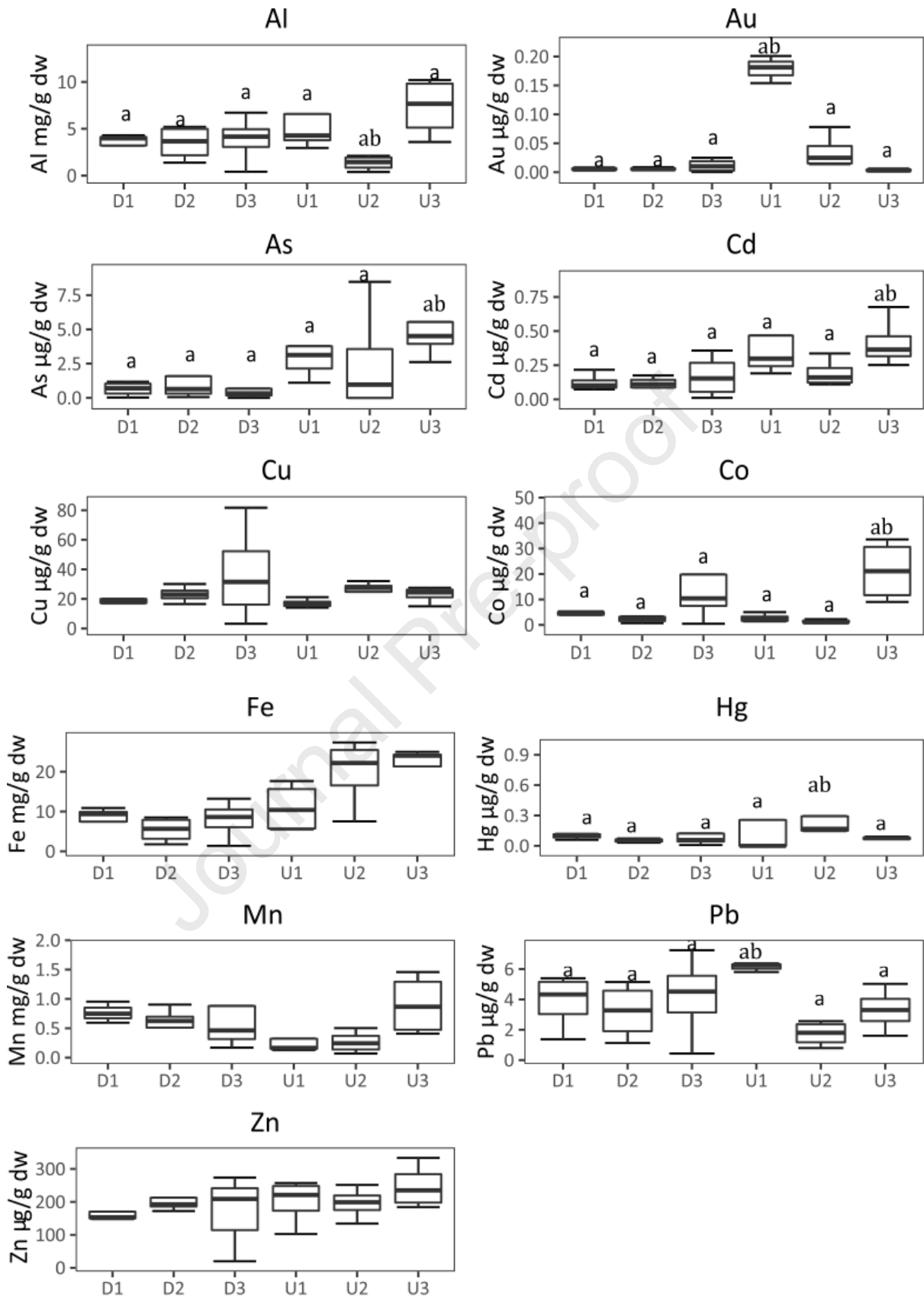
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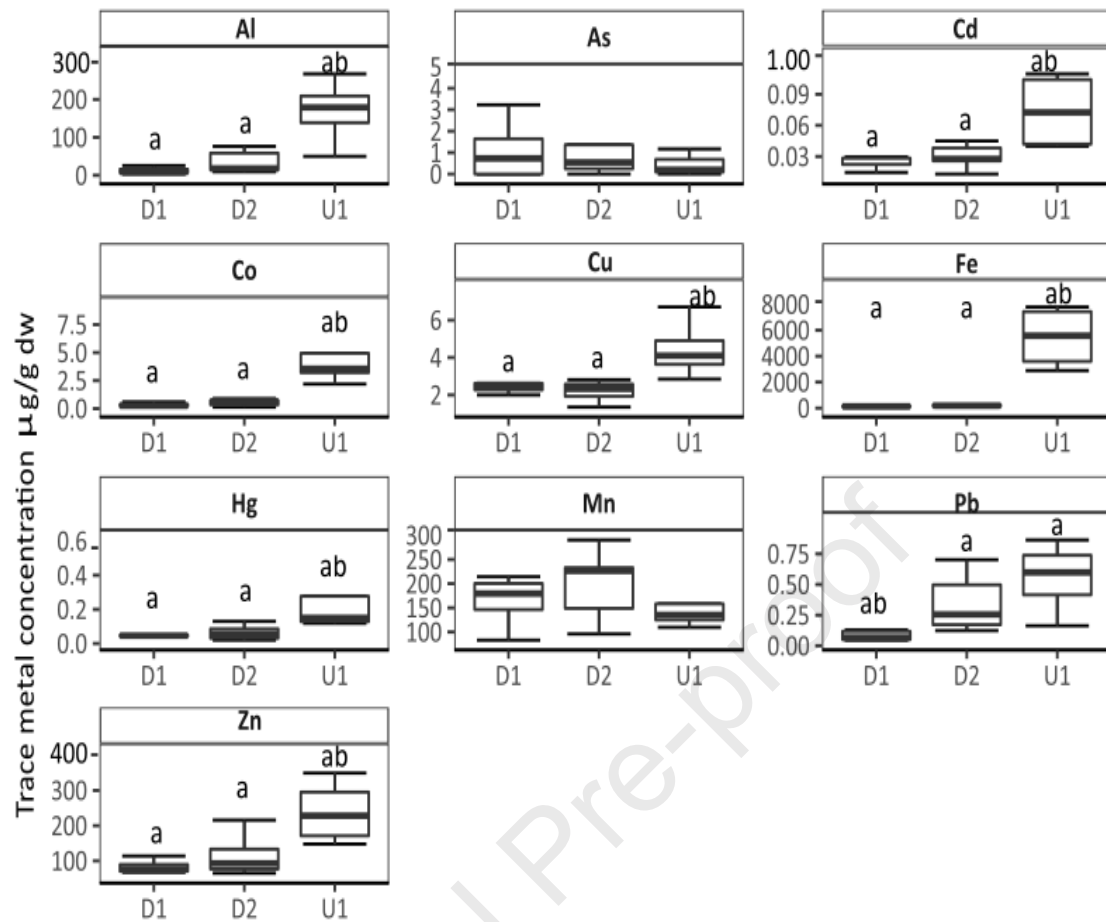
893 Figure 2: Trace metal concentrations in the sediment samples from the six sites in the
 894 Rwizi River. U1 to D3: sampled sites (table 1). Different letters indicate significant
 895 differences among sites, N=4



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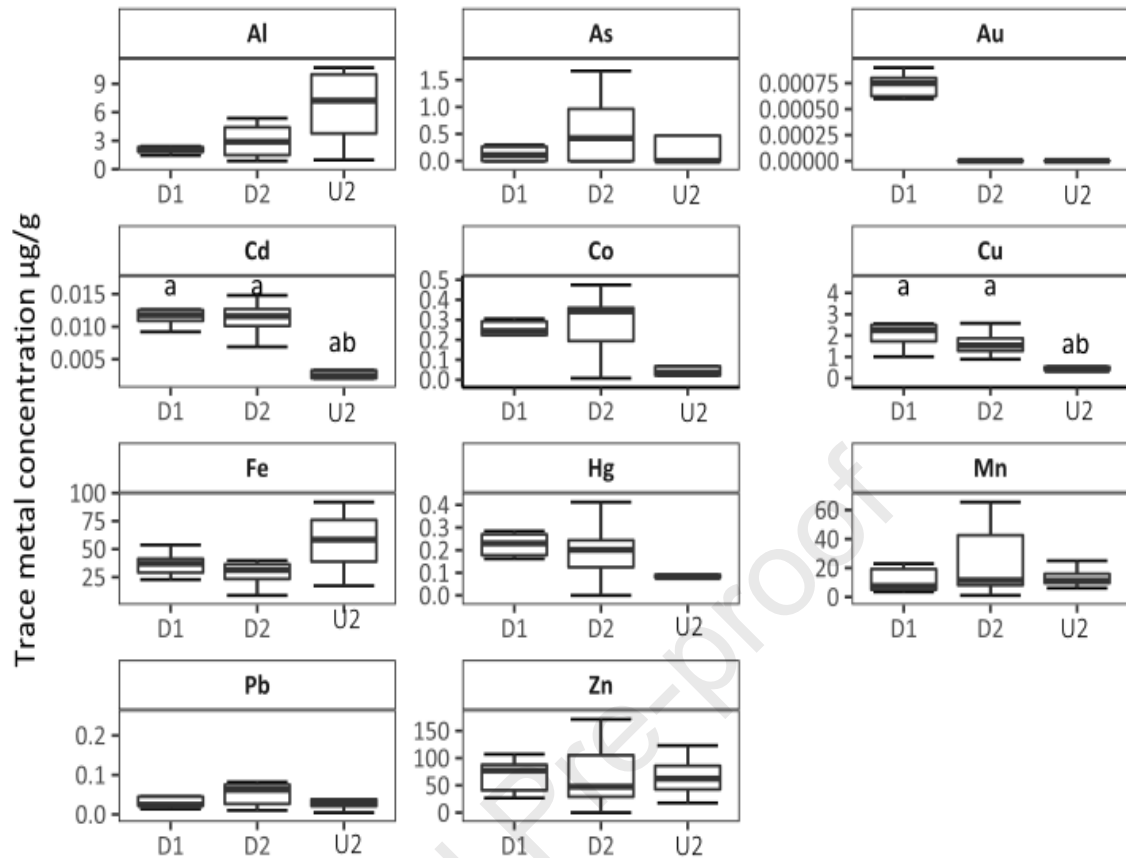
898 Figure 3: Trace metal concentrations in the damselfly larvae from the six sites in the
 899 Rwizi River. Different letters indicate significant differences among sites. N= 4. U1 to D3:
 900 sampling sites (table 1).



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902 Figure 4. Trace metal concentration in the gill tissue of *Barbus altianalis* from Rwebikona
 903 (D1, n=5) and Katete (D2, n=9) and *Brycinus sadleri* from Kayanja (U1, n=5). Different
 904 letters indicate significant differences among sites.

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907 Figure 5. Trace metal concentrations in the muscle of *Barbus altianalis* collected at Katete (D2,
 908 n=9), Rwebikona (D1, n=5) and *Brycinus sadleri* from Kayanja (U2, n = 5). Different letters
 909 indicate significant differences among sites.

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916 Table 1: Number of replicate samples collected for each sampled variable

	Replicate samples obtained							
	Water	Sediment	Invertebrates	<i>Barbus altianalis</i>			<i>Brycinus sadleri</i>	
				Liver	Gills	Muscle	Gills	Muscle
Katenga (U1)	4	4	4	0	0	0	0	0
Kayanja (U2)	4	4	4	0	0	0	5	5
Sheema (U3)	4	4	4	0	0	0	0	0
Rwebikona (D1)	4	4	4	5	5	5	0	0
Katete (D2)	4	4	4	9	9	9	0	0
Buleba (D3)	4	4	4	0	0	0	0	0

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920 Table2: Range in trace metal concentrations in the surface water ($\mu\text{g/L}$) at the six locations in the Rwizi River.

Metals	U1	U2	U3	D1	D2	D3	Water quality standards	
							Drinking water	Environmental standards
Al	58.2 – 349 ^{ab}	7.9 - 15.4 ^a	4.3 - 13.4 ^a	6.2 - 9.2 ^a	4.65 – 110 ^a	3.2 - 11.2 ^a	-	-
As	BDL - 1.6	BDL - 5.72	0 - 5.43	BDL - 5.9	0 - 2.89	BDL - 6.2	10	3
Au	0.003 - 0.01	BDL	BDL	BDL	BDL	BDL	-	-
Cd	0.03 - 0.1 ^a	0.028 - 0.03 ^{ab}	0.19 - 0.25 ^{abc}	0.03 - 0.05 ^a	0.03 - 0.06 ^a	0.03 - 0.04 ^a	3	0.08-0.45
Co	0.2 - 8.1 ^a	0.27 - 0.3 ^a	9.7 - 10.8 ^{abc}	0.3 - 0.4 ^a	0.29 - 1.3 ^a	0.32 - 0.4 ^a	-	0.5
Cu	0.2- 0.9 ^a	BDL - 0.13 ^a	BDL - 0.23 ^{ab}	0.48 - 0.83 ^a	0.58 - 2.49 ^a	1.1 - 4.44 ^{ab}	2,000*	7.2
Fe	156 – 67267 ^a	424 – 535 ^{ab}	61 – 91 ^{ab}	110 – 160 ^a	89 – 433 ^a	79 – 107 ^a	50,000*	-
Hg	0.06 - 0.3	0 - 0.01	BDL	BDL	BDL	BDL	6.0*	0.05
Mn	36.6- 3751 ^a	87 – 629 ^a	740 – 831 ^{ab}	24.5 - 36.8 ^a	28.8 – 101 ^a	29.4 - 47.3 ^a	400*	-
Pb	0.06 - 0.22 ^{ab}	0.02 - 0.09 ^a	0.017 - 0.06 ^a	0.03 - 0.05 ^a	0.027 - 0.27 ^a	0.02 - 0.04 ^a	10*	7.2
Zn	3.14 - 11.9 ^a	3.48 - 10.2 ^a	89.25 – 102 ^{ab}	0.71 - 53.3 ^a	0.77 - 18.7 ^a	5.0 – 144 ^a	3,000*	20

921 BDL: Below Detection Limit. N=6.

922 U1-D3: Sampled sites along the River Rwizi (Table 1). Different superscript letters in a row indicate significant differences ($p < 0.05$) among sites
923 for the trace metals; * drinking water standards WHO; environmental standards: EU 2013

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925 Table 3. Results of the multiple regression analyses between metal(loid) concentrations
 926 in damselfly larvae (*Ceriatrigon glabrum*) and in sediment or water.

Metals	Sediment	Water
Au	$Au_{invert} = 0.0022 + 0.267 * Au_{sed} + 0.049 * TOC$ $R^2 = 0.99, p < 0.001$	NS
	$Au_{invert} = 0.0034 - 986.9_{wat} + 3.984_{sed}, p = 0.00003, R^2 = 0.99, \text{Significant}$	
Cd	NS	$Cd_{invert} = 0.471 + (1.39 * Cd_{wat}) - 0.0043 * Cond$ $R^2 = 0.81, P = 0.038$
	$Cu_{(invert)} = 0.0687 - 9.4_{wat} + 6.62_{sed}, p = 0.009, R^2 = 0.958$	
Cu	NS	$Cu_{invert} = 19.4 + (5.46 * Cu_{wat}), R^2 = 0.49, p = 0.05$
Fe	NS	$Fe_{invert} = 12.27 - (0.329 * Fe_{wat}) - 17.5 * pH$ $R^2 = 0.89, p = 0.016$
Hg	$Hg_{invert} = 0.029 + 0.711 * Hg_{sed} + 0.479 * TOC$ $R^2 = 0.97, P = 0.015$	NS
Pb	$Pb_{invert} = -4.92 + (0.99 * Pb_{sed})$ $R^2 = 0.79, NS$ $p = 0.011$ $Pb_{invert} = -4.21 + (1.02 * Pb_{sed}) - (3.01 * TOC)$ $R^2 = 0.92, P = 0.011$	
	$Pb_{(invert)} = -4.62 + 8.953_{wat} + 0.9_{sed}, p = 0.004, R^2 = 0.85$	

927 Invert=invertebrates, sed= sediment; wat = water; TOC= total organic carbon, NS= not
 928 significant

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931 Table 4: Maximal daily amount (Q, g) of fish that could be eaten without health risk by
 932 a person of 70 kg average body weight and Hazard Quotient (HQ) for *Barbus altianalis*
 933 (*B.a*, n=14) and *Brycinus sadleri* (*B.s*, n=4) obtained from River Rwizi at 50 (median, mg/g
 934 ww) and 95 percentile (mg/g ww) trace metal levels

	Al	As	Cd	Co	Cu	Hg	Pb	Zn
MRL (mg/kg/day)	1	0.0003	0.00001	0.01	0.01	0.0003	0.00036	0.3
MRL for 70 kg person (mg/kg/day)	70	0.021	0.007	0.7	0.7	0.02	0.02	21
Concentration in <i>B.a</i> at 50 %	0.2	0.03	0.001	0.026	0.2	0.017	0.005	6.8
Concentration in <i>B.a</i> at 95 %	1.0	0.13	0.002	0.101	0.5	0.06	0.028	34
Q for <i>B.a</i> at 50 %	329739	820	5448	26788	4648	1271	4949	3076
Q for <i>B.a</i> at 95 %	69142	162	3189	6938	1340	350	911	614
HQ for <i>B.a</i> at 50 %	0.0	0.04	0.006	0.001	0.01	0.025	0.01	0.010
HQ for <i>B.a</i> at 95 %	0.0	0.19	0.010	0.005	0.02	0.09	0.04	0.051
Concentration in <i>B.s</i> at 50 %	9.7		0.003	0.037	0.4	0.08	0.03	62
Concentration in <i>B.s</i> at 95 %	121	1.5	0.01	0.17	0.8	0.2	0.1	116
Q for <i>B.s</i> at 50 %	7183		2755	19077	1821	259	857	337
Q for <i>B.s</i> at 95 %	577	14	729	4161	892	89	256	182
HQ for <i>B.s</i> at 50 %	0.0	0.00	0.01	0.002	0.02	0.1	0.04	0.09
HQ for <i>B.s</i> at 95 %	0.1	2.2	0.04	0.008	0.04	0.4	0.1	0.17

935 *B.a*: *Barbus altianalis*, *B.s*: *Brycinus sadleri*, HQ: Hazard Quotient, MRL: Minimum Risk
 936 Levels (ATSDR, 2018), Q: Maximum health risk free quantity

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1 Table 1: Number of replicate samples collected for each sampled variable

	Replicate samples obtained							
	Water	Sediment	Invertebrates	<i>Barbus altianalis</i>			<i>Brycinus sadleri</i>	
				Liver	Gills	Muscle	Gills	Muscle
Katenga (U1)	4	4	4	0	0	0	0	0
Kayanja (U2)	4	4	4	0	0	0	5	5
Sheema (U3)	4	4	4	0	0	0	0	0
Rwebikona (D1)	4	4	4	5	5	5	0	0
Katete (D2)	4	4	4	9	9	9	0	0
Buleba (D3)	4	4	4	0	0	0	0	0

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4

5 Table2: Range in trace metal concentrations in the surface water ($\mu\text{g/L}$) at the six locations in the Rwizi River.

Metals	U1	U2	U3	D1	D2	D3	Water quality standards	
							Drinking water	Environmental standards
Al	58.2 – 349 ^{ab}	7.9 - 15.4 ^a	4.3 - 13.4 ^a	6.2 - 9.2 ^a	4.65 – 110 ^a	3.2 - 11.2 ^a	-	-
As	BDL - 1.6	BDL - 5.72	0 - 5.43	BDL - 5.9	0 - 2.89	BDL - 6.2	10	3
Au	0.003 - 0.01	BDL	BDL	BDL	BDL	BDL	-	-
Cd	0.03 - 0.1 ^a	0.028 - 0.03 ^{ab}	0.19 - 0.25 ^{abc}	0.03 - 0.05 ^a	0.03 - 0.06 ^a	0.03 - 0.04 ^a	3	0.08-0.45
Co	0.2 - 8.1 ^a	0.27 - 0.3 ^a	9.7 - 10.8 ^{abc}	0.3 - 0.4 ^a	0.29 - 1.3 ^a	0.32 - 0.4 ^a	-	0.5
Cu	0.2- 0.9 ^a	BDL - 0.13 ^a	BDL - 0.23 ^{ab}	0.48 - 0.83 ^a	0.58 - 2.49 ^a	1.1 - 4.44 ^{ab}	2,000*	7.2
Fe	156 – 67267 ^a	424 – 535 ^{ab}	61 – 91 ^{ab}	110 – 160 ^a	89 – 433 ^a	79 – 107 ^a	50,000*	-
Hg	0.06 - 0.3	0 - 0.01	BDL	BDL	BDL	BDL	6.0*	0.05
Mn	36.6- 3751 ^a	87 – 629 ^a	740 – 831 ^{ab}	24.5 - 36.8 ^a	28.8 – 101 ^a	29.4 - 47.3 ^a	400*	-
Pb	0.06 - 0.22 ^{ab}	0.02 - 0.09 ^a	0.017 - 0.06 ^a	0.03 - 0.05 ^a	0.027 - 0.27 ^a	0.02 - 0.04 ^a	10*	7.2
Zn	3.14 - 11.9 ^a	3.48 - 10.2 ^a	89.25 – 102 ^{ab}	0.71 - 53.3 ^a	0.77 - 18.7 ^a	5.0 – 144 ^a	3,000*	20

6 BDL: Below Detection Limit. N=6.

7 U1-D3: Sampled sites along the River Rwizi (Table 1). Different superscript letters in a row indicate significant differences ($p < 0.05$) among ~~between~~

8 sites for the trace metals; * drinking water standards WHO; environmental standards: EU 2013

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Table2: Range in trace metal concentrations in the surface water ($\mu\text{g/L}$) at the six locations in the Rwizi River.

Metals	U1	U2	U3	D1	D2	D3
Al	58.2 – 349 ^{ab}	7.9 - 15.4 ^a	4.3 - 13.4 ^a	6.2 - 9.2 ^a	4.65 – 110 ^a	3.2 - 11.2 ^a
As	BDL - 1.6	BDL - 5.72	0 - 5.43	BDL - 5.9	0 - 2.89	BDL - 6.2
Au	0.003 - 0.01	BDL	BDL	BDL	BDL	BDL
Cd	0.03 - 0.1 ^a	0.028 - 0.03 ^{ab}	0.19 - 0.25 ^{abc}	0.03 - 0.05 ^a	0.03 - 0.06 ^a	0.03 - 0.04 ^a
Co	0.2 - 8.1 ^a	0.27 - 0.3 ^a	9.7 - 10.8 ^{abc}	0.3 - 0.4 ^a	0.29 - 1.3 ^a	0.32 - 0.4 ^a
Cu	0.2- 0.9 ^a	BDL - 0.13 ^a	BDL - 0.23 ^{ab}	0.48 - 0.83 ^a	0.58 - 2.49 ^a	1.1 - 4.44 ^{ab}
Fe	156 – 67267 ^a	424 – 535 ^{ab}	61 – 91 ^{ab}	110 – 160 ^a	89 – 433 ^a	79 – 107 ^a
Hg	0.06 - 0.3	0 - 0.01	BDL	BDL	BDL	BDL
Mn	36.6- 3751 ^a	87 – 629 ^a	740 – 831 ^{ab}	24.5 - 36.8 ^a	28.8 – 101 ^a	29.4 - 47.3 ^a
Pb	0.06 - 0.22 ^{ab}	0.02 - 0.09 ^a	0.017 - 0.06 ^a	0.03 - 0.05 ^a	0.027 - 0.27 ^a	0.02 - 0.04 ^a
Zn	3.14 - 11.9 ^a	3.48 - 10.2 ^a	89.25 – 102 ^{ab}	0.71 - 53.3 ^a	0.77 - 18.7 ^a	5.0 – 144 ^a

BDL: Below Detection Limit. N=6.

U1-D3: Sampled sites along the River Rwizi (Table 1). Different superscript letters in a row indicate significant differences between sites for the trace metals.

Table 3. Results of the multiple regression analyses between metal(loid) concentrations in damselfly larvae (*Ceriatrigon glabrum*) and in sediment or water.

Metals	Sediment	Water
Au	$Au_{invert} = 0.0022 + 0.267 * Au_{sed} + 0.049 * TOC$ $R^2 = 0.99, p < 0.001$	NS
	$Au_{invert} = 0.0034 - 986.9_{wat} + 3.984_{sed}, p = 0.00003, R^2 = 0.99, \text{Significant}$	
Cd	NS	$Cd_{invert} = 0.471 + (1.39 * Cd_{wat}) - 0.0043 * Cond$ $R^2 = 0.81, P = 0.038$
	$Cu_{(inverts)} = 0.0687 - 9.4_{wat} + 6.62_{sed}, p = 0.009, R^2 = 0.958$	
Cu	NS	$Cu_{invert} = 19.4 + (5.46 * Cu_{wat}), R^2 = 0.49, p = 0.05$
Fe	NS	$Fe_{invert} = 12.27 - (0.329 * Fe_{wat}) - 17.5 * pH$ $R^2 = 0.89, p = 0.016$
Hg	$Hg_{invert} = 0.029 + 0.711 * Hg_{sed} + 0.479 * TOC$ $R^2 = 0.97, P = 0.015$	NS
Pb	$Pb_{invert} = -4.92 + (0.99 * Pb_{sed})$ $R^2 = 0.79, p = 0.011$ $Pb_{invert} = -4.21 + (1.02 * Pb_{sed}) - (3.01 * TOC)$ $R^2 = 0.92, P = 0.011$	NS
	$Pb_{(inverts)} = -4.62 + 8.953_{wat} + 0.9_{sed}, p = 0.0.04, R^2 = 0.85$	

Invert=invertebrates, sed= sediment; wat = water; TOC= total organic carbon, NS= not significant

Table 4: Maximal daily amount (Q, g) of fish that could be eaten without health risk by a person of 70 kg average body weight and Hazard Quotient (HQ) for *Barbus altianalis* (*B.a*, n=14) and *Brycinus sadleri* (*B.s*, n=4) obtained from River Rwizi at 50 (median, mg/g ww) and 95 percentile (mg/g ww) trace metal levels

	Al	As	Cd	Co	Cu	Hg	Pb	Zn
MRL (mg/kg/day)	1	0.0003	0.00001	0.01	0.01	0.0003	0.00036	0.3
MRL for 70 kg person (mg/kg/day)	70	0.021	0.007	0.7	0.7	0.02	0.02	21
Concentration in <i>B.a</i> at 50 %	0.2	0.03	0.001	0.026	0.2	0.017	0.005	6.8
Concentration in <i>B.a</i> at 95 %	1.0	0.13	0.002	0.101	0.5	0.06	0.028	34
Q for <i>B.a</i> at 50 %	329739	820	5448	26788	4648	1271	4949	3076
Q for <i>B.a</i> at 95 %	69142	162	3189	6938	1340	350	911	614
HQ for <i>B.a</i> at 50 %	0.0	0.04	0.006	0.001	0.01	0.025	0.01	0.010
HQ for <i>B.a</i> at 95 %	0.0	0.19	0.010	0.005	0.02	0.09	0.04	0.051
Concentration in <i>B.s</i> at 50 %	9.7		0.003	0.037	0.4	0.08	0.03	62
Concentration in <i>B.s</i> at 95 %	121	1.5	0.01	0.17	0.8	0.2	0.1	116
Q for <i>B.s</i> at 50 %	7183		2755	19077	1821	259	857	337
Q for <i>B.s</i> at 95 %	577	14	729	4161	892	89	256	182
HQ for <i>B.s</i> at 50 %	0.0	0.00	0.01	0.002	0.02	0.1	0.04	0.09
HQ for <i>B.s</i> at 95 %	0.1	2.2	0.04	0.008	0.04	0.4	0.1	0.17

B.a: *Barbus altianalis*, *B.s*: *Brycinus sadleri*, HQ: Hazard Quotient, MRL: Minimum Risk Levels (ATSDR, 2018), Q: Maximum health risk free quantity

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1 Table 1: Number of replicate samples collected for each sampled variable

	Replicate samples obtained							
	Water	Sediment	Invertebrates	<i>Barbus altianalis</i>			<i>Brycinus sadleri</i>	
				Liver	Gills	Muscle	Gills	Muscle
Katenga (U1)	4	4	4	0	0	0	0	0
Kayanja (U2)	4	4	4	0	0	0	5	5
Sheema (U3)	4	4	4	0	0	0	0	0
Rwebikona (D1)	4	4	4	5	5	5	0	0
Katete (D2)	4	4	4	9	9	9	0	0
Buleba (D3)	4	4	4	0	0	0	0	0

2

3

4

5 Table2: Range in trace metal concentrations in the surface water ($\mu\text{g/L}$) at the six locations in the Rwizi River.

Metals	U1	U2	U3	D1	D2	D3	Water quality standards	
							Drinking water	Environmental standards
Al	58.2 – 349 ^{ab}	7.9 - 15.4 ^a	4.3 - 13.4 ^a	6.2 - 9.2 ^a	4.65 – 110 ^a	3.2 - 11.2 ^a	-	-
As	BDL - 1.6	BDL - 5.72	0 - 5.43	BDL - 5.9	0 - 2.89	BDL - 6.2	10	3
Au	0.003 - 0.01	BDL	BDL	BDL	BDL	BDL	-	-
Cd	0.03 - 0.1 ^a	0.028 - 0.03 ^{ab}	0.19 - 0.25 ^{abc}	0.03 - 0.05 ^a	0.03 - 0.06 ^a	0.03 - 0.04 ^a	3	0.08-0.45
Co	0.2 - 8.1 ^a	0.27 - 0.3 ^a	9.7 - 10.8 ^{abc}	0.3 - 0.4 ^a	0.29 - 1.3 ^a	0.32 - 0.4 ^a	-	0.5
Cu	0.2- 0.9 ^a	BDL - 0.13 ^a	BDL - 0.23 ^{ab}	0.48 - 0.83 ^a	0.58 - 2.49 ^a	1.1 - 4.44 ^{ab}	2,000*	7.2
Fe	156 – 67267 ^a	424 – 535 ^{ab}	61 – 91 ^{ab}	110 – 160 ^a	89 – 433 ^a	79 – 107 ^a	50,000*	-
Hg	0.06 - 0.3	0 - 0.01	BDL	BDL	BDL	BDL	6.0*	0.05
Mn	36.6- 3751 ^a	87 – 629 ^a	740 – 831 ^{ab}	24.5 - 36.8 ^a	28.8 – 101 ^a	29.4 - 47.3 ^a	400*	-
Pb	0.06 - 0.22 ^{ab}	0.02 - 0.09 ^a	0.017 - 0.06 ^a	0.03 - 0.05 ^a	0.027 - 0.27 ^a	0.02 - 0.04 ^a	10*	7.2
Zn	3.14 - 11.9 ^a	3.48 - 10.2 ^a	89.25 – 102 ^{ab}	0.71 - 53.3 ^a	0.77 - 18.7 ^a	5.0 – 144 ^a	3,000*	20

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B.a: *Barbus altianalis*, *B.s*: *Brycinus sadleri*, HQ: Hazard Quotient, MRL: Minimum Risk Levels (ATSDR, 2018), Q: Maximum health risk free quantity

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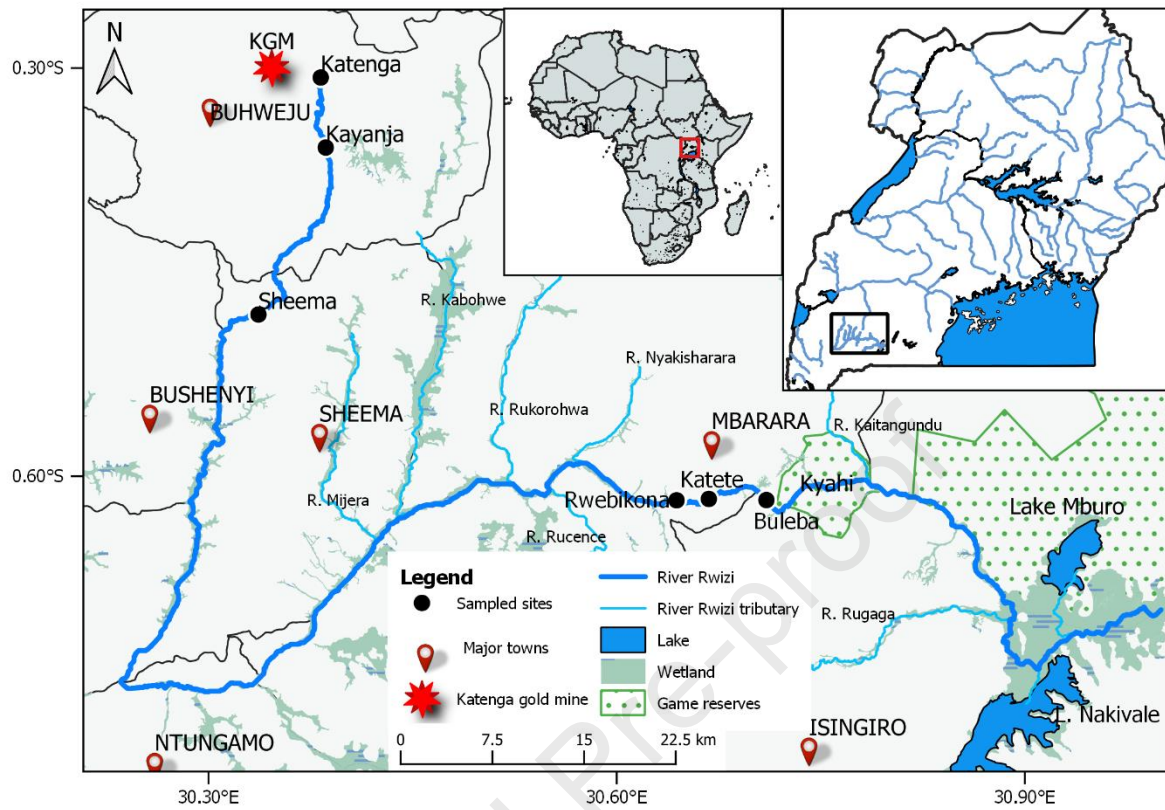


Figure 1. Study sites; upstream sites; Katenga (U1) near the gold mine; Kayanja (U2) and Sheema (U3); downstream sites; Rwebikona (D1); Katete (D2) and Buleba (D3)

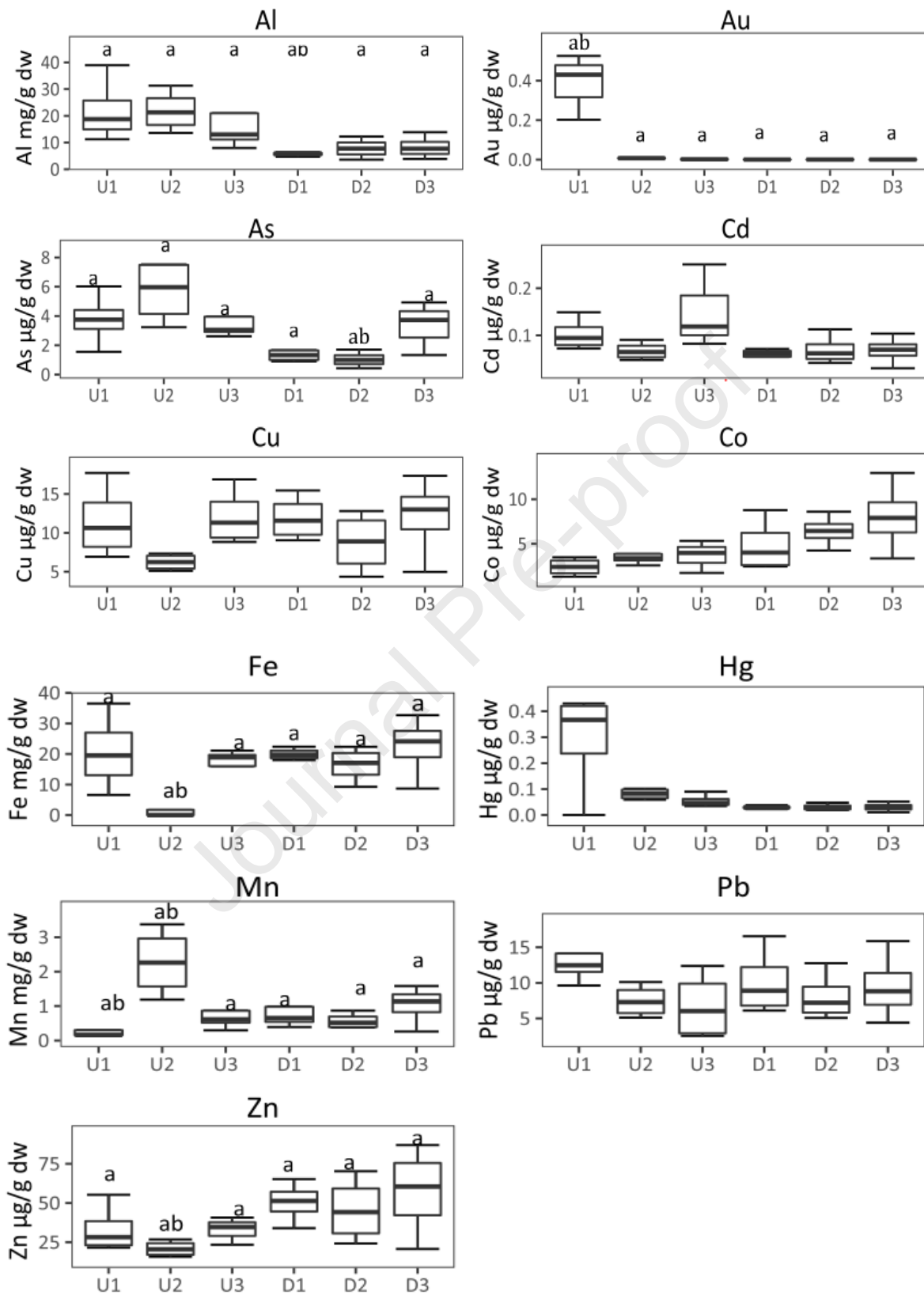


Figure 2: Trace metal concentrations in the sediment samples from the six sites in the Rwizi River. U1 to D3: sampled sites (table 1). Different letters indicate significant differences among sites, N=4

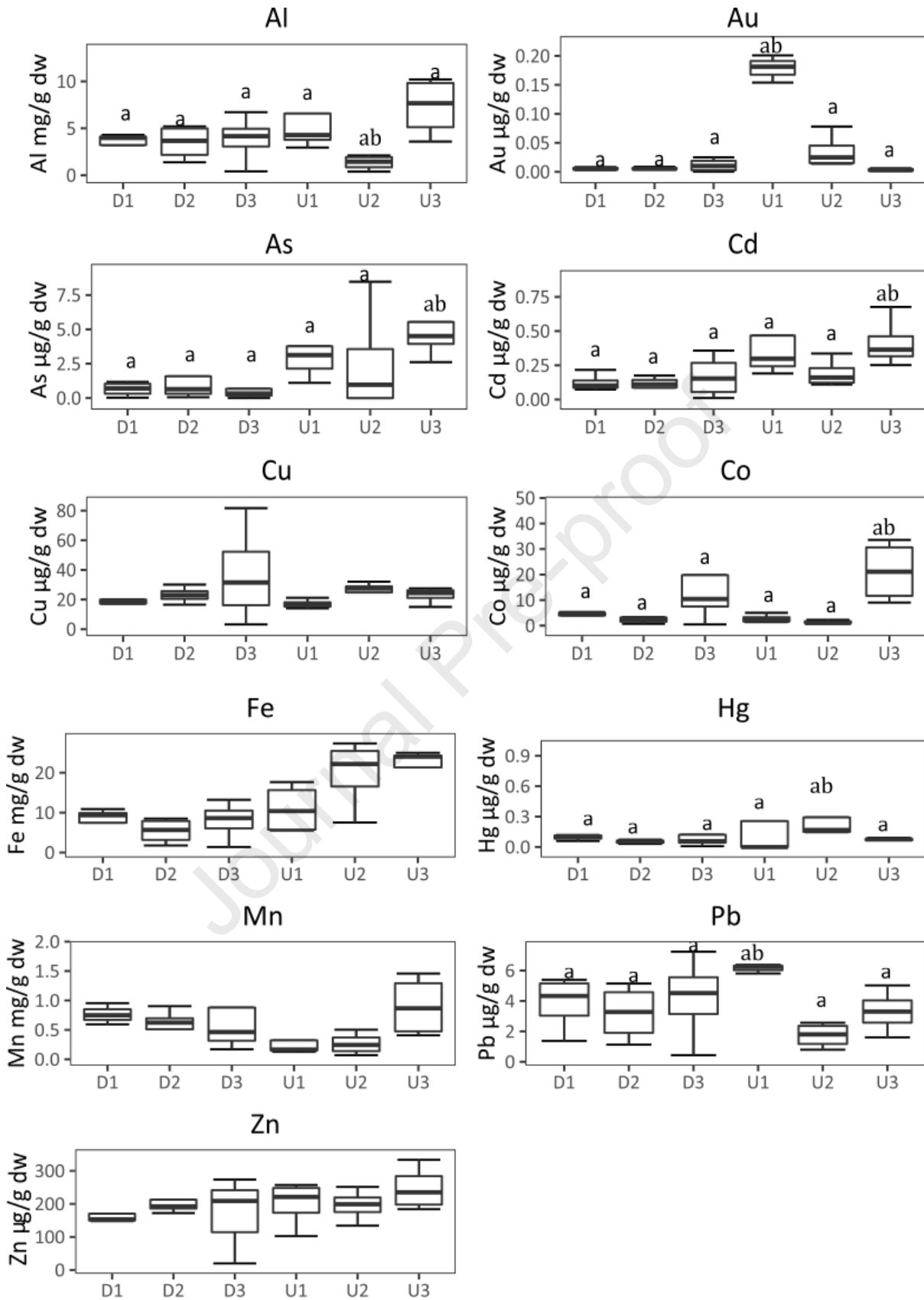


Figure 3: Trace metal concentrations in the damselfly larvae from the six sites in the Rwizi River. Different letters indicate significant differences among sites. N= 4. U1 to D3: sampling sites (table 1).

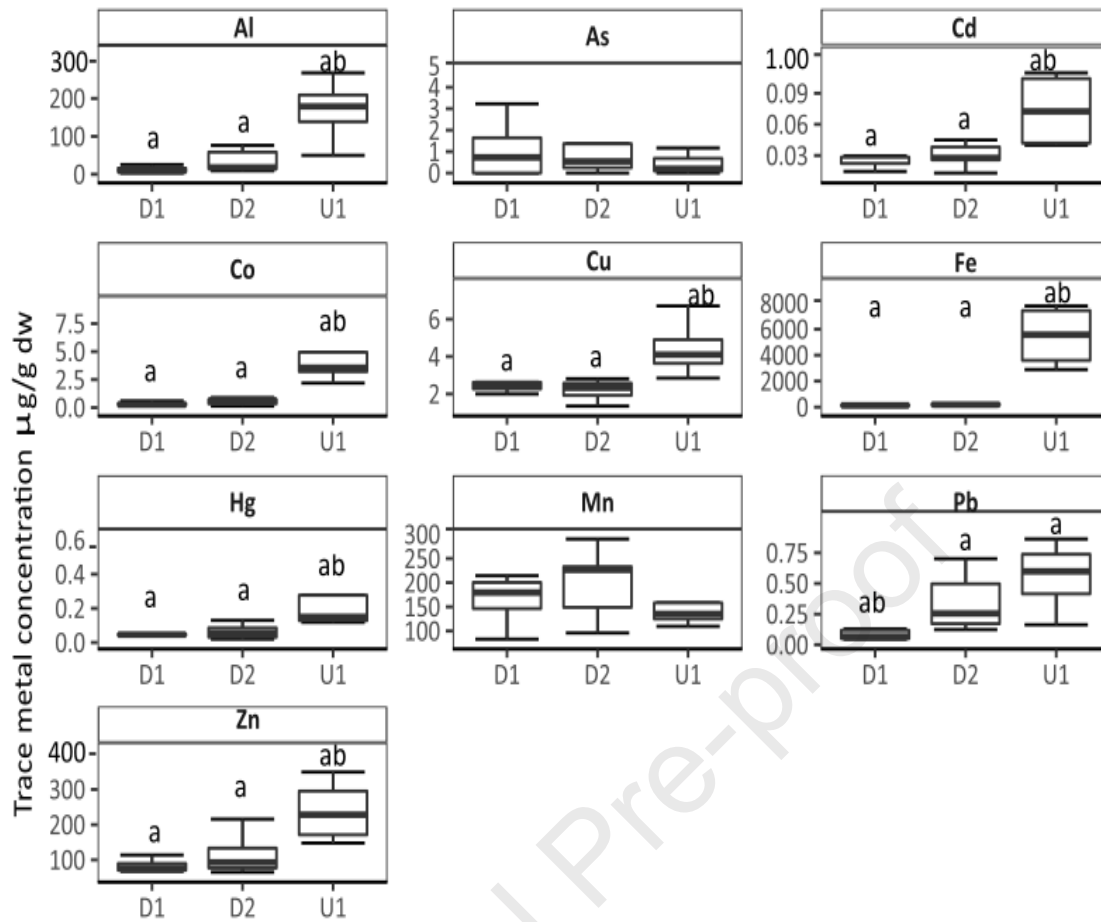


Figure 4. Trace metal concentration in the gill tissue of *Barbus altianalis* from Rwebikona (D1, n=5) and Katete (D2, n=9) and *Brycinus sadleri* from Kayanja (U1, n=5). Different letters indicate significant differences among sites.

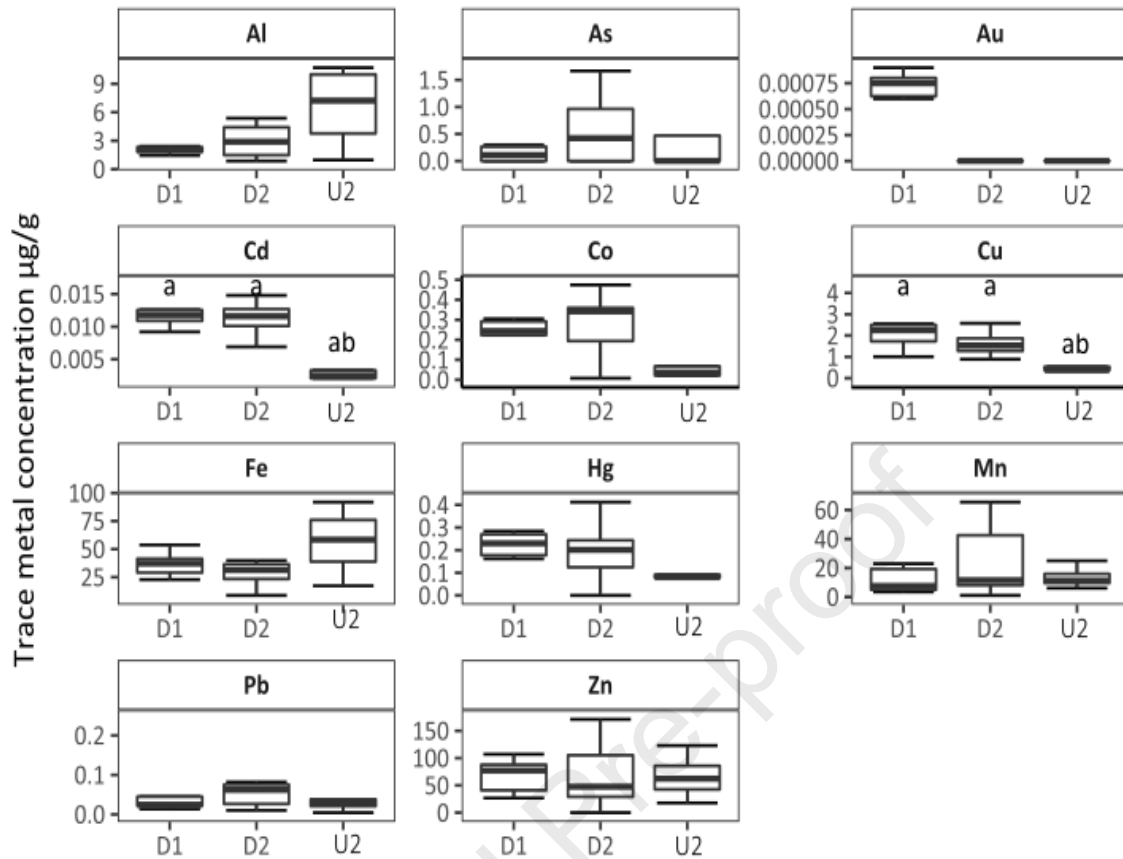


Figure 5. Trace metal concentrations in the muscle of *Barbus altianalis* collected at Katete (D2, n=9), Rwebikona (D2, n=5) and *Brycinus sadleri* from Kayanja (U2, n = 5). Different letters indicate significant differences among sites.

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